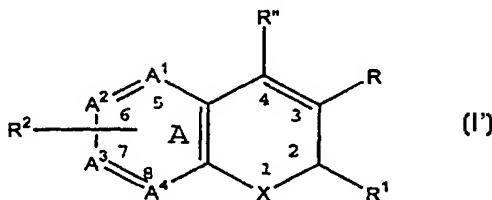




INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification ⁷ : C07D 311/58, 335/06, 215/54, 311/92, 493/04, 409/04, 491/04, A61K 31/35 // (C07D 491/04, 311:00, 221:00) (C07D 493/04, 311:00, 307:00) (C07D 493/04, 317:00, 311:00)</p>	A1	<p>(11) International Publication Number: WO 00/23433</p> <p>(43) International Publication Date: 27 April 2000 (27.04.00)</p>
<p>(21) International Application Number: PCT/US99/21460</p> <p>(22) International Filing Date: 15 October 1999 (15.10.99)</p> <p>(30) Priority Data: 09/175,584 20 October 1998 (20.10.98) US</p> <p>(63) Related by Continuation (CON) or Continuation-in-Part (CIP) to Earlier Application US Not furnished (CON) Filed on Not furnished</p> <p>(71) Applicant (for all designated States except US): G.D. SEARLE & CO. [US/US]; Corporate Patent Dept., P.O. Box 5110, Chicago, IL 60680 (US).</p> <p>(72) Inventors; and (75) Inventors/Applicants (for US only): CARTER, Jeffrey, S. [US/US]; 15321 Grantley Drive, Chesterfield, MO 63017 (US). DEVADAS, Balekudru [-/US]; 2175 Parasol Drive, Chesterfield, MO 63017 (US). TALLEY, John, J. [US/US]; 4701 Clifton Avenue, St. Louis, MO 63109</p>		<p>(US). BROWN, David, L. [US/US]; 15504 Twingate, Chesterfield, MO 63017 (US). GRANETO, Matthew, J. [US/US]; 352 Hartwell Court, Chesterfield, MO 63017 (US). ROGIER, Donald, J., Jr. [US/US]; 1828 Westmeade Drive, Chesterfield, MO 63017 (US). NAGARAJAN, Srinivasan, R. [-/US]; 16209 Forest Meadows Drive, Chesterfield, MO 63017 (US). HANAU, Cathleen, E. [US/US]; 15551 Wendimill Drive, St. Louis, MO 63017 (US). HARTMANN, Susan, J. [US/US]; 449 Colony Woods Drive, Kirkwood, MO 63122 (US). LUDWIG, Cindy, L. [US/US]; 1312 Dautel Lane, St. Louis, MO 63017 (US). METZ, Suzanne [US/US]; 525 Westernmill Drive, Chesterfield, MO 63017 (US). KORTE, Donald, E. [US/US]; 5 Birch Court, Mundelein, IL 60062 (US). BERTENSHAW, Stephen, R. [US/US]; 43 Lansdown, Cheshire, CT 06410 (US). OBUKOWICZ, Mark, G. [US/US]; 655 North Kirkwood Road, Kirkwood, MO 63122 (US).</p> <p>(74) Agents: BULOCK, Joseph, W. et al.; G.D. Searle & Co., Corporate Patent Dept., P.O. Box 5110, Chicago, IL 60680 (US).</p> <p>(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).</p> <p>Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i></p>

(54) Title: SUBSTITUTED BENZOPYRAN ANALOGS FOR THE TREATMENT OF INFLAMMATION



(57) Abstract

A class of benzopyrans, benzothiopyrans, dihydroquinolines, dihydronaphthalenes, and analogs thereof, is described for use in treating cyclooxygenase-2 mediated disorders. Compounds of particular interest are defined by Formula (I') wherein X, A¹, A², A³, A⁴, R, R', R'' and R² are as described in the specification.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece			TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
CA	Canada	IT	Italy	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	NZ	New Zealand		
CM	Cameroon			PL	Poland		
CN	China	KR	Republic of Korea	PT	Portugal		
CU	Cuba	KZ	Kazakstan	RO	Romania		
CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
DE	Germany	LI	Liechtenstein	SD	Sudan		
DK	Denmark	LK	Sri Lanka	SE	Sweden		
EE	Estonia	LR	Liberia	SG	Singapore		

**SUBSTITUTED BENZOPYRAN ANALOGS
FOR THE TREATMENT OF INFLAMMATION**

Related Cases

5

This is a continuation in part of application Serial No. 09/062,537, filed April 17, 1998, which is a continuation in part of application Serial No. 60/044,485, filed April 21, 1997.

10

FIELD OF THE INVENTION

This invention is in the field of anti-inflammatory pharmaceutical agents and specifically relates to
15 compounds, compositions and methods for treating cyclooxygenase-2 mediated disorders, such as inflammation and inflammation-related disorders.

BACKGROUND OF THE INVENTION

20

Prostaglandins play a major role in the inflammation process and the inhibition of prostaglandin production, especially production of PGG₂, PGH₂ and PGE₂, has been a common target of
25 antiinflammatory drug discovery. However, common non-steroidal antiinflammatory drugs (NSAIDs) that are active in reducing the prostaglandin-induced pain and swelling associated with the inflammation process are also active in affecting other
30 prostaglandin-regulated processes not associated with the inflammation process. Thus, use of high doses of most common NSAIDs can produce severe side effects, including life threatening ulcers, that limit their therapeutic potential. An
35 alternative to NSAIDs is the use of corticosteroids, which have even more drastic side effects, especially when long term therapy is involved.

Previous NSAIDs have been found to prevent the production of prostaglandins by inhibiting enzymes in the human arachidonic acid/prostaglandin pathway, including the enzyme
5 cyclooxygenase (COX). The recent discovery of an inducible enzyme associated with inflammation (named "cyclooxygenase-2 (COX-2)" or "prostaglandin G/H synthase II") provides a viable target of inhibition which more
10 effectively reduces inflammation and produces fewer and less drastic side effects.

Recently, there has been significant research into some of the roles of cyclooxygenase-2. It has been found that COX-2
15 is upregulated in benign and malignant tumors (K. Subbaramaiah et al., Proc. Soc. Exp. Biol. Med., 216, 201 (1997)) including lung cancer (T. Hida et al., Anticancer Res., 18, 775-82 (1998)), Barrett's esophagus (K. Wilson, Cancer Res., 58, 2929-34 (1998)) and skin cancer (S. Buckman et al., Carcinogenesis, 19, 723-29 (1998)). It is expressed in airway cells with implication in asthma (P. Barnes et al., Lung Biol. Health Dis., 114, 111-27 (1998)). Cox-2 also has a role in
25 pre-term labor, angiogenesis (M. Tsujii et al. Cell, 93, 705-16 (1998)), vascular rejection (M. Bustos, J. Clin. Invest., 100, 1150-58 (1997)), HIV induced apoptosis (G. Bagetta et al., Biochem. Biophys. Res. Commun., 244, 819-24 (1998)), neurodegeneration (T. Sandhya et al., Brain Res., 788, 223-31 (1998)), inflammatory
30 bowel disease, colitis, (I. Singer et al., Gastroenterology, 115, 297-306 (1998)), cerebral ischemia (S. Nogawa et al., Proc. Natl. Acad. Sci., 95, 10966-71 (1998)), hypertension (A. Nasjletti, Hypertension, 31, 194-200 (1997)), among others.

Drugs that inhibit cyclooxygenase affect colon cancer (T. Kawamori et al., Cancer Res., 58, 409-12 (1998)), allergic neuritis (K. Miyamoto et al., Neuro Report, 9, 2331-4 (1998)),
5 dementia, burn infections (M. Shoup, J. Trauma: Inj., Infec., Crit care, 45, 215-21 (1998)), cytomegalovirus infectivity (E. Speir et al., Circ. Res., 83, 210-16 (1998)), lumbago (H. Bosch, Curr. Med. Res. Opin., 14, 29-38 (1997)),
10 among others.

The references below that disclose antiinflammatory activity, show continuing efforts to find a safe and effective antiinflammatory agent. The novel benzopyran, dihydroquinoline,
15 benzothiopyran and dihydronaphthalene derivatives disclosed herein are such safe and also effective antiinflammatory agents furthering such efforts. The substituted benzopyran, dihydroquinoline, benzothiopyran and dihydronaphthalene derivatives
20 disclosed herein preferably selectively inhibit cyclooxygenase-2 over cyclooxygenase-1.

US Patent No. 5,618,843, to Fisher et al., generically describes acid substituted bicyclic moieties as IIb/IIIA antagonists. WO 94/13659,
25 published June 23, 1994, describes fused benzo compounds for the treatment of CNS disorders. Manrao et al. (J. Indian. Counc. Chem., 12, 38-41 (1996)) describes carboxy coumarinimide derivatives and their antifungal activity. US Patent No.
30 5,348,976, to Shibata et al., describes amide substituted benzopyrans as antifungals.

WO96/40110, published December 19, 1996, describes benzopyran derivatives as tyrosine kinase modulators. Loiodice et al. (Tetrahedron,
35 6, 1001-11 (1995)) describe the preparation of 6-chloro-2,3-dihydro-4H-1-benzopyran carboxylic acids.

Clemence et al. (J. Med. Chem., 31, 1453-62, (1988)) describe 4-hydroxy-3-quinolinecarboxylic acids as starting material in the preparation of antiinflammatories. Lazer, et al. (J. Med. Chem., 40, 980-89 (1997)) describe benzothiopyran carboxylates as starting material in the preparation of antiinflammatories.

US 5,281,720, to Young et al., describes naphthoic acids as lipxygenase inhibitors. US 5,348,976, to Shibata et al., describes amide substituted benzopyrans as antifungals. US 5,004,744, to Weissmiller et al., describes 2H-benzopyran-3-carboxylic acid as an intermediate for pesticides. US 4,814,346, to Albert et al., describes 3-phenylbenzopyrans as 5-lipxygenase inhibitors. US 4,761,425, to Girard and Rokach, describes 4-oxo-benzopyrans as leukotriene antagonists. US 4,609,744, to Young et al., describes 4-oxo-benzopyran-carboxylic acids as leukotriene antagonists. US 5,082,849, to Huang et al., describes 4-oxo-benzopyrans as leukotriene antagonists. WO95/07274, published March 16, 1996, describes 2H-benzopyran-3-carboxylic acid as intermediates. WO88/04654, published June 30, 1988, describes 2H-benzopyran-3-carboxylic acid as intermediates. EP412,939, published February 13, 1991, describes substituted chromenes as 5-lipxygenase inhibitors. JP2-22272 describes benzopyran-3-carboxylic acids. JP59-29681 describes 8-methoxy-benzopyran-3-carboxylic acid as an intermediate. Bunting et al (Can. J. Chem., 62, 1301-07 (1984)) describes the synthesis of 2-hydroxy-1,2-dihydro quinolines. Ukhin et al (Izv. Akad. Nauk. Ser. Khim., 5, 1222-28 (1996)) describe the synthesis of [2-morpholino-6-nitrobenzopyran]-3-carboxylate. Gupta et al. (Indian J. Chem., 21B, 344-347 (1982)) describe chromene-3-carboxylic acid as an intermediate in the preparation of centrally acting muscle relaxants. Rene and Royer (Eur. J. Med. Chem. - Chim. Ther., 10, 72-78 (1975)) describe the preparation

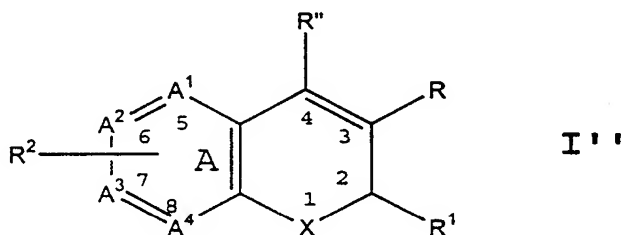
of chromene-3-carboxylic acid. US Patent No. 4,665,202, to Rimbault et al., describes 2-phenyl substituted flavenes and thioflavenes as 5-lipoxygenase inhibitors. U.S. Patent No. 5,250,547, to Lochead et al., describes benzopyran derivatives as 5-lipoxygenase inhibitors. 5 Satoh et al. [J. Med. Chem., 36, 3580-94 (1993)] describe substituted chromenes as 5-lipoxygenase inhibitors. U.S. Patent No. 5,155,130, to Stanton et al. describes substituted chromenes as 5-lipoxygenase 10 inhibitors, and specifically 6-benzyloxy-2H-benzopyran-3-carboxylic acid as an intermediate.

However, compounds of the current invention have not been described as antiinflammatory cyclooxygenase inhibitors.

15

DESCRIPTION OF THE INVENTION

A class of compounds useful in treating cyclooxygenase-2 medicated disorders is defined by 20 Formula I'':



wherein X is selected from O, S, CR^cR^b and NR^a;
 25 wherein R^a is selected from hydrido, C₁-C₃-alkyl, (optionally substituted phenyl)-C₁-C₃-alkyl, acyl and carboxy-C₁-C₆-alkyl;
 wherein each of R^b and R^c is independently selected from hydrido, C₁-C₃-alkyl, phenyl-C₁-C₃-alkyl, C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl; or wherein CR^bR^c forms a 3-30 6 membered cycloalkyl ring;

- wherein R is selected from carboxyl, aminocarbonyl, C₁-C₆-alkylsulfonylaminocarbonyl and C₁-C₆-alkoxycarbonyl;
- 5 wherein R" is selected from hydrido, phenyl, thienyl, C₁-C₆-alkyl and C₂-C₆-alkenyl;
- wherein R¹ is selected from C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl;
- 10 wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl, aryl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, methylenedioxy, C₁-C₆-alkylthio, C₁-C₆-alkylsulfinyl, aryloxy, arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-alkoxy-C₁-C₆-alkyl, aryl-C₁-C₆-alkyloxy, heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-haloalkylthio, C₁-C₆-haloalkylsulfinyl, C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl, hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino, arylamino, aryl-C₁-C₆-alkylamino, heteroarylamino, heteroaryl-C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, C₁-C₆-alkylaminosulfonyl, arylaminosulfonyl, heteroarylaminosulfonyl, aryl-C₁-C₆-alkylaminosulfonyl, heteroaryl-C₁-C₆-alkylaminosulfonyl, heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-alkylsulfonyl, optionally substituted aryl, optionally substituted heteroaryl, aryl-C₁-C₆-alkylcarbonyl, heteroaryl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, arylcarbonyl, aminocarbonyl, C₁-C₆-alkoxycarbonyl, formyl, C₁-C₆-
- 15
- 20
- 25
- 30
- 35

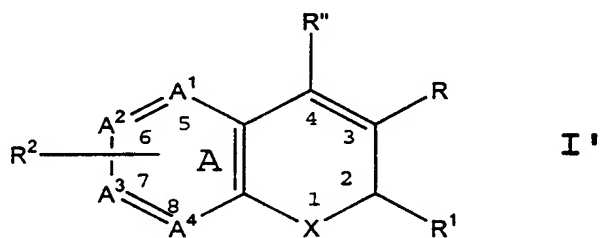
haloalkylcarbonyl and C₁-C₆-alkylcarbonyl;
and

wherein the A ring atoms A¹, A², A³ and A⁴ are
independently selected from carbon and
5 nitrogen with the proviso that at least two
of A¹, A², A³ and A⁴ are carbon;

or wherein R² together with ring A forms a
radical selected from naphthyl, quinolyl,
isoquinolyl, quinoliziny, quinoxaliny and
10 dibenzofuryl;

or an isomer or pharmaceutically acceptable salt
thereof.

A related class of compounds useful in treating
15 cyclooxygenase-2 mediated disorders is defined by
Formula I':



20 wherein X is selected from O, S, CR^cR^b and NR^a;
wherein R^a is selected from hydrido, C₁-C₃-alkyl,
(optionally substituted phenyl)-C₁-C₃-alkyl,
alkylsulfonyl, phenylsulfonyl, benzylsulfonyl, acyl and
carboxy-C₁-C₆-alkyl;

25 wherein each of R^b and R^c is independently selected
from hydrido, C₁-C₃-alkyl, phenyl-C₁-C₃-alkyl, C₁-C₃-
perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy,
nitro, cyano and cyano-C₁-C₃-alkyl;

or wherein CR^cR^b form a cyclopropyl ring;

30 wherein R is selected from carboxyl, aminocarbonyl,
C₁-C₆-alkylsulfonylaminocarbonyl and C₁-C₆-
alkoxycarbonyl;

wherein R" is selected from hydrido, phenyl, thienyl, C₂-C₆-alkynyl and C₂-C₆-alkenyl;

wherein R¹ is selected from C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and
5 cyano-C₁-C₃-alkyl;

wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl, aryl-C₂-C₆-alkenyl, C₁-C₆-alkoxy,
10 methylenedioxy, C₁-C₆-alkylthio, C₁-C₆-alkylsulfinyl, -O(CF₂)₂O-, aryloxy, arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-alkoxy-C₁-C₆-alkyl, aryl-C₁-C₆-alkyloxy, heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-
15 haloalkylthio, C₁-C₆-haloalkylsulfinyl, C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl, hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino, arylamino, aryl-C₁-C₆-alkylamino, heteroarylamino, heteroaryl-C₁-C₆-alkylamino, nitro,
20 cyano, amino, aminosulfonyl, C₁-C₆-alkylaminosulfonyl, arylaminosulfonyl, heteroarylaminosulfonyl, aryl-C₁-C₆-alkylaminosulfonyl, heteroaryl-C₁-C₆-alkylaminosulfonyl, heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-alkylsulfonyl, optionally substituted aryl, optionally
25 substituted heteroaryl, aryl-C₁-C₆-alkylcarbonyl, heteroaryl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, arylcarbonyl, aminocarbonyl, C₁-C₆-alkoxycarbonyl, formyl, C₁-C₆-haloalkylcarbonyl and C₁-C₆-alkylcarbonyl;
and

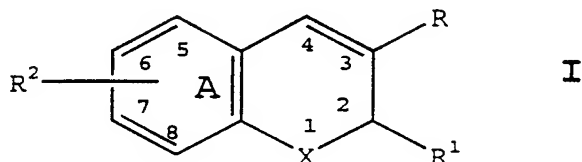
30 wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least two of A¹, A², A³ and A⁴ are carbon;

or wherein R² together with ring A forms a radical
35 selected from naphthyl, quinolyl, isoquinolyl, quinoliziny, quinoxaliny and dibenzofuryl;

or an isomer or pharmaceutically acceptable salt thereof.

A related class of compounds useful in treating cyclooxygenase-2 mediated disorders is defined by Formula I:

5



wherein X is selected from O or S or NR^a;

wherein R^a is alkyl;

10 wherein R is selected from carboxyl, aminocarbonyl, alkylsulfonylaminocarbonyl and alkoxycarbonyl;

 wherein R¹ is selected from haloalkyl, alkyl, aralkyl, cycloalkyl and aryl optionally substituted with one or more radicals selected from alkylthio, nitro and alkylsulfonyl; and

15 wherein R² is one or more radicals selected from hydrido, halo, alkyl, aralkyl, alkoxy, aryloxy, heteroaryloxy, aralkyloxy, heteroaralkyloxy, haloalkyl, haloalkoxy, alkylamino, arylamino, aralkylamino, heteroarylamine, heteroarylalkylamine, nitro, amino, aminosulfonyl, alkylaminosulfonyl, arylaminosulfonyl, heteroarylaminosulfonyl, aralkylaminosulfonyl, heteroaralkylaminosulfonyl, heterocyclosulfonyl, alkylsulfonyl, optionally substituted aryl, optionally substituted heteroaryl, aralkylcarbonyl, heteroarylcarbonyl, arylcarbonyl, aminocarbonyl, and alkylcarbonyl;

25 or wherein R² together with ring A forms a naphthyl radical;

30 or an isomer or pharmaceutically acceptable salt thereof.

 Compounds of the present invention would be useful for, but not limited to, the treatment of inflammation in a subject, and for treatment of other cyclooxygenase-2 mediated disorders, such as, as an analgesic in the

35

treatment of pain and headaches, or as an antipyretic for the treatment of fever. For example, compounds of the invention would be useful to treat arthritis, including but not limited to rheumatoid arthritis, spondyloarthropathies, gouty arthritis, osteoarthritis, systemic lupus erythematosus and juvenile arthritis. Such compounds of the invention would be useful in the treatment of asthma, bronchitis, menstrual cramps, preterm labor, tendinitis, bursitis, allergic neuritis, cytomegalovirus infectivity, apoptosis including HIV induced apoptosis, lumbago, liver disease including hepatitis, skin-related conditions such as psoriasis, eczema, acne, UV damage, burns and dermatitis, and from post-operative inflammation including from ophthalmic surgery such as cataract surgery and refractive surgery. Compounds of the invention also would be useful to treat gastrointestinal conditions such as inflammatory bowel disease, Crohn's disease, gastritis, irritable bowel syndrome and ulcerative colitis. Compounds of the invention would be useful in treating inflammation in such diseases as migraine headaches, periarteritis nodosa, thyroiditis, aplastic anemia, Hodgkin's disease, scleroderma, rheumatic fever, type I diabetes, neuromuscular junction disease including myasthenia gravis, white matter disease including multiple sclerosis, sarcoidosis, nephrotic syndrome, Behcet's syndrome, polymyositis, gingivitis, nephritis, hypersensitivity, swelling occurring after injury including brain edema, myocardial ischemia, and the like. The compounds would also be useful in the treatment of ophthalmic diseases, such as retinitis, conjunctivitis, retinopathies, uveitis, ocular photophobia, and of acute injury to the eye tissue. The compounds would also be useful in the treatment of pulmonary inflammation, such as that associated with viral infections and cystic fibrosis, and in bone reorption such as associated with osteoporosis.

The compounds would also be useful for the treatment of certain central nervous system disorders, such as cortical dementias including Alzheimer's disease, neurodegeneration, and central nervous system damage resulting from stroke, ischemia and trauma. The term "treatment" includes partial or total inhibition of the dementia, including Alzheimer's disease, vascular dementia, multi-infarct dementia, pre-senile dementia, alcoholic dementia, and senile dementia.

The compounds of the invention are useful as anti-inflammatory agents, such as for the treatment of arthritis, with the additional benefit of having significantly less harmful side effects. These compounds would also be useful in the treatment of allergic rhinitis, respiratory distress syndrome, endotoxin shock syndrome, and liver disease. The compounds would also be useful in the treatment of pain, but not limited to postoperative pain, dental pain, muscular pain, and pain resulting from cancer.

The method above would be useful for, but not limited to, treating and preventing inflammation-related cardiovascular disorders in a subject. The method would be useful for treatment and prevention of vascular diseases, coronary artery disease, aneurysm, vascular rejection, arteriosclerosis, atherosclerosis including cardiac transplant atherosclerosis, myocardial infarction, embolism, stroke, thrombosis, including venous thrombosis, angina including unstable angina, coronary plaque inflammation, bacterial-induced inflammation including *Chlamydia*-induced inflammation, viral induced inflammation, and inflammation associated with surgical procedures such as vascular grafting including coronary artery bypass surgery, revascularization procedures including angioplasty, stent placement, endarterectomy, or other invasive procedures involving arteries, veins and capillaries.

The compounds would be useful for, but not limited to, the treatment of angiogenesis-related disorders in a

subject. According to the present invention, the compounds can be administered to a subject in need of angiogenesis inhibition. The method would be useful for treatment of neoplasia, including metastasis;
5 ophthalmological conditions such as corneal graft rejection, ocular neovascularization, retinal neovascularization including neovascularization following injury or infection, diabetic retinopathy, macular degeneration, retrolental fibroplasia and
10 neovascular glaucoma; ulcerative diseases such as gastric ulcer; pathological, but non-malignant, conditions such as hemangiomas, including infantile hemangiomas, angiofibroma of the nasopharynx and avascular necrosis of bone; and disorders of the female
15 reproductive system such as endometriosis.

Compounds of the invention would be useful for the prevention or treatment of benign and malignant tumors/neoplasia including cancer, such as colorectal cancer, brain cancer, bone cancer, epithelial cell-
20 derived neoplasia (epithelial carcinoma) such as basal cell carcinoma, adenocarcinoma, gastrointestinal cancer such as lip cancer, mouth cancer, esophageal cancer, small bowel cancer and stomach cancer, colon cancer, liver cancer, bladder cancer, pancreas cancer, ovary
25 cancer, cervical cancer, lung cancer, breast cancer and skin cancer, such as squamous cell and basal cell cancers, prostate cancer, renal cell carcinoma, and other known cancers that effect epithelial cells throughout the body. Preferably, neoplasia is selected
30 from gastrointestinal cancer, Barrett's esophagus, liver cancer, bladder cancer, pancreas cancer, ovary cancer, prostate cancer, cervical cancer, lung cancer, breast cancer and skin cancer, such as squamous cell and basal cell cancers. The compounds can also be used to treat
35 the fibrosis which occurs with radiation therapy. The method can be used to treat subjects having adenomatous polyps, including those with familial adenomatous polyposis (FAP). Additionally, the method can be used

to prevent polyps from forming in patients at risk of FAP.

The administration of compounds of the present invention may be used alone or in conjunction with additional therapies known to those skilled in the art in the prevention or treatment of neoplasia. Alternatively, the compounds described herein may be used in conjunctive therapy. By way of example, the compounds may be administered alone or in conjunction with other antineoplastic agents or other growth inhibiting agents or other drugs or nutrients.

There are large numbers of antineoplastic agents available in commercial use, in clinical evaluation and in pre-clinical development, which could be selected for treatment of neoplasia by combination drug chemotherapy. Such antineoplastic agents fall into several major categories, namely, antibiotic-type agents, alkylating agents, antimetabolite agents, hormonal agents, immunological agents, interferon-type agents and a category of miscellaneous agents. Alternatively, other anti-neoplastic agents, such as metallomatrix proteases (MMP), SOD mimics or $\alpha_2\beta_1$ inhibitors may be used.

A first family of antineoplastic agents which may be used in combination with compounds of the present invention consists of antimetabolite-type antineoplastic agents. Suitable antimetabolite antineoplastic agents may be selected from the group consisting of 5-FU-fibrinogen, acanthifolic acid, aminothiadiazone, brequinar sodium, carmofur, Ciba-Geigy CGP-30694, cyclopentyl cytosine, cytarabine phosphate stearate, cytarabine conjugates, Lilly DATHF, Merrel Dow DDFC, dezaguanine, dideoxycytidine, dideoxyguanosine, didox, Yoshitomi DMDC, doxifluridine, Wellcome EHNA, Merck & Co. EX-015, fazarabine, floxuridine, fludarabine phosphate, 5-fluorouracil, N-(2'-furanidyl)-5-fluorouracil, Daiichi Seiyaku FO-152, isopropyl pyrrolizine, Lilly LY-188011, Lilly LY-264618, methobenzaprim, methotrexate, Wellcome MZPES,

norspermidine, NCI NSC-127716, NCI NSC-264880, NCI NSC-39661, NCI NSC-612567, Warner-Lambert PALA, pentostatin, piritrexim, plicamycin, Asahi Chemical PL-AC, Takeda TAC-788, thioguanine, tiazofurin, Erbamont TIF, 5 trimetrexate, tyrosine kinase inhibitors, tyrosine protein kinase inhibitors, Taiho UFT and uricytin.

A second family of antineoplastic agents which may be used in combination with compounds of the present invention consists of alkylating-type antineoplastic 10 agents. Suitable alkylating-type antineoplastic agents may be selected from the group consisting of Shionogi 254-S, aldo-phosphamide analogues, altretamine, anaxirone, Boehringer Mannheim BBR-2207, bestrabucil, budotitane, Wakunaga CA-102, carboplatin, carmustine, 15 Chinoin-139, Chinoin-153, chlorambucil, cisplatin, cyclophosphamide, American Cyanamid CL-286558, Sanofi CY-233, cyplatate, Degussa D-19-384, Sumimoto DACHP(Myrr)2, diphenylspiromustine, diplatinum cytostatic, Erba distamycin derivatives, Chugai DWA- 20 2114R, ITI E09, elmustine, Erbamont FCE-24517, estramustine phosphate sodium, fotemustine, Unimed G-6-M, Chinoin GYKI-17230, hepsul-fam, ifosfamide, iproplatin, lomustine, mafosfamide, mitolactol, Nippon Kayaku NK-121, NCI NSC-264395, NCI NSC-342215, 25 oxaliplatin, Upjohn PCNU, prednimustine, Proter PTT-119, ranimustine, semustine, SmithKline SK&F-101772, Yakult Honsha SN-22, spiromustine, Tanabe Seiyaku TA-077, tauromustine, temozolomide, teroxirone, tetraplatin and trimelamol.

30 A third family of antineoplastic agents which may be used in combination with compounds of the present invention consists of antibiotic-type antineoplastic agents. Suitable antibiotic-type antineoplastic agents may be selected from the group consisting of Taiho 4181- 35 A, aclarubicin, actinomycin D, actinoplanone, Erbamont ADR-456, aerophysinin derivative, Ajinomoto AN-201-II, Ajinomoto AN-3, Nippon Soda anisomycins, anthracycline, azino-mycin-A, bisucaberin, Bristol-Myers BL-6859,

Bristol-Myers BMY-25067, Bristol-Myers BMY-25551,
Bristol-Myers BMY-26605, Bristol-Myers BMY-27557,
Bristol-Myers BMY-28438, bleomycin sulfate, bryostatin-
1, Taiho C-1027, caliche mycin, chromoximycin,
5 dactinomycin, daunorubicin, Kyowa Hakko DC-102, Kyowa
Hakko DC-79, Kyowa Hakko DC-88A, Kyowa Hakko DC89-A1,
Kyowa Hakko DC92-B, ditrisarubicin B, Shionogi DOB-41,
doxorubicin, doxorubicin-fibrinogen, elsamicin-A,
epirubicin, erbstatin, esorubicin, esperamicin-A1,
10 esperamicin-Alb, Erbamont FCE-21954, Fujisawa FK-973,
fostriecin, Fujisawa FR-900482, glidobactin, gregatin-A,
grincamycin, herbimycin, idarubicin, illudins,
kazusamycin, kesarirhodins, Kyowa Hakko KM-5539, Kirin
Brewery KRN-8602, Kyowa Hakko KT-5432, Kyowa Hakko KT-
15 5594, Kyowa Hakko KT-6149, American Cyanamid LL-D49194,
Meiji Seika ME 2303, menogaril, mitomycin, mitoxantrone,
SmithKline M-TAG, neoenactin, Nippon Kayaku NK-313,
Nippon Kayaku NKT-01, SRI International NSC-357704,
oxalysine, oxaunomycin, peplomycin, pilatin,
20 pirarubicin, porothramycin, pyrindamycin A, Tobishi RA-
I, rapamycin, rhizoxin, rodorubicin, sibanomycin,
siwenmycin, Sumitomo SM-5887, Snow Brand SN-706, Snow
Brand SN-07, sorangicin-A, sparsomycin, SS
Pharmaceutical SS-21020, SS Pharmaceutical SS-7313B, SS
25 Pharmaceutical SS-9816B, steffimycin B, Taiho 4181-2,
talisomycin, Takeda TAN-868A, terpentecin, thrazine,
tricrozarin A, Upjohn U-73975, Kyowa Hakko UCN-10028A,
Fujisawa WF-3405, Yoshitomi Y-25024 and zorubicin.

A fourth family of antineoplastic agents which may
30 be used in combination with compounds of the present
invention consists of a miscellaneous family of
antineoplastic agents selected from the group consisting
of alpha-carotene, alpha-difluoromethyl-arginine,
acitretin, Biotec AD-5, Kyorin AHC-52, alstonine,
35 amonafide, amphetamine, amsacrine, Angiostat,
ankinomycin, anti-neoplaston A10, antineoplaston A2,
antineoplaston A3, antineoplaston A5, antineoplaston
AS2-1, Henkel APD, aphidicolin glycinate, asparaginase,

Avarol, baccharin, batracylin, benfluron, benzotript, Ipsen-Beaufour BIM-23015, bisantrene, Bristo-Myers BMY-40481, Vestar boron-10, bromofosfamide, Wellcome BW-502, Wellcome BW-773, caracemide, carmethizole hydrochloride, 5 Ajinomoto CDAF, chlorsulfaquinoxalone, Chemes CHX-2053, Chemex CHX-100, Warner-Lambert CI-921, Warner-Lambert CI-937, Warner-Lambert CI-941, Warner-Lambert CI-958, clanfenur, claviridenone, ICN compound 1259, ICN compound 4711, Contracan, Yakult Honsha CPT-11, 10 crisnatol, curaderm, cytochalasin B, cytarabine, cytocytin, Merz D-609, DABIS maleate, dacarbazine, datelliptinium, didemnin-B, dihaematoporphyrin ether, dihydrolenperone, dinaline, distamycin, Toyo Pharmar DM-341, Toyo Pharmar DM-75, Daiichi Seiyaku DN-9693, 15 elliprabin, elliptinium acetate, Tsumura EPMTTC, ergotamine, etoposide, etretinate, fenretinide, Fujisawa FR-57704, gallium nitrate, genkwadaphnin, Chugai GLA-43, Glaxo GR-63178, grifolan NMF-5N, hexadecylphosphocholine, Green Cross HO-221, 20 homoharringtonine, hydroxyurea, BTG ICRF-187, ilmofofosine, isoglutamine, isotretinoin, Otsuka JI-36, Ramot K-477, Otsuak K-76COONa, Kureha Chemical K-AM, MECT Corp KI-8110, American Cyanamid L-623, leukoregulin, lonidamine, Lundbeck LU-23-112, Lilly LY-25 186641, NCI (US) MAP, marycin, Merrel Dow MDL-27048, Medco MEDR-340, merbarone, merocyanine derivatives, methylanilinoacridine, Molecular Genetics MGI-136, minactivin, mitonafide, mitoquidone, mopidamol, motretinide, Zenyaku Kogyo MST-16, N-(retinoyl)amino 30 acids, Nisshin Flour Milling N-021, N-acylated-dehydroalanines, nafazatrom, Taisho NCU-190, nocodazole derivative, Normosang, NCI NSC-145813, NCI NSC-361456, NCI NSC-604782, NCI NSC-95580, octreotide, Ono ONO-112, oquizanocine, Akzo Org-10172, pancratistatin, 35 pazelliptine, Warner-Lambert PD-111707, Warner-Lambert PD-115934, Warner-Lambert PD-131141, Pierre Fabre PE-1001, ICRT peptide D, piroxantrone, polyhaematoporphyrin, polypreic acid, Efamol porphyrin,

probimane, procarbazine, proglumide, Invitron protease
nexin I, Tobishi RA-700, razoxane, Sapporo Breweries
RBS, restrictin-P, retelliptine, retinoic acid, Rhone-
Poulenc RP-49532, Rhone-Poulenc RP-56976, SmithKline
5 SK&F-104864, Sumitomo SM-108, Kuraray SMANCS, SeaPharm
SP-10094, spatol, spirocyclopropane derivatives,
spirogermanium, Unimed, SS Pharmaceutical SS-554,
strypoldinone, Stypoldione, Suntory SUN 0237, Suntory
SUN 2071, superoxide dismutase, Toyama T-506, Toyama T-
10 680, taxol, Teijin TEI-0303, teniposide, thaliblastine,
Eastman Kodak TJB-29, tocotrienol, Topostin, Teijin TT-
82, Kyowa Hakko UCN-01, Kyowa Hakko UCN-1028, ukrain,
Eastman Kodak USB-006, vinblastine sulfate, vincristine,
vindesine, vinestramide, vinorelbine, vintriptol,
15 vinzolidine, withanolides and Yamanouchi YM-534.

Examples of radioprotective agents which may be
used in combination with compounds of the present
invention are AD-5, adchnon, amifostine analogues,
detox, dimesna, l-102, MM-159, N-acylated-
20 dehydroalanines, TGF- Genentech, tiprotimod, amifostine,
WR-151327, FUT-187, ketoprofen transdermal, nabumetone,
superoxide dismutase (Chiron) and superoxide dismutase
Enzon.

Besides being useful for human treatment, these
25 compounds are also useful for veterinary treatment
of companion animals, exotic animals and farm
animals, including mammals, rodents, and the like.
More preferred animals include horses, dogs, and
cats.

30 The present compounds may also be used in
co-therapies, partially or completely, in
addition to other antiinflammatories, such as
together with steroids, NSAIDs, iNOS inhibitors,
p-38 inhibitors, TNF inhibitors, 5-lipoxygenase
35 inhibitors, LTB₄ receptor antagonists and LTA₄
hydrolase inhibitors.

Suitable LTA₄ hydrolase inhibitors include
RP-64966, (S,S)-3-amino-4-(4-benzyloxyphenyl) -

2-hydroxybutyric acid benzyl ester (Scripps Res. Inst.), N-(2(R)-(cyclohexylmethyl)-3-(hydroxycarbamoyl)propionyl)-L-alanine (Searle), 7-(4-(4-ureidobenzyl)phenyl)heptanoic acid
5 (Rhone-Poulenc Rorer), and 3-(3-(1E,3E-tetradecadienyl)-2-oxiranyl)benzoic acid lithium salt (Searle).

Suitable LTB₄ receptor antagonists include, among others, ebselen, linazolast, ontazolast,
10 Bayer Bay-x-1005, Ciba Geigy compound CGS-25019C, Leo Denmark compound ETH-615, Merck compound MAFP, Terumo compound TMK-688, Tanabe compound T-0757, Lilly compounds LY-213024, LY-210073, LY223982, LY233469, and LY255283, LY-
15 293111, 264086 and 292728, ONO compounds ONO-LB457, ONO-4057, and ONO-LB-448, Shionogi compound S-2474, calcitrol, Lilly compounds Searle compounds SC-53228, SC-41930, SC-50605 and SC-51146, Warner Lambert compound BPC 15,
20 SmithKline Beecham compound SB-209247 and SK&F compound SKF-104493. Preferably, the LTB₄ receptor antagonists are selected from calcitrol, ebselen, Bayer Bay-x-1005, Ciba Geigy compound CGS-25019C, Leo Denmark compound ETH-
25 615, Lilly compound LY-293111, Ono compound ONO-4057, and Terumo compound TMK-688.

Suitable 5-LO inhibitors include, among others, Abbott compounds A-76745, 78773 and ABT761, Bayer Bay-x-1005, Cytomed CMI-392, Eisai
30 E-3040, Scotia Pharmaceutica EF-40, Fujirebio F-1322, Merckle ML-3000, Purdue Frederick PF-5901, 3M Pharmaceuticals R-840, rilopirox, flobufen, linasolast, lonapolene, masoprocol, ontasolast, tenidap, zileuton, pranlukast, tepoxalin,
35 rilopirox, flezelastine hydrochloride, enazadrem phosphate, and bunaprolast.

The present compounds may also be used in combination therapies with opioids and other

analgesics, including narcotic analgesics, Mu receptor antagonists, Kappa receptor antagonists, non-narcotic (i.e. non-addictive) analgesics, monoamine uptake inhibitors, adenosine regulating
5 agents, cannabinoid derivatives, Substance P antagonists, neurokinin-1 receptor antagonists and sodium channel blockers, among others. More preferred would be combinations with compounds selected from morphine, meperidine, codeine,
10 pentazocine, buprenorphine, butorphanol, dezocine, meptazinol, hydrocodone, oxycodone, methadone, Tramadol [(+) enantiomer], DuP 747, Dynorphine A, Enadoline, RP-60180, HN-11608, E-2078, ICI-204448, acetaminophen (paracetamol), propoxyphene,
15 nalbuphine, E-4018, filenadol, mirfentanil, amitriptyline, DuP631, Tramadol [(-) enantiomer], GP-531, acadesine, AKI-1, AKI-2, GP-1683, GP-3269, 4030W92, tramadol racemate, Dynorphine A, E-2078, AXC3742, SNX-111, ADL2-1294, ICI-204448, CT-3, CP-
20 99,994, and CP-99,994.

The compounds can be used in co-therapies, in place of other conventional antiinflammatories, in combination with one or more antihistamines, decongestants, diuretics, antitussive agents or
25 with other agents previously known to be effective in combination with antiinflammatory agents.

The term "prevention" includes either preventing the onset of clinically evident cardiovascular disorders altogether or preventing the onset of a preclinically
30 evident stage of cardiovascular disorder in individuals. This includes prophylactic treatment of those at risk of developing a disease, such as a cardiovascular disorder, dementia or cancer, for example.

The phrase "therapeutically-effective" is intended
35 to qualify the amount of each agent which will achieve the goal of improvement in disorder severity and the frequency of incidence over treatment of each agent by

itself, while avoiding adverse side effects typically associated with alternative therapies.

The present invention preferably includes compounds which selectively inhibit
5 cyclooxygenase-2 over cyclooxygenase-1.
Preferably, the compounds have a cyclooxygenase-2 IC_{50} of less than about 0.5 μM , and also have a selectivity ratio of cyclooxygenase-2 inhibition over cyclooxygenase-1 inhibition of at least 50,
10 and more preferably of at least 100. Even more preferably, the compounds have a cyclooxygenase-1 IC_{50} of greater than about 5 μM . Such preferred selectivity may indicate an ability to reduce the incidence of common NSAID-induced side effects.

15 A preferred class of compounds consists of those compounds of Formula I wherein X is oxygen or sulfur; wherein R is selected from carboxyl, lower alkyl, lower aralkyl and lower alkoxycarbonyl; wherein R^1 is selected from lower haloalkyl, lower cycloalkyl and phenyl; and
20 wherein R^2 is one or more radicals selected from hydrido, halo, lower alkyl, lower alkoxy, lower haloalkyl, lower haloalkoxy, lower alkylamino, nitro, amino, aminosulfonyl, lower alkylaminosulfonyl, 5- or 6-membered heteroarylalkylaminosulfonyl, lower
25 aralkylaminosulfonyl, 5- or 6- membered nitrogen containing heterocyclosulfonyl, lower alkylsulfonyl, optionally substituted phenyl, lower aralkylcarbonyl, and lower alkylcarbonyl; or wherein R^2 together with ring A forms a naphthyl radical; or an isomer or
30 pharmaceutically acceptable salt thereof.

A more preferred class of compounds consists of those compounds of Formula I wherein X is oxygen or sulfur; wherein R is selected from carboxyl; wherein R^1 is selected from lower haloalkyl; and wherein R^2 is one
35 or more radicals selected from hydrido, halo, lower alkyl, lower haloalkyl, lower haloalkoxy, lower alkylamino, amino, aminosulfonyl, lower alkylaminosulfonyl, 5- or 6- membered

heteroarylalkylaminosulfonyl, lower
aralkylaminosulfonyl, lower alkylsulfonyl, 6- membered
nitrogen containing heterocyclosulfonyl, optionally
substituted phenyl, lower aralkylcarbonyl, and lower
5 alkylcarbonyl; or wherein R^2 together with ring A forms
a naphthyl radical; or an isomer or pharmaceutically
acceptable salt thereof.

An even more preferred class of compounds consists
of those compounds of Formula I wherein R is carboxyl;
10 wherein R^1 is selected from fluoromethyl, chloromethyl,
dichloromethyl, trichloromethyl, pentafluoroethyl,
heptafluoropropyl, difluoroethyl, difluoropropyl,
dichloroethyl, dichloropropyl, difluoromethyl, and
trifluoromethyl; and wherein R^2 is one or more radicals
15 selected from hydrido, chloro, fluoro, bromo, iodo,
methyl, ethyl, isopropyl, tert-butyl, butyl, isobutyl,
pentyl, hexyl, methoxy, ethoxy, isopropoxy,
tertbutyloxy, trifluoromethyl, difluoromethyl,
trifluoromethoxy, amino, N,N-dimethylamino, N,N-
20 diethylamino, N-phenylmethylaninosulfonyl, N-
phenylethylaninosulfonyl, N-(2-
furylmethyl)aminosulfonyl, nitro, N,N-
dimethylaninosulfonyl, aminosulfonyl, N-
methylaninosulfonyl, N-ethylsulfonyl, 2,2-
25 dimethylethylaninosulfonyl, N,N-dimethylaninosulfonyl,
N-(2-methylpropyl)aminosulfonyl, N-morpholinosulfonyl,
methylsulfonyl, benzylcarbonyl, 2,2-
dimethylpropylcarbonyl, phenylacetyl and phenyl; or
wherein R^2 together with ring A forms a naphthyl
30 radical; or an isomer or pharmaceutically acceptable
salt thereof.

An even more preferred class of compounds consists
of those compounds of Formula I wherein R is carboxyl;
wherein R^1 is trifluoromethyl or pentafluoroethyl; and
35 wherein R^2 is selected from one or more radicals hydrido,
chloro, fluoro, bromo, iodo, methyl, ethyl, isopropyl,
tert-butyl, methoxy, trifluoromethyl, trifluoromethoxy,
N-phenylmethylaninosulfonyl, N-phenylethylaninosulfonyl,

N-(2-furylmethyl)aminosulfonyl, N,N-dimethylaminosulfonyl, N-methylaminosulfonyl, N-(2,2-dimethylethyl)aminosulfonyl, dimethylaminosulfonyl, 2-methylpropylaminosulfonyl, N-morpholinosulfonyl,
 5 methylsulfonyl, benzylcarbonyl, and phenyl; or wherein R² together with ring A forms a naphthyl radical; or an isomer or pharmaceutically acceptable salt thereof.

A preferred class of compounds consists of those compounds of Formula I' wherein X is selected from O, S,
 10 CR^cR^d and NR^a; wherein R^a is selected from hydrido, C₁-C₃-alkyl, (optionally substituted phenyl)-C₁-C₃-alkyl, acyl and carboxy-C₁-C₆-alkyl; wherein each of R^b and R^b is independently selected from hydrido, C₁-C₃-alkyl, phenyl-C₁-C₃-alkyl, C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-
 15 alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl; wherein R is selected from carboxyl, aminocarbonyl, C₁-C₆-alkylsulfonylaminocarbonyl and C₁-C₆-alkoxycarbonyl; wherein R' is selected from hydrido, phenyl, thienyl and C₂-C₆-alkenyl; wherein R¹ is selected
 20 from C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl; wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl,
 25 aryl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, methylenedioxy, C₁-C₆-alkylthio, C₁-C₆-alkylsulfinyl, aryloxy, arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-alkoxy-C₁-C₆-alkyl, aryl-C₁-C₆-alkyloxy, heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy,
 30 C₁-C₆-haloalkylthio, C₁-C₆-haloalkylsulfinyl, C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl, hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino, arylamino, aryl-C₁-C₆-alkylamino, heteroarylamino, heteroaryl-C₁-C₆-alkylamino, nitro,
 35 cyano, amino, aminosulfonyl, C₁-C₆-alkylaminosulfonyl, arylaminosulfonyl, heteroarylaminosulfonyl, aryl-C₁-C₆-alkylaminosulfonyl, heteroaryl-C₁-C₆-alkylaminosulfonyl, heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-

alkylsulfonyl, optionally substituted aryl, optionally substituted heteroaryl, aryl-C₁-C₆-alkylcarbonyl, heteroaryl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, arylcarbonyl, aminocarbonyl, C₁-C₆-alkoxycarbonyl, 5 formyl, C₁-C₆-haloalkylcarbonyl and C₁-C₆-alkylcarbonyl; and wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a 10 naphthyl or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

A more preferred class of compounds consists of those compounds of Formula I' wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, C₁- 15 C₃-alkyl and (optionally substituted phenyl)methyl; wherein R' is selected from hydrido and C₂-C₆-alkenyl; wherein R is carboxyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, 20 C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, phenyl-C₁-C₆-alkyl, phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₃-alkoxy, methylenedioxy, C₁-C₃-alkoxy-C₁-C₃-alkyl, C₁-C₃-alkylthio, C₁-C₃-alkylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, C₁-C₃-haloalkyl-C₁-C₃-hydroxyalkyl, 25 phenyl-C₁-C₃-alkyloxy-C₁-C₃-alkyl, C₁-C₃-haloalkyl, C₁-C₃-haloalkoxy, C₁-C₃-haloalkylthio, C₁-C₃-hydroxyalkyl, C₁-C₃-alkoxy-C₁-C₃-alkyl, hydroxyimino-C₁-C₃-alkyl, C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, N-alkylaminosulfonyl, N-arylaminosulfonyl, N- 30 heteroarylaminosulfonyl, N-(phenyl-C₁-C₆-alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, phenyl-C₁-C₃-alkylsulfonyl, 5- to 8-membered heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, optionally substituted phenyl, optionally substituted 5- 35 to 9-membered heteroaryl, phenyl-C₁-C₆-alkylcarbonyl, phenylcarbonyl, 4-chlorophenylcarbonyl, 4-hydroxyphenylcarbonyl, 4-trifluoromethylphenylcarbonyl, 4-methoxyphenylcarbonyl, aminocarbonyl, formyl, and C₁-

C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl, benzofurylphenyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds consists of those compounds of Formula I' wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, methyl, ethyl, (4-trifluoromethyl)benzyl, (4-chloromethyl)benzyl, (4-methoxy)benzyl, and (4-cyano)benzyl, (4-nitro)benzyl; wherein R is carboxyl; wherein R' is selected from hydrido and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethoxy, trifluoromethylthio, hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl, phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, optionally substituted phenylcarbonyl, aminocarbonyl, formyl and methylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen

with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

- 5 Within Formula I' there is a subclass of chromene compounds wherein X is O; wherein R is carboxyl; wherein R'' is selected from hydrido and C₂-C₆-alkenyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals independently selected from
- 10 hydrido, halo, C₁-C₆-alkyl, phenyl-C₁-C₆-alkyl, phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, phenyloxy, 5- or 6-membered heteroaryloxy, phenyl-C₁-C₆-alkyloxy, 5- or 6-membered heteroaryl-C₁-C₆-alkyloxy, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, N-(C₁-C₆-alkyl)amino, N,N-di-(C₁-C₆-
- 15 alkyl)amino, N-phenylamino, N-(phenyl-C₁-C₆-alkyl)amino, N-heteroarylamino, N-(heteroaryl-C₁-C₆-alkyl)amino, nitro, amino, aminosulfonyl, N-(C₁-C₆-alkyl)aminosulfonyl, N,N-di-(C₁-C₆-alkyl)aminosulfonyl, N-arylaminosulfonyl, N-heteroarylaminosulfonyl, N-(phenyl-C₁-C₆-
- 20 alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, 5- to 8-membered heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, optionally substituted phenyl, optionally substituted 5- or 6-membered heteroaryl, phenyl-C₁-C₆-alkylcarbonyl,
- 25 heteroarylcarbonyl, phenylcarbonyl, aminocarbonyl, and C₁-C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or an isomer or pharmaceutically
- 30 acceptable salt thereof.

- An even more preferred class of compounds consists of those compounds of Formula I' wherein X is O; wherein R is carboxyl; wherein R'' is selected from hydrido and ethenyl; wherein R¹ is selected from
- 35 trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-

dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, pyridyloxy, thienyloxy, furyloxy, phenylmethoxy, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethoxy, trifluoromethylthio, hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, N-phenylamino, N-(benzyl)amino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl, phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, furylcarbonyl, phenylcarbonyl, aminocarbonyl, formyl, and methylcarbonyl; and wherein one of the A ring atoms A¹, A², A³ and A⁴ is nitrogen and the other three are carbon; or an isomer or pharmaceutically acceptable salt thereof.

Another even more preferred class of compounds consists of those compounds of Formula I' wherein X is O; wherein R is carboxyl; wherein R'' is selected from hydrido and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, pyridyloxy, thienyloxy, furyloxy, phenylmethoxy, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethoxy, trifluoromethylthio,

hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, N-phenylamino, N-(benzyl)amino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl, phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, furylcarbonyl, phenylcarbonyl, aminocarbonyl, formyl, and methylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are carbon; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I' there is another subclass of benzothiopyran compounds wherein X is S; wherein R is carboxyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, phenyl-C₁-C₆-alkyl, phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, phenyloxy, 5- or 6-membered heteroaryloxy, phenyl-C₁-C₆-alkyloxy, 5- or 6-membered heteroaryl-C₁-C₆-alkyloxy, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-alkylamino, N-phenylamino, N-(phenyl-C₁-C₆-alkyl)amino, N-heteroarylamino, N-(heteroaryl-C₁-C₆-alkylamino, nitro, amino, aminosulfonyl, N-alkylaminosulfonyl, N-arylaminosulfonyl, N-heteroarylaminosulfonyl, N-(phenyl-C₁-C₆-alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, 5- to 8-membered heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, optionally substituted phenyl, optionally substituted 5- or 6-membered heteroaryl, phenyl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, phenylcarbonyl, aminocarbonyl, and C₁-C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A²,

A³ and A⁴ are carbon; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds consists of those compounds of Formula I' wherein X is S;
5 wherein R is carboxyl; wherein R'' is selected from hydrido and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl,
10 ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, pyridyloxy,
15 thienyloxy, furyloxy, phenylmethoxy, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethoxy, trifluoromethylthio, hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, N-phenylamino, N-
20 (benzyl)amino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl,
25 phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, furylcarbonyl, phenylcarbonyl,
30 aminocarbonyl, formyl, and methylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are carbon; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I' there is a third subclass of dihydroquinoline compounds wherein X is NR^a; wherein R^a
35 is selected from hydrido, C₁-C₃-alkyl, phenyl-C₁-C₃-alkyl, acyl and carboxy-C₁-C₃-alkyl; wherein R is carboxyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals

independently selected from hydrido, halo, C₁-C₆-alkyl, phenyl-C₁-C₆-alkyl, phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, phenyloxy, 5- or 6-membered heteroaryloxy, phenyl-C₁-C₆-alkyloxy, 5- or 6-membered
 5 heteroaryl-C₁-C₆-alkyloxy, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-alkylamino, N-phenylamino, N-(phenyl-C₁-C₆-alkyl)amino, N-heteroarylamino, N-(heteroaryl-C₁-C₆-alkylamino, nitro, amino, aminosulfonyl, N-alkylaminosulfonyl, N-arylaminosulfonyl, N-
 10 heteroarylaminosulfonyl, N-(phenyl-C₁-C₆-alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, 5- to 8-membered heterocyclisulfonyl, C₁-C₆-alkylsulfonyl, optionally substituted phenyl, optionally substituted 5- or 6-
 15 membered heteroaryl, phenyl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, phenylcarbonyl, aminocarbonyl, and C₁-C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A²,
 20 A³ and A⁴ are carbon; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds consists of those compounds of Formula I' wherein X is NR^a; wherein R^a is selected from hydrido, methyl, ethyl, (4-trifluoromethyl)benzyl, (4-chloromethyl)benzyl, (4-methoxy)benzyl, (4-cyano)benzyl, and (4-nitro)benzyl; wherein R is carboxyl; wherein R" is selected from hydrido and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one
 30 or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl,
 35 phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, pyridyloxy, thienyloxy, furyloxy, phenylmethoxy, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl,

pentafluoroethyl, trifluoromethoxy, trifluoromethylthio, hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, N-phenylamino, N-(benzyl)amino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl, phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, furylcarbonyl, phenylcarbonyl, aminocarbonyl, formyl, and methylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are carbon; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I' there is a fourth subclass of compounds wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, C₁-C₃-alkyl, phenyl-C₁-C₃-alkyl, acyl and carboxy-C₁-C₃-alkyl; wherein R is selected from carboxyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; and wherein R² together with ring A forms a naphthyl or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds consists of those compounds of Formula I' wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, methyl, ethyl, (4-trifluoromethyl)benzyl, (4-chloromethyl)benzyl, (4-methoxy)benzyl, and (4-cyano)benzyl, (4-nitro)benzyl; wherein R is carboxyl; wherein R' is selected from hydrido and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are

carbon; or wherein R² together with ring A forms a naphthyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

A preferred class of compounds consists of those
 5 compounds of Formula I'' wherein X is selected from O, S, CR^cR^b and NR^a; wherein R^a is selected from hydrido, C₁-C₃-alkyl, (optionally substituted phenyl)-C₁-C₃-alkyl, acyl and carboxy-C₁-C₆-alkyl; wherein each of R^b and R^c is independently selected from hydrido, C₁-C₃-alkyl,
 10 phenyl-C₁-C₃-alkyl, C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl; or wherein CR^bR^c forms a cyclopropyl ring; wherein R is selected from carboxyl, aminocarbonyl, C₁-C₆-alkylsulfonylaminocarbonyl and C₁-C₆-alkoxycarbonyl;
 15 wherein R'' is selected from hydrido, phenyl, thienyl, C₁-C₄-alkyl and C₂-C₄-alkenyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl; wherein R² is one or more radicals independently selected from
 20 hydrido, halo, C₁-C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl, aryl-C₂-C₆-alkenyl, C₁-C₆-alkoxy, methylenedioxy, C₁-C₆-alkylthio, C₁-C₆-alkylsulfinyl, aryloxy, arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-alkoxy-C₁-C₆-alkyl,
 25 aryl-C₁-C₆-alkyloxy, heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-haloalkylthio, C₁-C₆-haloalkylsulfinyl, C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl, hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino, arylamino, aryl-C₁-C₆-alkylamino,
 30 heteroarylamino, heteroaryl-C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, C₁-C₆-alkylaminosulfonyl, arylaminosulfonyl, heteroarylaminosulfonyl, aryl-C₁-C₆-alkylaminosulfonyl, heteroaryl-C₁-C₆-alkylaminosulfonyl,
 35 heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-alkylsulfonyl, optionally substituted aryl, optionally substituted heteroaryl, aryl-C₁-C₆-alkylcarbonyl, heteroaryl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl,

arylcarbonyl, aminocarbonyl, C₁-C₆-alkoxycarbonyl, formyl, C₁-C₆-haloalkylcarbonyl and C₁-C₆-alkylcarbonyl; and wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

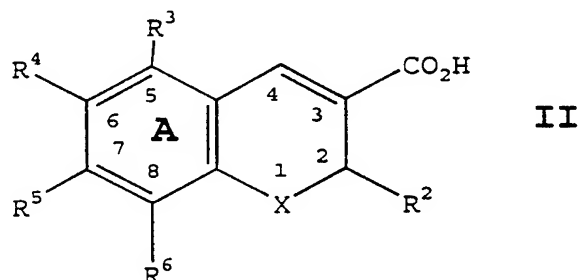
A more preferred class of compounds of Formula I'' consists of compounds wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, C₁-C₃-alkyl and (optionally substituted phenyl)methyl; wherein R is carboxyl; wherein R'' is selected from hydrido, C₁-C₃-alkyl and C₂-C₃-alkenyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals independently selected from hydrido, halo, C₁-C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, optionally substituted phenyl-C₁-C₆-alkyl, optionally substituted phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₃-alkoxy, methylenedioxy, C₁-C₃-alkoxy-C₁-C₃-alkyl, C₁-C₃-alkylthio, C₁-C₃-alkylsulfinyl, optionally substituted phenyloxy, optionally substituted phenylthio, optionally substituted phenylsulfinyl, C₁-C₃-haloalkyl-C₁-C₃-hydroxyalkyl, phenyl-C₁-C₃-alkyloxy-C₁-C₃-alkyl, C₁-C₃-haloalkyl, C₁-C₃-haloalkoxy, C₁-C₃-haloalkylthio, C₁-C₃-hydroxyalkyl, C₁-C₃-alkoxy-C₁-C₃-alkyl, hydroxyimino-C₁-C₃-alkyl, C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, N-alkylaminosulfonyl, N-arylaminosulfonyl, N-heteroarylaminosulfonyl, N-(phenyl-C₁-C₆-alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, phenyl-C₁-C₃-alkylsulfonyl, 5- to 8-membered heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, optionally substituted phenyl, optionally substituted 5- to 9-membered heteroaryl, phenyl-C₁-C₆-alkylcarbonyl, phenylcarbonyl, 4-chlorophenylcarbonyl, 4-hydroxyphenylcarbonyl, 4-trifluoromethylphenylcarbonyl, 4-methoxyphenylcarbonyl, aminocarbonyl, formyl, and C₁-C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and

A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl, benzofurylphenyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds of Formula I'' consists of compounds wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, methyl, ethyl, (4-trifluoromethyl)benzyl, (4-chloromethyl)benzyl, (4-methoxy)benzyl, and (4-cyano)benzyl, (4-nitro)benzyl; wherein R is carboxyl; wherein R'' is selected from hydrido, ethyl and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfonyl, phenyloxy, phenylthio, phenylsulfonyl, methylenedioxy, benzyloxymethyl, trifluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethoxy, trifluoromethylthio, hydroxymethyl, hydroxy-trifluoroethyl, methoxymethyl, hydroxyiminomethyl, N-methylamino, nitro, cyano, amino, aminosulfonyl, N-methylaminosulfonyl, N-phenylaminosulfonyl, N-furylaminosulfonyl, N-(benzyl)aminosulfonyl, N-(furylmethyl)aminosulfonyl, benzylsulfonyl, phenylethylaminosulfonyl, furylsulfonyl, methylsulfonyl, phenyl, phenyl substituted with one or more radicals selected from chloro, fluoro, bromo, methoxy, methylthio and methylsulfonyl, benzimidazolyl, thienyl, thienyl substituted with chloro, furyl, furyl substituted with chloro, benzylcarbonyl, optionally substituted phenylcarbonyl, aminocarbonyl, formyl and methylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴

are carbon; or wherein R² together with ring A forms a naphthyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I there is a subclass of compounds
 5 of high interest represented by Formula II:

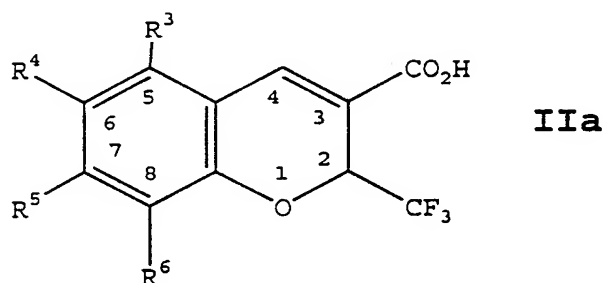


wherein X is selected from O, NR^a and S;
 10 wherein R² is lower haloalkyl;
 wherein R³ is selected from hydrido, and halo;
 wherein R⁴ is selected from hydrido, halo, lower
 alkyl, lower haloalkoxy, lower alkoxy, lower
 aralkylcarbonyl, lower dialkylaminosulfonyl, lower
 15 alkynyl, phenyl-lower alkynyl, lower alkylaminosulfonyl,
 lower aralkylaminosulfonyl, lower
 heteroaralkylaminosulfonyl, and 5- or 6- membered
 nitrogen containing heterocyclosulfonyl;
 wherein R⁵ is selected from hydrido, lower alkyl,
 20 lower hydroxyalkyl, halo, lower alkenyl, lower alkynyl,
 lower alkoxy, phenyloxy, and aryl; and
 wherein R⁶ is selected from hydrido, halo, lower
 alkyl, lower alkoxy, and aryl;
 or an isomer or pharmaceutically acceptable salt
 25 thereof.

A class of compounds of particular interest
 consists of those compounds of Formula II wherein R² is
 trifluoromethyl or pentafluoroethyl; wherein R³ is
 selected from hydrido, chloro, and fluoro; wherein R⁴ is
 30 selected from hydrido, chloro, bromo, fluoro, iodo,
 methyl, tert-butyl, trifluoromethoxy, methoxy, ethynyl,
 phenylethynyl, benzylcarbonyl, dimethylaminosulfonyl,

isopropylaminosulfonyl, methylaminosulfonyl, benzylaminosulfonyl, phenylethylaminosulfonyl, methylpropylaminosulfonyl, methylsulfonyl, and morpholinosulfonyl; wherein R⁵ is selected from hydrido, methyl, ethyl, isopropyl, tert-butyl, chloro, methoxy, diethylamino, ethynyl, ethenyl, 2-hydroxy-1,1-dimethylethyl, phenyloxy and phenyl; and wherein R⁶ is selected from hydrido, chloro, bromo, fluoro, methyl, ethyl, tert-butyl, methoxy, and phenyl; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I there is a subclass of compounds of high interest represented by Formula IIa:



wherein R³ is selected from hydrido, C₁-C₃-alkyl, C₁-C₃-hydroxyalkyl, C₁-C₃-alkoxy and halo; wherein R⁴ is selected from hydrido, halo, C₁-C₄-alkyl, C₁-C₃-alkylthio, C₁-C₃-haloalkyl, amino, aminosulfonyl, C₁-C₃-alkylsulfonyl, C₁-C₃-alkylsulfinyl, C₁-C₃-alkoxy-C₁-C₃-alkyl, C₁-C₃-alkylcarbonyl, formyl, cyano, C₁-C₃-haloalkylthio, substituted or unsubstituted phenylcarbonyl, C₁-C₃-haloalkoxy, C₁-C₃-alkoxy, aryl-C₁-C₃-alkylcarbonyl, di-C₁-C₃-alkylaminosulfonyl, C₁-C₃-alkylaminosulfonyl, aryl-C₁-C₃-alkylaminosulfonyl, 5- or 6-heteroaryl-C₁-C₃-alkylaminosulfonyl, 5- or 6-membered heteroaryl, C₁-C₃-hydroxyalkyl, substituted or unsubstituted phenyl and 5- or 6-membered nitrogen-containing heterocyclosulfonyl; wherein R⁵ is selected from hydrido, C₁-C₃-alkyl, halo, C₁-C₃-haloalkyl, C₁-C₄-hydroxyalkyl, C₂-C₃-

alkynyl, C₂-C₃-alkenyl, C₁-C₃-alkoxy, phenoxy, phenoxy independently substituted with one or more radicals selected from C₁-C₃-haloalkyl, nitro, carboxy, C₁-C₃-haloalkoxy, C₁-C₃-alkoxy, cyano, C₁-C₃-alkyl and halo, naphthyloxy, naphthyloxy substituted with one or more halo radicals, phenylthio, phenylthio substituted with one or more halo radicals, phenylsulfinyl, phenylsulfinyl substituted with one or more halo radicals, phenylsulfonyl, phenylsulfonyl substituted with one or more halo radicals, pyridinyloxy, pyridinyloxy substituted with one or more halo radicals, and phenyl; and wherein R⁶ is selected from hydrido, halo, cyano, hydroxyiminomethyl, C₁-C₃-hydroxyalkyl, C₂-C₃-alkynyl, phenyl-C₂-C₃-alkynyl, C₁-C₃-alkyl, C₁-C₃-alkoxy, formyl and phenyl; or an isomer or pharmaceutically acceptable salt thereof.

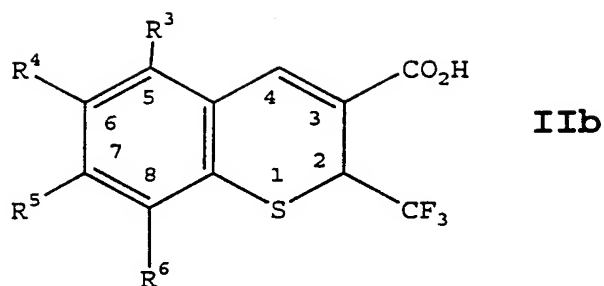
A class of compounds of particular interest consists of those compounds of Formula IIa wherein R³ is selected from hydrido, and chloro; wherein R⁴ is selected from chloro, methyl, tert-butyl, methylthio, trifluoromethyl, difluoromethyl, pentafluoromethyl, trifluoromethylthio, trifluoromethoxy, cyano, substituted or unsubstituted phenylcarbonyl, and substituted or unsubstituted phenyl; wherein R⁵ is selected from hydrido, methyl, tert-butyl, 2,2,2-trifluoroethoxy, 2-hydroxy-1,1-dimethylethyl, phenoxy, 4-methoxyphenoxy, 4-chlorophenoxy, 3-chlorophenoxy, 2-chlorophenoxy, 4-cyanophenoxy, 2,6-dimethylphenoxy, 2,4-dichlorophenoxy, 3,4-difluorophenoxy, 4-chloro-3-fluorophenoxy, 4-(trifluoromethyl)phenoxy, 4-nitrophenoxy, 4-carboxyphenoxy, 3-carboxyphenoxy, 2-chloro-4-carboxyphenoxy, 4-(trifluoromethoxy)phenoxy, 2-bromo-4-chlorophenoxy, (6-bromo-2-naphthalenyl)oxy, phenylthio, (4-methoxyphenyl)thio, (4-chlorophenyl)thio, (4-chlorophenyl)sulfinyl, (4-chlorophenyl)sulfonyl, (6-

chloro-2-pyridinyl)oxy, (2-chloro-3-pyridinyl)oxy, (3-pyridinyl)oxy, (2-pyridinyl)oxy, iodo, ethenyl, ethynyl, chloro; and wherein R⁶ is selected from hydrido, chloro, thienyl, hydroxyiminomethyl, substituted or
5 unsubstituted phenylethynyl, and substituted or unsubstituted phenyl; or an isomer or pharmaceutically acceptable salt thereof.

A class of compounds of more particular interest consists of those compounds of Formula IIa wherein R³ is
10 hydrido; wherein R⁴ is chloro or hydrido; and wherein R⁶ is hydrido; or an isomer or pharmaceutically acceptable salt thereof.

An even more preferred class of compounds of Formula IIa consists of compounds wherein R⁵ is selected
15 from 2,2,2-trifluoroethoxy, 2-hydroxy-1,1-dimethylethyl, phenoxy, 4-methoxyphenoxy, 4-chlorophenoxy, 3-chlorophenoxy, 2-chlorophenoxy, 4-cyanophenoxy, 2,6-dimethylphenoxy, 2,4-dichlorophenoxy, 3,4-difluorophenoxy, 4-chloro-3-fluorophenoxy, 4-
20 (trifluoromethyl)phenoxy, 4-nitrophenoxy, 4-carboxyphenoxy, 3-carboxyphenoxy, 2-chloro-4-carboxyphenoxy, 4-(trifluoromethoxy)phenoxy, 2-bromo-4-chlorophenoxy, (6-bromo-2-naphthalenyl)oxy, phenylthio, (4-methoxyphenyl)thio, (4-chlorophenyl)thio, (4-
25 chlorophenyl)sulfinyl, (4-chlorophenyl)sulfonyl, (6-chloro-2-pyridinyl)oxy, (2-chloro-3-pyridinyl)oxy, (3-pyridinyl)oxy, and (2-pyridinyl)oxy; or an isomer or pharmaceutically acceptable salt thereof.

Within Formula I there is a subclass of compounds
30 of high interest represented by Formula IIb



wherein R^3 is selected from hydrido, lower alkyl, lower hydroxyalkyl, lower alkoxy and halo;

5 wherein R^4 is selected from hydrido, halo, lower alkyl, lower alkylthio, lower haloalkyl, amino, aminosulfonyl, lower alkylsulfonyl, lower alkylsulfinyl, lower alkoxyalkyl, lower alkylcarbonyl, formyl, cyano, lower haloalkylthio, substituted or unsubstituted
10 phenylcarbonyl, lower haloalkoxy, lower alkoxy, lower aralkylcarbonyl, lower dialkylaminosulfonyl, lower alkylaminosulfonyl, lower aralkylaminosulfonyl, lower heteroaralkylaminosulfonyl, 5- or 6- membered heteroaryl, lower hydroxyalkyl, optionally substituted
15 phenyl and 5- or 6- membered nitrogen containing heterocyclosulfonyl;

wherein R^5 is selected from hydrido, lower alkyl, halo, lower haloalkyl, lower alkoxy, and phenyl; and

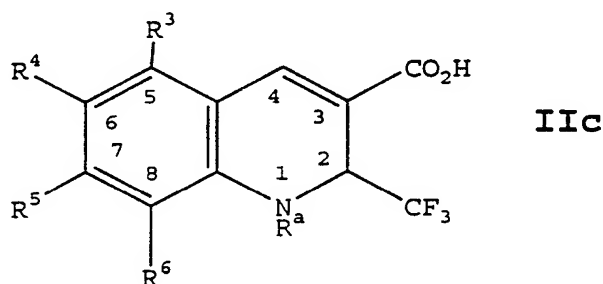
20 wherein R^6 is selected from hydrido, halo, cyano, hydroxyiminomethyl, lower hydroxyalkyl, lower alkynyl, phenylalkynyl, lower alkyl, lower alkoxy, formyl and phenyl;

or an isomer or pharmaceutically acceptable salt thereof.

25 A class of compounds of particular interest consists of those compounds of Formula IIb wherein R^3 is selected from hydrido, and chloro; wherein R^4 is selected from chloro, methyl, tert-butyl, methylthio, trifluoromethyl, difluoromethyl, pentafluoromethyl,
30 trifluoromethylsulfide, trifluoromethoxy, cyano, substituted or unsubstituted phenylcarbonyl, and substituted or unsubstituted phenyl; wherein R^5 is

selected from hydrido, methyl, tert-butyl, chloro; and
 wherein R⁶ is selected from hydrido, chloro, thienyl,
 hydroxyiminomethyl, substituted or unsubstituted
 phenylethynyl, and substituted or unsubstituted phenyl;
 5 or an isomer or pharmaceutically acceptable salt
 thereof.

Within Formula I there is a subclass of compounds
 of high interest represented by Formula IIc:



10

wherein R^a is selected from hydrido and lower
 aralkyl;

wherein R³ is selected from hydrido, lower alkyl,
 15 lower hydroxyalkyl, lower alkoxy and halo;

wherein R⁴ is selected from hydrido, halo, lower
 alkyl, lower alkylthio, lower haloalkyl, amino,
 aminosulfonyl, lower alkylsulfonyl, lower alkylsulfinyl,
 lower alkoxyalkyl, lower alkylcarbonyl, formyl, cyano,
 20 lower haloalkylthio, substituted or unsubstituted
 phenylcarbonyl, lower haloalkoxy, lower alkoxy, lower
 alkynyl, phenyl-lower alkynyl, lower aralkylcarbonyl,
 lower dialkylaminosulfonyl, lower alkylaminosulfonyl,
 lower aralkylaminosulfonyl, lower
 25 heteroaralkylaminosulfonyl, 5- or 6- membered
 heteroaryl, lower hydroxyalkyl, optionally substituted
 phenyl and 5- or 6- membered nitrogen containing
 heterocyclosulfonyl;

wherein R⁵ is selected from hydrido, lower alkyl,
 30 halo, lower haloalkyl, lower alkoxy, and phenyl; and

wherein R⁶ is selected from hydrido, halo, cyano,
 hydroxyiminomethyl, lower hydroxyalkyl, lower alkynyl,

phenylalkynyl, lower alkyl, lower alkoxy, formyl and phenyl;

or an isomer or pharmaceutically acceptable salt thereof.

5 A class of compounds of particular interest consists of those compounds of Formula IIc wherein R³ is selected from hydrido, and chloro; wherein R⁴ is selected from chloro, methyl, tert-butyl, methylthio, trifluoromethyl, difluoromethyl, pentafluoromethyl, 10 trifluoromethylsulfide, trifluoromethoxy, ethynyl, phenylethynyl, substituted or unsubstituted phenylcarbonyl, and substituted or unsubstituted phenyl; wherein R⁵ is selected from hydrido, methyl, tert-butyl, chloro; and wherein R⁶ is selected from hydrido, chloro, 15 thienyl, hydroxyiminomethyl, substituted or unsubstituted phenylethynyl, and substituted or unsubstituted phenyl; or an isomer or pharmaceutically acceptable salt thereof.

 A class of compounds of more particular interest 20 consists of those compounds of Formula IIc wherein R³ is hydrido or chloro; wherein R⁴ is selected from ethynyl, and optionally substituted phenylethynyl; or an isomer or pharmaceutically acceptable salt thereof.

 A family of specific compounds of particular 25 interest within Formula I consists of compounds and pharmaceutically-acceptable salts thereof as follows:
6-chloro-7-[(2-pyridyl-N-oxide)thio]-2-
 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-7-[(3-chloro-2-pyridyl-(N-oxide))thio]-2-
30 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-1,2-dihydro-7-[(2-pyridyl-N-oxide)oxy]-2-
 (trifluoromethyl)-2H-quinoline-3-carboxylic acid;
6-chloro-7-[(3-chloro-2-pyridyl-(N-oxide))oxy]-1,2-
 dihydro-2-(trifluoromethyl)-2H-quinoline-3-carboxylic
35 acid;
6-chloro-1,2-dihydro-7-[(2-pyridyl-N-oxide)thio]-2-
 (trifluoromethyl)-2H-quinoline-3-carboxylic acid;

- 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})thio]-1,2-dihydro-2-(trifluoromethyl)-2H-quinoline-3-carboxylic acid;
- 5 6-chloro-7-[(2-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})oxy]-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 10 6-chloro-7-[(2-pyridyl-N-oxide)thio]-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})thio]-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 15 6-chloro-7-(3,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-bromo-4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(4-bromo-3-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3,4-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[4-chloro-3-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-[3-chloro-4-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,6-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,6-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-(2,6-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,5-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-(2,5-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,5-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-(2,3-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,3-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-(2,3-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-cyanophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-(2-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-(4-chloro-2-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-chloro-3-tert-butyl phenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-ethynyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-cyanophenoxy)-2-(trifluoromethyl)-2H-
1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-[4-(trifluoromethyl)phenoxy]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-[4-(trifluoromethoxy)phenoxy]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
10 acid;
- 6-chloro-7-(2-bromo-4-chlorophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-[(6-bromo-2-naphthyl)oxy]-2-
15 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-(2,6-dimethylphenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 20 6-chloro-7-[(4-methoxyphenyl)thio]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-[(4-chlorophenyl)thio]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
25 acid;
- 6-chloro-7-[(4-chlorophenyl)sulfinyl]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-[(4-chlorophenyl)sulfonyl]-2-
30 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 6-chloro-7-phenylthio-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-chlorophenoxy)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-(2,4-dichlorophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;

- 6-chloro-7-(3,4-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-[(6-chloro-2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(2-chloro-3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-[(3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-nitrophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-(2-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-fluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-chloro-4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-(3-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-1,2-dihydro-8-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 1,2-dihydro-5-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 35 1,2-dihydro-6-(4-fluorophenyl)-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 1,2-dihydro-6-ethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;

8-bromo-1,2-dihydro-6-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;

1,2-dihydro-6-phenylethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;

5 6-chloro-7-[(3-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

6-chloro-7-[(2-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

10 6-chloro-7-[(3-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

6-chloro-7-[(4-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

15 6-chloro-7-[(5-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

2 6-chloro-7-[(6-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

20

6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

2

25 7-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

7-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

30

2,7-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

30 7-bromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

6-chloro-7-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

35

35 8-(1-methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

6-chloro-7-(1,1-dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- 8-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6,7-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6,8-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 10 6,8-dibromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6,8-dimethoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 15 6-nitro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-amino-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- ethyl 6-amino-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate;
- 20 6-chloro-8-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 8-chloro-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 8-chloro-6-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6,8-difluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-bromo-8-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 8-bromo-6-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 8-bromo-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 35 8-bromo-5-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- 6-bromo-8-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 7-(N,N-diethylamino)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 5 6-[(phenylmethyl)amino]sulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-[(dimethylamino)sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-aminosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 10 6-(methylamino)sulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-[(4-morpholino)sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 15 6-[(1,1-dimethylethyl)aminosulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-[(2-methylpropyl)aminosulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-methylsulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 20 8-chloro-6-[(phenylmethyl)amino]sulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-N,N-diethylaminosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 6-phenylacetyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6-(2,2-dimethylpropylcarbonyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 6,8-dichloro-7-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-2-trifluoromethyl-2H-1-benzothiopyran-3-carboxylic acid;
- 6-[(2-furanylmethyl)amino]sulfonyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-[(phenylmethyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-[[(phenylethyl) amino] sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
5 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
8-bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
10 6-chloro-8-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-bromo-7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
15 5,6-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-hydroxymethyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
20 6-(difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
2,6-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
25 5,6,7-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6,7,8-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-(methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
30 6-(methylsulfinyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
5,8-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
35 6-(pentafluoroethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-[[(phenylethyl) amino] sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
5 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
8-bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
10 6-chloro-8-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-bromo-7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
15 5,6-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-hydroxymethyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
20 6-(difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
2,6-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
25 5,6,7-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6,7,8-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-(methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
30 6-(methylsulfinyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
5,8-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
35 6-(pentafluoroethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 2-(trifluoromethyl)-6-[(trifluoromethyl)thio]-2H-1-benzopyran-3-carboxylic acid;
- 6,8-dichloro-7-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-2,7-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5-methoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-benzoyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-(4-chlorobenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(4-hydroxybenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 8-chloro-6-(4-chlorophenoxy)-2-trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- 20 6-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(3-chloro-4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 8-chloro-2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-8-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-[(hydroxyimino)methyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(hydroxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 8-(1H-benzimidazol-2-yl)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 7-(1,1-dimethylethyl)-2-(pentafluoroethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(methoxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-8-(benzyloxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-ethynyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-8-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(2-furanyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-8-(5-chloro-1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(phenylethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-8-(3,3-dimethyl-1-butynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-[(4-chlorophenyl)ethynyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-8-[(4-methoxyphenyl)ethynyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(phenylethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-8-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(3-methoxyphenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-8-[(4-methylthio)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-8-[(4-methylsulfonyl)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-8-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-bromo-8-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(4-fluorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 8-chloro-6-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6,8-diiodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-(5-chloro-2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(4-bromophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-(ethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-8-(4-methoxyphenyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-2-(trifluoromethyl)-4-ethenyl-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-2-(trifluoromethyl)-4-phenyl-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-4-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-(2,2,2-trifluoro-1-hydroxyethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
6,8-dimethyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
5 6-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
6,7-dimethyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
10 8-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
15 6-chloro-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
7-chloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
6,7-dichloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
20 2-(trifluoromethyl)-6-[(trifluoromethyl)thio]-2H-1-benzothiopyran-3-carboxylic acid;
6,8-dichloro-2-trifluoromethyl-2H-1-benzothiopyran-3-carboxylic acid;
25 6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6,8-dichloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
30 6,7-difluoro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-bromo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
35 1,2-dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylic acid;

- 6-(trifluoromethyl)-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
5 6-chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-chloro-1,2-dihydro-2-(trifluoromethyl)-1-[[4-(trifluoromethyl)phenyl]methyl]-3-quinolinecarboxylic acid;
10 6-chloro-1-[(4-chlorophenyl)methyl]-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-chloro-1,2-dihydro-2-(trifluoromethyl)-1-[[4-(methoxy)phenyl]methyl]-3-quinolinecarboxylic acid;
6-chloro-1-[(4-cyanophenyl)methyl]-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
15 6-chloro-1,2-dihydro-1-[(4-nitrophenyl)methyl]-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
6-chloro-1,2-dihydro-1-ethyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
20 6-chloro-2-(trifluoromethyl)-1,2-dihydro[1,8]naphthyridine-3-carboxylic acid;

6-chloro-7-[(3-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid
25 6-chloro-7-[(2-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-7-[(3-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-7-[(4-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
30 6-chloro-7-[(5-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-7-[(6-chloro-2-pyridyl-(N-oxide))oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
35 6-chloro-7-[(2-pyridyl-N-oxide)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
6-chloro-7-[(3-chloro-2-pyridyl-(N-oxide))thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-[(2-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-
2H-1-dihydroquinoline-3-carboxylic acid;
- 5 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})oxy]-2-
(trifluoromethyl)-2H-1-dihydroquinoline-3-carboxylic
acid;
- 6-chloro-7-[(2-pyridyl-N-oxide)thio]-2-
(trifluoromethyl)-2H-1-dihydroquinoline-3-carboxylic
acid;
- 10 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})thio]-2-
(trifluoromethyl)-2H-1-dihydroquinoline-3-carboxylic
acid;
- 6-chloro-7-[(2-pyridyl-N-oxide)oxy]-2-(trifluoromethyl)-
15 2H-1-benzothiopyran-3-carboxylic acid;
- 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})oxy]-2-
(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic
acid;
- 6-chloro-7-[(2-pyridyl-N-oxide)thio]-2-
20 (trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic
acid;
- 6-chloro-7-[(3-chloro-2-pyridyl-{N-oxide})thio]-2-
(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic
acid;
- 25 7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-
carboxylic acid;
- 6-chloro-7-phenoxy-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 30 6-chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
- 7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-
1-benzopyran-3-carboxylic;
- 35 6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 6-chloro-7-ethynyl-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;

- 6-chloro-7-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-cyanophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-[4-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[4-(trifluoromethoxy)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-(2-bromo-4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(6-bromo-2-naphthyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(2,6-dimethylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-[(4-methoxyphenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(4-chlorophenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-[(4-chlorophenyl)sulfinyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-[(4-chlorophenyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-phenylthio-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-(2,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3,4-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-[(6-chloro-2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(2-chloro-3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-[(3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(4-nitrophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-(4-chloro-3-fluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-(2-chloro-4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 1,2-dihydro-6-ethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 1,2-dihydro-6-phenylethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 5 6-chloro-7-(3,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-bromo-4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-bromo-3-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-(3,4-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[4-chloro-3-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-[3-chloro-4-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,6-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(2,6-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,6-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-(2,5-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,5-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,5-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-(2,3-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,3-dibromophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-(2,3-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-cyanophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-(3-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-(4-chloro-3-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-2-tert-butylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-(2-chloro-3-tert-butyl phenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-(3-chloro-2-tert-butyl phenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(2-chloro-3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(5-chloro-3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-(2,5-dichloro-3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,4-dichloro-3-thienyloxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-[(3-chloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-[(4-chloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-[(3-chloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-[(5-chloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(6-chloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-[(5,6-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(3,4-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-[(4,5-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(3,5-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-[(3,6-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-[(4,5-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 6-chloro-7-[(4,6-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(5,6-dichloro-2-pyridyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 6-chloro-7-[(2-quinolyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

2-trifluoromethyl-2H-naphtho[1,2-b]pyran-3-carboxylic
acid;
2-trifluoromethyl-3H-naphtho[2,1-b]pyran-3-carboxylic
5 acid;
2-trifluoromethyl-2H-naphtho[2,3-b]pyran-3-carboxylic
acid;
5-(hydroxymethyl)-8-methyl-2-(trifluoromethyl)-2H-
pyrano[2,3-c]pyridine-3-carboxylic acid;
10 6-(trifluoromethyl)-6h-1,3-dioxolo[4,5-g][1]benzopyran-
7-carboxylic acid; and
3-(trifluoromethyl)-3H-benzofuro[3,2-f][1]benzopyran-2-
carboxylic acid.

15 A preferred family of specific compounds of
particular interest within Formulas I-I' consists of
compounds as follows:

(S)-7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-
3-carboxylic acid;
20 (S)-6-chloro-7-phenoxy-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
(S)-6-chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;
25 (S)-7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic;
(S)-6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
(S)-6-chloro-7-ethynyl-2-(trifluoromethyl)-2H-1-
30 benzopyran-3-carboxylic acid;
(S)-6-chloro-7-ethenyl-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
(S)-6-chloro-7-(4-methoxyphenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
35 acid;
(S)-6-chloro-7-(4-chlorophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid;

- (S)-6-chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-chloro-7-(4-cyanophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-[4-(trifluoromethyl)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-6-chloro-7-[4-(trifluoromethoxy)phenoxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-(2-bromo-4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-6-chloro-7-[(6-bromo-2-naphthyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-(2,6-dimethylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-chloro-7-[(4-methoxyphenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-chloro-7-[(4-chlorophenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-[(4-chlorophenyl)sulfinyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-chloro-7-[(4-chlorophenyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-phenylthio-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-chloro-7-(3-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-chloro-7-(2,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-chloro-7-(3,4-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-[(6-chloro-2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-6-chloro-7-[(2-chloro-3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-[(3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-6-chloro-7-[(2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-(4-nitrophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-chloro-7-(2-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-(4-chloro-3-fluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-chloro-7-(4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-chloro-7-(2-chloro-4-carboxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-(3-carboxyphenoxy)-2-
- 35 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-chloro-1,2-dihydro-8-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-1,2-dihydro-5-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-1,2-dihydro-6-(4-fluorophenyl)-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 10 (S)-1,2-dihydro-6-ethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-8-bromo-1,2-dihydro-6-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-1,2-dihydro-6-phenylethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 15 (S)-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-7-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-2,7-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-bromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-chloro-7-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-(1-methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-chloro-7-(1,1-dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(1-methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-8-ethoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-(1,1-dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-bromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-8-bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-trifluoromethoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-5,7-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7,8-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-7-isopropoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7,8-dimethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6,8-bis(1,1-dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-7-(1-methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-7-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-8-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-chloro-7-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,7-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- (S)-6,8-dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,8-dibromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6,8-dimethoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-nitro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-amino-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-ethyl 6-amino-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate;
- (S)-6-chloro-8-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-8-chloro-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-6-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,8-difluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-bromo-8-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-bromo-6-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-8-bromo-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-bromo-5-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-bromo-8-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-7-(N,N-diethylamino)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-[[(phenylmethyl) amino] sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-[(dimethylamino) sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-aminosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(methylamino)sulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-[(4-morpholino)sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-[(1,1-dimethylethyl)aminosulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-[(2-methylpropyl)aminosulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-6-methylsulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-6-[(phenylmethyl)aminosulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-6-N,N-diethylaminosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-phenylacetyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(2,2-dimethylpropylcarbonyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6,8-dichloro-7-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-2-trifluoromethyl-2H-1-benzothiopyran-3-carboxylic acid;
- 25 (S)-6-[(2-furanylmethyl)aminosulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-[(phenylmethyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-[(phenylethyl)aminosulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-bromo-7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-5,6-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid ;
- 10 (S)-6-hydroxymethyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-2,6-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-5,6,7-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,7,8-trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-(methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(methylsulfinyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-5,8-dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(pentafluoroethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-2-(trifluoromethyl)-6-[(trifluoromethyl)thio]-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-6,8-dichloro-7-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-chloro-2,7-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-5-methoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-benzoyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(4-chlorobenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-(4-hydroxybenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-6-(4-chlorophenoxy)-2-trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-6-(3-chloro-4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-chloro-8-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-[(hydroxyimino)methyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-chloro-8-(hydroxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-(1H-benzimidazol-2-yl)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-7-(1,1-dimethylethyl)-2-(pentafluoroethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(methoxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 35 (S)-6-chloro-8-(benzyloxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-chloro-8-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-ethynyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-chloro-8-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(2-furanyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(5-chloro-1-pentynyl)-2-
- 10 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(phenylethynyl)-2-(trifluoromethyl)-2H-1-
- 15 benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(3,3-dimethyl-1-butynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-[(4-chlorophenyl)ethynyl]-2-
- 20 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-[(4-methoxyphenyl)ethynyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-(phenylethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(3-methoxyphenyl)-2-(trifluoromethyl)-2H-
- 30 1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-[(4-methylthio)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-[(4-methylsulfonyl)phenyl]-2-
- 35 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- (S)-6-bromo-8-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(4-fluorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 (S)-6-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-8-chloro-6-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,8-diiodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 (S)-6-(5-chloro-2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 (S)-6-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(4-bromophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-(ethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 (S)-6-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-8-(4-methoxyphenyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid;
- 25 (S)-6-chloro-2-(trifluoromethyl)-4-ethenyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-2-(trifluoromethyl)-4-phenyl-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-chloro-4-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30 (S)-6-(2,2,2-trifluoro-1-hydroxyethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- (S)-6-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 35 (S)-6,8-dimethyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;

- (S)-6-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 5 (S)-6,7-dimethyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-8-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 10 (S)-6-chloro-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-7-chloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- 15 (S)-6,7-dichloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid;
- (S)-2-(trifluoromethyl)-6-[(trifluoromethyl)thio]-2H-1-benzopyran-3-carboxylic acid;
- (S)-6,8-dichloro-2-trifluoromethyl-2H-1-benzothiopyran-3-carboxylic acid;
- 20 (S)-6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6,8-dichloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 25 (S)-6,7-difluoro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 30 (S)-6-bromo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-1,2-dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-(trifluoromethyl)-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- 35 (S)-6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;

- (S)-6-chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-chloro-1,2-dihydro-2-(trifluoromethyl)-1-[[4-(trifluoromethyl)phenyl]methyl]-3-quinolinecarboxylic acid;
- (S)-6-chloro-1-[(4-chlorophenyl)methyl]-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-chloro-1,2-dihydro-2-(trifluoromethyl)-1-[[4-(methoxy)phenyl]methyl]-3-quinolinecarboxylic acid;
- (S)-6-chloro-1-[(4-cyanophenyl)methyl]-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-chloro-1,2-dihydro-1-[(4-nitrophenyl)methyl]-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-chloro-1,2-dihydro-1-ethyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid;
- (S)-6-chloro-2-(trifluoromethyl)-1,2-dihydro[1,8]naphthyridine-3-carboxylic acid;
- (S)-2-trifluoromethyl-2H-naphtho[1,2-b]pyran-3-carboxylic acid;
- (S)-2-trifluoromethyl-3H-naphtho[2,1-b]pyran-3-carboxylic acid;
- (S)-2-trifluoromethyl-2H-naphtho[2,3-b]pyran-3-carboxylic acid; and
- (S)-5-(hydroxymethyl)-8-methyl-2-(trifluoromethyl)-2H-pyrano[2,3-c]pyridine-3-carboxylic acid.

The term "hydrido" denotes a single hydrogen atom (H). This hydrido radical may be attached, for example, to an oxygen atom to form a hydroxyl radical or two hydrido radicals may be attached to a carbon atom to form a methylene ($-\text{CH}_2-$) radical. Where the term "alkyl" is used, either alone or within other terms such as "haloalkyl" and "alkylsulfonyl", it embraces linear or branched radicals having one to about twenty carbon atoms or, preferably, one to about twelve carbon atoms. More preferred alkyl radicals are "lower alkyl" radicals having one to about six carbon atoms. Examples of such

radicals include methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, sec-butyl, tert-butyl, pentyl, iso-amyl, hexyl and the like. Even more preferred are lower alkyl radicals having one to three carbon atoms. The

5 term "alkenyl" embraces linear or branched radicals having at least one carbon-carbon double bond of two to about twenty carbon atoms or, preferably, two to about twelve carbon atoms. More preferred alkenyl radicals are "lower alkenyl" radicals having two to about six carbon

10 atoms. Examples of alkenyl radicals include ethenyl, propenyl, allyl, propenyl, butenyl and 4-methylbutenyl. The term "alkynyl" denotes linear or branched radicals having two to about twenty carbon atoms or, preferably, two to about twelve carbon atoms. More preferred alkynyl

15 radicals are "lower alkynyl" radicals having two to about ten carbon atoms. Most preferred are lower alkynyl radicals having two to about six carbon atoms. Examples of such radicals include propargyl, butynyl, and the like. The terms "alkenyl" and "lower alkenyl",

20 embrace radicals having "cis" and "trans" orientations, or alternatively, "E" and "Z" orientations. The term "halo" means halogens such as fluorine, chlorine, bromine or iodine atoms. The term "haloalkyl" embraces radicals wherein any one or more of the alkyl carbon

25 atoms is substituted with halo as defined above. Specifically embraced are monohaloalkyl, dihaloalkyl and polyhaloalkyl radicals. A monohaloalkyl radical, for one example, may have either an iodo, bromo, chloro or fluoro atom within the radical. Dihalo and polyhaloalkyl

30 radicals may have two or more of the same halo atoms or a combination of different halo radicals. "Lower haloalkyl" embraces radicals having 1-6 carbon atoms. Examples of haloalkyl radicals include fluoromethyl, difluoromethyl, trifluoromethyl, chloromethyl,

35 dichloromethyl, trichloromethyl, pentafluoroethyl, heptafluoropropyl, difluorochloromethyl, dichlorofluoromethyl, difluoroethyl, difluoropropyl, dichloroethyl and dichloropropyl. "Perfluoroalkyl"

means alkyl radicals having all hydrogen atoms replaced with fluoro atoms. Examples include trifluoromethyl and pentafluoroethyl. The term "hydroxyalkyl" embraces linear or branched alkyl radicals having one to about
5 ten carbon atoms any one of which may be substituted with one or more hydroxyl radicals. More preferred hydroxyalkyl radicals are "lower hydroxyalkyl" radicals having one to six carbon atoms and one or more hydroxyl radicals. Examples of such radicals include
10 hydroxymethyl, hydroxyethyl, hydroxypropyl, hydroxybutyl and hydroxyhexyl. Even more preferred are lower hydroxyalkyl radicals having one to three carbon atoms. The term "cyanoalkyl" embraces linear or branched alkyl radicals having one to about ten carbon atoms any one of
15 which may be substituted with one cyano radicals. More preferred cyanoalkyl radicals are "lower cyanoalkyl" radicals having one to six carbon atoms and one cyano radical. Even more preferred are lower cyanoalkyl radicals having one to three carbon atoms. Examples of
20 such radicals include cyanomethyl. The terms "alkoxy" embrace linear or branched oxy-containing radicals each having alkyl portions of one to about ten carbon atoms. More preferred alkoxy radicals are "lower alkoxy" radicals having one to six carbon atoms. Examples of
25 such radicals include methoxy, ethoxy, propoxy, butoxy and tert-butoxy. Even more preferred are lower alkoxy radicals having one to three carbon atoms. The "alkoxy" radicals may be further substituted with one or more halo atoms, such as fluoro, chloro or bromo, to provide
30 "haloalkoxy" radicals. Even more preferred are lower haloalkoxy radicals having one to three carbon atoms. Examples of such radicals include fluoromethoxy, chloromethoxy, trifluoromethoxy, trifluoroethoxy, fluoroethoxy and fluoropropoxy. The term "aryl", alone
35 or in combination, means a carbocyclic aromatic system containing one or two rings wherein such rings may be attached together in a pendent manner or may be fused. The term "aryl" embraces aromatic radicals such as

phenyl, naphthyl, tetrahydronaphthyl, indane and biphenyl. More preferred aryl is phenyl. Said "aryl" group may have 1 to 3 substituents such as lower alkyl, hydroxy, halo, haloalkyl, nitro, cyano, alkoxy and lower alkylamino. The term "heterocyclyl" embraces saturated, partially saturated and unsaturated heteroatom-containing ring-shaped radicals, where the heteroatoms may be selected from nitrogen, sulfur and oxygen. Examples of saturated heterocyclic radicals include saturated 3 to 6-membered heteromonocyclic group containing 1 to 4 nitrogen atoms [e.g. pyrrolidinyl, imidazolidinyl, piperidino, piperazinyl]; saturated 3 to 6-membered heteromonocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms [e.g. morpholinyl]; saturated 3 to 6-membered heteromonocyclic group containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms [e.g., thiazolidinyl]. Examples of partially saturated heterocyclyl radicals include dihydrothiophene, dihydropyran, dihydrofuran and dihydrothiazole. Examples of unsaturated heterocyclic radicals, also termed "heteroaryl" radicals, include unsaturated 5 to 6 membered heteromonocyclyl group containing 1 to 4 nitrogen atoms, for example, pyrrolyl, pyrrolinyl, imidazolyl, pyrazolyl, 2-pyridyl, 3-pyridyl, 4-pyridyl, pyrimidyl, pyrazinyl, pyridazinyl, triazolyl [e.g., 4H-1,2,4-triazolyl, 1H-1,2,3-triazolyl, 2H-1,2,3-triazolyl]; unsaturated condensed heterocyclic group containing 1 to 5 nitrogen atoms, for example, indolyl, isoindolyl, indolizinyl, benzimidazolyl, quinolyl, isoquinolyl, indazolyl, benzotriazolyl, tetrazolopyridazinyl [e.g., tetrazolo [1,5-b]pyridazinyl]; unsaturated 3 to 6-membered heteromonocyclic group containing an oxygen atom, for example, pyranyl, 2-furyl, 3-furyl, etc.; unsaturated 5 to 6-membered heteromonocyclic group containing a sulfur atom, for example, 2-thienyl, 3-thienyl, etc.; unsaturated 5- to 6-membered heteromonocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen

atoms, for example, oxazolyl, isoxazolyl, oxadiazolyl [e.g., 1,2,4-oxadiazolyl, 1,3,4-oxadiazolyl, 1,2,5-oxadiazolyl]; unsaturated condensed heterocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms [e.g. benzoxazolyl, benzoxadiazolyl]; unsaturated 5 to 6-membered heteromonocyclic group containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms, for example, thiazolyl, thiadiazolyl [e.g., 1,2,4- thiadiazolyl, 1,3,4-thiadiazolyl, 1,2,5-thiadiazolyl]; unsaturated condensed heterocyclic group containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms [e.g., benzothiazolyl, benzothiadiazolyl] and the like. The term also embraces radicals where heterocyclic radicals are fused with aryl radicals. Examples of such fused bicyclic radicals include benzofuran, benzothiophene, and the like. Said "heterocyclyl" group may have 1 to 3 substituents such as lower alkyl, hydroxy, oxo, amino and lower alkylamino. Preferred heterocyclic radicals include five to ten membered fused or unfused radicals. More preferred examples of heteroaryl radicals include benzofuryl, 2,3-dihydrobenzofuryl, benzothienyl, indolyl, dihydroindolyl, chromanyl, benzopyran, thiochromanyl, benzothiopyran, benzodioxolyl, benzodioxanyl, pyridyl, thienyl, thiazolyl, oxazolyl, furyl, and pyrazinyl. Even more preferred heteroaryl radicals are 5- or 6-membered heteroaryl, containing one or two heteroatoms selected from sulfur nitrogen and oxygen, selected from thienyl, furanyl, pyrrolyl, thiazolyl, oxazolyl, imidazolyl, pyrazolyl, isoxazolyl, isothiazolyl, pyridyl, piperidinyl and pyrazinyl. The term "sulfonyl", whether used alone or linked to other terms such as alkylsulfonyl, denotes respectively divalent radicals $-SO_2-$. "Alkylsulfonyl" embraces alkyl radicals attached to a sulfonyl radical, where alkyl is defined as above. More preferred alkylsulfonyl radicals are "lower alkylsulfonyl" radicals having one to six carbon atoms. Even more preferred are lower alkylsulfonyl radicals having one to three carbon atoms.

Examples of such lower alkylsulfonyl radicals include methylsulfonyl, ethylsulfonyl and propylsulfonyl. "Haloalkylsulfonyl" embraces haloalkyl radicals attached to a sulfonyl radical, where haloalkyl is defined as

5 above. More preferred haloalkylsulfonyl radicals are "lower haloalkylsulfonyl" radicals having one to six carbon atoms. Even more preferred are lower haloalkylsulfonyl radicals having one to three carbon atoms. Examples of such lower haloalkylsulfonyl

10 radicals include trifluoromethylsulfonyl. The term "arylalkylsulfonyl" embraces aryl radicals as defined above, attached to an alkylsulfonyl radical. Examples of such radicals include benzylsulfonyl and phenylethylsulfonyl. The term "heterocyclosulfonyl"

15 embraces heterocyclo radicals as defined above, attached to a sulfonyl radical. More preferred heterocyclosulfonyl radicals contain 5-7 membered heterocyclo radicals containing one or two heteroatoms. Examples of such radicals include

20 tetrahydropyrrolylsulfonyl morpholinylsulfonyl and azepinylsulfonyl. The terms "sulfamyl," "aminosulfonyl" and "sulfonamidyl," whether alone or used with terms such as "N-alkylaminosulfonyl", "N-arylamino-sulfonyl", "N,N-dialkylaminosulfonyl" and "N-alkyl-N-

25 arylaminosulfonyl", denotes a sulfonyl radical substituted with an amine radical, forming a sulfonamide ($-\text{SO}_2\text{NH}_2$). The term "alkylaminosulfonyl" includes "N-alkylaminosulfonyl" and "N,N-dialkylaminosulfonyl" where sulfamyl radicals are substituted, respectively, with

30 one alkyl radical, or two alkyl radicals. More preferred alkylaminosulfonyl radicals are "lower alkylaminosulfonyl" radicals having one to six carbon atoms. Even more preferred are lower alkylaminosulfonyl radicals having one to three carbon atoms. Examples of

35 such lower alkylaminosulfonyl radicals include N-methylaminosulfonyl, N-ethylaminosulfonyl and N-methyl-N-ethylaminosulfonyl. The terms "N-arylamino-sulfonyl" and "N-alkyl-N-arylamino-sulfonyl" denote sulfamyl

radicals substituted, respectively, with one aryl radical, or one alkyl and one aryl radical. More preferred N-alkyl-N-arylaminosulfonyl radicals are "lower N-alkyl-N-arylsulfonyl" radicals having alkyl radicals of one to six carbon atoms. Even more preferred are lower N-alkyl-N-arylsulfonyl radicals having one to three carbon atoms. Examples of such lower N-alkyl-N-aryl-aminosulfonyl radicals include N-methyl-N-phenylaminosulfonyl and N-ethyl-N-phenylaminosulfonyl. Examples of such N-aryl-aminosulfonyl radicals include N-phenylaminosulfonyl. The term "arylalkylaminosulfonyl" embraces aralkyl radicals as described above, attached to an aminosulfonyl radical. More preferred are lower arylalkylaminosulfonyl radicals having one to three carbon atoms. The term "heterocyclylaminosulfonyl" embraces heterocyclyl radicals as described above, attached to an aminosulfonyl radical. The terms "carboxy" or "carboxyl", whether used alone or with other terms, such as "carboxyalkyl", denotes $\text{-CO}_2\text{H}$. The term "carboxyalkyl" embraces radicals having a carboxy radical as defined above, attached to an alkyl radical. The term "carbonyl", whether used alone or with other terms, such as "alkylcarbonyl", denotes -(C=O)- . The term "acyl" denotes a radical provided by the residue after removal of hydroxyl from an organic acid. Examples of such acyl radicals include alkanoyl and aroyl radicals. Examples of such lower alkanoyl radicals include formyl, acetyl, propionyl, butyryl, isobutyryl, valeryl, isovaleryl, pivaloyl, hexanoyl, trifluoroacetyl. The term "aroyl" embraces aryl radicals with a carbonyl radical as defined above. Examples of aroyl include benzoyl, naphthoyl, and the like and the aryl in said aroyl may be additionally substituted. The term "alkylcarbonyl" embraces radicals having a carbonyl radical substituted with an alkyl radical. More preferred alkylcarbonyl radicals are "lower alkylcarbonyl" radicals having one to six carbon

atoms. Even more preferred are lower alkylcarbonyl radicals having one to three carbon atoms. Examples of such radicals include methylcarbonyl and ethylcarbonyl. The term "haloalkylcarbonyl" embraces radicals having a carbonyl radical substituted with an haloalkyl radical. More preferred haloalkylcarbonyl radicals are "lower haloalkylcarbonyl" radicals having one to six carbon atoms. Even more preferred are lower haloalkylcarbonyl radicals having one to three carbon atoms. Examples of such radicals include trifluoromethylcarbonyl. The term "arylcarbonyl" embraces radicals having a carbonyl radical substituted with an aryl radical. More preferred arylcarbonyl radicals include phenylcarbonyl. The term "heteroarylcarbonyl" embraces radicals having a carbonyl radical substituted with a heteroaryl radical. Even more preferred are 5- or 6-membered heteroarylcarbonyl radicals. The term "arylalkylcarbonyl" embraces radicals having a carbonyl radical substituted with an arylalkyl radical. More preferred radicals are phenyl-C₁-C₃-alkylcarbonyl, including benzylcarbonyl. The term "heteroarylalkylcarbonyl" embraces radicals having a carbonyl radical substituted with a heteroarylalkyl radical. Even more preferred are lower heteroarylalkylcarbonyl radicals having 5-6-membered heteroaryl radicals attached to alkyl portions having one to three carbon atoms. The term "alkoxycarbonyl" means a radical containing an alkoxy radical, as defined above, attached via an oxygen atom to a carbonyl radical. Preferably, "lower alkoxycarbonyl" embraces alkoxy radicals having one to six carbon atoms. Examples of such "lower alkoxycarbonyl" ester radicals include substituted or unsubstituted methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl and hexyloxycarbonyl. Even more preferred are lower alkoxycarbonyl radicals having alkoxy portions of one to three carbon atoms. The term "aminocarbonyl" when used by itself or with other terms such as "aminocarbonylalkyl", "N-alkylaminocarbonyl", "N-

arylaminoacarbonyl", "N,N-dialkylaminocarbonyl", "N-alkyl-N-arylaminoacarbonyl", "N-alkyl-N-hydroxyaminocarbonyl" and "N-alkyl-N-hydroxyaminocarbonylalkyl", denotes an amide group of the formula $-C(=O)NH_2$. The terms "N-alkylaminocarbonyl" and "N,N-dialkylaminocarbonyl" denote aminocarbonyl radicals which have been substituted with one alkyl radical and with two alkyl radicals, respectively. More preferred are "lower alkylaminocarbonyl" having lower alkyl radicals as described above attached to an aminocarbonyl radical. The terms "N-arylaminoacarbonyl" and "N-alkyl-N-arylaminoacarbonyl" denote aminocarbonyl radicals substituted, respectively, with one aryl radical, or one alkyl and one aryl radical. The term "N-cycloalkylaminocarbonyl" denoted aminocarbonyl radicals which have been substituted with at least one cycloalkyl radical. More preferred are "lower cycloalkylaminocarbonyl" having lower cycloalkyl radicals of three to seven carbon atoms, attached to an aminocarbonyl radical. The term "aminoalkyl" embraces alkyl radicals substituted with amino radicals. The term "alkylaminoalkyl" embraces aminoalkyl radicals having the nitrogen atom substituted with an alkyl radical. Even more preferred are lower alkylaminoalkyl radicals having one to three carbon atoms. The term "heterocyclylalkyl" embraces heterocyclic-substituted alkyl radicals. More preferred heterocyclylalkyl radicals are "5- or 6- membered heteroarylalkyl" radicals having alkyl portions of one to six carbon atoms and a 5- or 6- membered heteroaryl radical. Even more preferred are lower heteroarylalkyl radicals having alkyl portions of one to three carbon atoms. Examples include such radicals as pyridylmethyl and thienylmethyl. The term "aralkyl" embraces aryl-substituted alkyl radicals. Preferable aralkyl radicals are "lower aralkyl" radicals having aryl radicals attached to alkyl radicals having one to six carbon atoms. Even more preferred are lower aralkyl radicals

phenyl attached to alkyl portions having one to three carbon atoms. Examples of such radicals include benzyl, diphenylmethyl and phenylethyl. The aryl in said aralkyl may be additionally substituted with halo, alkyl, alkoxy, haloalkyl and haloalkoxy. The term "arylalkenyl" embraces aryl-substituted alkenyl radicals. Preferable arylalkenyl radicals are "lower arylalkenyl" radicals having aryl radicals attached to alkenyl radicals having two to six carbon atoms. Examples of such radicals include phenylethenyl. The aryl in said arylalkenyl may be additionally substituted with halo, alkyl, alkoxy, haloalkyl and haloalkoxy. The term "arylalkynyl" embraces aryl-substituted alkynyl radicals. Preferable arylalkynyl radicals are "lower arylalkynyl" radicals having aryl radicals attached to alkynyl radicals having two to six carbon atoms. Examples of such radicals include phenylethynyl. The aryl in said aralkyl may be additionally substituted with halo, alkyl, alkoxy, haloalkyl and haloalkoxy. The terms benzyl and phenylmethyl are interchangeable. The term "alkylthio" embraces radicals containing a linear or branched alkyl radical, of one to ten carbon atoms, attached to a divalent sulfur atom. Even more preferred are lower alkylthio radicals having one to three carbon atoms. An example of "alkylthio" is methylthio, (CH₃-S-). The term "haloalkylthio" embraces radicals containing a haloalkyl radical, of one to ten carbon atoms, attached to a divalent sulfur atom. Even more preferred are lower haloalkylthio radicals having one to three carbon atoms. An example of "haloalkylthio" is trifluoromethylthio. The term "alkylsulfinyl" embraces radicals containing a linear or branched alkyl radical, of one to ten carbon atoms, attached to a divalent -S(=O)- atom. More preferred are lower alkylsulfinyl radicals having one to three carbon atoms. The term "arylsulfinyl" embraces radicals containing an aryl radical, attached to a divalent -S(=O)- atom. Even more preferred are optionally

substituted phenylsulfinyl radicals. The term "haloalkylsulfinyl" embraces radicals containing a haloalkyl radical, of one to ten carbon atoms, attached to a divalent $-S(=O)-$ atom. Even more preferred are lower haloalkylsulfinyl radicals having one to three carbon atoms. The terms "N-alkylamino" and "N,N-dialkylamino" denote amino groups which have been substituted with one alkyl radical and with two alkyl radicals, respectively. More preferred alkylamino radicals are "lower alkylamino" radicals having one or two alkyl radicals of one to six carbon atoms, attached to a nitrogen atom. Even more preferred are lower alkylamino radicals having one to three carbon atoms. Suitable "alkylamino" may be mono or dialkylamino such as N-methylamino, N-ethylamino, N,N-dimethylamino, N,N-diethylamino or the like. The term "arylamino" denotes amino groups which have been substituted with one or two aryl radicals, such as N-phenylamino. The "arylamino" radicals may be further substituted on the aryl ring portion of the radical. The term "heteroarylamino" denotes amino groups which have been substituted with one or two heteroaryl radicals, such as N-thienylamino. The "heteroarylamino" radicals may be further substituted on the heteroaryl ring portion of the radical. The term "aralkylamino" denotes amino groups which have been substituted with one or two aralkyl radicals. More preferred are phenyl- C_1-C_3 -alkylamino radicals, such as N-benzylamino. The "aralkylamino" radicals may be further substituted on the aryl ring portion of the radical. The terms "N-alkyl-N-arylamino" and "N-aralkyl-N-alkylamino" denote amino groups which have been substituted with one aralkyl and one alkyl radical, or one aryl and one alkyl radical, respectively, to an amino group. The term "arylthio" embraces aryl radicals of six to ten carbon atoms, attached to a divalent sulfur atom. An example of "arylthio" is phenylthio. The term "aralkylthio" embraces aralkyl radicals as described above, attached

to a divalent sulfur atom. More preferred are phenyl-C₁-C₃-alkylthio radicals. An example of "aralkylthio" is benzylthio. The term "aralkylsulfonyl" embraces aralkyl radicals as described above, attached to a divalent
5 sulfonyl radical. More preferred are phenyl-C₁-C₃-alkylsulfonyl radicals. The term "aryloxy" embraces optionally substituted aryl radicals, as defined above, attached to an oxygen atom. Examples of such radicals include phenoxy. The term "aralkoxy" embraces oxy-
10 containing aralkyl radicals attached through an oxygen atom to other radicals. More preferred aralkoxy radicals are "lower aralkoxy" radicals having optionally substituted phenyl radicals attached to lower alkoxy radical as described above.

15 The present invention comprises a pharmaceutical composition comprising a therapeutically-effective amount of a compound of Formula I-I" in association with at least one pharmaceutically-acceptable carrier, adjuvant or
20 diluent.

The present invention also comprises a method of treating cyclooxygenase-2 mediated disorders, such as inflammation, in a subject, the method comprising treating the subject having or
25 susceptible to such disorder with a therapeutically-effective amount of a compound of Formula I-I".

Also included in the family of compounds of Formula I-I" are the stereoisomers thereof. Compounds of the
30 present invention can possess one or more asymmetric carbon atoms and are thus capable of existing in the form of optical isomers as well as in the form of racemic or nonracemic mixtures thereof. Accordingly, some of the compounds of this invention may be present
35 in racemic mixtures which are also included in this invention. The optical isomers can be obtained by resolution of the racemic mixtures according to conventional processes, for example by formation of

diastereoisomeric salts by treatment with an optically active base and then separation of the mixture of diastereoisomers by crystallization, followed by liberation of the optically active bases from these salts. Examples of appropriate bases are brucine, strychnine, dehydroabietylamine, quinine, cinchonidine, ephedrine, α -methylbenzylamine, amphetamine, deoxyephedrine, chloramphenicol intermediate, 2-amino-1-butanol, and 1-(1-naphthyl)ethylamine. A different process for separation of optical isomers involves the use of a chiral chromatography column optimally chosen to maximize the separation of the enantiomers. Still another available method involves synthesis of covalent diastereoisomeric molecules. The synthesized diastereoisomers can be separated by conventional means such as chromatography, distillation, crystallization or sublimation, and then hydrolyzed to deliver the enantiomerically pure compound. The optically active compounds of Formula I can likewise be obtained by utilizing optically active starting materials. These isomers may be in the form of a free acid, a free base, an ester or a salt. Additional methods for resolving optical isomers, known to those skilled in the art may be used, for example, those discussed by J. Jaques et al in Enantiomers, Racemates, and Resolutions, John Wiley and Sons, New York (1981).

Also included in the family of compounds of Formula I-I'' are the protected acids thereof, such as the esters, hydroxyamino derivatives, amides and sulfonamides. Thus primary and secondary amines can be reacted with the chromene-3-carboxylic acids of Formula I-I'' to form amides which can be useful as prodrugs. Preferred amines heterocyclicamines, including optionally substituted aminothiazoles, optionally substituted amino-isoxazoles, and optionally substituted aminopyridines; aniline derivatives; sulfonamides; aminocarboxylic acids; and the like. Additionally, 1-acyldihydroquinolines can behave as prodrugs for the 1H-

dihydroquinolines. The esters, hydroxyamino derivatives and sulfonamides can be prepared from the acids by methods known to one skilled in the art.

Also included in the family of compounds of Formula I-I'' are the pharmaceutically-acceptable salts thereof. The term "pharmaceutically-acceptable salts" embraces salts commonly used to form alkali metal salts and to form addition salts of free acids or free bases. The nature of the salt is not critical, provided that it is pharmaceutically-acceptable. Suitable pharmaceutically-acceptable acid addition salts of compounds of Formula I-I'' may be prepared from an inorganic acid or from an organic acid. Examples of such inorganic acids are hydrochloric, hydrobromic, hydroiodic, nitric, carbonic, sulfuric and phosphoric acid. Appropriate organic acids may be selected from aliphatic, cycloaliphatic, aromatic, araliphatic, heterocyclic, carboxylic and sulfonic classes of organic acids, example of which are formic, acetic, propionic, succinic, glycolic, gluconic, lactic, malic, tartaric, citric, ascorbic, glucuronic, maleic, fumaric, pyruvic,

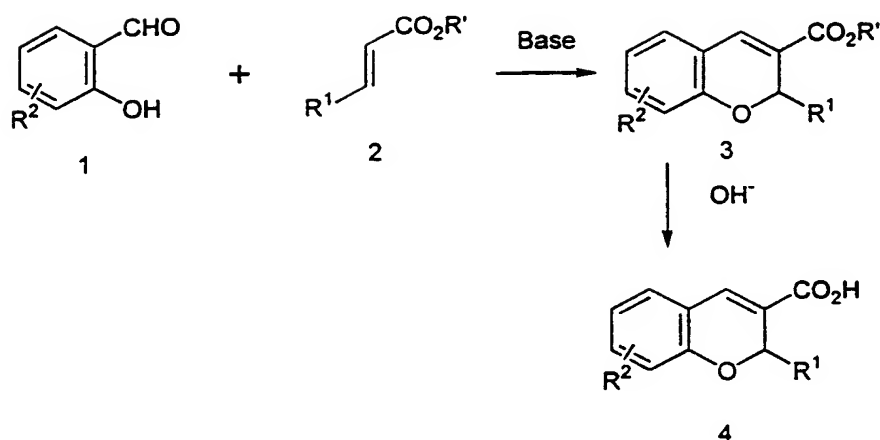
aspartic, glutamic, benzoic, anthranilic, mesylic,
salicyclic, salicyclic, 4-hydroxybenzoic, phenylacetic,
mandelic, embonic (pamoic), methanesulfonic,
ethanesulfonic, benzenesulfonic, pantothenic, 2-
5 hydroxyethanesulfonic, toluenesulfonic, sulfanilic,
cyclohexylaminosulfonic, stearic, algenic, β -
hydroxybutyric, salicyclic, galactaric and galacturonic
acid. Suitable pharmaceutically-acceptable base
addition salts of compounds of Formula I-I'' include
10 metallic salts, such as salts made from aluminum,
calcium, lithium, magnesium, potassium, sodium and zinc,
or salts made from organic bases including primary,
secondary and tertiary amines, substituted amines
including cyclic amines, such as caffeine, arginine,
15 diethylamine, N-ethyl piperidine, histidine, glucamine,
isopropylamine, lysine, morpholine, N-ethyl morpholine,
piperazine, piperidine, triethylamine, trimethylamine.
All of these salts may be prepared by conventional means
from the corresponding compound of the invention by
20 reacting, for example, the appropriate acid or base with
the compound of Formula I-I''.

GENERAL SYNTHETIC PROCEDURES

25 The compounds of the invention can be
synthesized according to the following procedures
of Schemes 1-16, wherein the R¹-R⁶ substituents are
as defined for Formulas I-II, above, except where
further noted.

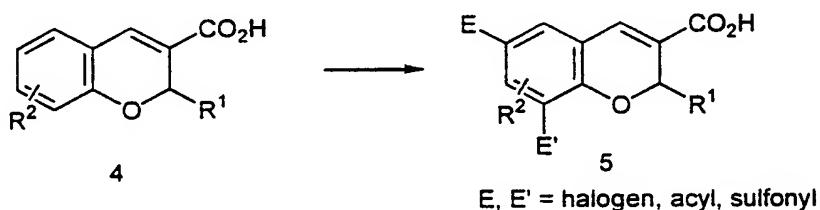
30

SCHEME 1

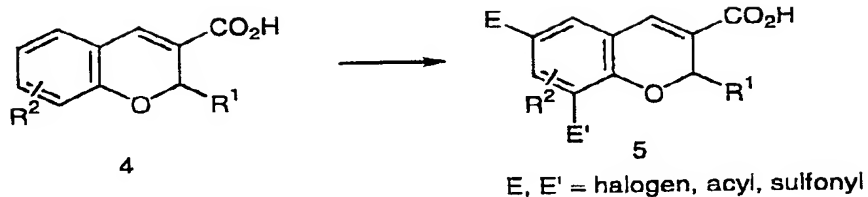


Synthetic Scheme 1 illustrates the general method for the preparation of a wide variety of substituted 2H-1-benzopyran derivatives 3 and 4. In step 1, a representative ortho-hydroxybenzaldehyde (salicylaldehyde) derivative 1 is condensed with an acrylate derivative 2 in the presence of base, such as potassium carbonate in a solvent such as dimethylformamide, to afford the desired 2H-1-benzopyran ester 3. An alternative base-solvent combination for this condensation includes an organic base such as triethylamine and a solvent such as dimethyl sulfoxide. In step 2 the ester is hydrolyzed to the corresponding acid, such as by treatment with aqueous base (sodium hydroxide) in a suitable solvent such as ethanol to afford after acidification the substituted 2H-1-benzopyran-3-carboxylic acid 4.

SCHEME 2



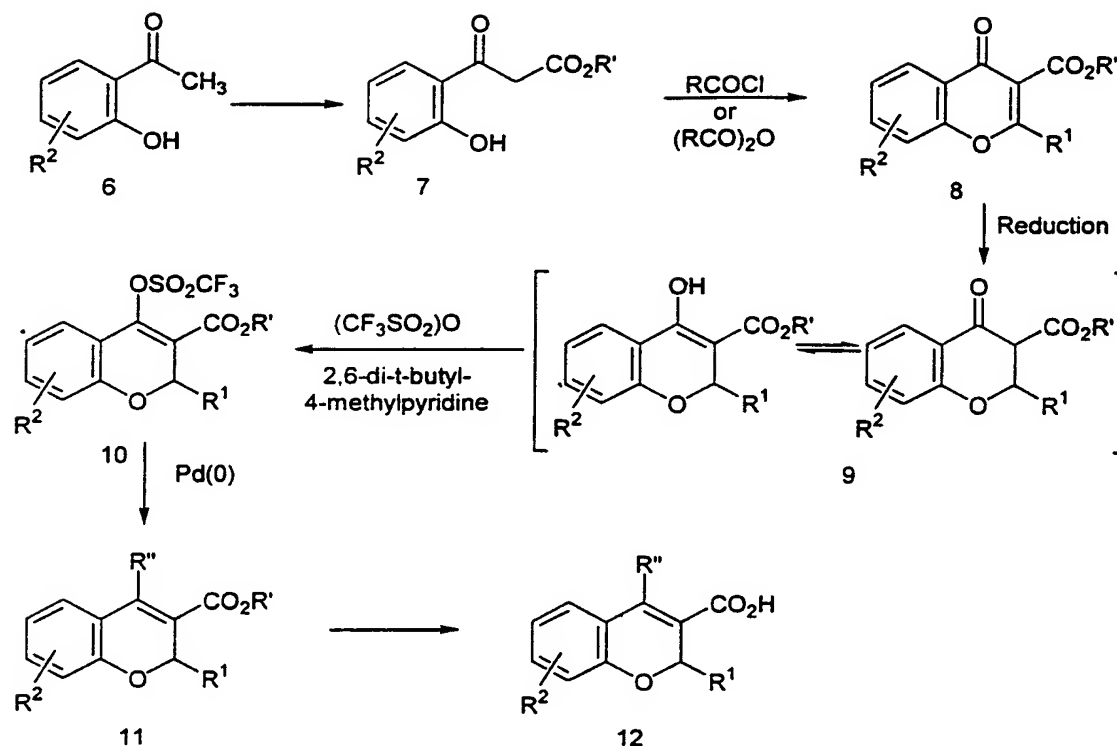
SCHEME 2



Synthetic Scheme 2 shows the general method for functionalizing selected 2H-1-benzopyrans. Treatment of the 2H-1-benzopyran carboxylic acid 4 or ester 3 with an electrophilic agent makes a 6-substituted 2H-1-benzopyran 5. A wide variety of electrophilic agents react selectively with 2H-1-benzopyrans 4 in the 6-position to provide new analogs in high yield.

Electrophilic reagents such as halogen (chlorine or bromine) give the 6-halo derivatives. Chlorosulfonic acid reacts to afford the 6-position sulfonyl chloride that can further be converted to a sulfonamide or sulfone. Friedel-Crafts acylation of 4 provides 6-acylated 2H-1-benzopyrans in good to excellent yield. A number of other electrophiles can be used to selectively react with these 2H-1-benzopyrans in a similar manner. A 6-position substituted 2H-1-benzopyran can react with an electrophilic reagent at the 8-position using similar chemistries to that described for electrophilic substitution of the 6-position. This yields an 2H-1-benzopyran which is substituted at both the 6 and 8 positions.

SCHEME 3



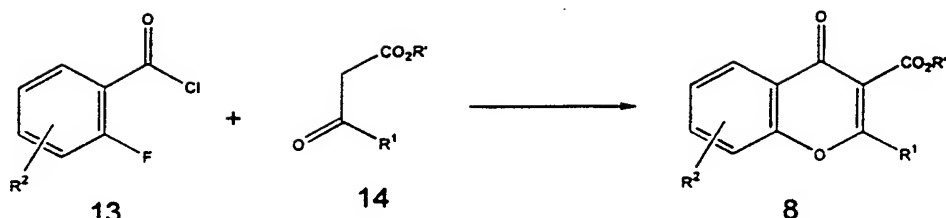
- Synthetic Scheme 3 illustrates a second general synthesis of substituted 2H-1-benzopyran-3-carboxylic acids which allows substitution at position 4 of the 2H-1-benzopyran. In this case a commercially or synthetically available substituted ortho-hydroxy acetophenone 6 is treated with two or more equivalents of a strong base such as lithium bis(trimethylsilyl)amide in a solvent such as tetrahydrofuran (THF), followed by reaction with diethyl carbonate to afford the beta-keto ester 7. Ester 7 is condensed with an acid chloride or anhydride in the presence of a base such as potassium carbonate in a solvent such as toluene with heat to afford 4-oxo-4H-1-benzopyran 8. Reduction of the olefin can be accomplished by a variety of agents including sodium borohydride (NaBH₄) in solvent mixtures such as ethanol and tetrahydrofuran (THF), or by use of triethylsilane

in a solvent such as trifluoroacetic acid, or by catalytic reduction using palladium on charcoal and hydrogen gas in a solvent such as ethanol to yield the new beta-keto ester 9 (two tautomeric structures shown).

- 5 Acylation of the oxygen of the ketone enolate in the presence of a base such as 2,6-di-tert-butyl-4-methylpyridine, an acylating agent such as trifluoromethanesulfonic anhydride, and using a solvent such as methylene chloride yields the enol-triflate 10.
- 10 Triflate 10 can be reduced with reagents such as tri-n-butyltin hydride, lithium chloride and a palladium (0) catalyst such as tetrakis(triphenylphosphine)palladium (0) in a solvent such as tetrahydrofuran to yield 2H-1-benzopyran ester 11 where R'' is hydrogen. The ester 11
- 15 can be saponified with a base such as 2.5 N sodium hydroxide in a mixed solvent such as tetrahydrofuran-ethanol-water (7:2:1) to yield the desired substituted 2H-1-benzopyran-3-carboxylic acid.

- To incorporate a carbon fragment R³ one can treat
- 20 triflate 10 with reagents known to undergo "cross-coupling" chemistries such as tributylethylenyltin, lithium chloride and a palladium(0) catalyst such as tetrakis(triphenylphosphine)palladium (0) in a solvent such as tetrahydrofuran to yield 2H-1-benzopyran ester
- 25 11 where R³ is a vinyl moiety. The ester 6 can be saponified with a base such as 2.5 N sodium hydroxide in a mixed solvent such as tetrahydrofuran-ethanol-water (7:2:1) to yield the desired 4-vinyl-2H-1-benzopyran-3-carboxylic acid (12, R'' = CH₂CH-). Similarly triflate
- 30 10 can be converted under similar conditions using tri-n-butylphenyltin to 2H-1-benzopyran where R³ = phenyl and by hydrolysis of the ester converted to the carboxylic acid 12 where R³ = phenyl. Using a similar strategy, substituents which be incorporated as
- 35 substituent R³ can be substituted olefins, substituted aromatics, substituted heteroaryl, acetylenes and substituted acetylenes.

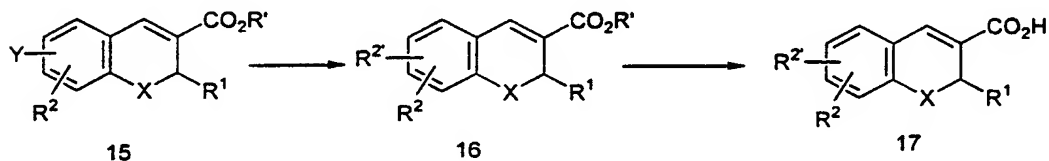
SCHEME 4



Synthetic Scheme 4 shows an alternative general procedure for the preparation of 4-oxo-4H-1-benzopyran **8**. Treatment of an ortho-fluorobenzoyl chloride with an appropriately substituted beta-keto ester **14** with a base such as potassium carbonate in a solvent such as toluene provides 4-oxo-4H-1-benzopyran **8**. 4-Oxo-4H-1-benzopyran **8** can be converted to 2H-1-benzopyran **12** as described in Scheme 3.

10

SCHEME 5



Y = Br, I, CF₃SO₃

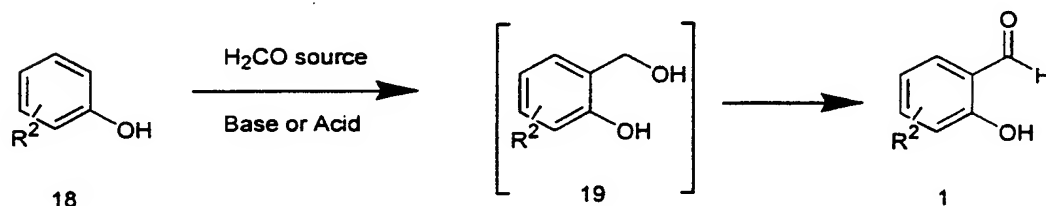
Synthetic Scheme 5 shows a general method for substitution of the aromatic ring of the 2H-1-benzopyran. This can be accomplished through organo-palladium mediated "cross-coupling" chemistries using a palladium (0) catalyst to couple benzopyran **15** at position Y, where Y is iodide, bromide or triflate, with an acetylene, olefin, nitrile, or aryl coupling agent. Substituted acetylenes as the coupling agent will provide the corresponding substituted acetylene. Substituted aryl moieties can be incorporated using arylboronic acids or esters; nitriles can be incorporated by use of zinc (II) cyanide. The resulting

ester **16** can be converted to carboxylic acid **17** as described in Scheme 1.

Another approach to substitution of the aryl moiety of the benzopyran **15** is to convert Y, where Y is iodide or bromide, to a perfluoroalkyl moiety. Exemplary of this transformation is the conversion of **15** (Y = iodide) to **16** (R^{2'} = pentafluoroethyl) using a potassium pentafluoropropionate and copper (I) iodide in hexamethylphosphoramide (HMPA). The resulting ester **16** can be converted to carboxylic acid **15** as described in Scheme 1.

A similar method adds substitution of the aromatic ring in dihydroquinoline-3-carboxylates. This can be accomplished through organopalladium couplings with aryl iodides, bromides, or triflates and various coupling agents (R. F. Heck, *Palladium Reagents in Organic Synthesis*. Academic Press 1985). When using a suitable palladium catalyst such as tetrakis(triphenylphosphine)palladium(0) in this reaction, coupling agents such as alkynes provide disubstituted alkynes, phenyl boronic acids afford biphenyl compounds, and cyanides produce arylcyano compounds. A number of other palladium catalysts and coupling reagents could be used to selectively react with appropriately substituted dihydroquinoline-3-carboxylates in a similar manner.

SCHEME 6



Synthetic Scheme 6 shows a general synthetic route for conversion of a commercially or synthetically available substituted phenol into a substituted

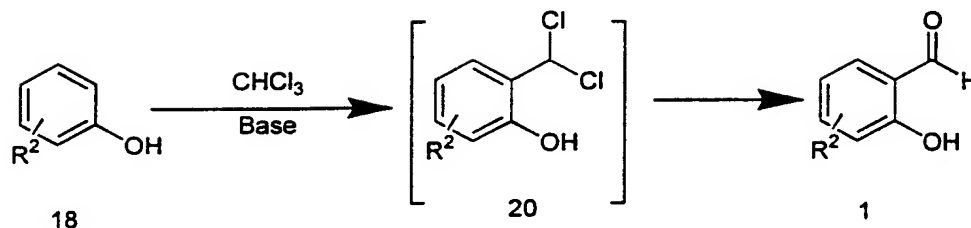
salicylaldehyde. Several different methods which utilize formaldehyde or a chemically equivalent reagent are described in detail below.

Reaction of an appropriately substituted phenol **18** in basic media with formaldehyde (or chemical equivalent) will yield the corresponding salicylaldehyde **1**. The intermediate, ortho-hydroxymethylphenol **19**, will under appropriate reaction conditions be oxidized to the salicylaldehyde **1** in situ. The reaction commonly employs ethyl magnesium bromide or magnesium methoxide (one equivalent) as the base, toluene as the solvent, paraformaldehyde (two or more equivalents) as the source of formaldehyde, and employs hexamethylphosphoramide (HMPA) or *N,N,N',N'*-tetramethylethylenediamine (TMEDA). (See: Casiraghi, G. et al., J.C.S. Perkin I, **1978**, 318-321.)

Alternatively an appropriately substituted phenol **18** may react with formaldehyde under aqueous basic conditions to form the substituted ortho-hydroxybenzyl alcohol **19** (See: a) J. Leroy and C. Wakselman, J. Fluorine Chem., 40, 23-32 (**1988**). b) A. A. Moshfegh, et al., Helv. Chim. Acta., 65, 1229-1232 (**1982**)). Commonly used bases include aqueous potassium hydroxide or sodium hydroxide. Formalin (38% formaldehyde in water) is commonly employed as the source of formaldehyde. The resulting ortho-hydroxybenzyl alcohol **19** can be converted to the salicylaldehyde **1** by an oxidizing agent such as manganese (IV) dioxide in a solvent such as methylene chloride or chloroform (See: R-G. Xie, et al., Synthetic Commun. 24, 53-58 (**1994**)).

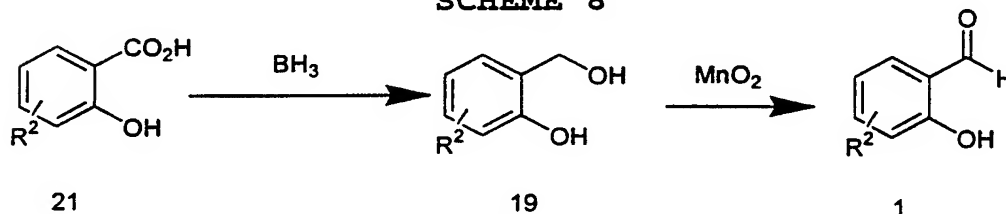
An appropriately substituted phenol **18** can be treated under acidic conditions with hexamethylenetetramine (HMTA) to prepare the salicylaldehyde **1** (Duff Reaction; See: Y. Suzuki, and H. Takahashi, Chem. Pharm. Bull., 31, 1751-1753 (**1983**)). This reaction commonly employs acids such as acetic acid, boric acid, methanesulfonic acid, or trifluoromethanesulfonic acid. The source of formaldehyde commonly used is hexamethylenetetramine.

SCHEME 7



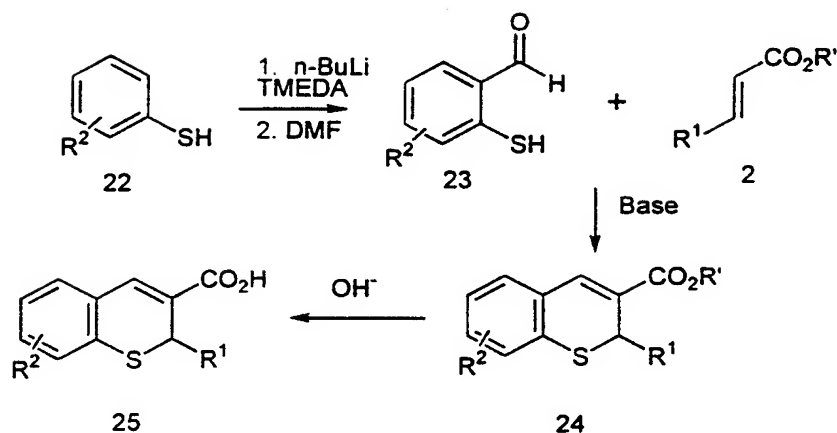
Synthetic Scheme 7 shows the Reimer-Tiemann reaction in which an commercially or synthetically available appropriately substituted phenol **18** will under basic conditions react with chloroform to yield a substituted salicylaldehyde **1** (See: Cragoe, E.J.; Schultz, E.M., U.S. Patent 3 794 734, 1974).

SCHEME 8



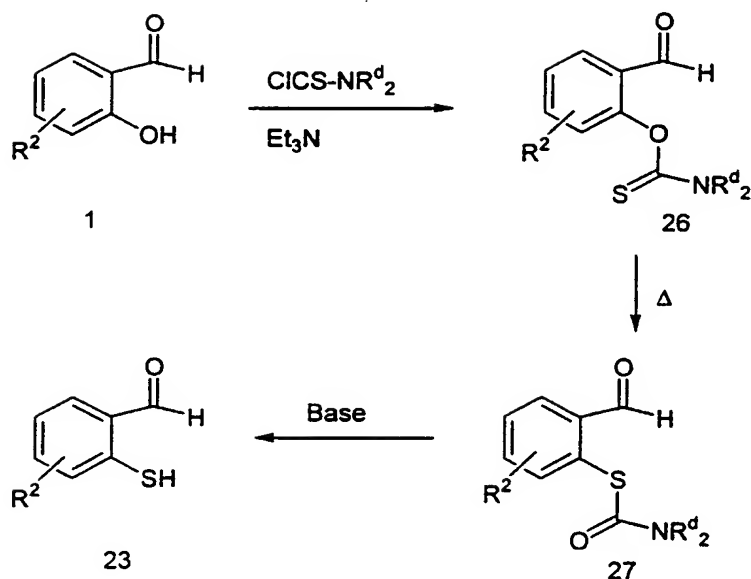
Synthetic Scheme 8 shows the conversion of a commercially or synthetically available appropriately substituted salicylic acid **21** to its respective salicylaldehyde **1** via an intermediate 2-hydroxybenzyl alcohol **19**. Reduction of the salicylic acid **21** can be accomplished with a hydride reducing agent such as borane in a solvent such as tetrahydrofuran. Treatment of the intermediate 2-hydroxybenzyl alcohol **19** with an oxidizing agent such as manganese (IV) oxide in a solvent such as methylene chloride or chloroform provides salicylaldehyde **1**.

SCHEME 9



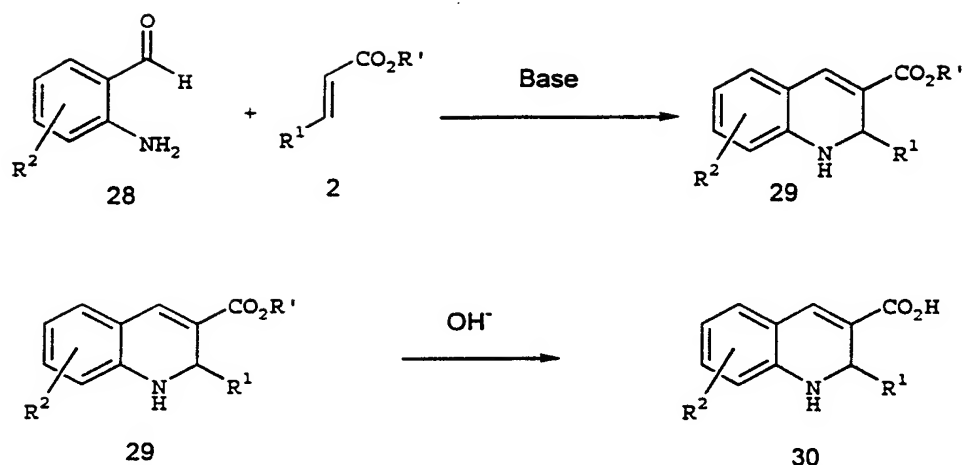
Synthetic Scheme 9 illustrates a general synthetic method for preparation of a wide variety of substituted
 5 2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acids (25). In step 1, an appropriately commercially or synthetically available substituted thiophenol 22 is ortho-metallated with a base such as n-butyllithium employing TMEDA (N,N,N',N'-tetramethylethylenediamine)
 10 followed by treatment with dimethylformamide to provide the 2-mercaptobenzaldehyde 23. Condensation of the 2-mercaptobenzaldehyde 23 with an acrylate 2 in the presence of base provides ester 24 which can be saponified in the presence of aqueous base to afford the
 15 substituted 2H-1-benzothiopyran-3-carboxylic acids 25.

SCHEME 10



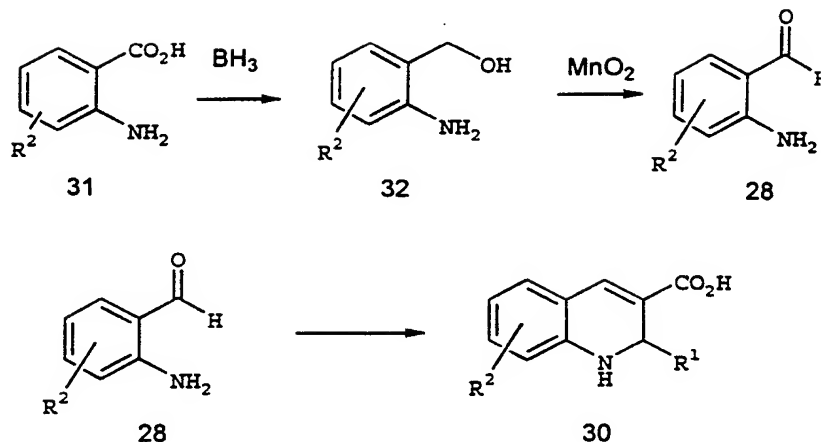
Synthetic Scheme 10 shows a method for preparing a substituted 2-mercaptobenzaldehyde from an appropriate commercially or synthetically available substituted salicylaldehyde. In step 1, the phenolic hydroxyl of salicylaldehyde **1** is converted to the corresponding O-aryl thiocarbamate **26** by acylation with an appropriately substituted thiocarbamoyl chloride such as *N,N*-dimethylthiocarbamoyl chloride in a solvent such as dimethylformamide using a base such as triethylamine. In Step 2, O-aryl thiocarbamate **26** rearranges to S-aryl thiocarbamate **27** when heated sufficiently such as to 200 °C using either no solvent or a solvent such as *N,N*-dimethylaniline (See: A. Levai, and P. Sebok, *Synth. Commun.*, 22 1735-1750 (1992)). Hydrolysis of S-aryl thiocarbamate **27** with a base such as 2.5 N sodium hydroxide in a solvent mixture such as tetrahydrofuran and ethanol yields the substituted 2-mercaptobenzaldehyde **23** which can be converted to the substituted 2H-1-benzothiopyran-3-carboxylic acids **25** as described in Scheme 9.

SCHEME 11



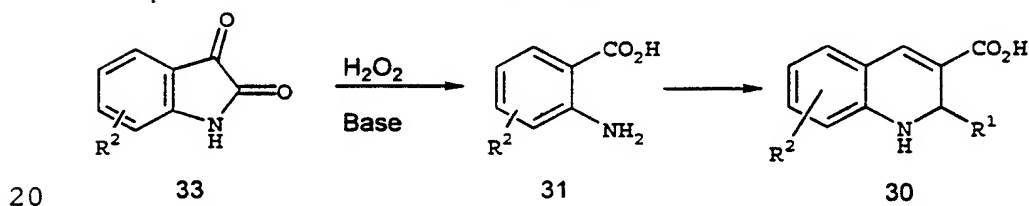
Synthetic Scheme 11 illustrates the general method for the preparation of a wide variety of dihydroquinoline-3-carboxylic acid derivatives 30. R^2 represents the aromatic substitution of commercially and synthetically available 2-aminobenzaldehydes 28. The 2-amino-benzaldehyde derivative 28, where R^2 represents various substitutions, is condensed with an acrylate derivative 2 in the presence of a base such as potassium carbonate, triethylamine, or diazabicyclo[2.2.2]undec-7-ene in solvents such as dimethylformamide to afford the dihydroquinoline-3-carboxylate esters 29. The ester 29 can be saponified to the corresponding acid, such as by treatment with aqueous inorganic base such as 2.5 N sodium hydroxide in a suitable solvent such as ethanol to afford after acidification the desired dihydroquinoline-3-carboxylic acid 30.

SCHEME 12



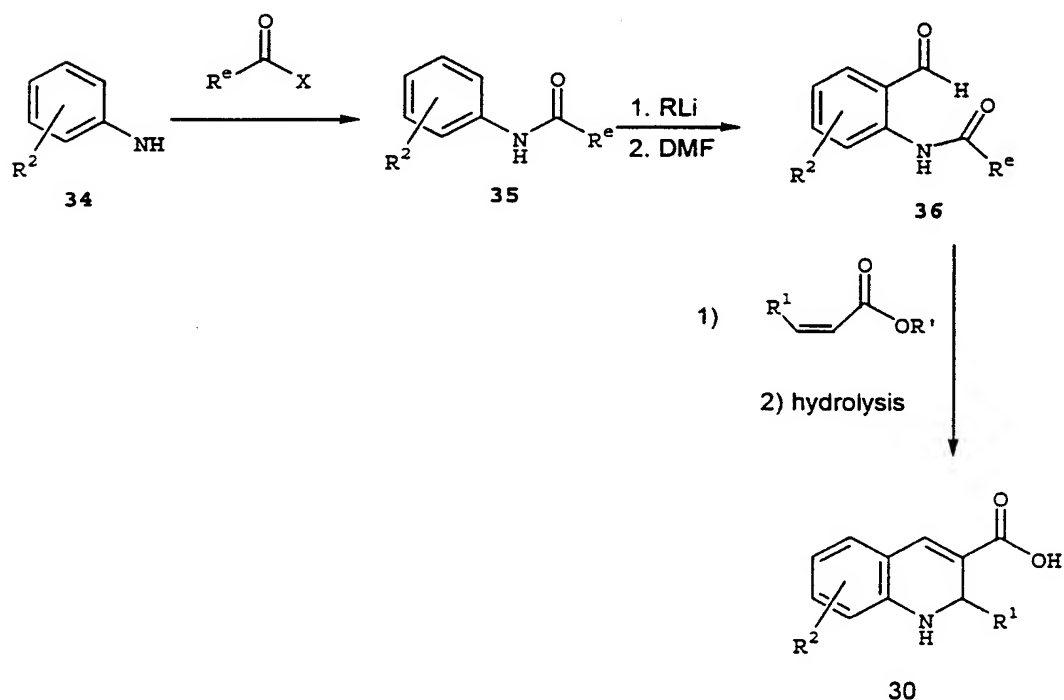
Synthetic Scheme 12 illustrates the preparation of dihydroquinoline-3-carboxylic acid **30** from 2-aminobenzoic acids **31**. R^2 represents the aromatic substitution of commercially and synthetically available 2-aminobenzoic acids **31**. Reduction of the representative 2-aminobenzoic acid **31** to the desired 2-aminobenzyl alcohol **32** was accomplished with a hydride reducing agent such as borane in a solvent such as tetrahydrofuran. Treatment of the desired 2-aminobenzyl alcohol **32** with an oxidizing agent such as manganese(IV)oxide in a solvent such as methylene chloride provides the representative 2-aminobenzaldehydes **28**. (C. T. Alabaster, et al. *J. Med. Chem.* 31, 2048-2056 (1988)) The 2-aminobenzaldehydes were converted to the desired dihydroquinoline-3-carboxylic acid **30** as described in Scheme 11.

SCHEME 13



for the preparation of a wide variety of dihydroquinoline-3-carboxylic acid derivatives **30** from isatins **33**. R^2 represents the aromatic substitution of commercially and synthetically available isatins **33**. A representative isatin **33** was treated with basic peroxide generated from hydrogen peroxide and a base such as sodium hydroxide to afford the desired representative 2-aminobenzoic acids **31**. (M. S. Newman and M. W. Lougue, J. Org. Chem., 36, 1398-1401 (1971)) The 2-aminobenzoic acids **31** are subsequently converted to the desired dihydroquinoline-3-carboxylic acid derivatives **30** as described in synthetic Scheme 12.

SCHEME 14

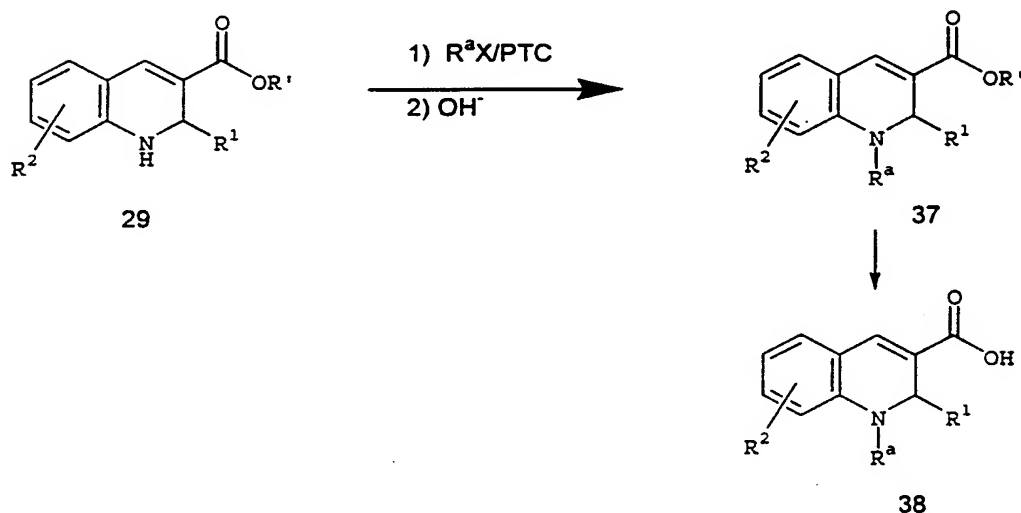


15

Synthetic Scheme 14 is another general method for the preparation of dihydroquinoline-3-carboxylic acid derivatives **30**. In step 1, an appropriate commercially or synthetically available substituted aniline **34** can be

chloride yielding an amide **35**. The *ortho*-dianion of amide **35** is prepared by treating amide **35** with organo-lithium bases such as *n*-butyllithium or *tert*-butyllithium in tetrahydrofuran at low temperature. The dianion is quenched with dimethylformamide to afford the acylated-2-amino-benzaldehydes **36**. (J. Turner, *J. Org. Chem.*, 48, 3401-3408 (1983)) Reaction of these aldehydes in the presence of bases such as lithium hydride with an acrylate followed by work up with aqueous inorganic bases and hydrolysis, such as by treatment with aqueous base (sodium hydroxide) in a suitable solvent such as ethanol affords, after acidification, a dihydroquinoline-3-carboxylic acid **30**.

SCHEME 15

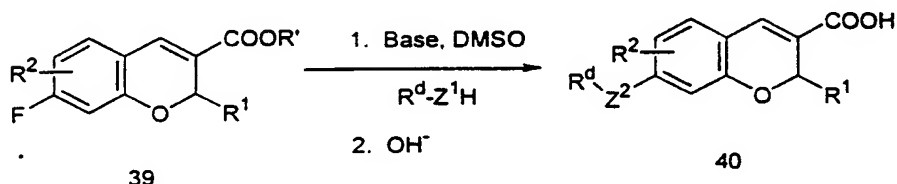


Synthetic Scheme 15 shows a general method for alkylation of the nitrogen of dihydroquinoline-3-carboxylate ester derivatives **29**. The step involves treatment of dihydroquinoline-3-carboxylate ester derivatives **29** with alkyl halides such as iodoethane in the presence of phase transfer catalysts such as tetrabutylammonium iodide, and a base such as caustic (50% aqueous sodium hydroxide) in a solvent such as dichloromethane. These conditions afford the N-

alkylated dihydroquinoline-3-carboxylate esters **37**. Saponification of **37** with aqueous base provides N-alkylated-dihydroquinoline-3-carboxylic acid derivatives **38**.

5

SCHEME 16



Synthetic Scheme 16 shows a general method for the preparation of a 7-ether ($Z^1=O$) or thioether ($Z^1=S$) substituted benzopyran-3-carboxylic ester. An appropriately substituted phenol, thiophenol, hydroxy-heterocycle, mercaptoheterocycle, alcohol, or alkylthiol can be condensed under basic conditions using a base such as potassium carbonate in a solvent such as dimethylsulfoxide at a temperature above room temperature such as 100 °C, with an appropriately substituted 7-fluorobenzopyran derivative **30** to yield the corresponding ether or thioether. Hydrolysis of the ester with an aqueous base such as lithium hydroxide or sodium hydroxide in a solvent mixture such as tetrahydrofuran-ethanol-water yields acid. When appropriate a thioether ($Z^2=S$) can be oxidized to the sulfoxide ($Z^2=SO$) or sulfone ($Z^2=SO_2$) with an oxidant such as OXONE® or m-CPBA either before or after ester hydrolysis. In this chemistry R^d can include aryl, heteroaryl, heterocyclic, alicyclic, branched or linear aliphatic branched or linear perfluoroaliphatic moiety.

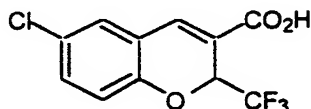
The following examples contain detailed descriptions of the methods of preparation of compounds of Formulas I-II. These detailed descriptions fall within the scope, and serve to exemplify, the above described General Synthetic Procedures which form part of the invention. These detailed descriptions are presented for illustrative purposes only and are not intended as a restriction on the scope of the invention. All parts are by weight and temperatures are in Degrees centigrade unless otherwise indicated. All compounds showed NMR spectra consistent with their assigned structures.

The following abbreviations are used:

- HCl - hydrochloric acid
- 15 MgSO₄ - magnesium sulfate
- Na₂SO₄ - sodium sulfate
- DMF - dimethylformamide
- THF - tetrahydrofuran
- NaOH - sodium hydroxide
- 20 EtOH - ethanol
- K₂CO₃ - potassium carbonate
- CDCl₃ - deuterated chloroform
- CD₃OD - deuterated methanol
- Et₂O - diethyl ether
- 25 EtOAc - ethyl acetate
- NaHCO₃ - sodium bicarbonate
- KHSO₄ - potassium sulfate
- NaBH₄ - sodium borohydride
- TMEDA - tetramethylethylenediamine
- 30 HMTA - hexamethylenetetraamine
- DMSO - dimethyl sulfoxide
- HMPA hexamethyl phosphoric triamide
- h - hour
- P₂O₅ - phosphorous pentoxide
- 35 HOAc - acetic acid
- NaOD - deuterated sodium hydroxide
- n-BuLi - n-butyllithium

CH₂Cl₂ - methylene chloride
TFA - trifluoroacetic acid
OXONE - potassium peroxymonosulfate

5

EXAMPLE 1

10

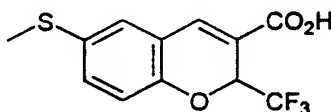
6-Chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acidStep 1. Preparation of ethyl 6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

A mixture of 5-chlorosalicylaldehyde (20.02 g, 0.128 mole) and ethyl 4,4,4-trifluorocrotonate (23.68 g, 0.14 mole) was dissolved in anhydrous DMF, warmed to 60 °C and treated with anhydrous K₂CO₃ (17.75 g, 0.128 mole). The solution was maintained at 60 °C for 20 hours, cooled to room temperature, and diluted with water. The solution was extracted with ethyl acetate. The combined extracts were washed with brine, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to afford 54.32 g of an oil. The oil was dissolved in 250 mL of methanol and 100 mL of water, whereupon a white solid formed that was isolated by filtration, washed with water and dried *in vacuo*, to afford the ester as a yellow solid (24.31 g, 62%): mp 62-64 °C. ¹H NMR (CDCl₃/90 MHz) 7.64 (s, 1H), 7.30-7.21 (m, 2H), 6.96 (d, 1H, *J* = Hz), 5.70 (q, 1H, *J* = Hz), 4.30 (q, 2H, *J* = 7.2 Hz), 1.35 (t, 3H, *J* = 7.2 Hz).

Step 2. Preparation of 6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

A solution of the ester from Step 1 (13.02 g, 42 mmole) was dissolved in 200 mL of methanol and 20 mL of water, treated with lithium hydroxide (5.36 g, 0.128 mole) and stirred at room temperature for 16 hours. The reaction mixture was acidified with 1.2 N HCl, whereupon a solid formed that was isolated by filtration. The solid was washed with 200 mL of water and 200 mL of hexanes and dried *in vacuo* to afford the title compound as a yellow solid (10.00 g, 85%): mp 181-184 °C.

EXAMPLE 2



6-(Methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of 5-(methylthio)salicylaldehyde.

Ethyl magnesium bromide (38 mL of a 3.0 M solution in diethyl ether, 113.8 mmole) was chilled with an ice-water bath. To the chilled solution was added a solution of 4-(methylthio)phenol (15.95 g, 113.8 mmole) in diethyl ether (30 mL) over 0.15 hour during which time gas was evolved. The reaction was held at 0 °C for 0.5 hour, at room temperature for 0.5 hour, and the addition funnel replaced with a distillation head. Toluene (250 mL) and the diethyl ether were distilled out of the reactor. The reaction was cooled, toluene (250 mL) and hexamethylphosphoramide (HMPA) (19.8 mL, 20 .4 g, 113.8 mmole) were added, and the resulting mixture

was stirred for 0.25 hours. The distillation head was replaced with a condenser and paraformaldehyde (8.5 g, 284.4 mmole) was added. The reaction was heated to 90 °C for 3 hours. The reaction mixture was cooled to room temperature, was acidified with 1N HCl and the layers separated. The organic phase was washed with water, and with brine, dried over MgSO₄, filtered, and concentrated *in vacuo* to yield a solid. This solid was purified by silica chromatography (hexanes-ethyl acetate, 5:1) yielding the salicylaldehyde as a yellow crystalline solid (6.01 g) of suitable purity to be used in the next reaction without further purification.

15 Step 2. Preparation of ethyl 6-(methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

5-Methylthiosalicylaldehyde (Step 1) (2.516 g, 14.96 mmole) was added to dimethylformamide (3.5 mL), potassium carbonate (2.27 g, 16.45 mmole) and ethyl 4,4,4-trifluorocrotonate (3.3 mL, 3.8 g, 22.4 mmole). The mixture was heated to 65 °C for 3 h. The reaction was cooled to room temperature, poured into H₂O (50 mL), and extracted with diethyl ether (2 X 75 mL). The combined ethereal phases were washed with aqueous NaHCO₃ solution (3 X 50 mL), aqueous 2 N HCl solution (3 X 50 mL), and brine (3 X 50 mL), dried over MgSO₄, filtered, diluted with isooctane and partially concentrated *in vacuo* causing the precipitation of the ethyl ester (2.863 g, 60 %) as a yellow powder: mp 87.8-89.6 °C. This ester was of suitable purity to use without further purification.

Step 3. Preparation of 6-(methylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

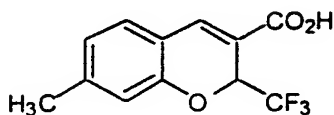
35 The ester (Step 2) was hydrolyzed to form the carboxylic acid via a method similar to that described in Example 1, Step 2: mp 166.3-167.9 °C. ¹H NMR (acetone-d₆/300 MHz) 7.87 (s, 1H), 7.43 (d, 1H, J = 2.2

Hz), 7.33 (dd, 1H, $J = 8.5, 2.4$ Hz), 6.98 (d, 1H, $J = 8.5$ Hz), 5.79 (q, 1H, $J = 7.0$ Hz), 2.48 (s, 3H).

FABLRMS m/z 291 (M+H). ESHRMS m/z 289.0152 (M-H, Calc'd 289.0146). Anal. Calc'd for $C_{12}H_9F_3O_3S_1$: C,

5 49.66; H, 3.13; S, 11.05. Found: C, 49.57; H, 3.02; S, 11.37.

EXAMPLE 3

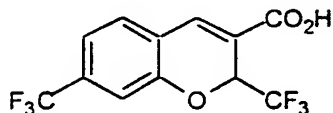


7-Methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

15 3-Methylphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 202.1-203.1 °C. 1H NMR ($CDCl_3$ /300 MHz) 7.84 (s, 1H), 7.12 (d, 1H, $J = 8.3$ Hz), 6.82 (m, 2H), 5.65 (q, 1H, $J =$
20 6.8 Hz), 2.35 (s, 3H). FABLRMS m/z 259 (M+H). FABHRMS m/z 259.0576 (M+H, Calc'd 259.0582). Anal. Calc'd for $C_{12}H_9F_3O_3$: C, 55.82; H, 3.51. Found: C, 55.93; H, 3.59.

25

EXAMPLE 4

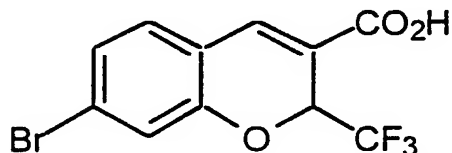


2,7-bis(Trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

30 3-(Trifluoromethyl)phenol was converted to the title compound by a procedure similar to that described in Example 2: mp 190.3-193.5

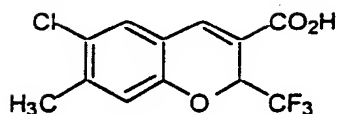
°C. ^1H NMR (acetone- d_6 /300 MHz) 7.98 (s, 1H), 7.73 (d, 1H, $J = 7.9$ Hz), 7.46 (d, 1H, $J = 7.9$ Hz), 7.36 (s, 1H), 5.93 (q, 1H, $J = 7.1$ Hz). FABLRMS m/z 313 (M+H). FABHRMS m/z 313.0267 (M+H, Calc'd 313.0299). Anal. Calc'd for $\text{C}_{12}\text{H}_6\text{F}_6\text{O}_3$: C, 46.17; H, 1.94. Found: C, 46.25; H, 2.00.

EXAMPLE 5



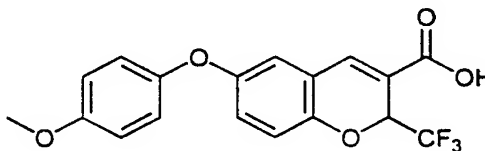
7-Bromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

3-Bromophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 198.4-199.5 °C. ^1H NMR (acetone- d_6 /300 MHz) 7.89 (s, 1H), 7.43 (d, 1H, $J = 8.1$ Hz), 7.31 (s, 1H), 7.30 (d, 1H, $J = 8.1$ Hz), 5.84 (q, 1H, $J = 7.1$ Hz). FABLRMS m/z 323 (M+H). Anal. Calc'd for $\text{C}_{11}\text{H}_6\text{BrF}_3\text{O}_3$: C, 40.90; H, 1.87. Found: C, 41.00; H, 1.85.

EXAMPLE 6

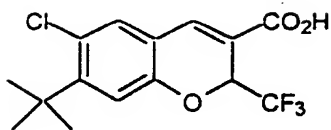
6-Chloro-7-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

4-Chloro-3-methylphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 207.5-209.3 °C. ¹H NMR (CDCl₃/300 MHz) 7.77 (s, 1H), 7.23 (s, 1H), 6.90 (s, 1H), 5.65 (q, 1H, J = 6.8 Hz), 2.37 (s, 3H). FABLRMS m/z 292 (M+H). FABHRMS m/z 299.0287 (M+Li, Calc'd 299.0274). Anal. Calc'd for C₁₂H₈ClF₃O₃: C, 49.25; H, 2.76; Cl, 12.11. Found: C, 49.37; H, 2.82; Cl, 12.17.

EXAMPLE 7

6-(4-Methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

4-(4-Methoxyphenyl)phenol was converted to the title compound by a procedure similar to that described in Example 2: mp 181.7-182.9 °C. ¹H NMR (acetone-d₆/300 MHz) 7.87 (s, 1H), 7.11 (m, 1H), 7.02 (m, 2H), 6.98 (m, 4H), 5.81 (q, 1H, J = 7.0 Hz), 3.80 (s, 3H). FABLRMS m/z 365 (M-H). FABHRMS m/z 367.0809 (M+H, Calc'd 367.0793). Anal. Calc'd for C₁₈H₁₃F₃O₅: C, 59.02; H, 3.58. Found: C, 59.10; H, 3.61.

EXAMPLE 8

5 **6-Chloro-7-(1,1-dimethylethyl)-2-**
 trifluoromethyl-2H-1-benzopyran-3-carboxylic
 acid

Step 1. Preparation of 4-tert-butylsalicylaldehyde.

10 A five liter three-neck round bottom flask equipped
with overhead mechanical stirrer and condenser was
charged with trifluoroacetic acid (2.4 L). A mixture of
3-tert-butylphenol (412 g, 2.8 mole) and HMTA (424 g,
3.0 mole) was added portion-wise causing an exotherm.
With cooling, the temperature was maintained under 80
15 °C. The reaction was heated at 80 °C for one hour, then
cooled, and water (2 L) added. After 0.5 hour
additional water (4 L) was added and the mixture was
extracted with ethyl acetate (6 L). The organic extract
was washed with water and brine. The resulting organic
20 phase was divided into 2 L volumes and each diluted with
water (1 L), and solid NaHCO₃ added until the mixture
was neutralized. The organic phases were isolated and
combined, dried over MgSO₄, filtered and concentrated *in*
vacuo yielding an oil. This oil was distilled at 95 °C
25 (0.8 mm) yielding the desired salicylaldehyde as an oil
(272.9 g, 56 %) which was of sufficient purity to be
used without further purification.

Step 2. Preparation of ethyl 7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A one liter three-neck flask was charged with 4-tert-butylsalicylaldehyde (Step 1) (100.0 g, 0.56 mole),
5 dimethylformamide (110 mL), and potassium carbonate (79.9 g, 0.58 mole) causing the temperature of the mixture to rise to 40 °C. Ethyl 4,4,4-trifluorocrotonate (118.0 g, 0.70 mole) in dimethylformamide (110 mL) was added and the mixture
10 heated to 60 °C at which time the reaction temperature rose to 70 °C. The reaction was cooled to 60 °C, maintained at 60 °C (with added heating) for 8.5 hours and cooled to room temperature. Ethyl acetate (600 mL) and 3 N HCl (600 mL) were added, mixed, and the layers
15 separated. The aqueous phase was extracted with ethyl acetate and the organic phases were combined. The combined organic phases were washed with brine-water (1:1), brine, dried over MgSO₄, filtered and concentrated *in vacuo*, yielding a semi-solid. Hexane
20 (600 mL) was added with mixing and the mixture was filtered. The filtrate was washed with brine, dried over MgSO₄, filtered and concentrated *in vacuo* yielding a solid. This solid was dissolved in hot ethanol (600 mL). Water (190 mL) was added which induced
25 crystallization. Filtration of the mixture and drying of the product provided the desired ester as a crystalline solid (131.3 g, 71%): mp 91.0-94.9 °C. This material was of suitable purity to be used in subsequent steps without further purification.

30

Step 3. Preparation of ethyl 6-chloro-7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A one liter three-neck flask equipped with
35 mechanical stirrer and gas inlet tube was charged with the ester (Step 2) (100 g, 0.3 mole) and acetic acid (300 mL). While cooling (water bath) the reaction

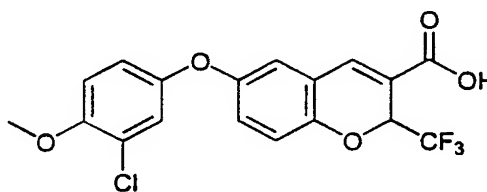
mixture, chlorine gas (37.6 g, 0.53 mole) was added which caused the temperature to rise to 48 °C. After stirring for two hours, the reaction was cooled in an ice-water bath to 15 °C. Zinc powder (19.5 g, 0.3 mole) was added in one portion which caused the temperature to rise to 72 °C. After cooling to room temperature additional zinc powder (5.0 g, 0.08 mole) was added and the mixture was stirred for 0.5 hour longer. The crude mixture was filtered through diatomaceous earth and was concentrated in vacuo yielding an oil. The oil was dissolved in ethyl acetate (700 mL) washed with brine-water (1:1, 1 L) and brine (0.5 L). The resulting aqueous phase was extracted with ethyl acetate (700 mL). This ethyl acetate phase was washed with brine-water (1:1, 1 L) and brine (0.5 L). The combined organic phases were dried over MgSO₄, filtered and concentrated in vacuo yielding the title compound as a yellow oil (116 g, 106 %). This material, which contained some entrained ethyl acetate, was of suitable purity to be used in subsequent steps without further purification.

Step 4. Preparation of 6-chloro-7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

To a solution of the ester (Step 3) (116 g, 0.3 mole) in methanol (500 mL) and tetrahydrofuran (500 mL) in a one liter flask was added aqueous sodium hydroxide (2.5 N, 240 mL, 0.6 mole). After stirring overnight, the pH of the solution was adjusted to 1 with concentrated hydrochloric acid and the solution was extracted with ethyl acetate. The ethyl acetate phase was dried over MgSO₄, filtered and concentrated in vacuo yielding a solid. This solid was dissolved in hot ethanol (500 mL). Water (500 mL) was added and upon cooling to room temperature crystals formed which were collected by vacuum

filtration. The crystals were washed with ethanol-water (3:7, 3 X 200 mL) and dried providing the title acid as a crystalline solid (91.6 g, 91 %): mp 194.9-196.5 °C. ¹H NMR (acetone-d₆/300 MHz) 7.86 (s, 1H), 7.52 (s, 1H), 7.12 (s, 1H), 5.83 (q, 1H, *J* = 7.1 Hz), 1.48 (s, 9H). Anal. Calc'd for C₁₅H₁₄ClF₃O₃: C, 53.83; H, 4.22; Cl, 10.59. Found: C, 53.92; H, 4.24; Cl, 10.50.

10

EXAMPLE 9

6-(3-Chloro-4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

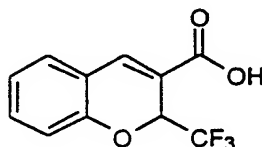
15

To a stirred solution of chlorine in acetic acid (3.5 mL of 0.24 M solution, 0.84 mmol) was added 6-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (0.31 g, 0.85 mmol) (Example 7). After 1 hour additional chlorine in acetic acid (1.5 mL of 0.24 M solution, 0.36 mmol) was added. After three additional hours additional chlorine in acetic acid (0.25 mL of 0.25 M solution, 0.06 mmol) was added. After 2.5 hours the reaction was quenched with aqueous 10% sodium bisulfite solution and the resulting mixture extracted with ethyl acetate. The organic phase was washed with water, brine, dried over MgSO₄, filtered, and concentrated *in vacuo* yielding a brown oil. The oil was dissolved in a minimum of hexanes which induced crystallization. Vacuum filtration of the mixture provided the title compound as yellow crystals (0.18 g, 53%): mp 205-207 °C. ¹H NMR (acetone-d₆/300 MHz) 7.89 (s, 1H), 6.97-7.18 (m, 6H), 5.83 (q, 1H, *J* = 7.0 Hz),

30

3.90 (s, 3H). FABLRMS m/z 400 (M+). FABHRMS m/z 399.0249 (M-H, Calc'd 399.0247). Anal. Calc'd for $C_{18}H_{12}ClF_3O_5$: C, 53.95; H, 3.02; Cl, 8.85. Found: C, 53.78; H, 3.08; Cl, 8.98.

5

EXAMPLE 10**2-Trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

10

Step 1. Preparation of ethyl 2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

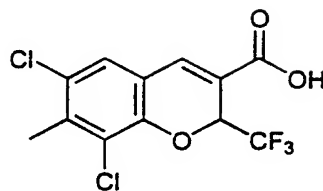
The ester was prepared from salicylaldehyde by a procedure similar to the method described in Example 1, Step 1: bp 107 °C 2mm. ^1H NMR (acetone- d_6 /300 MHz) 7.89 (s, 1H), 7.52-7.38 (m, 2H), 7.09 (dt, 1 J = 1.0, 7.7 Hz), 7.03 (d, 1H, J = 8.3 Hz), 5.84 (q, 1H, J = 7.3 Hz), 4.39-4.23 (m, 2H), 1.33 (t, 3H, J = 7.0 Hz). FABLRMS m/z 273 (M+H). ESHRMS (m/z 273.0720 (M+H Calcd 273.0739)

20

Step 2. Preparation of 2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The acid was prepared from the ethyl ester (Step 1) by a procedure similar to the method described in Example 1, Step 2: mp 152.2-153.3 °C. ^1H NMR (acetone- d_6 /300 MHz) 7.89 (s, 1H), 7.39-7.49 (m, 2H), 7.11-7.01 (m, 2H), 5.81 (q $H-F$, 1H, J = 7.2 Hz). FABHRMS m/z 245.0422 (M+H, Calc'd 245.0426). Anal. Calc'd for $C_{11}H_7F_3O_3$: C, 54.11; H, 2.89. Found: C, 54.22; H, 2.97.

30

EXAMPLE 11

5 **6,8-Dichloro-7-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

Step 1. Preparation of 3,5-dichloro-4-methylsalicylaldehyde.

10 2,4-Dichloro-3-methylphenol (25.0 g, 141.2 mmol) was added to methanesulfonic acid (100 mL). With stirring, hexamethylenetetramine (HMTA) (39.8g, 282.4 mmol) and additional methanesulfonic acid (100 mL) was added portion-wise during which time the reaction began

15 to froth and exotherm. The resulting mixture was heated to 100 °C for 3 hours. The crude ocher colored suspension was cooled to 50 °C and poured over a mechanically stirred mixture of ice-water (2 L). A yellow precipitate was formed which was collected by

20 vacuum filtration. This solid was purified by flash chromatography (silica, hexanes-methylene chloride, 9:10) yielding the salicylaldehyde as a pale yellow powder (6.17 g, 21%; mp 94.0-95.1 °C) of suitable purity to use without further purification.

25

Step 2. Preparation of ethyl 6,8-dichloro-7-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

 A mixture of 3,5-dichloro-4-methylsalicylaldehyde (Step 1) (5.94 g, 29.0 mmol) and ethyl 4,4,4-trifluorocrotonate (7.67 g, 45.6 mmol) dissolved in anhydrous DMSO (10 mL) was treated with triethylamine (5.88 g, 58.1 mmol). The reaction was stirred at 85 °C

30 for 49 hours then cooled in ice and filtered to give an

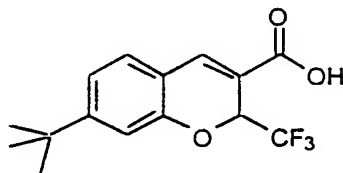
orange solid. The solid was dissolved in ethyl acetate (100 mL), washed with 3 N HCl (2 x 50 mL), saturated NaHCO₃, washed with brine, dried over MgSO₄, and concentrated *in vacuo* to give a yellow solid (8.63 g,

5 84%): mp 117.1-119.5 °C. ¹H NMR (CDCl₃/300 MHz) 7.63 (s, 1H), 7.17 (s, 1H), 5.80 (q, 1H, *J* = 6.6 Hz), 4.33 (m, 2H), 2.48 (s, 3H), 1.35 (t, 3H, *J* = 7.1 Hz).

10 Step 3. Preparation of 6,8-dichloro-7-methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 2 (8.39 g 23.6 mmol) was dissolved in THF (30 mL) and ethanol (20 mL), treated with 2.5 N sodium hydroxide (20 mL, 50 mmol), and stirred at room temperature for 3.5 hours. The reaction
15 mixture was concentrated *in vacuo*, acidified with 3 N HCl, filtered, and recrystallized from ethanol/ water to yield a yellow solid (6.0 g, 78%): mp 229.9-230.9 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.90 (s, 1H), 7.58 (s, 1H), 6.00 (q, 1H, *J* = 6.8 Hz), 2.50 (s, 3H). FABLRMS *m/z* 325
20 (M-H). FABHRMS *m/z* 324.9636 (M-H, Calc'd 324.9646). Anal. Calc'd for C₁₂H₇Cl₂F₃O₃: C, 44.07; H, 2.16; Cl, 21.68. Found: C, 44.06; H, 2.21; Cl, 21.74.

EXAMPLE 12



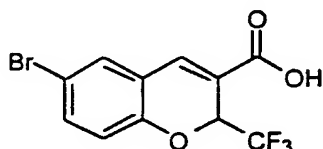
25

7-(1,1-Dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

Ethyl 7-(1,1-dimethylethyl)-2-(trifluoromethyl)-2H-
30 1-benzopyran-3-carboxylate (Example 8, Step 2) was hydrolyzed to the carboxylic acid via a procedure similar to that described in Example 1, Step 2: mp 165.6-166.8 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.86 (s,

1H), 7.38 (d, 1H, $J = 8.1$ Hz), 7.15 (dd, 1H, $J = 1.8$ Hz, and $J = 7.8$ Hz), 7.05 (bs, 1H), 5.79 (q $H-F$, 1H, $J = 7.2$ Hz), 1.32 (s, 9H). FABHRMS m/z 301.1033 ($M+H$, Calc'd 301.1051). Anal. Calc'd for $C_{15}H_{15}F_3O_3$: C, 60.00; H, 5.04. Found: C, 59.80; H, 5.10.

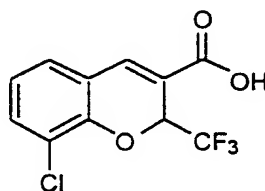
EXAMPLE 13



10 **6-Bromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

5-Bromosalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 189.6-190.9 °C. 1H NMR (acetone- d_6 /300 MHz) 7.89 (s, 1H), 7.70 (d, 1H, $J = 2.1$ Hz), 7.55 (dd, 1H, $J = 2.4$ Hz, and $J = 8.7$ Hz), 7.02 (d, 1H, $J = 8.7$ Hz), 5.86 (q $H-F$, 1H, $J = 7.2$ Hz). FABHRMS m/z 322.9519 ($M+H$, Calc'd 322.9531). Anal. Calc'd for $C_{11}H_6BrF_3O_3$: C, 40.90; H, 1.87; Br, 24.73. Found: C, 40.87; H, 1.92; Br, 24.80.

EXAMPLE 14

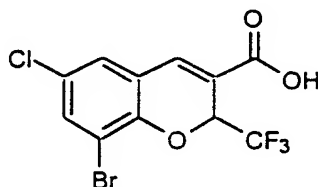


25 **8-Chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

2-Chlorophenol was converted to the title compound by a procedure similar to that

described in Example 2: mp 224.5-225.6 °C. ¹H
NMR (acetone-d₆/300 MHz) 7.91 (s, 1H), 7.49
(m, 2H), 7.11 (t, 1H, J = 7.8 Hz), 5.96 (q H-
F, 1H, J = 7.2 Hz). FABHRMS m/z 279.0027
5 (M+H, Calc'd 279.0036). Anal. Calc'd for
C₁₁H₆ClF₃O₃: C, 47.42; H, 2.17. Found: C,
47.33; H, 2.17.

EXAMPLE 15

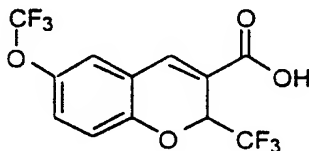


10

**8-Bromo-6-chloro-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

2-Bromo-4-chlorosalicylaldehyde was
15 converted to the title compound by a similar
procedure to that described in Example 1: mp
227.8-228.9 °C. ¹H NMR (acetone-d₆/300 MHz)
7.90 (s, 1H), 7.65 (dd, 2H, J = 2.4 and J =
28.8 Hz), 6.00 (q H-F, 1H, J = 7.2 Hz).
20 FABHRMS m/z 356.9134 (M+H, Calc'd 356.9141).
Anal. Calc'd for C₁₁H₅BrClF₃O₃: C, 36.96; H,
1.41. Found: C, 37.05; H, 1.33.

EXAMPLE 16

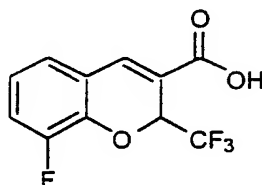


25

**6-Trifluoromethoxy-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

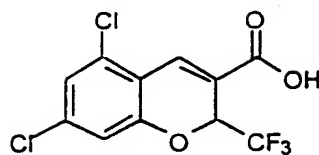
5-(Trifluoromethoxy)salicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 118.4-119.5 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.95 (s, 1H), 7.54 (d, 1H, *J* = 2.1 Hz), 7.39 (dd, 1H, *J* = 2.4 Hz, and *J* = 9.0 Hz), 7.02 (d, 1H, *J* = 9.0 Hz), 5.88 (q _{H-F}, 1H, *J* = 7.2 Hz). FABHRMS *m/z* 329.0228 (M+H, Calc'd 329.0249). Anal. Calc'd for C₁₂H₆F₆O₄: C, 43.92; H, 1.84. Found: C, 43.84; H, 1.87.

EXAMPLE 17



15 8-Fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

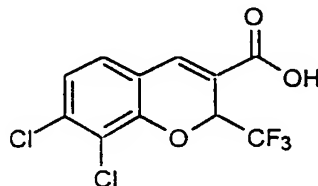
3-Fluorosalicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 197.7-210.1 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.94 (s, 1H), 7.30 (m, 2H), 7.11 (m 1H), 5.93 (q _{H-F}, 1H, *J* = 7.2 Hz). FABHRMS *m/z* 263.0341 (M+H, C₁₁H₆F₄O₃ Calc'd 263.0331). Anal. Calc'd for C₁₁H₆F₄O₃: C, 50.40; H, 2.31. Found: C, 50.48; H, 2.25.

EXAMPLE 18

5 **5,7-Dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

4,6-Dichlorosalicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 190.1-191.2 °C. ¹H NMR (acetone-d₆/300 MHz) 8.01 (s, 1H), 7.3 (bs, 1H), 7.16 (bs, 1H), 5.94 (q H-F, 1H, J = 7.2 Hz). FABHRMS m/z 312.9636 (M+H, Calc'd 312.9646). Anal. Calc'd for C₁₁H₅Cl₂F₃O₃: C, 42.20; H, 1.61. Found: C, 42.27; H, 1.56.

10
15

EXAMPLE 19

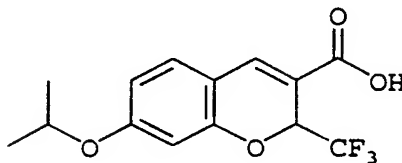
20 **7,8-Dichloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

3,4-Dichlorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 219.5-220.9 °C. ¹H NMR (acetone-d₆/300 MHz) 7.94 (s, 1H), 7.51 (d, 1H, J = 8.4 Hz), 7.34 (d, 1H, J = 8.4 Hz), 6.02 (q H-F, 1H, J = 7.2 Hz). FABHRMS m/z 318.9709 (M+Li, C₁₁H₅Cl₂F₃O₃ Calc'd

25

318.9728). Anal. Calc'd for $C_{11}H_5Cl_2F_3O_3$: C, 42.20; H, 1.61. Found: C, 42.15; H, 1.68.

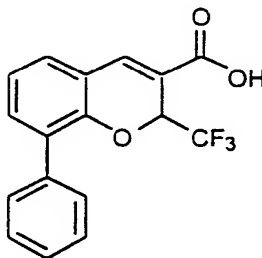
EXAMPLE 20



7-Isopropoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

2,4-Dihydroxybenzaldehyde was alkylated to prepare 4-(1-methylethoxy)salicylaldehyde. This salicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 161-163 °C. 1H NMR ($CD_3OD/300$ MHz) 7.73 (s, 1H), 7.21 (d, 1H, $J = 8.5$ Hz), 6.57 (dd, 1H, $J = 8.5, 2.2$ Hz). FABHRMS m/z 301.0688 ($M-H^+$, $C_{11}H_{12}F_3O_4$ requires 301.0687). Anal. Calc'd for $C_{11}H_{13}F_3O_4$: C, 55.63; H, 4.34. Found: C, 55.72; H, 4.34.

EXAMPLE 21

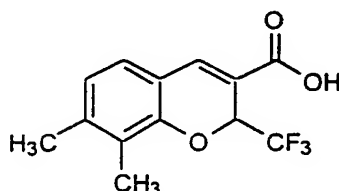


8-Phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

2-Phenylphenol was converted to the title compound by a procedure similar to that described

in Example 2: mp 171.6-175.0 °C. ^1H NMR
(acetone- d_6 /300 MHz) 7.95 (s, 1H), 7.46 (m, 7H),
7.18 (t, 1H, $J = 7.5$ Hz), 5.81 (q H-F , 1H, $J =$
7.2 Hz). FABHRMS m/z 327.0816 ($\text{M}+\text{Li}$, Calc'd
5 327.0820). Anal. Calc'd for $\text{C}_{17}\text{H}_{11}\text{F}_3\text{O}_3$: C,
63.76; H, 3.46. Found: C, 63.52; H, 3.55.

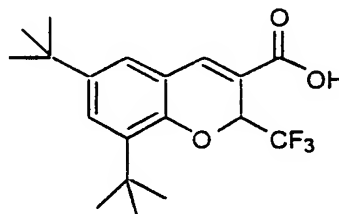
EXAMPLE 22



10 **7,8-Dimethyl-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

2,3-Dimethylphenol was converted to the
title compound by a procedure similar to that
15 described in Example 2: mp 245.2-247.3 °C. ^1H
NMR (acetone- d_6 /300 MHz) 7.83 (s, 1H), 7.17
(d, 1H, $J = 7.8$ Hz), 6.89 (d, 1H, $J = 7.8$ Hz),
5.82 (q H-F , 1H, $J = 7.2$ Hz), 2.30 (s, 3H),
2.17 (s, 3H). Anal. Calc'd for $\text{C}_{13}\text{H}_{11}\text{F}_3\text{O}_3 +$
20 1.56 % H_2O : C, 56.46; H, 4.18. Found: C,
56.46; H, 4.15.

EXAMPLE 23

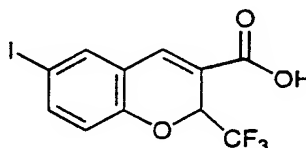


25 **6,8-bis(1,1-Dimethylethyl)-2-trifluoromethyl-
2H-1-benzopyran-3-carboxylic acid**

3,5-Di-*tert*-butylsalicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 171.6-175.0 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.65 (s, 1H), 7.34 (d, 1H, *J* = 2.4 Hz), 7.15 (d, 1H, *J* = 2.4 Hz), 6.02 (q H-F, 1H, *J* = 7.2 Hz). FABHRMS *m/z* 363.1743 (M+Li, Calc'd 363.1759). Anal. Calc'd for C₁₉H₂₃BrF₃O₃: C, 64.03; H, 6.50. Found: C, 64.13; H, 6.49.

10

EXAMPLE 24



6-Iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

15

Step 1: Preparation of 2-hydroxy-5-iodobenzyl alcohol.

A solution of 5-iodosalicylic acid (25.0 g, 94.6 mmol) in tetrahydrofuran (500 mL) was cooled to 0°C. With vigorous mixing, borane-methyl sulfide complex (15.1 ml of 10 M solution, 151.0 mmol) was added dropwise over 0.25 hours. The solution was warmed to room temperature and then heated at reflux for 4 h. A white precipitate formed during the reflux. The solution was cooled to room temperature and 10% aqueous hydrochloric acid (100 mL) was added over 15 min and the solution stirred at room temperature for 2 h. The precipitate dissolved and the solvent was concentrated *in vacuo* to a volume of approximately 200 mL. The solution was poured into ethyl acetate (300 mL) and washed with water (2 x 200 mL), saturated sodium bicarbonate (2 x 200 mL), and saturated ammonium chloride (2 x 200 mL). The organic layer was dried over sodium sulfate and concentrated *in vacuo*. The 2-hydroxy-5-iodobenzyl alcohol was isolated

20

25

30

as a white solid (21.3 g, 85.2 mmol) from hexanes. (90% yield): mp 105-110°C. ¹H NMR (CDCl₃/300 MHz) 8.21 (s, 1H), 7.30-7.33 (M, 2H), 6.57 (d, 1H, *J* = 8.3 Hz), 4.97 (bs, 1H), 4.62 (s, 2H). EIHRMS *m/z* = 249.9492 (M⁺, Calc'd 249.9491).

Step 2: Preparation of 2-hydroxy-5-iodobenzaldehyde

To a stirred solution of 2-hydroxy-5-iodobenzyl alcohol (43.5 g, 174.0 mmol) in acetone (700 mL) was added 85% activated manganese(IV) oxide (5 micron, 50 g, 494.0 mmol) and the solution stirred at room temperature for 16 hours. The manganese oxide was removed by filtration through diatomaceous earth and the filtrate concentrated *in vacuo*. The product was purified by flash silica chromatography (0-20% ethyl acetate in hexanes) The 2-hydroxy-5-iodobenzaldehyde was obtained as a greenish-yellow solid (24.3 g, 58%). A small amount of the 2-hydroxy-5-iodobenzaldehyde was recrystallized from methanol/water to afford an analytical sample and the remainder of the compound was used without further purification: mp 99-101 °C. ¹H NMR (CDCl₃/300 MHz) 9.83 (s, 1H), 7.79 (d, 1H, *J* = 2.2 Hz), 7.77 (dd, 1H, *J* = 8.7 Hz, *J* = 2.2 Hz), 6.81 (d, 1H, *J* = 8.7 Hz). ESHRMS 246.9229 (M-H Calc'd 246.9256).

Step 3: Preparation of ethyl 6-iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

A mixture of 5-iodosalicylaldehyde (16.2 g, 65.3 mmol), ethyl 4,4,4-trifluorocrotonate (22.4 g, 133 mmol) and triethylamine (50 mL, 395 mmol) were combined, stirred at 70°C for 8 h and then heated at reflux for 48 h. The solution was poured into ethyl acetate (300 mL) and washed with 1N hydrochloric acid (3 x 200 mL). The aqueous layers were combined and extracted with ethyl acetate (1 x 100 mL). The combined ethyl acetate extracts were washed with saturated ammonium chloride (2 x 200 mL), dried over magnesium sulfate and concentrated

in vacuo yielding a dark red oil. This oil was purified by flash chromatography using ethyl acetate-hexanes (3:7) yielding a red oil. Crystallization of this oil from hexanes yielded the title compound as light red

5 crystals (8.3 g, 31%): mp 105-106 °C. ¹H NMR (CDCl₃/300 MHz) 7.63 (s, 1H), 7.58 (dd, 2H, J = 8.6, J = 2.1 Hz, 7.54 (d, 1H, J = 2.1 Hz), 6.77 (d, 1H, J = 8.6 Hz), 5.70 (q, 1H, J = 6.7 Hz), 4.20-4.38 (m 2H), 1.35 (t, 3H, J = 7.2 Hz). ESHRMS 415.9926 (M+NH₄⁺ Calc'd 396.9746)

10

Step 4: Preparation of 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

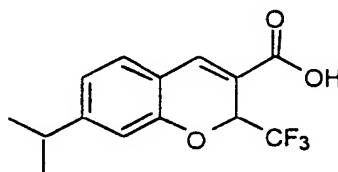
Hydrolysis of the ester (Step 3), using a procedure similar to Example 1, Step 2, yielded the carboxylic acid:

15

mp 168-170 °C. ¹H NMR (CD₃OD/300 MHz) 7.57 (s, 1H), 7.70 (d, 1H, J = 2.2 Hz), 7.64 (dd, 1H, J = 8.5, 2.2 Hz), 6.79 (d, 1H, J = 8.5 Hz) 5.78 (q, 1H, J = 7.0 Hz). ESHRMS m/z 368.9222 (Calc'd for M-H 368.9235). Anal. Calc'd for C₁₁H₆F₃IO₃: C, 35.70; H, 1.63. Found C, 35.67; H, 1.63.

20

EXAMPLE 25



25

7-(1-Methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

3-(1-Methylethyl)phenol was converted to the title compound by a procedure similar to that described in Example 2: mp 158.3-159.7 °C. ¹H NMR (acetone-d₆/300 MHz) 7.86 (s, 1H), 7.37 (d, 1H, J = 7.8 Hz), 7.00 (d, 1H, J = 7.8

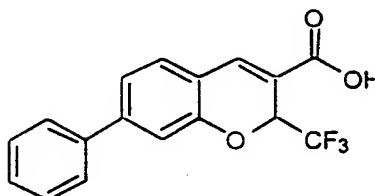
30

Hz), 6.91 (s, 1H), 5.78 (q, 1H, $J = 6.9$ Hz),
2.93 (m, 1H), 1.24 (d, 6H, $J = 6.9$ Hz).

FABLRMS m/z 287 (M+H). Anal. Calc'd for
 $C_{14}H_{13}F_3O_3$: C, 58.74; H, 4.58. Found: C,

5 57.37; H, 4.49.

EXAMPLE 26



10 **7-Phenyl-2-trifluoromethyl-2H-1-benzopyran-3-
carboxylic acid**

3-Phenylphenol was converted to the title
compound by a procedure similar to that

described in Example 2: mp 209.4-211.7 °C. 1H

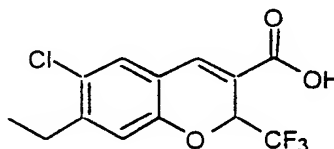
15 NMR (acetone- d_6 /300 MHz) 7.94 (s, 1H), 7.74

(m, 2H), 7.47 (m, 5H), 7.33 (s, 1H), 5.86 (q,
1H, $J = 7.2$ Hz). FABLRMS m/z 321 (M+H).

Anal. Calc'd for $C_{17}H_{11}F_3O_3$: C, 63.76; H,
3.46. Found: C, 64.17; H, 3.61.

20

EXAMPLE 27



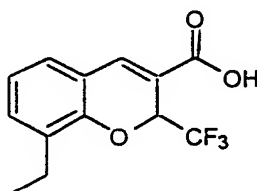
25 **6-Chloro-7-ethyl-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

4-Chloro-3-ethylphenol was converted to
the title compound by a procedure similar to
that described in Example 2: mp 170.7-172.1

°C. ^1H NMR ($\text{CDCl}_3/300\text{ MHz}$) 7.78 (s, 1H), 7.26 (s, 1H), 6.90 (s, 1H), 5.67 (q, 1H, $J = 6.9\text{ Hz}$), 2.73 (q, 2H, $J = 7.8\text{ Hz}$), 1.24 (t, 3H, $J = 7.8\text{ Hz}$). FABLRMS m/z 307 (M+H). Anal.

5 Calc'd for $\text{C}_{13}\text{H}_{10}\text{F}_3\text{O}_3$: C, 50.92; H, 3.29.
Found: C, 51.00; H, 3.33.

EXAMPLE 28



10

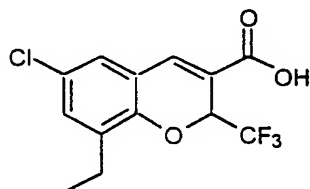
8-Ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

2-Ethylphenol was converted to the title

15 compound by a procedure similar to that described in Example 2: mp 185.4-186.8 °C. ^1H NMR (acetone- $d_6/300\text{ MHz}$) 7.85 (s, 1H), 7.28 (d, 2H, $J = 7.5\text{ Hz}$), 7.00 (t, 1H, $J = 7.5\text{ Hz}$), 5.84 (q, 1H, $J = 7.2\text{ Hz}$), 2.65 (m, 2H), 1.18

20 (t, 3H, $J = 7.5\text{ Hz}$). FABLRMS m/z 273 (M+H). Anal. Calc'd for $\text{C}_{13}\text{H}_{11}\text{F}_3\text{O}_3$: C, 57.36; H, 4.07. Found: C, 57.15; H, 4.11.

EXAMPLE 29

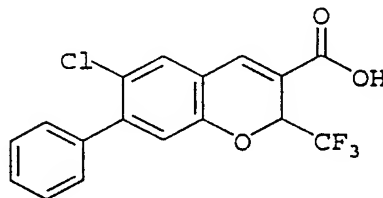


25

6-Chloro-8-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

8-Ethyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 28) (0.68 g, 2.5 mmol) was dissolved in trimethylphosphate (5 mL) and was treated with sulfuryl chloride (0.35 g, 2.62 mmol) at 0 °C. After stirring at 0 °C for 45 minutes and 1 hour at room temperature, the reaction was diluted with cold water (15 mL). The resulting oily mixture was extracted with hexanes-ethyl acetate. The organic phase was washed with brine, dried, and concentrated in vacuo yielding the title compound as a solid (0.9 g, 117 %): mp 197.2-199.1 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.86 (s, 1H), 7.38 (d, 1H, *J* = 2.7 Hz), 7.30 (d, 1H, *J* = 2.4 Hz), 5.88 (q, 1H, *J* = 7.2 Hz), 2.65 (m, 2H), 1.19 (t, 3H, *J* = 7.5 Hz). FABLRMS *m/z* 307 (M+H). Anal. Calc'd for C₁₃H₁₀ClF₃O₃: C, 50.92; H, 3.29. Found: C, 51.00; H, 3.23.

EXAMPLE 30

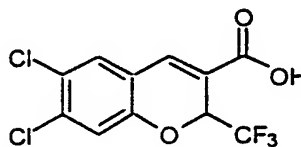


6-Chloro-7-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

7-Phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 26) was converted to the title compound by a procedure similar to that described in Example 29: mp

185.3-187.8 °C. ^1H NMR (acetone- d_6 /300 MHz)
7.94 (s, 1H), 7.68 (s, 1H), 7.47 (m, 5H), 7.06
(s, 1H), 5.87 (q, 1H, $J = 6.9$ Hz). FABLRMS m/z
355 (M+H). Anal. Calc'd for $\text{C}_{17}\text{H}_{10}\text{ClF}_3\text{O}_3$: C,
5 57.56; H, 2.84. Found: C, 58.27; H, 3.11.

EXAMPLE 31

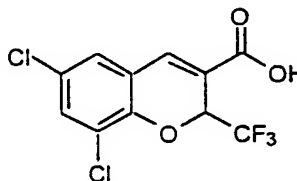


10

6,7-Dichloro-2-trifluoromethyl-2H-1-benzopyran- 3-carboxylic acid

3,4-Dichlorophenol was converted to the
15 title compound by a procedure similar to that
described in Example 2: mp 196.1-198.3 °C. ^1H
NMR (acetone- d_6 /300 MHz) 7.90 (s, 1H), 7.74 (s,
1H), 7.30 (s, 1H), 5.88 (q, 1H, $J = 6.9$ Hz).
FABLRMS m/z 314 (M+H). Anal. Calc'd for
20 $\text{C}_{11}\text{H}_5\text{Cl}_2\text{F}_3\text{O}_3$: C, 42.20; H, 1.61. Found: C,
42.31; H, 1.65.

EXAMPLE 32

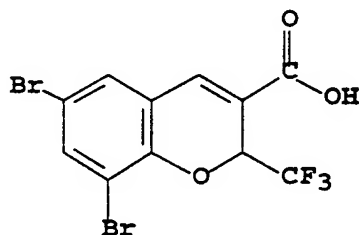


25

6,8-Dichloro-2-trifluoromethyl-2H-1-benzopyran- 3-carboxylic acid

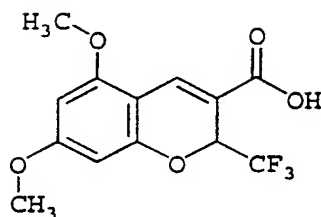
3,5-Dichlorosalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 11, Steps 2 & 3: mp 212.8-216.8 °C. ¹H NMR (CDCl₃/300 MHz) 7.77 (s, 1H), 7.41 (d, 1H, *J* = 2.4 Hz), 7.18 (d, 1H, *J* = 2.2 Hz), 5.82 (q, 1H, *J* = 6.7 Hz). FABLRMS *m/z* 311 (M-H). FABHRMS *m/z* 312.9644 (M+H, Calc'd 312.9646). Anal. Calc'd for C₁₁H₅F₃Cl₂O₃: C, 42.20; H, 1.61. Found: C, 42.50; H, 1.71.

EXAMPLE 33



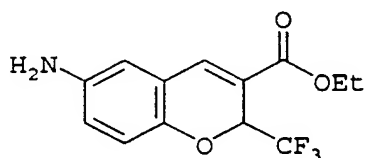
6,8-Dibromo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

3,5-Dibromosalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 225-226 °C. ¹H NMR (CD₃OD/300 MHz) 7.76 (s, 1H), 7.74 (d, 1H, *J* = 2.2 Hz), 7.55 (d, 1H, *J* = 2.2 Hz), 5.91 (q H-F, 1H, *J* = 7.2 Hz). FABHRMS *m/z* 400.8648 (M+H⁺, Calc'd 400.8636). Anal. Calcd for C₁₁H₅Br₂F₃O₃: C, 32.87; H, 1.25. Found: C, 33.47; H, 1.38.

EXAMPLE 34

5 **6,8-Dimethoxy-2-trifluoromethyl-2H-1-benzopyran-
3-carboxylic acid**

4,6-Dimethoxysalicylaldehyde was converted to the title compound by a procedure similar to
 10 that described in Example 1: mp 215-217 °C. ¹H NMR (CD₃OD/300 MHz) 7.95 (s, 1H), 6.18-6.20 (m, 2H), 5.65 (q H-F, 1H, J = 7.2 Hz), 3.87 (s, 1H), 3.81 (s, 1H). FABHRMS m/z 303.0497 (M-H⁺, Calc'd 303.0380). Anal. Calc'd for C₁₃H₁₁F₃O₅: C, 51.33; H, 3.64. Found: C, 51.19; H, 3.71.
 15

EXAMPLE 35

20 **Ethyl 6-amino-2-trifluoromethyl-2H-1-benzopyran-
3-carboxylate**

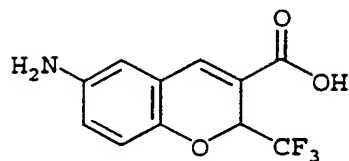
Step 1. Preparation of ethyl 6-nitro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

25 A mixture of 5-nitrosalicylaldehyde (4.80g, 28.7 mmol) and ethyl 4,4,4-trifluorocrotonate (6.6 g, 39.4 mol) in anhydrous DMF was warmed to 60 °C and treated with anhydrous K₂CO₃ (3.90 g, 28.9 mol). The solution was maintained at 60 °C

for 20 hours, cooled to room temperature, diluted with water, and extracted with ethyl acetate. The organic extracts were washed with brine, dried over anhydrous MgSO_4 , filtered and concentrated *in vacuo* to afford an oil. The oil was dissolved in diethyl ether (5 mL). Hexanes was added until the solution became cloudy. Upon standing at room temperature overnight the ester was obtained as yellow crystals (0.856 g, 7% yield). This material was of sufficient purity to be used in subsequent steps without further purification. ^1H NMR (CDCl_3 /300 MHz) 8.15-8.19 (m, 2H), 7.74 (s, 1H), 7.09 (d, 1H, $J = 8.9$ Hz), 5.81 (q, 1H, $J = 5.8$ Hz), 4.29-4.39 (m, 2H), 1.35 (t, 3H, $J = 6.0$ Hz),

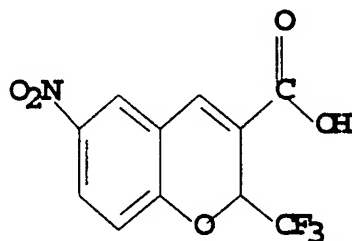
Step 2. Preparation of ethyl 6-amino-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

The ester (Step 1) (0.345 g, 1.08 mmol) was stirred in ethanol (10.0 mL) with 10% palladium on charcoal (15 mg) with hydrogen at 1 atmosphere for 1 hour. The catalyst was removed by filtration and the solvent removed *in vacuo* to afford the title compound as an orange-yellow solid (0.298 g, 95 %): mp 111-115 °C. (CD_3OD /300 MHz) 7.69 (s, 1H), 6.69-6.74 (m, 3H), 5.65 (q H-F , 1H, $J = 7.2$ Hz), 4.26-4.37 (m, 2H), 1.34 (t, 3H, $J = 7$ Hz). FABHRMS m/z 288.0860 ($\text{M}+\text{H}^+$, $\text{C}_{13}\text{H}_{13}\text{F}_3\text{NO}_3$ requires 288.0847). Anal. Calc'd for $\text{C}_{13}\text{H}_{12}\text{F}_3\text{NO}_3$: C, 54.36; H, 4.21; N, 4.88. Found: C, 54.46; H, 4.27; N, 4.83.

EXAMPLE 36

5 **6-Amino-2-trifluoromethyl-2H-1-benzopyran-3-**
 carboxylic acid

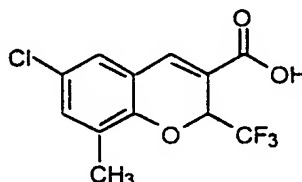
Ethyl 6-amino-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 35, Step 2)
10 was hydrolyzed to the carboxylic acid (title compound) by a procedure similar to that
described in Example 1, Step 2: mp 126-133 °C.
¹H NMR (CD₃OD/300 MHz) 6.81-6.90 (m, 3H), 5.66
(q H-F, 1H, J=7.2 Hz). FABHRMS m/z 260.0535
15 (M+H⁺, C₁₁H₉F₃NO₅ requires 260.0534).

EXAMPLE 37

20 **6-Nitro-2-trifluoromethyl-2H-1-benzopyran-3-**
 carboxylic acid

Ethyl 6-nitro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 35, Step 1)
25 was hydrolyzed to the carboxylic acid (title compound) by a procedure similar to that
described in Example 1, Step 2: mp 187-189 °C.

¹H NMR (CD₃OD/300 MHz) 8.34 (d, 1H, *J* = 2.6 Hz), 8.27 (dd, 1H, *J* = 8.7, 2.6 Hz), 7.90 (s, 1H), 7.09 (s, 1H, *J* = 8.7 Hz), 5.81 (q H-F, 1H, *J* = 7.2 Hz). EIHRMS *m/z* 289.0177 (Calc'd 289.0198).
5 Anal. Calc'd for C₁₁H₆F₃NO₅: C, 45.69; H, 2.09; N 4.84 Found: C, 45.71; H, 2.08; N 4.75.

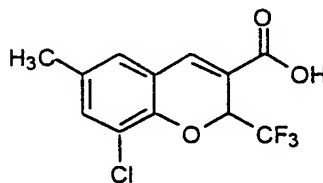
EXAMPLE 38

10

6-Chloro-8-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

4-Chloro-2-methylphenol was converted to
15 the title compound by a procedure similar to that described in Example 2: mp 231.9-233.2 °C.
¹H NMR (CDCl₃/300 MHz) 7.76 (s, 1H), 7.19 (d, 1H, *J* = 1.8 Hz), 7.09 (d, 1H, *J* = 2.4 Hz), 5.72 (q, 1H, *J* = 6.9 Hz), 2.24 (s, 3H). ¹⁹F NMR
(CDCl₃/282 MHz) -79.2 (d, *J* = 6.5 Hz). FABLRMS
20 *m/z* 299 (M+Li). FABHRMS *m/z* 293.0196 (M+H, Calc'd 293.0192). Anal. Calc'd for C₁₂H₈ClF₃O₃: C, 49.25; H, 2.76. Found: C, 49.37; H, 2.86.

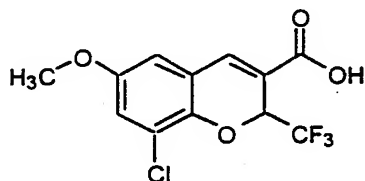
25

EXAMPLE 39**8-Chloro-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

2-Chloro-4-methylphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 226.4-227.4 °C.

5 ¹H NMR (CDCl₃/300 MHz) 7.79 (s, 1H), 7.23 (d, 1H, *J* = 1.4 Hz), 6.97 (d, 1H, *J* = 1.4 Hz), 5.77 (q, 1H, *J* = 6.8 Hz), 2.29 (s, 3H). ¹⁹F NMR (CDCl₃/282 MHz) -79.1 (d, *J* = 7.3 Hz). FABLRMS *m/z* 291 (M-H). EIHRMS *m/z* 292.0118 (M+,
10 C₁₂H₈ClF₃O₃ Calc'd 292.0114).

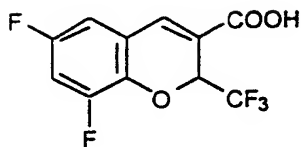
EXAMPLE 40



15

8-Chloro-6-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

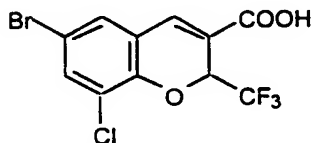
2-Chloro-4-methoxyphenol was converted to
20 the title compound by a procedure similar to that described in Example 2: mp 204.5-206.9 °C.
¹H NMR (CDCl₃/300 MHz) 7.78 (s, 1H), 6.98 (d, 1H, *J* = 2.8 Hz), 6.71 (d, 1H, *J* = 2.8 Hz), 5.74 (q, 1H, *J* = 6.9 Hz), 3.79 (s, 3H). FABLRMS *m/z*
25 326 (M+NH₄). EIHRMS *m/z* 308.0053 (M+ Calc'd 308.0063). Anal. Calc'd for C₁₂H₈ClF₃O₄: C, 46.70; H, 2.61. Found: C, 46.60; H, 2.68.

EXAMPLE 41

5 **6,8-Difluoro-2-trifluoromethyl-2H-1-benzopyran-
3-carboxylic acid**

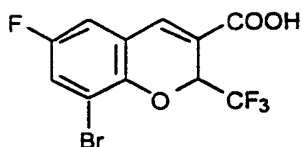
2,4-Difluorophenol was converted to the
title compound by a procedure similar to that
10 described in Example 2: mp 207-211 °C. ¹H NMR
(CDCl₃) 7.63 (s, 1H), 6.89-6.72 (m, 2H), 5.69
(q, 1H, J = 6.7 Hz). Anal. Calc'd for C₁₁H₅O₃F₅:
C, 47.16; H, 1.80. Found: C, 47.28; H, 1.87.

15

EXAMPLE 42

20 **6-Bromo-8-chloro-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

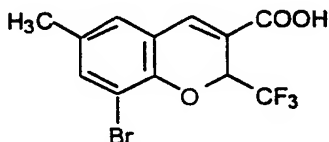
4-Bromo-2-chlorophenol was converted to the
title compound by a procedure similar to that
described in Example 2: mp 220.7-221.7 °C. ¹H
25 NMR (CDCl₃) 7.58 (s, 1H), 7.44 (d, 1H, J = 2.2
Hz), 7.22 (d, 1H, J = 2.2 Hz), 5.74 (q, 1H, J
= 6.8 Hz). Anal. Calc'd for C₁₁H₅O₃F₃BrCl: C,
36.96; H, 1.41. Found: C, 37.03; H, 1.44.

EXAMPLE 43

5 **8-Bromo-6-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

2-Bromo-4-fluorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp >300 °C. ¹H NMR (CDCl₃) 7.58 (s, 1H), 7.22 (dd, 1H, J = 6.3, 3 Hz), 6.88 (dd, 1H, J = 6.1, 3.1 Hz), 5.72 (q, 1H, J = 6.7 Hz). Anal. Calc'd for C₁₁H₅O₃F₄Br: C, 38.74; H, 1.48. Found: C, 38.82; H, 1.56.

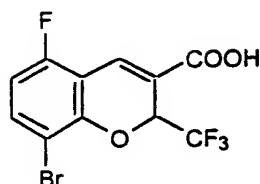
15

EXAMPLE 44

20 **8-Bromo-6-methyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

2-Bromo-4-methylphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 237-238 °C. ¹H NMR (CDCl₃) 7.59 (s, 1H), 7.27 (m, 1H), 6.91 (d, 1H, J = 1.4 Hz), 5.69 (q, 1H, J = 6.9 Hz), 2.20 (s, 3H). Anal. Calc'd for C₁₂H₈O₃F₃Br: C, 42.76; H, 2.39. Found: C, 43.34; H, 2.56.

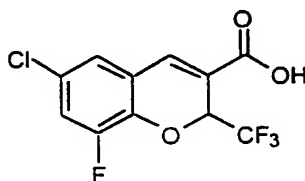
30

EXAMPLE 45

5 **8-Bromo-5-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

2-Bromo-5-fluorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 221.7-223.3 °C. ¹H NMR (CDCl₃) 7.81 (s, 1H), 7.38 (dd, 1H, J = 7.3, 5.8 Hz), 6.58 (t, 1H, J = 8.9 Hz), 5.71 (q, 1H, J = 6.7 Hz). Anal. Calc'd for C₁₁H₅O₃F₄Br: C, 38.74; H, 1.48. Found: C, 38.70; H, 1.54.

15

EXAMPLE 46

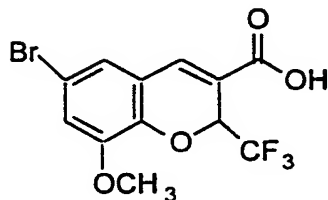
20 **6-Chloro-8-fluoro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

4-Chloro-2-fluorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 190.8-193.0 °C. ¹H NMR (CDCl₃/300 MHz) 7.77 (s, 1H), 7.19 (d of d, 1H, J = 2.2 and 9.7 Hz), 7.07 (t, 1H, J = 1.8 Hz), 5.76 (q, 1H, J = 6.7 Hz). FABLRMS m/z 295 (M-H). EIHRMS m/z 295.9876 (M+ Calc'd

295.9863). Anal. Calc'd for $C_{11}H_5ClF_4O_3$: C, 44.54; H, 1.70. Found: C, 44.36; H, 1.85.

EXAMPLE 47

5



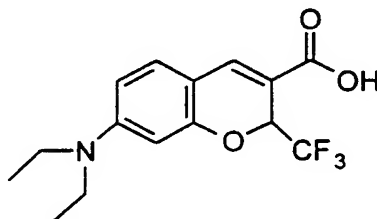
6-Bromo-8-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

10

4-Bromo-2-methoxysalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp dec. at 244 °C. 1H NMR ($CD_3OD/300$ MHz) 7.71 (s, 1H), 7.18 (d, 1H, $J = 2.2$ Hz), 7.11 (d, 1H, $J = 2.2$ Hz), 5.77 (q H-F, 1H, $J=7.2$ Hz), 3.84 (s, 3H). FABLRMS m/z 351 (m-H). Anal. Calc'd for $C_{12}H_8BrF_3O_5$: C, 40.82; H, 2.28. Found: C, 40.83; H, 2.30.

20

EXAMPLE 48

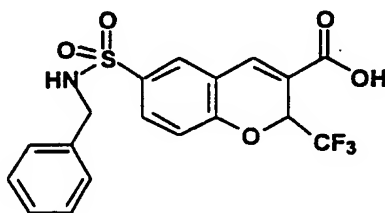


25

7-(N,N-Diethylamino)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

4-(*N,N*-Diethylamino)salicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 214.4-215.4 °C. ¹H NMR (CD₃OD/300 MHz) 7.67 (s, 1H), 7.06 (d, 1H, *J* = 8.6 Hz), 6.34 (dd, 1H, *J* = 8.6, 2.3 Hz), 5.60 (q H-F, 1H, *J*=7.2 Hz), 3.38 (q, 4H, *J* = 7.1 Hz), 1.16 (t, 6H, *J* = 7.1 Hz). ESLRMS *m/z* 316 (M + H). FABHRMS *m/z* 316.1145 (M + H⁺, Calc'd 316.1161). Anal. Calc'd for C₁₅H₁₆F₃NO₃: C, 57.14; H, 5.11; N, 4.44. Found: C, 57.14; H, 5.08; N, 4.44.

EXAMPLE 49



6-[[[(Phenylmethyl)amino]sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl 6-chlorosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

Chlorosulfonic acid (50.0 mL) was cooled to 15 °C and ethyl 2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (Example 10, Step 2) (6.21 g, 22.83 mmol) was added. After stirring at -15 °C for 1 hour, the solution was warmed to room temperature and stirred for 16 hours. The solution was added dropwise onto ice (500 mL) with vigorous stirring and extracted with diethyl ether (2 x 250 mL). The ether layers were combined, washed with water (2 x 250 mL),

saturated sodium bicarbonate (2 x 250 mL), and
brine (2 x 250 mL). Hexanes (50 mL) were added
and the solution was dried over sodium sulfate.
The solvent was removed *in vacuo* to afford the
5 ester as a yellow solid (7.41 g, 87%): mp 97.2-
98.4 °C. ¹H NMR (CDCl₃, 300 MHz) 7.97 (dd, 1H,
J = 8.6, 2.2 Hz), 7.92 (d, 1H, *J* = 2.2 Hz),
7.73 (s, 1H), 7.17 (d, 1H, *J* = 2.2 Hz), 5.82 (q
H-F, 1H, *J* = 7.2 Hz), 4.28-4.39 (m, 2H), 1.35 (t,
10 3H, *J* = 7.0 Hz). FABLRMS *m/z* 376 (M+Li⁺).

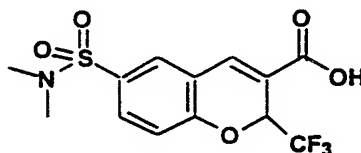
Step 2. Preparation of ethyl 6-
[[(phenylmethyl)amino]sulfonyl]-2-
trifluoromethyl-2H-1-benzopyran-3-carboxylate.

15 The sulfonyl chloride from Step 1 (451.0
mg, 1.22 mmol) and benzylamine (600 mg, 5.62
mmol) were mixed in diethyl ether (25 mL) for 1
hour at room temperature. The solution was
washed with 1N HCl (2 x 25 mL), saturated sodium
20 bicarbonate (2 x 25 mL), and brine (2 x 25 mL).
The solution was dried over sodium sulfate and
dried *in vacuo*. The aminosulfonyl was obtained
by crystallization from hexanes (431 mg, 84%):
mp 128.2-131.9 °C. ¹H NMR (CDCl₃, 300 MHz) 7.76
25 (dd, 1H, *J* = 8.4, 2.2 Hz), 7.70 (d, 1H, *J* = 2.2
Hz), 7.67 (s, 1H), 7.12-7.30 (m, 5H), 7.05 (d,
1H, *J* = 8.4 Hz), 5.78 (q H-F, 1H, *J* = 7.2 Hz),
4.68 (m, 2H), 4.19-4.32 (m, 2H), 1.37 (t, 3H, *J*
= 7.0 Hz). FABLRMS *m/z* 442 (M + H⁺). FABHRMS
30 *m/z* 442.0936 (M + H⁺, C₂₀H₁₉F₃NO₅S Calc'd
442.0916).

Step 3. Preparation of 6-
[[(phenylmethyl)amino]sulfonyl]-2-
35 trifluoromethyl-2H-1-benzopyran-3-carboxylic
acid.

The acid was converted from the ester (step 2) via the method similar to that described in Example 1, step 2: mp 223.3-224.4 °C ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$) 7.31-7.80 (m, 3H), 7.15-7.25 (m, 5H), 7.06 (d, 1H, $J = 8.3\text{ Hz}$), 5.87 (q H-F , 1H, $J=7.2\text{ Hz}$), 4.11 (s, 2H). FABLRMS m/z 420 ($\text{M} + \text{Li}^+$). FABHRMS m/z 414.0589 ($\text{M} + \text{H}^+$ Calc'd 414.0623). Anal. Calc'd for $\text{C}_{18}\text{H}_{14}\text{F}_3\text{NO}_5\text{S}$: C, 52.30; H, 3.41; N, 3.39. Found: C, 5.16; H, 3.44; N, 3.32.

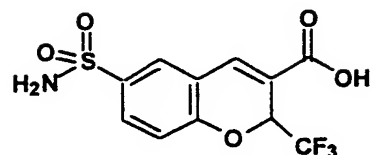
EXAMPLE 50



15

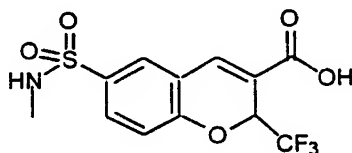
6-[(Dimethylamino)sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared by a similar procedure to that described in Example 49: mp 201.2-202.5 °C. ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$) 7.90 (s, 1H), 7.82 (d, 1H, $J = 2.2\text{ Hz}$), 7.76 (dd, 1H, $J = 8.6, 2.2\text{ Hz}$), 7.19 (d, 1H, $J = 8.6\text{ Hz}$), 5.91 (q H-F , 1H, $J=7.2\text{ Hz}$), 2.70 (s, 6H). FABLRMS m/z 352 ($\text{M} + \text{H}^+$). FABHRMS m/z 352.0466 ($\text{M} + \text{H}^+$ Calc'd 352.0467). Anal. Calc'd for $\text{C}_{13}\text{H}_{12}\text{F}_3\text{NO}_5\text{S}$: C, 44.45; H, 3.44; N, 3.99. Found: C, 4.42; H, 3.45; N, 3.96.

EXAMPLE 51

5 **6-Aminosulfonyl-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

The title compound was prepared by a similar procedure to that described in Example
 10 49: mp 187.9-189.8 °C. ¹H NMR (CD₃OD/300 MHz) 7.58-7.88 (m, 3H), 7.12 (d, J = 8.3 Hz), 5.87 (q H-F, 1H, J=7.2 Hz). FABLRMS m/z 324 (M + H⁺). FABHRMS m/z 324.0156 (M + H⁺ Calc'd 324.0154). Anal. Calc'd for C₁₁H₈F₃NO₅S * 0.74 H₂O: C,
 15 39.26; H, 2.84; N, 4.16. Found: C, 39.33; H, 2.82; N, 4.11.

EXAMPLE 52

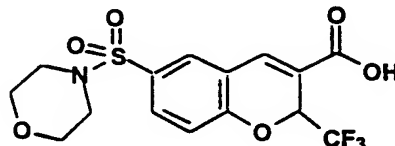
20

**6-(Methylamino)sulfonyl-2-trifluoromethyl-2H-1-
benzopyran-3-carboxylic acid**

25 The title compound was prepared by a similar procedure to that described in Example 49: mp 207.6-208.6 °C. ¹H NMR (CD₃OD/300 MHz) 7.83-7.97 (m, 3H), 7.19 (d, 1H, J = 8.5 Hz), 5.91 (q H-F, 1H, J=7.2 Hz), 3.11 (s, 3H).

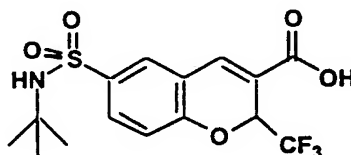
FABLRMS m/z 338 ($M + H^+$). FABHRMS m/z 338.0331
($M + H^+$ Calc'd 338.0310). Anal. Calc'd for
 $C_{12}H_{11}F_3NO_5S$: C, 42.73; H, 2.99; N, 4.15.
Found: C, 42.91; H, 3.06; N, 4.04.

5

EXAMPLE 53

10 **6-[(4-Morpholino)sulfonyl]-2-trifluoromethyl-2H-
1-benzopyran-3-carboxylic acid**

The title compound was prepared by a
similar procedure to that described in Example
15 49: mp 215.2-219.3 °C 1H NMR ($CD_3OD/300$ MHz)
7.88 (s, 1H), 7.81 (d, 1H, $J = 2.2$ Hz), 7.74
(dd, 1H, $J = 8.6, 2.2$ Hz), 5.90 (q $H-F$, 1H,
 $J=7.2$ Hz), 3.54-3.70 (m, 4H), 2.94-2.97 (m, 4H).
FABLRMS m/z 394 ($M + H^+$). FABHRMS 394.0567 ($M +$
20 H^+ , $C_{15}H_{15}F_3NO_6S$ Calc'd 394.0572).

EXAMPLE 54

25

**6-[(1,1-Dimethylethyl)aminosulfonyl]-2-
trifluoromethyl-2H-1-benzopyran-3-carboxylic
acid**

A.A.

The title compound was prepared by a similar procedure to that described in Example

49: mp 229.3-233.5 °C ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$)

7.82-7.87 (m, 3H), 7.12 (d, 1H, $J = 8.6\text{ Hz}$),

5 5.87 (q H-F , 1H, $J=7.2\text{ Hz}$), 1.18 (s, 9H).

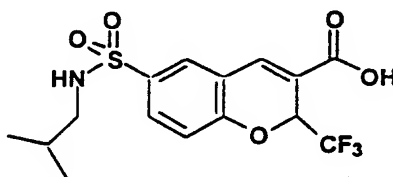
FABLRMS m/z 380 ($\text{M} + \text{H}^+$). Anal. Calc'd for

$\text{C}_{15}\text{H}_{16}\text{F}_3\text{NO}_5\text{S}$: C, 47.49; H, 4.25; N, 3.69.

Found: C, 47.95; H, 4.48; N, 3.55.

10

EXAMPLE 55



15

6-[(2-Methylpropyl)aminosulfonyl]-2-
trifluoromethyl-2H-1-benzopyran-3-carboxylic
acid

The title compound was prepared by a similar procedure to that described in Example

20 49: mp 190.6-192.4 °C. ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$)

7.77-7.84 (m, 3H), 7.13 (d, 1H, $J = 8.4\text{ Hz}$),

5.86 (q H-F , 1H, $J=7.2\text{ Hz}$), 2.64 (d, 2H, $J = 6.8$

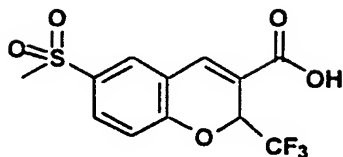
Hz), 1.66 (sept, 1H, $J = 6.6\text{ Hz}$), 0.84 (d, 6H, J

= 6.6 Hz). FABLRMS m/z 380 ($\text{M} + \text{H}^+$). Anal.

25 Calc'd for $\text{C}_{15}\text{H}_{16}\text{F}_3\text{NO}_5\text{S}$: C, 47.49; H, 4.25; N,

3.69. Found: C, 47.61; H, 3.34; N, 3.55.

EXAMPLE 56



5 **6-Methylsulfonyl-2-trifluoromethyl-2H-1-**
 benzopyran-3-carboxylic acid

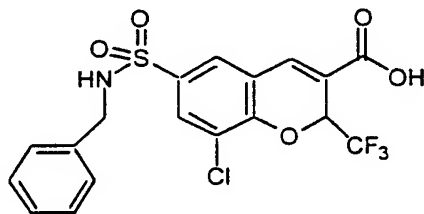
Step 1. Preparation of 6-chlorosulfonyl-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
10 acid.

To chlorosulfonic acid (50.0 mL) chilled to
-15 °C was added 2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid (Example 10) (4.0
g, 16.7 mmol). After stirring at -15 °C for 1
15 hour, the solution was warmed to room
temperature and stirred for 16 hours. The
resulting solution was added dropwise over ice
(100 mL) with two diethyl ether (2 x 75 mL)
extractions. The diethyl ether layers were
20 combined, washed with water (2 x 75 mL), and
brine (2 x 75 mL, dried over sodium sulfate and
concentrated *in vacuo*. The resulting solids
were triturated with hexane-ethyl acetate (9:1,
100 mL). The 6-chlorosulfonyl-2-
25 trifluoromethyl-2H-1-benzopyran-3-carboxylic
acid was isolated as a white solid: mp 169-174.
¹H NMR (CD₃OD/300 MHz) 8.18 (d, 1H, *J* = 2.7 Hz),
8.06 (dd, 1H, *J* = 8.7, 2.7 Hz), 7.93 (s, 1H),
7.28 (d, 1H, *J* = 8.7 Hz), 6.00 (q, 1H, *J* = 6.6
30 Hz). EIHRMS *m/z* 324.9977 (M⁺, Calcd 324.9994).

Step 2. Preparation of 6-methylsulfonyl-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid

A slurry of the chlorosulfonyl intermediate (Example 49, Step 1) (493 mg, 1.44 mmol), sodium bicarbonate (362 mg, 4.32 mmol), and sodium bisulfite (181 mg, 1.44 mmol) in water (1.5 mL) was heated to 60 °C for 1.5 h, followed by the addition of bromoacetic acid (212 mg, 1.55 mmol). The resulting suspension was heated to reflux, followed by the addition of sodium hydroxide solution (50% NaOH soln., 0.10 mL) and water (3.0 mL). The solution was refluxed for 8 hours, cooled to room temperature, and acidified to pH 1 with 1N aqueous hydrochloric acid. The solution was extracted with ethyl acetate (2 x 25 mL). The combined ethyl acetate layers were washed with 1N aqueous hydrochloric acid (2 x 25 mL), water (2 x 25 mL), and brine (2 x 25 mL), dried over sodium sulfate, filtered and concentrated *in vacuo* yielding the title compound as an off white solid. (231 mg, 50 % yield): mp 208.3–212.4°C. ¹H NMR (CD₃OD, 300 MHz) 7.97 (d, 1H, 2.2 Hz), 7.91 (1H, dd, *J* = 8.7, 2.2 Hz), 7.19 (d, 1H, *J* = 8.7 Hz), 5.91 (q H-F, 1H, *J* = 7.2 Hz), 3.11 (s, 1H) HRLRMS *m/z* 321 (M - H) FABLRMS *m/z* 321 (M - H). Anal. Calc'd for C₁₂H₉F₃O₅S*0.61 H₂O: C, 43.26; H, 3.09. Found: C, 43.24; H, 3.09.

EXAMPLE 57



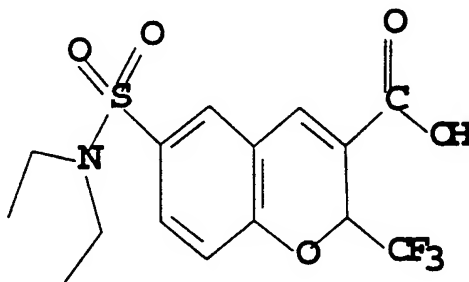
8-Chloro-6-[[(phenylmethyl) amino] sulfonyl]-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared by a similar procedure to that described in Example

49: mp 167.0-173.8 °C ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$)

- 5 7.78 (s, 1H), 7.72 (d, 1H, $J = 2.0\text{ Hz}$), 7.64 (d, 1H, $J = 2.0\text{ Hz}$). 7.44 (s, 1H), 7.15-7.23 (m, 5H), 6.01 (q H-F , 1H, $J=7.2\text{ Hz}$), 4.08-4.15 (m, 2H). FABLRMS m/z 454 ($\text{M} + \text{Li}^+$). Anal. Calc'd for $\text{C}_{16}\text{H}_{13}\text{ClF}_3\text{NO}_5\text{S}$: C, 48.28; H, 2.93; N, 3.13.
- 10 Found: C, xx; H, xx; N, xx .

EXAMPLE 58

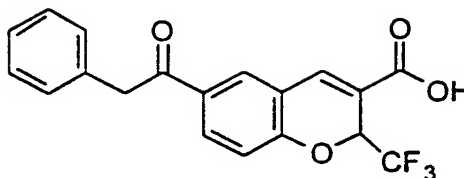


- 15 **6-N,N-Diethylaminosulfonyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

The title compound was prepared by a similar procedure to that described in Example

- 20 49: mp 238-240 °C. ^1H NMR ($\text{CD}_3\text{OD}/300\text{ MHz}$) 7.88 (s, 1H), 7.85 (d, 1H, $J = 2.2\text{ Hz}$), 7.79 (dd, 1H, $J = 8.5, 2.2\text{ Hz}$), 7.14 (d, 1H, $J = 8.5\text{ Hz}$), 5.88 (q H-F , 1H, $J = 7.2\text{ Hz}$), 3.24 (q, 2H, $J = 7.3\text{ Hz}$), 1.11 (t, 3H, $J = 7.3\text{ Hz}$). FABHRMS m/z
- 25 380.0763 ($\text{M} + \text{H}^+$, Calc'd 380.0780). Anal. Calc'd for $\text{C}_{15}\text{H}_{16}\text{F}_3\text{NO}_4\text{S}$: C, 47.49; H, 4.25; H, 3.69. Found: C, 47.62; H, 4.30; N, 3.72.

EXAMPLE 59



6-Phenylacetyl-2-trifluoromethyl-2H-1-benzopyran-3-
carboxylic acid

Step 1. Preparation of ethyl 6-phenylacetyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

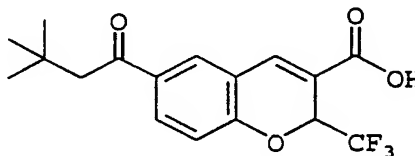
2-Trifluoromethyl-2H-1-benzopyran-3-carboxylic acid (Example 10) (1.32 g, 4.85 mmol) was cooled to 0 °C in dichloromethane (50 mL). Aluminum chloride (2.58 g, 19.5 mmol) was added and a dark red solution resulted. A solution of phenylacetyl chloride (1.8 g, 12.1 mmol) in dichloromethane (10.0 mL) was added dropwise over 40 minutes. The solution was warmed to room temperature and stirred for 16 hours. The solution was poured onto ice (200 mL) and extracted with diethyl ether (2 x 100 mL). The diethyl ether layers were combined, extracted with water (1 x 100 mL), 1 N HCl (2 x 100 mL), and saturated sodium bicarbonate (3 x 100 mL). Hexanes (20 mL) were added and the solution was extracted with brine (1 x 100 mL). The solution was dried over sodium sulfate and solvent was removed *in vacuo*. The crude ester was purified by flash chromatography over silica gel (with ethyl acetate as eluant) to afford the ester that was crystallized from diethyl ether/hexanes (830 mg, 44%): mp 136.2-138.0 °C. ¹H NMR (CDCl₃/300 MHz) 7.98 (dd, 2H, *J* = 8.4, 2.0 Hz), 7.90 (d, 1H, *J* = 2.0 Hz), 7.29 (s, 1H), 7.22 - 7.38 (m, 5H), 7.02 (d, 1H, *J* =

8.4 Hz), 5.75 (q H-F, 1H, $J=7.2$ Hz), 4.25 - 4.40 (m, 2H), 4.21 (s, 2H), 1.34 (t, 3H, $J = 7.0$ Hz). FABLRMS m/z 391 ($M + H^+$).

5 Step 2. Preparation of 6-phenylacetyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

The acid was converted from the ester (Step 1) via a method similar to that described in Example 1, step 2: mp 159.0-164.0 °C 1H NMR ($CD_3OD/300$ MHz) 8.04-8.16 (m, 3H), 7.87 (s, 1H), 7.05-7.30 (m, 5H), 5.86 (q H-F, 1H, $J=7.2$ Hz), 4.31 (s, 2H). FABLRMS m/z 363 ($M + H^+$). Anal. Calc'd for $C_{19}H_{13}F_3O_4 \cdot 0.29 H_2O$: C, 62.08; H, 3.73. Found: C, 62.04; H, 4.03.

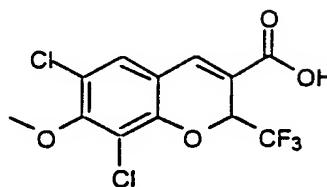
EXAMPLE 60



20 **6-(2,2-Dimethylpropylcarbonyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid**

The title compound was prepared by a similar procedure to that described in Example 59: mp 198-200 °C. 1H NMR ($CD_3OD/300$ MHz) 7.98-8.06 (m, 2H), 7.88 (s, 1H), 7.07 (d, 1H, $J = 8.9$ Hz), 5.86 (q H-F, 1H, $J = 7.2$ Hz), 2.88 (s, 2H), 1.05 (s, 9H). FABHRMS m/z 343.1175 ($M+H^+$, $C_{17}H_{18}F_3O_4$ requires 343.1157). Anal. Calc'd for $C_{17}H_{17}F_3O_4$: C, 59.65; H, 5.01. Found: C, 59.70; H, 4.97.

EXAMPLE 61



6,8-Dichloro-7-methoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

5

Step 1: Preparation of ethyl 7-methoxy-2-trifluoromethyl-benzopyran-2H-3-carboxylate.

4-Methoxysalicylaldehyde (2.38 g, 15.64 mmol), K_2CO_3 (2.16 g, 15.64 mmol) and ethyl
 10 4,4,4-trifluorocrotonate (2.8 mL, 3.16 g, 18.77 mmol) were dissolved in DMF (10 mL). The reaction was stirred at room temperature for 24 hours, diluted with water and extracted with Et_2O . The combined Et_2O phases were washed
 15 with water, dried over $MgSO_4$, filtered and concentrated *in vacuo* yielding an oil. Trituration with hexanes induced crystallization. Collection of the solid by vacuum filtration yielded the ester as a light
 20 brown crystalline solid (1.80 g, 38%): mp 78-80 °C. 1H NMR ($CDCl_3$ /300 MHz) δ 7.69 (s, 1H), 7.14 (d, 1H, J = 8.1 Hz), 6.59-6.50 (m, 2H), 5.68 (q, 1H, J = 7.1 Hz), 4.39-4.24 (m, 2H), 3.82 (s, 3H), 1.34 (t, 3H, J = 7.3 Hz).
 25 FABLRMS m/z 303 (M+H). FABHRMS m/z 303.0849 (M+H Calc'd 303.0844). Anal. Calc'd for $C_{14}H_{13}F_3O_4$: C, 55.63; H, 4.34. Found: C, 55.47; H, 4.31.

30 Step 2. Preparation of ethyl 6,8-dichloro-7-methoxy-2-trifluoromethyl-benzopyran-2H-3-carboxylate.

Chlorine gas (excess) was added to a stirred solution of the ester (Step 1) (1.35 g,

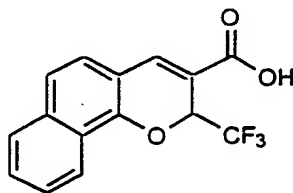
4.47 mmol) in HOAc (30 mL) until the yellow color persisted. After 20 minutes, the reaction was sparged with nitrogen causing the reaction to become straw colored. Zinc (0.86 g, 13.40 mmol) was added to this solution with vigorous stirring. After 45 minutes, additional zinc (0.86 g, 13.40 mmol) was added and the reaction was stirred overnight. The crude mixture was diluted with EtOH and filtered through diatomaceous earth. The filtrate was concentrated *in vacuo* yielding a crystalline mass. This solid was dissolved in EtOAc, washed with 2N HCl, brine, dried over MgSO₄, filtered and concentrated *in vacuo* yielding an oil. The oil was dissolved in a minimum of isooctane, inducing crystallization. Vacuum filtration of the suspension yielded tan needles (1.078 g) which were recrystallized from isooctane yielding the dichloro ester as tan crystals (0.71 g, 43%) of suitable purity to use in the next step: mp 113.3-115.1 °C. ¹H NMR (acetone-d₆/300 MHz) 7.88 (s, 1H), 7.63 (s, 1H), 6.02 (q, 1H, *J* = 6.8 Hz), 4.38-4.22 (m, 2H), 3.93 (s, 3H), 1.31 (t, 3H, *J* = 7.1 Hz). ¹⁹F NMR (acetone-d₆/282 MHz) -80.00 (d, *J* = 7.2 Hz).

Step 3: Preparation of 6,8-dichloro-7-methoxy-2-trifluoromethyl-benzopyran-2H-3-carboxylic acid.

To a stirred solution of the dichloro ester from Step 2 (0.686 g, 1.848 mmol) in THF (10 mL) and EtOH (3 mL) was added NaOH (0.81 mL of 2.5 M aqueous solution, 2.03 mmol) in one portion. After stirring overnight the reaction was partially concentrated, diluted with H₂O and washed with diethyl ether. The resulting aqueous phase was sparged with nitrogen and

acidified with 2N HCl solution causing the solution to become turbid. Filtration of this suspension yielded the title compound as a white powder (0.559 g, 88 %): mp 195.6-199.1 °C. ¹H NMR (CDCl₃/300 MHz) 7.90 (s, 1H), 7.64 (s, 1H), 6.01 (q, 1H, *J* = 6.8 Hz), 3.94 (s, 3H). ¹⁹F NMR (CDCl₃/282 MHz) -79.63 (d, *J* = 7.1 Hz). FABLRMS *m/z* 349 (M+Li). EIHRMS *m/z* 341.9681 (M+, Calc'd 341.9673). Anal. Calc'd for C₁₂H₇Cl₂F₃O₄: C, 42.01; H, 2.06. Found: C, 41.76; H, 2.14.

EXAMPLE 62



2-Trifluoromethyl-2H-naphtho[1,2-b]pyran-3-carboxylic acid

Step 1. Preparation of ethyl 2-trifluoromethyl-3H-naphthopyran-carboxylate.

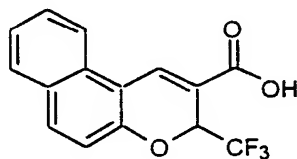
A mixture of 2-hydroxy-1-naphthaldehyde (8.6 g, 0.050 mol) and ethyl 4,4,4-trifluorocrotonate (9.2 g, 0.055 mol) dissolved in anhydrous dimethylformamide (DMF) and treated with anhydrous K₂CO₃ (13.8 g, 0.100 mol). The solution was maintained at room temperature for 50 hours and diluted with water. The solution was extracted with ethyl acetate, and the combined extracts were washed with brine, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to afford 4.8 g of an oil. The oil was purified by HPLC, eluting with hexanes:ethyl acetate (30:1). The appropriate fractions were concentrated to

afford 1.6 g (10%) of the naphthopyran ester as a yellow solid.

- Step 2. Preparation of 2-trifluoromethyl-3H-naphthopyran-carboxylic acid. A solution of the ester from Step 1 (0.8 g, 2.5 mmol) was dissolved in 40 mL of ethanol and 10 mL of tetrahydrofuran, treated with sodium hydroxide (2.5 N, 10 mL, 25 mmol) and stirred at room temperature for 16 hours. The reaction mixture was acidified with 1.0 N HCl, whereupon a solid formed that was isolated by filtration. The solid was washed with 20 mL of water to afford 0.7 g (95%) of the title compound as a yellow solid: mp 245.9-248.6 °C. ¹H NMR (acetone-*d*₆/300 MHz) 8.57 (s, 1H), 8.28 (d, 1H, *J* = 8.7 Hz), 8.03 (d, 1H, *J* = 9.0 Hz), 7.93 (d, 1H, *J* = 8.7), 7.67 (m, 1H), 7.50 (m, 1H), 7.28 (d, 1H, *J* = 9.0), 5.96 (q H-F, 1H, *J* = 7.2 Hz). FABHRMS *m/z* 295.0561 (M+H, Calc'd 295.0582). Anal. Calc'd for C₁₅H₉O₃F₃ + 3.31 % H₂O: C, 59.21; H, 3.35. Found: C, 59.17; H, 3.07.

EXAMPLE 63

25



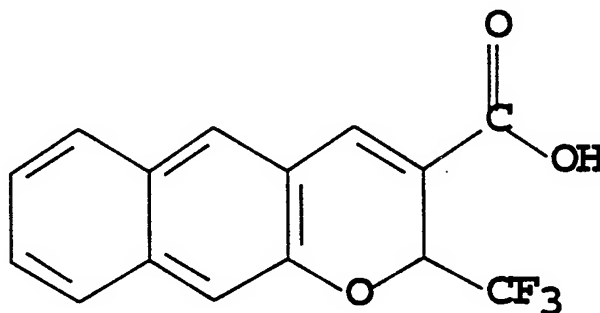
2-Trifluoromethyl-3H-naphtho[2,1-b]pyran-3-carboxylic acid

30

2-Hydroxy-naph-1-aldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 244.7-249.8 °C. ¹H NMR (CDCl₃/300 MHz) 8.61 (s, 1H), 8.09 (d,

1H, $J = 8.3$ Hz), 7.90 (d, 1H, $J = 8.9$ Hz), 7.82 (d, 1H, $J = 8.3$ Hz), 7.63 (t, 1H, $J = 8.1$ Hz), 7.47 (t, 1H, $J = 8.1$ Hz), 7.23 (d, 1H, $J = 9.1$ Hz), 5.84 (q, 1H, $J = 6.8$ Hz). ^{19}F NMR (CDCl₃/282 MHz) -79.56 (d, $J = 7.3$ Hz).
FABLRMS m/z 295 (M+H). FABHRMS m/z 295.0560 (M+H, Calc'd 295.0582). Anal. Calc'd for C₁₅H₉F₃O₃: C, 61.23; H, 3.08. Found: C, 60.85; H, 3.12.

10

EXAMPLE 64

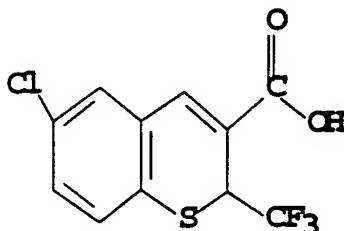
2-Trifluoromethyl-2H-naphtho[2,3-b]pyran-3-carboxylic acid

15

3-Hydroxynaphthalene-2-carboxylic acid was converted to 3-hydroxynaphthalene-2-carboxaldehyde by a similar procedure to that described in Example 24, Steps 1 & 2. The 3-hydroxynaphthalene-2-carboxaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp decompose >300 °C. ^1H NMR (CD₃OD/300 MHz) 7.99 (s, 1H), 7.90 (s, 1H), 7.84 (d, 1H, $J = 8.2$ Hz), 7.74 (d, 1H, $J = 8.2$ Hz), 7.50 (t, 1H, $J = 8.2$ Hz), 7.39 (t, 1H, $J = 8.2$ Hz), 7.34 (s, 1H), 5.77 (q, 1H, $J = 6.6$ Hz). EIHRMS m/z 294.0474 (M+, Calc'd 294.0504).

25

EXAMPLE 65



**6-Chloro-2-trifluoromethyl-2H-1-benzothiopyran-
3-carboxylic acid**

5

Step 1: Synthesis of 5-chloro-
thiosalicylaldehyde.

Tetramethylethylenediamine (TMEDA) (10.44 mL, 8.035 g, 69.15 mmol) was added via syringe to n-BuLi (43.22 mL of 1.6 M in hexanes, 69.15 mmol) and the solution was chilled to 0 °C. A solution of 4-chlorothiophenol (5.00 g, 34.57 mmol) in cyclohexane (25 mL) was added with stirring over 1 hour. The resulting tan slurry was stirred overnight at room temperature, chilled to 0 °C, and DMF (2.94 mL, 2.78 g, 38.03 mmol) was added via syringe over 2 minutes. The resulting gummy slurry was stirred at room temperature for 30 hours and became a powdery suspension. A mixture of 2 N HCl and ice was added to the reaction mixture until the pH became acidic (pH = 1). During this addition, the mixture warmed and became first red and then pale yellow. This mixture was extracted with ethyl acetate. The combined organic layers were washed with brine, dried over MgSO₄, filtered and concentrated *in vacuo* yielding a clear red-brown oil. This oil was triturated with hexanes yielding a red-brown semisolid. This semisolid was purified by plug flash chromatography over silica gel, eluting with 1:1, hexanes:dichloromethane to afford 5-chloro-thiosalicylaldehyde (0.858 g, 14%) as an intensely yellow solid suitable for use without further purification.

Step 2: Preparation of ethyl 6-chloro-2-trifluoromethyl-benzo-1-thiopyran-2-H-3-carboxylate.

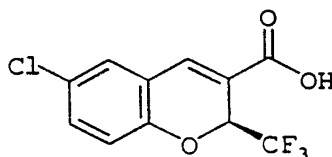
5-Chloro-thiosalicylaldehyde (Step 1) (0.84 g, 4.86 mmol) was added to DMF (3 mL) and ethyl 4,4,4-trifluorocrotonate (1.10 mL, 1.22 g). With stirring, K₂CO₃ (0.67 g, 4.86 mmol) was added causing the reaction to become a deep red. After stirring overnight at room temperature, the reaction was diluted with diethyl ether and washed with water, saturated NaHCO₃ solution, aqueous KHSO₄ solution (0.25 M), brine, dried over MgSO₄, filtered and concentrated *in vacuo* yielding an oil. The oil was purified by flash chromatography (5:1; hexanes: ethyl acetate) yielding upon concentration ethyl 6-chloro-2-trifluoromethyl-benzo-1-thiopyran-2-H-3-carboxylate as a bright orange solid (0.492 g, 31%): mp 94.6-97.4 °C ¹H NMR (acetone d₆/300 MHz) δ 8.01 (s, 1H), 7.71 (d, 1H, J = 2.2 Hz), 7.50 (d, 1H, J = 8.5 Hz), 7.44 (d of d, 1H, J = 2.3, 8.3 Hz), 5.07 (q, 1H, J = 8.5 Hz), 4.42-4.23 (m, 2H), 1.35 (t, 3H, J = 7.1 Hz). FABLRMS m/z 329 (M+Li).

Step 3: Preparation of 6-chloro-2-trifluoromethyl-benzo-1-thiopyran-2-H-3-carboxylic acid.

To a stirred solution of the ester from Step 2 (0.413 g, 1.280 mmol) in THF: EtOH : H₂O (7 : 2 : 1, 10 mL) was added NaOH solution (0.56 mL of 2.5 N solution, 1.408 mmol) with stirring. After stirring overnight, the reaction was partially concentrated *in vacuo* to remove the organic solvents, diluted

with H₂O and washed with several portions of diethyl ether. Acidification of the stirred aqueous phase with concentrated HCl caused precipitation of a flocculent yellow precipitate. Vacuum filtration of the suspension yielded 6-chloro-2-trifluoromethyl-
5 benzo-1-thiopyran-2H-3-carboxylic acid as a yellow powder (0.25 g, 66 %): mp 188.8-198.7 °C ¹H NMR (acetone d₆/300 MHz) δ 8.02 (s, 1H), 7.71 (d, 1H, J = 2.22 Hz), 7.50 (d, 1H, J = 8.5 Hz), 7.44 (d of d, 1 H, J = 2.2, 8.5 Hz), 5.05 (q, 1H, J = 8.6 Hz).
10 ¹⁹F NMR (Acetone d₆/282 MHz) δ -75.22 (d, J = 8.7 Hz). FABLRMS m/z 301 (M+Li); ESLRMS (neg. ion) m/z 293 (M-H).

15

EXAMPLE 66

20

(S)-6-Chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

To a solution of 6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid (Example 1, Step 2) (12.00 g, 43.07 mmol) and (S)(-)-α-methylbenzylamine (2.61 g, 21.54 mmol) in methyl-
25 tert-butyl ether (30 mL) was slowly added n-heptane (200 mL) until the mixture became cloudy. The mixture was heated (steam bath) to boiling and set

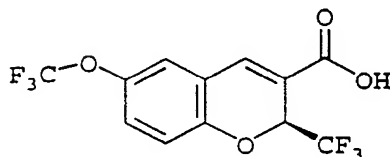
aside for 24 h during which time crystals formed. Filtration of the suspension yielded a crystalline product (5.5 g) which was recrystallized from methyl-tert-butyl ether (30 mL) and n-heptane (200 mL) yielding upon filtration a white solid (3.1 g). This solid was dissolved in EtOAc (100 mL) and washed with 1 N hydrochloric acid (50 mL) and brine (2 x 50 mL), dried over MgSO_4 and concentrated in vacuo yielding a white solid. Recrystallization of this solid from methyl-t-butyl ether/n-heptane yielded the title compound as the highly enriched isomer, a white solid (2.7 g, 45%): mp 126.7-128.9 °C. ^1H NMR (CDCl_3 /300 MHz) 7.78 (s, 1H), 7.3-7.1 (m, 3H), 6.94 (d, 1H, $J = 8.7$ Hz), 5.66 (q, 1H, $J = 6.9$ Hz). Anal. Calc'd for $\text{C}_{11}\text{H}_6\text{O}_3\text{F}_3\text{Cl}$: C, 47.42; H, 2.17; N, 0.0. Found: C, 47.53; H, 2.14; N, 0.0. This compound was determined to have an optical purity of greater than 90% ee.

20 Procedure for determining optical purity.

To a solution of the free acid (title compound) (0.005 g, 0.017 mmol) in ethyl acetate (1.5 mL) in a test tube was added (trimethylsilyl)diazomethane (30 μL of 2.0 N solution in hexanes, 60 mmol). The resulting yellow solution was warmed until the solution began to gently boil and then was allowed to cool to room temperature and stand for 0.08 hours. With vigorous mixing, the solution was quenched with aqueous 1 N HCl (1.5 mL). The layers were separated and a sample of the ethyl acetate fraction (0.3 mL) was transferred to a vial,

concentrated under a stream of nitrogen, was diluted with hexane (total of 1 mL) and a sample (10 μ L) analyzed by chiral chromatography. The HPLC utilized a Daicel ChiralPak AD column eluting with 10% isopropanol-hexane at 0.5 mL/min using a UV detector set at 254 nM.

EXAMPLE 67

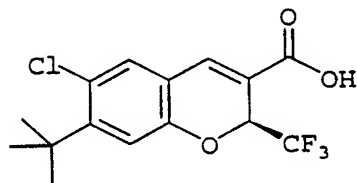


(S)-6-Trifluoromethoxy-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

15 To a solution of 6-trifluoromethoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 16) (17.72 g, 54.00 mmol) and (-)cinchonidine (7.95 g, 27.04 mmol) in methyl-tert-butyl ether (100 mL) heated on a steam-bath was
20 added n-heptane (200 mL). The mixture was heated on the steam bath to boiling and allowed to cool for 4 h during which time crystals formed. Filtration of the suspension yielded a crystalline solid (18.7 g). This solid was dissolved in 2-butanone (30 mL)
25 followed by the addition of n-heptane (500 mL). After standing for 16 hours, the resulting suspension was filtered yielded a white solid (10.3 g). This solid was dissolved in ethyl acetate (150

mL), washed with 1 N hydrochloric acid (100 mL) and
brine (2 x 50 mL), dried over MgSO_4 , filtered, and
concentrated *in vacuo* yielding a viscous yellow oil
(5.2 g, 59%): ^1H NMR (acetone- d_6 /300 MHz) 7.16 (s,
5 1H), 6.77 (d, 1H, $J = 2.7$ Hz), 6.94 (d, 1H, $J = 8.7$
Hz), 6.64 (m, 1H), 6.39 (d, 1H, $J = 8.7$ Hz) 5.13
(q, 1H, $J = 7.2$ Hz). Anal. Calc'd for $\text{C}_{12}\text{H}_6\text{O}_4\text{F}_6$: C,
43.92; H, 1.84; N, 0.0. Found: C, 43.79; H, 1.83;
N, 0.0. This compound was determined to have an
10 optical purity of greater than 90% ee. Chiral
purity was determined as describe in Example 66.

EXAMPLE 68

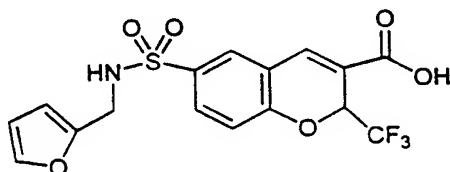


(S)-6-Chloro-7-(1,1-dimethylethyl)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

20 To a solution of 6-chloro-7-(1,1-
dimethylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-
carboxylic acid (Example 8) (11.4 g, 34.1 mmol) and
(S)(-)-2-amino-3-phenyl-1-propanol (2.57 g, 17.00
mmol) was added n-heptane (200 mL) and the mixture
25 set aside for 16 hours. The resulting suspension
was filtered yielding a solid (3.8 g). This solid
was recrystallized from 2-butanone (20 mL) and n-
heptane (200 mL) yielding upon filtration a white

solid (3.0 g). This solid was dissolved in ethyl acetate (100 mL) and washed with 1 N HCl (50 mL) and brine (2 x 50 mL), dried over MgSO₄ and concentrated in vacuo yielding a white solid. This solid was
5 recrystallized from n-heptane yielding the title compound of high optical purity as a crystalline solid (1.7 g, 30%): mp 175.4-176.9 °C. ¹H NMR (acetone-d₆/300 MHz) 7.86 (s, 1H), 7.52 (s, 1H), 7.12 (s, 1H), 5.83 (q, 1H, J = 7.1 Hz), 1.48 (s,
10 9H). Anal. Calc'd for C₁₅H₁₄O₃F₃Cl: C, 53.83; H, 4.22; N, 0.0; Cl, 10.59. Found: C, 53.78; H, 4.20; N, 0.0; Cl, 10.65. This compound was determined to have an optical purity of greater than 90% ee. Chiral purity was determined as describe in Example 66.

15

EXAMPLE 69

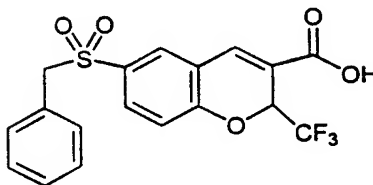
20 **6-[[[(2-Furanylmethyl)amino]sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The title compound was prepared by a similar procedure to that described in Example 49: mp 170-
25 173 °C. ¹H NMR (CD₃OD/300 MHz) 7.78 (s, 1H), 7.66-7.76 (m, 2H), 7.18-7.22 (m, 1H), 7.00-7.08 (m, 1H), 6.12-6.18 (m, 1H), 6.02-6.06 (m, 1H), 5.85 (q, 1H,

$J = 7.0$ Hz), 4.13 (s, 2H). EIHRMS m/z 403.0332 (M^+ , Calc'd 403.0337).

EXAMPLE 70

5



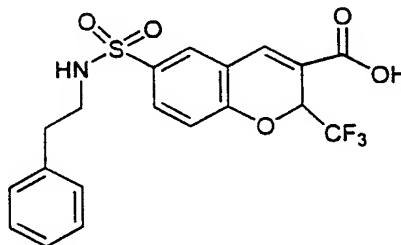
6-[(Phenylmethyl) sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

10.

The 2H-1-benzopyran-3-carboxylic acid was prepared analogous to the procedure described in Example 56: mp 172-176 °C. ^1H NMR ($\text{CD}_3\text{OD}/300$ MHz) 7.73 (s, 1H), 7.43-7.56 (m, 2H), 7.21-7.33 (m, 3H), 7.20-7.21 (m, 3H), 5.88 (q, 1H, $J = 7.0$ Hz), 4.83 (s, 2H). EIHRMS m/z 398.0399 (M^+ , Calc'd 398.0436).

15

EXAMPLE 71



20

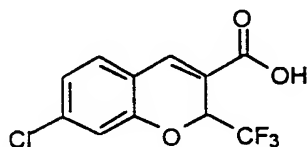
6-[[(Phenylethyl) amino] sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared analogous to the procedure described in

Example 49: mp 187-190 °C. ¹H NMR (CD₃OD/300 MHz)

5 7.82 (s, 1H), 7.74-7.90 (m, 2H), 7.08-7.29 (m, 6H),
5.89 (q, 1H, *J* = 6.8), 3.12 (t, 2H, *J* = 7.3 Hz),
2.72 (t, *J* = 7.3 Hz). EIHMS *m/z* 427.0675 (M⁺,
Calc'd 427.0701)

10

EXAMPLE 72

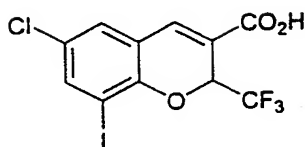
15

7-Chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

4-Chlorosalicylic acid was converted to 3-chlorosalicylaldehyde by a procedure similar to that described in Example 24, Steps 1 & 2. The 3-chlorosalicylaldehyde was converted to the title compound by a procedure similar to Example 1: mp 175.2-177.6 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.90 (s, 1H), 7.51 (d, 1H, *J* = 7.8 Hz), 7.12 (m, 2H), 5.86 (q H-F, 1H, *J* = 7.2 Hz). FABHMS *m/z* 285.0114
25 (M⁺Li, Calc'd 285.0118). Anal. Calc'd for C₁₁H₆ClF₃O₃: C, 47.42; H, 2.17; Cl, 12.72. Found: C, 47.54; H, 2.37; Cl, 12.85.

16A

EXAMPLE 73



5

**6-Chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-
3-carboxylic acid**

Step 1. Preparation of 3-iodo-5-

10 chlorosalicylaldehyde

N-Iodosuccinimide (144.0 g, 0.641 mole) was added to a solution of 5-chlorosalicylaldehyde (100 g, 0.638 mole) in dimethylformamide (400 mL). The reaction mixture was stirred for two days at room temperature. Additional N-iodosuccinimide (20 g, 0.089 mole) was added and the stirring was continued for an additional two days. The reaction mixture was diluted with ethyl acetate (1 liter), washed with hydrochloric acid (300 mL, 0.1 N), water (300 mL), sodium thiosulfate (300 mL, 5%), and brine (300 mL). It was dried over MgSO_4 , and was concentrated to dryness to afford the desired aldehyde as a pale yellow solid (162 g, 90%): mp 84.8-86.7 °C. ^1H NMR (CDCl_3 /300 MHz) 11.67 (s, 1H), 9.71 (s, 1H), 7.92 (d, 1H, $J = 2.5$ Hz), 7.54 (d, 1H, $J = 2.6$ Hz). FABLRMS m/z 281.0 (M-H). ESHRMS m/z 280.8851 (M-H, Calc'd. 280.88630).

Step 2. Preparation of ethyl 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate

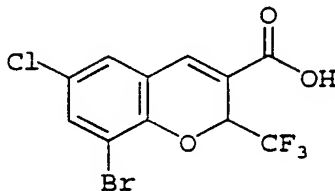
5-Chloro-3-iodosalicylaldehyde (20 g, 70.8 mmol), ethyl 4,4,4-trifluorocrotonate (17.85 g, 106 mmol), and triethylamine (14.33 g, 142 mmol) were dissolved in DMSO (200 mL). The reaction mixture was stirred at 90 °C for three days. The reaction mixture was poured into ethyl acetate (800 mL). It was extracted with 10% HCl (2 x 200 mL), saturated aqueous NaHCO₃ (2 x 200 mL), and water (2 x 200 mL). The ethyl acetate phase was dried over MgSO₄, filtered and evaporated to yield a brown solid. It was then run through a plug of silica with ethyl acetate-hexane (1:20). The solvent was evaporated to give a yellow solid, that was recrystallized in hexane to afford the ester as a white solid (19.61 g, 64%): mp 92.1-93.9 °C. ¹H NMR (CDCl₃/300 MHz) 7.71 (d, 1H, J = 2.2 Hz), 7.56 (s, 1H), 7.20 (d, 1H, J = 2.2 Hz), 5.81 (q, 1H, J = 6.7 Hz), 4.37-4.29 (m, 2H), 1.35 (t, 3H, J = 7.2 Hz). FABLRMS m/z 431.9 (M-H). EIHRMS m/z 431.9269 (M-H, Calc'd. 431.9237).

Step 3. Preparation of 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

25 The ester (Step 2) was converted to the acid by a procedure similar to the method described in Example 1, Step 2: mp 220-223 °C. ¹H NMR (CD₃OD/300 MHz) 7.77 (d, 1H, J = 2.2 Hz), 7.71 (s, 1H), 7.41 (d, 1H, J = 2.2 Hz), 5.87 (q, 1H, J = 7.0 Hz). EIHRMS m/z 403.8893 (M-H, Calc'd. 403.8924). Anal.

Calc'd for $C_{11}H_5ClF_3IO_3$: C, 32.66; H, 1.25. Found:
C, 33.13; H, 1.29.

EXAMPLE 74



5

8-Bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran- 3-carboxylic acid

10 Step 1. Preparation of ethyl 8-bromo-6-chloro-2- trifluoromethyl-2H-1-benzopyran-3-carboxylate

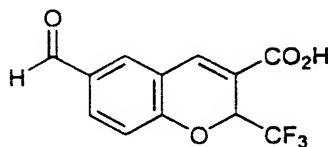
A mixture of 3-bromo-5-chlorosalicylaldehyde (1.9 g, 4.2 mmol), potassium carbonate (0.58 g, 4.2 mmol), and ethyl 4,4,4-trifluorocrotonate (0.79 g, 15 4.7 mmol) was stirred in *N,N*-dimethylformamide (5 mL) at 95° C for 18 h. Water (100 mL) was added and the mixture was extracted with ether (3 x 50 mL). The combined organic extracts were washed with sodium hydroxide (10 mL) and water (2 x 50 mL). 20 After drying over MgSO₄ and concentrating, the mixture filtered through of a pad of silica eluting with ethyl acetate-hexanes (1:4). The eluant was concentrated and a light yellow solid was crystallized from cold hexane (0.43 g, 26%): mp 25 101.0-102.2 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.90 (s, 1H), 7.65 (d, H, *J* = 2.4 Hz), 7.61 (d, H, *J* = 2.4 Hz), 6.03 (q _{H-F}, 1H, *J* = 6.9 Hz), 4.34 (m, 2H),

1.33 (t, 3H, $J = 7.5$ Hz). ESHRMS m/z 384.9435 (M-H, Calc'd 384.9454). Anal. Calc'd for $C_{13}H_9BrClF_3O_3$: C, 40.50; H, 2.35. Found: C, 40.61; H, 2.40.

5 Step 2. Preparation of 8-bromo-6-chloro-2-
trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

Ethyl 8-bromo-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (0.3 g), ethanol (15 mL), tetrahydrofuran (10 mL), and sodium hydroxide
10 solution (10 mL, 2.5 N) were stirred at room temperature for 16 h. Hydrochloric acid (1 N) was added until the mixture was acidic to pH paper. The addition of water (50 mL) caused the formation of a precipitate which was collected by filtration
15 yielding the title compound as a white solid (0.2 g, 72%): mp 227.8-228.9 °C. 1H NMR (acetone- d_6 /300 MHz) 7.90 (s, 1H), 7.65 (dd, 2H, $J = 2.4$ and $J = 28.8$ Hz), 6.00 (q $H-F$, 1H, $J = 7.2$ Hz). FABHRMS m/z 356.9134 (M+H, Calc'd 356.9141). Anal. Calc'd for
20 $C_{11}H_5BrClF_3O_3$: C, 36.96; H, 1.41. Found: C, 37.05; H, 1.33.

EXAMPLE 75



6-Formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

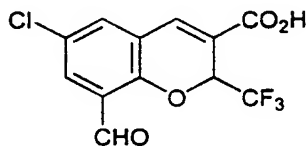
Step 1. Preparation of ethyl 6-formyl-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A 50 mL round bottom flask was charged with 5-
5 formylsalicylaldehyde (3.21 g, 21.39 mmol), ethyl
4,4,4-trifluorocrotonate (3.50 mL, 3.96 g, 23.53
mmol), dimethylformamide (15 mL) and potassium
carbonate (2.95 g, 21.39 mmol) and heated to 60 °C
for 12 hours. Additional ethyl 4,4,4-
10 trifluorocrotonate (3.50 mL, 3.96 g, 23.53 mmol) was
added and the reaction heated for 16 hours at 75 °C.
After cooling to room temperature, the reaction was
partitioned between H₂O and diethyl ether. The
organic phase was washed with saturated NaHCO₃,
15 solution, KHSO₄ solution (0.25 M), brine, treated
with decolorizing carbon (warmed gently). The
resulting black suspension was dried over MgSO₄,
vacuum filtered through diatomaceous earth, and
concentrated in vacuo yielding an orange crystalline
20 mass. This material was recrystallized from hot
hexanes yielding the ester (1.51 g, 24 %) as orange
crystals: mp 84.3-86.2 °C. ¹H NMR (acetone-d₆/300
MHz) 9.96 (s, 1H), 8.06 (d, 1H, *J* = 2Hz), 8.02 (s,
1H), 7.99 (dd, 1H, *J* = 8.5, 2.0Hz), 7.24 (d, 1H, *J* =
25 8.5 Hz), 5.99 (q, 1H, *J* = 7.1 Hz), 4.43-4.25 (m,
2H), 1.34 (t, 3H, *J* = 7.3 Hz). FABLRMS *m/z* 301
(M+H). EIHRMS *m/z* 300.0605 (M+, Calc'd 300.0609).
Anal. Calc'd for C₁₄H₁₁F₃O₄: C, 56.01; H, 3.69.
Found: C, 56.11; H, 3.73.

Step 2. Preparation of 6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester (Step 1) was converted to the acid via a method similar to that described in Example 1,
 5 Step 2: mp 211.3-215.7 °C. ¹H NMR (acetone-d₆/300 MHz) 9.97 (s, 1H), 8.07 (d, 1H, *J* = 2.0Hz), 8.03 (s, 1H), 8.00 (dd, 1H, *J* = 8.3, 2.0 Hz), 7.25 (d, 1H, *J* = 8.5 Hz), 5.98 (q, 1H, *J* = 6.9 Hz). FABLRMS *m/z* 273 (M+H). EIHRMS *m/z* 272.0266 (M+, Calc'd
 10 272.0296). Anal. Calc'd for C₁₂H₇F₃O₄: C, 52.95; H, 2.59. Found: C, 52.62; H, 2.58.

EXAMPLE 76



15

6-Chloro-8-Formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of 4-chloro-2,6-bis(hydroxymethyl)phenol.

Potassium hydroxide (84.82 g, 1.30 mole) was dissolved in H₂O (200 mL) in a two liter 3-neck round bottom flask fitted with thermocouple, mechanical stirrer, and stopper. With stirring, 4-chlorophenol (128.56 g, 1.0 mole) was added with
 25 cooling (ice bath) resulting in the temperature rising to 26 °C. Formalin (230 mL of 37% aqueous solution, 2.83 mole) was added portion-wise

maintaining the temperature below 25 °C. The reaction was warmed to 35 °C for 48 hours. To this solution was added aqueous acetic acid (80.0 mL, 84.1 g, 1.40 mole in 800 mL H₂O) causing the solution to become turbid. Vacuum filtration of the suspension yielded a tan solid. The solid was stirred with acetone (100 mL) and the insoluble product collected by vacuum filtration. The solution was diluted with hexanes yielding several crops of the diol as fine tan needles (35.0 g, 19%). mp 160.6-163.3 °C. ¹H NMR (acetone-d₆, NaOD, D₂O/300 MHz) 6.69 (s, 2H), 4.48 (s, 4H), 7.88 (d, 1H, J = 2.6 Hz), 7.75 (d, 1H, J = 2.6 Hz), 6.08 (q, 1H, J = 6.9 Hz). ESI/MS m/z 206 (M+NH₄⁺). ESI/MS m/z 187.0131 (M-H, Calc'd 187.0162).

Step 2. Preparation of 5-chloro-3-formyl-salicylaldehyde.

To a stirred suspension of diol (Step 1) (33.0 g, 0.18 mole) in chloroform (1.5 L) in a 2 L round bottom flask was added manganese dioxide (139 g, 1.60 mole) and the resulting suspension heated to a gentle reflux for 10 hours. The reaction was allowed to cool to room temperature, was filtered through diatomaceous earth, concentrated *in vacuo*, presorbed on silica gel and purified by flash chromatography (hexane/ethyl acetate) yielding the as a mustard colored powder dialdehyde (22.42 g, 67 %): mp 120.7-122.8 °C. This solid was of suitable purity to use in the next step without further purification.

Step 3. Preparation of ethyl 6-chloro-8-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A stirred solution of the dialdehyde (Step
5 2) (1.13 g, 6.14 mmol), dimethyl sulfoxide (6 mL),
ethyl 4,4,4-trifluorocrotonate (1.37 mL, 1.55g, 9.21
mmol) and triethylamine (1.71 mL, 1.24g, 12.28 mmol)
in a round bottom flask fitted with condenser was
heated to 80 °C for 8 h. Upon cooling to room
10 temperature the reaction was diluted with diethyl
ether (100 mL) and the resulting mixture washed with
aqueous sodium bicarbonate solution (3 X 75 mL), 1 N
HCl solution (3 X 70 mL), and brine (1 X 75 mL),
dried over MgSO₄, filtered and concentrated *in vacuo*
15 yielding a tan powder. This powder was taken up in
hot hexane-ethyl acetate and filtered to remove
insoluble matter. Upon cooling of the filtrate,
crystallization followed by vacuum filtration
yielded the desired ester as tan crystals (0.726 g,
20 35%): mp 118.1-119.7 °C. This material was of
suitable purity to use without further purification.

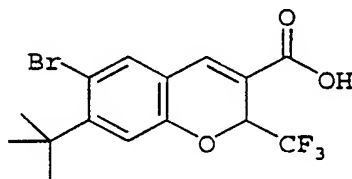
Step 4. Preparation of 6-chloro-8-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

25 To a stirred solution of the ester (Step
3) (0.284 g, 0.849 mmol) in THF:EtOH:H₂O (7:2:1, 5
mL) was added aqueous NaOH solution (0.41 mL of 2.5
M, 1.02 mmol). After stirring 40 hours, the
reaction was partially concentrated *in vacuo* to
30 remove the organic solvents, diluted with H₂O,
washed with diethyl ether, sparged with nitrogen to

remove trace diethyl ether, and acidified with concentrated HCl yielding a suspension. Vacuum filtration of the suspension yielded the title compound as a pale yellow powder (0.160 g, 23 %).

- 5 mp 243.3-252.4 °C. ^1H NMR (acetone- d_6 /300 MHz) 10.39 (s, 1H), 7.98 (s, 1H), 7.88 (d, 1H, $J = 2.6$ Hz), 7.75 (d, 1H, $J = 2.6$ Hz), 6.08 (q, 1H, $J = 6.9$ Hz). FABLRMS m/z 307 (M+H). ESHRMS m/z 304.9839 (M-H, Calc'd 304.9828). Anal. Calc'd for
10 $\text{C}_{12}\text{H}_6\text{Cl}_1\text{F}_3\text{O}_4$: C, 47.01; H, 1.97. Found: C, 46.64; H, 1.86.

EXAMPLE 77

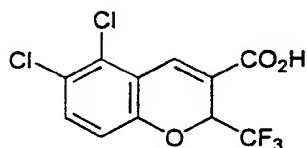


6-Bromo-7-(1,1-dimethylethyl)-2-(trifluoromethyl)- 2H-1-benzopyran-3-carboxylic acid

- 20 7-(1,1-Dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 12) (0.6 g, 2 mmol), chloroform (50 mL), iron filings (0.01 g, 0.2 mmol), and bromine (0.48 g, 3.00 mmol) were stirred at reflux for 16 h. The mixture was allowed to cool
25 and was washed with brine (2 x 50 mL). After drying over MgSO_4 , the mixture was filtered, concentrated *in vacuo*, and the residue crystallized from ether-

hexanes yielding the title compound as a white solid (0.5 g, 66%): mp 198.6-199.9 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.85 (s, 1H), 7.72 (s, 1H), 7.13 (s, 1H), 5.83 (q, 1H, *J* = 7.2 Hz), 1.5 (s, 9H). Anal. 5 Calc'd for C₁₅H₁₄O₃F₃Br: C, 47.52; H, 3.72; N, 21.07. Found: C, 47.42; H, 3.68; N, 21.15.

EXAMPLE 78

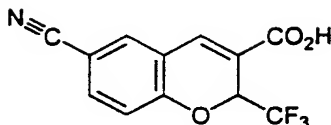


10

5,6-Dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

15 5,6-Dichlorosalicylaldehyde was prepared by the procedure described in Cragoe, E.J.; Schultz, E.M., U.S. Patent 3 794 734, 1974. This salicylaldehyde was converted to the title compound by a similar procedure to that described in Example 1: mp 211.5- 20 213.5 °C. ¹H NMR (acetone-*d*₆/300 MHz) 8.09 (s, 1H), 7.63 (d, 1H, *J* = 8.9 Hz), 7.12 (d, 1H, *J* = 8.9 Hz), 5.94 (q, 1H, *J* = 7.0 Hz). ESLRMS *m/z* 311 (M-H). EIHRMS *m/z* 311.9583 (M⁺, Calc'd 311.9568). Anal. Calc'd for C₁₁H₅Cl₂F₃O₃: C, 42.20; H, 1.61. Found: 25 C, 42.33; H, 1.67.

EXAMPLE 79



5 **6-Cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-**
 carboxylic acid

Step 1. Preparation of ethyl 6-
[(hydroxyimino)methyl]-2-(trifluoromethyl)-2H-1-
10 benzopyran-3-carboxylate.

A 50 mL round bottom flask was charged with hydroxylamine HCl (0.255 g, 3.67 mmol), ethyl 6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 75, Step 1) (1.00 g, 3.34
15 mmol), sodium acetate (0.301 g, 3.67 mmol), ethanol (10 mL), and H₂O (2 mL). The reaction was stirred at room temperature for 18 hours, then diluted with H₂O and diethyl ether. The layers were separated and the organic phase washed with H₂O, brine, dried
20 over MgSO₄, filtered, and concentrated *in vacuo* yielding an orange semi-crystalline mass. Recrystallization of this solid from hot ethyl acetate and isooctane yielded the oxime (0.578 g, 55%): mp 113.0-116.2 °C. ¹H NMR (acetone-*d*₆/300
25 MHz) 10.46 (s, ca.1 exch.), 8.11 (s, 2H), 7.92 (s, 1H), 7.72 (d, 1H, *J* = 2Hz), 7.68 (dd, 1H, *J* = 8.5, 2.0Hz), 7.07 (d, 1H, *J* = 8.5 Hz), 5.89 (q, 1H, *J* = 7.1 Hz), 4.43-4.22 (m, 2H), 1.34 9t, 3H, *J* = 7.3 Hz). FABLRMS *m/z* 316 (M+H). EIHRMS *m/z* 315.0719

(M+, Calc'd 315.0733). Anal. Calc'd for $C_{14}H_{12}F_3N_1O_4$: C, 53.34; H, 3.84; N 4.44. Found: C, 53.85; H, 3.90; N, 4.19.

5 Step 2. Preparation of ethyl 6-cyano-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

To a stirred solution of oxime (Step 1) (0.264 g, 0.840 mmol) in dioxane (4.5 mL) in a 25 mL pear-shaped flask was added trifluoroacetic anhydride
10 (0.130 mL, 0.194 g, 0.924 mmol) and triethylamine (0.140 mL, 0.102 g, 1.008 mmol). The reaction was stirred at room temperature for 12 hours, then heated to 85 °C for 4 hours. After cooling to room temperature, aqueous HCl (50 ml, 1 N HCL) was added,
15 and the resulting mixture extracted with ethyl acetate . The ethyl acetate phase was washed with chilled aqueous HCl (1 N), brine, dried over Na_2SO_4 , filtered, concentrated in vacuo yielding a pale yellow oil. This oil was resubmitted to similar
20 reaction conditions. After dissolution of the pale yellow oil in dioxane (4.5 mL), trifluoroacetic anhydride (0.130 mL, 0.194 g, 0.924 mmol) and triethylamine (0.140 mL, 0.102 g, 1.008 mmol) were then added. After stirring 3 hours at room
25 temperature, more triethylamine 0.50 mL, 0.36 g, 3.6 mmol) was added and then heated to 85 °C for 3 hours. After cooling to room temperature, aqueous HCl (50 ml, 1 N HCL) was added, and the resulting mixture extracted with ethyl acetate . The ethyl
30 acetate phase was washed with chilled aqueous HCl (1 N), brine, dried over Na_2SO_4 , filtered, concentrated

in vacuo yielding a pale yellow oil. Addition of hexanes induced crystallization followed by vacuum filtration yielded the title compound (0.101g, 40 %) as a yellow powder: mp 101.6-106.1 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.97 (d, 1H, *J* = 2.2 Hz), 7.95 (s, 1H), 7.82 (dd, 1H, *J* = 8.5, 2.0Hz), 7.24 (d, 1H, *J* = 8.5 Hz), 6.01 (q, 1H, *J* = 7.1 Hz), 4.38-4.24 (m, 2H), 1.34 (t, 3H, *J* = 7.3 Hz). FABLRMS *m/z* 298 (M+H). EIHRMS *m/z* 297.0575 (M+, Calc'd 297.0613).

10

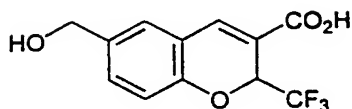
Step 3. Preparation of 6-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

To a stirred solution of the ester (Step 2) (0.077 g, 0.259 mmol) in THF-EtOH-H₂O (7:2:1, 2 mL) in a 5 mL pear-shaped flask was added aqueous NaOH (0.13 mL, 2.5 N solution) in one portion. After stirring for 6 hours at room temperature the solution was partially concentrated *in vacuo* to remove most of the THF and EtOH. The resulting solution was diluted with H₂O and washed with diethyl ether. The resulting aqueous phase was sparged with nitrogen to remove trace diethyl ether and was acidified with concentrated HCl yielding a sticky suspension. The suspension was extracted with diethyl ether and the ether was dried over MgSO₄, filtered and concentrated *in vacuo* yielding a pale yellow oil. This oil was crystallized from methylene chloride-hexanes yielding the title compound (0.041 g, 59 %) as a tan powder: mp 185.1-186.1 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.99-7.94 (m,

30

2H), 7.83 (dd, 1H, $J = 8.5, 2.0$ Hz), 7.25 (d, 1H, $J = 8.5$ Hz), 5.99 (q, 1H, $J = 7.0$ Hz). FABLRMS m/z 270 (M+H). EIHRMS m/z 269.0316 (M+, Calc'd 269.0300).

5

EXAMPLE 80

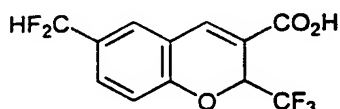
10 **6-Hydroxymethyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

To a chilled (ice bath), stirred solution of 6-formyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 75, Step 2) (0.133 g, 0.489 mmol) in THF (1 mL) and ethanol (1 mL) in a 10 mL round bottom flask was added NaBH₄ (0.020 g, 0.528 mmol) in two portions. The reaction was allowed to warm to room temperature and more NaBH₄ (0.050 g, 1.322 mmol) was added. The total reaction time was 3 hours. The reaction was quenched with aqueous HCl (1 N solution) and was extracted with chloroform. The organic phase was dried over MgSO₄, filtered and concentrated *in vacuo* yielding a foam.

25 This crude product was purified by flash chromatography (silica gel 60, eluant 1:1, hexane-ethyl Acetate with 2 % acetic acid). The product collected from the chromatography was recrystallized from hexanes and ethyl acetate, and collected by

vacuum filtration yielding the title compound (0.042 g, 31 %) as a very pale yellow powder: mp 177.5-180.8 °C. ¹H NMR (acetone-d₆/300 MHz) 7.89 (s, 1H), 7.44 (s, 1H), 7.41 (d, 1H, J = 8.3 Hz), 6.99 (d, 1H, J = 8.3 Hz), 5.80 (q, 1H, J = 7.3 Hz), 4.59 (s, 2H). FABLRMS m/z 275 (M+H). EIHRMS m/z 274.0417 (M+, Calc'd 274.0453). Anal. Calc'd for C₁₂H₉F₃O₄: C, 52.57; H, 3.31. Found: C, 52.43; H, 3.34.

10

EXAMPLE 81

15

6-(Difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl 6-(difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

Ethyl 6-formyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (Example 75, Step 1) (1.672 g, 5.569 mmol) in methylene chloride (1.5 mL) was added to methylene chloride (1.5 mL) and diethylaminosulfur trifluoride (DAST) (0.74 mL, 0.898 g, 5.569 mmol) over 0.07 hours via syringe. After stirring for 20 hours the reaction was poured into aqueous HCl (2.0 N) and the mixture was extracted with diethyl ether. The ethereal phase was washed with dilute aqueous HCl (2.0 N), saturated NaHCO₃ solution, brine, dried over MgSO₄,

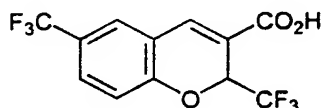
filtered and concentrated *in vacuo* yielding a clear colorless oil. This oil was purified by flash chromatography (Silica gel 60, Eluant (5:1; Hexanes : Ethyl Acetate) yielding ethyl 6-difluoromethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (0.96 g, 54 %) as an oil which solidified upon standing. This product was of sufficient purity to be used in the next step without further purification: ^1H NMR (acetone- d_6 /300 MHz) 7.97 (s, 1H), 7.74 (s, 1H), 7.65 (d, 1H, $J = 8.5$ Hz), 7.18 (d, 1H, $J = 8.5$ Hz), 6.90 (t, 1H, $J = 56.0$ Hz), 5.94 (q, 1H, $J = 7.0$ Hz), 4.40-4.25 (m, 2H), 1.34 (t, 3H, $J = 7.0$ Hz).

Step 2. Preparation of 6-(difluoromethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

Aqueous NaOH (1.31 mL, 3.277 mmol, 2.5 M solution) was added in one portion to the ester (Step 1) (0.880 g, 2.731 mmol) in THF:EtOH:H₂O (7:2:1, 10 mL). The resulting solution was stirred for 60 hours. The reaction mixture was partially concentrated *in vacuo* to remove the organic solvents and was diluted with H₂O. The resulting aqueous solution was washed with diethyl ether, sparged with nitrogen to remove trace ether, and acidified with concentrated HCl. The resulting oily suspension was extracted with diethyl ether. The combined organic phases were dried over MgSO₄, filtered and concentrated *in vacuo* yielding the title compound (0.483 g, 60%) as an oil which solidified as a white crystalline mass: mp 134.7-136.2 °C. ^1H NMR

(acetone- d_6 /300 MHz) 7.97 (s, 1H), 7.73 (s, 1H),
 7.67 (dd, 1H, $J = 8.5, 1.0$ Hz), 7.17 (d, 1H, $J = 8.5$
 Hz), 6.89 (t, 1H, $J = 56.2$ Hz), 5.90 (q, 1H, $J = 7.1$
 Hz). FAB-ESLRMS m/z 293 (M-H). EIHRMS m/z 293.0235
 5 (M-H, Calc'd 293.0237). Anal. Calc'd for $C_{12}H_7F_5O_3$:
 C, 49.00; H, 2.40. Found: C, 48.78; H, 2.21.

EXAMPLE 82



10

2,6-Bis(trifluoromethyl)-2H-1-benzopyran-3- carboxylic acid

15 Step 1. Preparation of Ethyl 2,6-
bis(trifluoromethyl)-4-oxo-4H-1-benzopyran-3-
carboxylate.

To a stirred solution of ethyl 4,4,4-
 trifluoroacetoacetate (3.22 mL, 4.06 g, 22.07 mmol)
 20 in toluene (100 mL) was added portion-wise sodium
 hydride (0.971 g, of 60 % oil dispersion reagent,
 22.07 mmol) causing gas evolution. After gas
 evolution has subsided, 2-fluoro-5-
 (trifluoromethyl)benzoyl chloride (5.00 g, 22.07
 25 mmol) was added. The reaction was stirred at room
 temperature for 24 hours, then heated to 105 °C for
 24 hours. After cooling to room temperature, the
 reaction was diluted with diethyl ether and the
 resulting solution was washed with H_2O and brine,

dried over MgSO_4 , filtered and concentrated *in vacuo* yielding a slightly sticky white solid. This solid was triturated with hexanes yielding the desired ester (3.05 g, 39 %) as a white powder: mp 116-120.1
5 °C. ^1H NMR (CDCl_3 /300 MHz) 8.52 (d, 2H, $J = 1.6$ Hz), 8.03 (dd, 1H, $J = 8.9, 2.2$ Hz), 7.71 (d, 1H, $J = 8.9$ Hz), 4.48 (q, 2H, $J = 7.3$ Hz), 1.39 (t, 3H, $J = 7.3$ Hz). FABLRMS m/z 355 (M+H). Anal. Calc'd for $\text{C}_{14}\text{H}_8\text{F}_6\text{O}_4$: C, 47.45; H, 2.28. Found: C, 47.59; H,
10 2.43.

Step 2. Preparation of ethyl 2,6-bis(trifluoromethyl)-4-oxo-dihydrobenzopyran-3-carboxylate.

15 A 250 mL round bottom flask was charged with ethyl 2,6-bis(trifluoromethyl)-benzopyran-4-one-3-carboxylate (Step 1) (2.307 g, 6.513 mmol) and THF (20 mL) yielding a pale yellow solution. Ethanol (20 mL) was added and the reaction chilled in an
20 ice-salt bath. While maintaining the reaction temperature at below 9 °C, NaBH_4 (0.246 g, 6.513 mmol) was added in two portions and the mixture stirred 1 h. The crude reaction mixture was poured into a vigorously stirred mixture of ice (200 mL)
25 and concentrated HCl (12 N, 5 mL) yielding a precipitate. Vacuum filtration of the resulting suspension yielded the desired keto ester (2.204 g, 87%) as faint pink powder of suitable purity to use in the next step without further purification: mp
30 71.8-76.9 °C. ^1H NMR (acetone- d_6 /300 MHz) 12.71 (br

s, 1H exch), 8.01 (d, 1H, $J = 2.0$ Hz), 8.01 (d, 1H, $J = 2.0$ Hz), 7.88 (dd, 1H, $J = 8.7, 1.8$ Hz), 7.31 (d, 1H, $J = 8.7$ Hz), 5.98 (q, 1H, $J = 6.6$ Hz), 4.51-4.28 (m, 2H), 1.35 (t, 3H, $J = 7.0$ Hz). FABLRMS m/z 355 (M-H). ESHRMS m/z 355.0394 (M-H, Calc'd 355.0405).
Anal. Calc'd for $C_{14}H_{10}F_6O_4$: C, 47.21; H, 2.83.
Found: C, 47.31; H, 2.97.

10 Step 3. Preparation of ethyl 2,6-bis(trifluoromethyl)-4-trifluoromethanesulfonato-2H-1-benzopyran-3-carboxylate.

A 50 mL 3-neck Morton flask fitted with addition funnel, 2 stoppers was charged with 2.6-di-
15 tert-butylpyridine (1.576 g, 1.50 mmol), methylene chloride (12 mL), and then via syringe was added trifluoromethanesulfonic anhydride (1.08 mL, 1.80 g, 1.25 mmol). To this solution was added dropwise a solution the keto ester (Step 2) (1.822 g, 5.115
20 mmol) in methylene chloride (10 mL) over 0.33 h and the reaction stirred for 48 h. The resulting off-white suspension was transferred to a 100 mL round bottom flask and was concentrated in vacuo. The residue was suspended in diethyl ether (50 mL) and
25 vacuum filtered to remove salts. The filtrate was further diluted with diethyl ether (50 mL) and was washed with ice cold HCl solution (2 N), brine, and dried over Na_2CO_3 , filtered and concentrated in vacuo yielding the desired triflate (1.64 g, 66%) as a tan
30 clumpy powder of suitable purity to use in the next step without further purification.

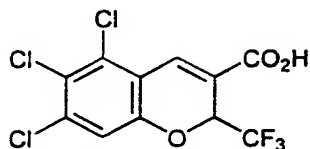
Step 4. Preparation of ethyl 2,6-
bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A 25 mL pear flask was charged with LiCl (0.136
5 g, 3.219 mmol), affixed to a high vacuum line and
heated with a heat gun removing superficial water.
The flask was allowed to cool to room temperature,
and tetrakis(triphenylphosphine)palladium(0) (0.124
g, 0.107 mmol) and THF (2 mL) were added. A reflux
10 condenser was affixed to the flask and the apparatus
was purged with nitrogen. A solution of the
triflate (Step 3) (0.524 g, 1.073 mmol) in THF (2 mL)
and tri-n-butyltin hydride (0.32 mL, 0.34 g, 1.18
mmol) were added sequentially via syringe. The
15 resulting light orange solution was heated to 50 °C
with stirring for 1 h, 60 °C for one hour, and 65 °C
for one hour. The reaction was allowed to cool to
room temperature and was poured into 2 N HCl,
stirred, and extracted with hexanes. The hexane
20 phase was dried over MgSO₄, filtered and
concentrated yielding a light brown oil. The oil
was dissolved in hexane and was washed with aqueous
ammonium fluoride solution. The resulting hexane
phase was dried over MgSO₄, filtered and
25 concentrated in vacuo yielding a dull yellow oily
solid which solidified as a flaky powder (0.443 g).
This solid was purified by flash silica
chromatography (eluant: hexanes-methylene chloride,
4:1) yielding ethyl 2,6-di-trifluoromethyl-2H-1-
30 benzopyran-3-carboxylate (0.069 g, 19 %) as a white

crystalline solid of suitable purity to proceed with the next step.

Step 5. Preparation of 2,6-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

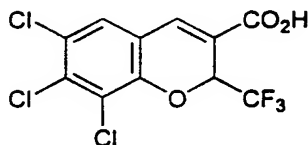
To a stirred solution of the ester (Step 4) (0.065 g, 0.191 mmol) in THF-EtOH-H₂O (7:2:1, 1 mL) was added NaOH solution (0.084 mL, 0.210 mmol) in one portion at room temperature and allowed to stir overnight. The reaction was partially concentrated *in vacuo* yielding a pale yellow clear syrup. The syrup was diluted with water (5 mL) and brine (1 mL) and was washed with diethyl ether (3 X 5 mL). The resulting aqueous phase was sparged with nitrogen to remove trace ether. With stirring, concentrated HCl was added to the aqueous phase causing the formation of a very fine white precipitate. This suspension was extracted with diethyl ether and the ether dried over Na₂SO₄, filtered, and concentrated by slow evaporation at atmospheric pressure. The resulting product was recrystallized from hexanes and ethyl acetate yielding the title compound (0.038 g, 64 %) as a fine tan powder: mp 143.5-145.2 °C. ¹H NMR (acetone-*d*₆/300 MHz) 11.97-11.67 (br s, 1H), 8.03 (s, 1H), 7.92 (s, 1H), 7.77 (d, 1H, *J* = 8.5 Hz), 7.26 (d, 1H, *J* = 8.7 Hz), 5.96 (q, 1H, *J* = 7.0 Hz). FABLRMS *m/z* 311 (M-H). ESHRMS *m/z* 311.0107 (M-H, Calc'd 311.0143).

EXAMPLE 83

5,6,7-Trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

5

3,4,5-Trichlorophenol was converted to 3-ethoxysalicylaldehyde via a procedure similar to that described in Example 11, Step 1. The 4,5,6-trichlorosalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 236.2-239.3 °C. ¹H NMR (acetone-*d*₆/300 MHz) 8.05 (s, 1H), 7.40 (s, 1H), 5.99 (q, 1H, *J* = 7.0 Hz). ESI/MS *m/z* 345 (M-H). ESI/MS *m/z* 344.9113 (M-H, Calc'd 344.9100). Anal. Calc'd for C₁₁H₄Cl₃F₃O₃ + 0.89 wt % H₂O: C, 37.68; H, 1.25; Cl, 30.33. Found: C, 37.48; H, 1.25; Cl, 30.33.

10
15**EXAMPLE 84**

20

6,7,8-Trichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

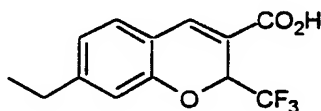
2,3,4-Trichlorophenol was converted to 3-ethoxysalicylaldehyde via a procedure similar to that described in Example 11, Step 1. The 3,4,5-

25

trichlorosalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 222.0-225.3 °C. ¹H NMR (acetone-d₆/300 MHz) 7.94 (s, 1H), 7.78 (s, 1H), 6.07 (q, 1H, J = 7.0 Hz). ESLRMS m/z 345 (M-H). EIHRMS m/z 344.9117 (M-H, Calc'd 344.9100). Anal. Calc'd for C₁₁H₄Cl₃F₃O₃ + 1.56 wt % H₂O: C, 37.43; H, 1.32; Cl, 30.13. Found: C, 37.79; H, 0.93; Cl, 29.55.

EXAMPLE 85

10

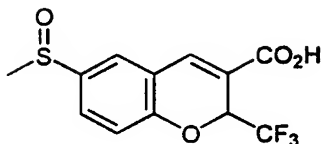


7-Ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

15

3-Ethylphenol was converted to the title compound by a procedure similar to that described in Example 2. : mp 167.0-168.6 °C. ¹H NMR (CDCl₃/300 MHz) 7.84 (s, 1H), 7.15 (d, 1H, J = 7.5 Hz), 6.84 (m, 2H), 5.66 (q, 1H, J = 6.8 Hz), 2.63 (q, 2H, J = 7.7 Hz, J = 7.7 Hz), 1.24 (t, 3H, J = 7.7 Hz). Anal. Calc'd for C₁₃H₁₁F₃O₃: C, 57.36; H, 4.07. Found: C, 57.25; H, 4.10.

25

EXAMPLE 86

5 **6-(Methylsulfinyl)-2-(trifluoromethyl)-2H-1-**
 benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl 6-
(methylsulfinyl)-2-(trifluoromethyl)-2H-1-
10 benzopyran-3-carboxylate.

Ethyl 6-(methylthio)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylate (Example 2, Step
2) (1.014 g, 3.18 mmol) in methylene chloride
was chilled to -50 °C (dry ice acetone). With
15 stirring, meta-chloroperbenzoic acid (0.91 g of
60 % reagent, 3.18 mmol) was added and reaction
allowed to proceed for 3 hours. Aqueous NaHSO₃
solution (40 mL 0.25 M) was poured into the
reaction. More methylene chloride was added
20 and the layers mixed, then separated. The
organic phase was washed with aqueous NaHSO₃
solution, aqueous saturated NaHCO₃ solution,
brine, dried over MgSO₄, filtered and
concentrated yielding an oil. The oil was
25 diluted with isooctane (2 mL) and concentrated
yielding an oil which upon standing
crystallized. Hexanes was added, the solution
was heated, and methylene chloride added until

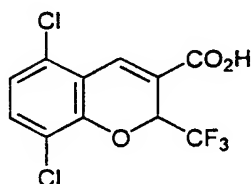
partial dissolution occurred. After cooling and standing overnight the suspension was vacuum filtered yielding the sulfoxide substituted ethyl ester (0.753 g, 71%) as white needles: mp 92.2-98.4 °C. This ester was of sufficient purity to be used without further purification.

10 Step 2. Preparation of 6-(methylsulfinyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

To a stirred solution of the ester (Step 1) (0.683 g, 2.043 mmol) in THF:EtOH:H₂O (7:2:1, 4 mL) was added aqueous NaOH solution (0.98 mL of 2.5 M, 2.45 mmol). After stirring 12 hours, the reaction was partially concentrated *in vacuo* to remove the organic solvents. The residue was diluted with H₂O, washed with diethyl ether, sparged with nitrogen to remove trace diethyl ether, and acidified with concentrated HCl yielding a oily suspension. The suspension was extracted with diethyl ether, and the resulting organic phase dried over MgSO₄, filtered, and diluted with hexanes. Upon concentration *in vacuo* the title acid was obtained as a sticky white powder (0.425 g, 68 %): mp 148.3-151.0 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.99 (s, 1H), 7.82 (s, 1H), 7.78-7.68 (m, 1H), 7.24 (d, 1H, *J* = 8.3 Hz), 5.92 (q, 1H, *J* = 7.1 Hz), 2.73 (s, 3H). FABLRMS *m/z* 307 (M+H). ESHRMS *m/z* 305.0098 (M-H, Calc'd 305.0095).

Anal. Calc'd for $C_{12}H_9F_3O_4S_1$: C, 47.06; H, 2.96; S, 10.47. Found: C, 46.69; H, 2.86; S, 10.45.

5

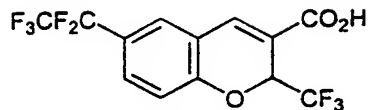
EXAMPLE 87

10

5,8-Dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

2,5-Dichlorophenol was converted to 3,6-dichlorosalicylaldehyde via a procedure similar to that described in Example 2, Step 1. The 3,6-dichlorosalicylaldehyde was converted to the title compound by a similar procedure to that described in Example 11, Steps 2 & 3: mp 205.7-207.1 °C. 1H NMR (acetone- d_6 /300 MHz) 8.02 (s, 1H), 7.53 (d, 1H, $J = 8.7$ Hz), 7.22 (d, 1H, $J = 8.7$ Hz), 6.04 (q, 1H, $J = 7.1$ Hz). FABLRMS m/z 311 (M-H). ESHRMS m/z 310.9506 (M-H, Calc'd 310.9490). Anal. Calc'd for $C_{11}H_5Cl_2F_3O_3 + 0.63$ wt % H_2O : C, 41.94; H, 1.67. Found: C, 41.54; H, 1.27.

25

EXAMPLE 88

5 **6-(Pentafluoroethyl)-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid**

Step 1. Preparation of ethyl 6-
(pentafluoroethyl)-2-(trifluoromethyl)-2H-1-
10 benzopyran-3-carboxylate.

Potassium pentafluoropropionate (0.476 g, 2.35 mmol) was dissolved in toluene (6 mL) and DMF (6 mL). The vessel was fitted with a distilling head, and CuI (0.471 g, 2.474 mmol)
15 was added with stirring. The reaction was heated to 120 °C, removing the toluene by distillation. Ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 72, Step 3) (0.469 g, 1.178 mmol) was
20 added and the reaction was heated to 150 °C for 2 hours. The reaction was allowed to cool to room temperature and was partitioned between diethyl ether and H₂O. The organic phase was dried over MgSO₄, filtered and concentrated in
25 vacuo. The resulting residue was purified by flash chromatography (silica gel 60, eluant: hexanes- ethyl acetate, 8:1) yielding, upon concentration of the solution, the desired ester (0.096 g, 21%) as a tan solid mass of

suitable purity to use without further

purification: ^1H NMR (acetone- d_6 /300 MHz) 8.04

(s, 1H), 7.91 (d, 1H, $J = 2.2$ Hz), 7.74 (dd,

1H, $J = 8.7, 2.2$ Hz), 6.00 (q, 1H, $J = 7.1$ Hz),

5 4.42-4.24 (m, 2H), 1.34 (t, 3H, $J = 7.3$ Hz).

Step 2. Preparation of 6-(pentafluoroethyl)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid.

10 To a stirred solution of the ethyl ester

(Step 1) (0.090 g, 0.231 mmol) in

THF:EtOH:H₂O (7:2:1) (4 mL) was added aqueous

NaOH solution (0.11 mL, 2.5 M). After stirring

16 hours, the reaction was partially

15 concentrated *in vacuo* to remove the organic

solvents, diluted with H₂O, and washed with

diethyl ether. The resulting aqueous phase

was acidified with concentrated HCl, extracted

with diethyl ether, dried over MgSO₄, filtered

20 and concentrated *in vacuo* yielding an oil. The

oil was purified by flash chromatography

(silica, hexanes-ethyl acetate, 3:1 with 5 %

acetic acid). This procedure yielded the title

acid (0.020 g, 24 %) as a white powder: mp

25 162.3-164.7 °C. ^1H NMR (acetone- d_6 /300 MHz)

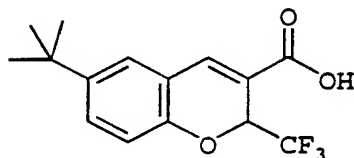
8.05 (s, 1H), 7.90 (s, 1H), 7.74 (d, 1H, $J =$

8.7 Hz), 7.29 (d, 1H, $J = 8.7$ Hz), 5.97 (q,

1H, $J = 6.8$ Hz). FABLRMS m/z 361 (M-H).

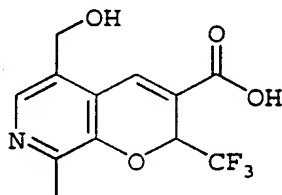
ESHRMS m/z 361.0111 (M-H, Calc'd 361.0094).

30

EXAMPLE 89

5 **6-(1,1-Dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

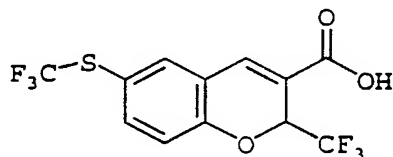
4-*tert*-Butylphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 170.6-173.2 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.89 (s, 1H), 7.5-7.4 (m, 2H), 6.93 (d, 1H, *J* = 8.4 Hz), 5.76 (q, 1H, *J* = 7.2 Hz), 1.3 (s, 9H). Anal. Calc'd for C₁₅H₁₅O₃F₃: C, 60.00; H, 5.04. Found: C, 59.93; H, 5.12.

EXAMPLE 90

20 **5-(Hydroxymethyl)-8-methyl-2-(trifluoromethyl)-2H-pyrano[2,3-c]pyridine-3-carboxylic acid**

3-Hydroxymethyl-5-methyl-4-formylpyridine was converted to the title compound by a procedure similar to that described in Example 1: mp 76.1-80.1 °C. ¹H NMR (acetone-*d*₆/300 MHz) 8.15 (s, 2H), 5.93 (q, 1H, *J* = 7.2 Hz), 1.3 (s, 9H) 5.30 (br s, 1H), 4.79 (br s, 1H), 2.41 (s, 3H). ESHRMS *m/z* 288.0485 (M+H, Calc'd 288.0483).

10

EXAMPLE 91

15

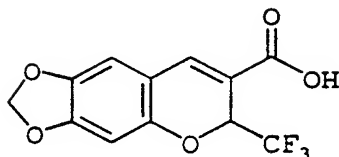
**2-(Trifluoromethyl)-6-[(trifluoromethyl)thio]-
2H-1-benzopyran-3-carboxylic acid**

4-(Trifluoromethoxy)phenol was converted to 5-(trifluoromethoxy)salicylaldehyde via a procedure similar to that described in Example 2, Step 1. The 5-(trifluoromethoxy)salicylaldehyde was converted to the title compound by a similar procedure to that described in Example 11, Steps 2 & 3: mp 139.1-143.2 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.95 (s, 1H), 7.88 (d, 2H, *J* = 2.4 Hz), 7.71-7.75 (m, 1H), 6.93 (d, 1H, *J* = 8.7 Hz), 5.91 (q, 1H, *J* = 6.9 Hz). Anal. Calc'd for

$C_{12}H_6O_3F_3S$: C, 41.87; H, 1.76. Found: C, 41.94;
H, 1.84.

EXAMPLE 92

5



6-(Trifluoromethyl)-6H-1,3-dioxolo[4,5-
g][1]benzopyran-7-carboxylic acid

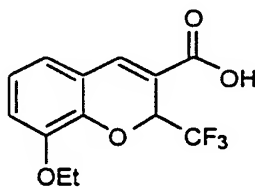
10

4-*tert*-Butylphenol was converted to the
title compound by a procedure similar to that
described in Example 2: mp 245.8-247.8 °C 1H NMR
(acetone- d_6 /300 MHz) 7.77 (s, 1H), 6.95 (s,
1H), 6.12 (s, 1H), 6.05 (d, 2H, J = 0.90 Hz),
5.91 (q, 1H, J = 7.2 Hz). Anal. Calc'd for
 $C_{12}H_7O_5F_3$: C, 50.01; H, 2.45. Found: C, 50.02;
H, 2.50.

15

20

EXAMPLE 93

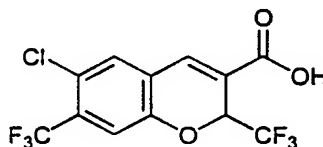


8-Ethoxy-2-trifluoromethyl-2H-1-benzopyran-3-
carboxylic acid

25

2-Ethoxyphenol was converted to 3-ethoxysalicylaldehyde via a procedure similar to that described in Example 11, Step 1. The
5 3-ethoxysalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 159.4-160.9 °C. ¹H NMR (acetone-d₆/300 MHz) 7.86 (s, 1H), 6.97-7.14 (m, 3H), 5.83 (q H-F, 1H, J = 7.2 Hz),
10 4.12 (q, 2H, J = 7.2 Hz), 1.38 (t, 3H, J = 7.2 Hz). FABHRMS m/z 289.0656 (M+H, Calc'd 289.0686). Anal. Calc'd for C₁₃H₁₁F₃O₄: C, 54.17; H, 3.85. Found: C, 54.06; H, 3.83.

15

EXAMPLE 94

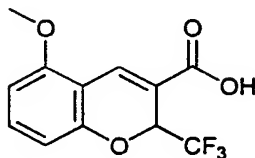
20

6-Chloro-2,7-bis(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

4-Chloro-3-(trifluoromethyl)phenol was converted to the title compound by a procedure similar to that described in Example 11: mp
25 180.9-182.4 °C. ¹H NMR (acetone-d₆/300 MHz) 7.96 (s, 1H), 7.84 (s, 1H), 7.47 (s, 1H), 5.96 (q, 1H, J = 6.8 Hz), 2.50 (s, 3H). FABLRMS m/z 345 (M-H). FABHRMS m/z 344.9767 (M-H, Calc'd

344.9753). Anal. Calc'd for $C_{12}H_5ClF_6O_3$: C, 41.58; H, 1.45; Cl, 10.23. Found: C, 41.57; H, 1.50; Cl, 10.33.

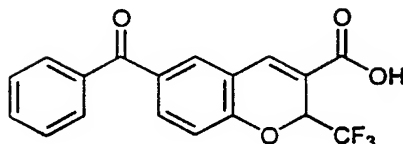
5

EXAMPLE 95

10 **5-Methoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

6-Methoxysalicylaldehyde was converted to the title compound by a similar procedure to that described in Example 11, Steps 2 & 3: mp
15 204.5-206.7 °C. 1H NMR (acetone- d_6 /300 MHz) 8.08 (s, 1H), 7.38 (dd, 1H, $J = 8.5$ Hz 8.3 Hz), 6.74 (d, 1H, $J = 8.5$ Hz), 6.65 (d, 1H, $J = 8.3$ Hz), 5.80 (q, 1H, $J = 7.2$ Hz), 3.94 (s, 3H). FABLRMS m/z 273 (M-H). EIHRMS m/z 274.0444
20 (M+, Calc'd 274.0453). Anal. Calc'd for $C_{12}H_9F_3O_4$: C, 52.57; H, 3.31. Found: C, 52.47; H, 3.34.

EXAMPLE 96



5 **6-Benzoyl-2-(trifluoromethyl)-2H-1-benzopyran-**
 3-carboxylic acid

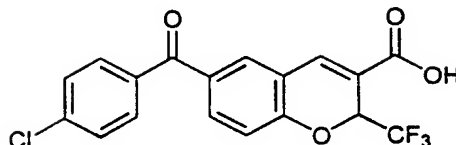
Step 1. Preparation of ethyl 6-benzoyl-2-
(trifluoromethyl)-2H-1-benzopyran-3-
10 carboxylate.

Ethyl 2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 10, Step 1) (1.59 g, 5.8 mmol) was dissolved in 1,2-dichloroethane (3 mL) and added to a 0 °C suspension of aluminum chloride (2.59 g, 19.4 mmol) in 1,2-
15 dichloroethane (3 mL). A solution of benzoyl chloride (1.01 g, 7.2 mmol) in 1,2-dichloroethane (3 mL) was added and the reaction was heated to 80 °C and stirred for 4
20 hours. The solution was poured onto 3 N HCl and ice and extracted with ethyl acetate. The ethyl acetate layers were combined, washed with 3N HCl, saturated sodium bicarbonate, brine, dried over MgSO₄ and concentrated *in vacuo*. The
25 crude ester was purified by flash chromatography over silica gel (with 1:9 ethyl acetate/hexane as eluant) to afford the ester as a white crystalline solid (0.26 g, 12%): mp

114.7-116.1 °C. ¹H NMR (CDCl₃/300MHz) 7.82 (dd, 1H, *J* = 8.5 Hz 2.0 Hz), 7.76 (m, 4H), 7.61 (m, 1H), 7.50 (m, 2H), 7.09 (d, 1H, *J* = 8.7 Hz), 5.79 (q, 1H, *J* = 6.8 Hz), 4.34 (m, 2H), 1.36 (t, 3H, *J* = 7.2 Hz).

Step 2. Preparation of 6-benzoyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

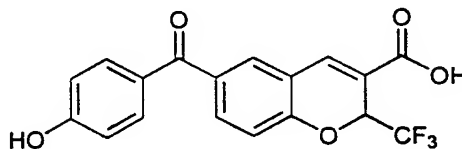
10 The ester from Step 1 (0.24 g, 0.64 mmol) was dissolved in THF (2 mL) and ethanol (2 mL), treated with 2.5 N sodium hydroxide (1.5 mL, 3.8 mmol), and stirred at room temperature for 4.3 hours. The reaction mixture was
15 concentrated *in vacuo*, acidified with 3N HCl yielding a solid. The solid was collected by filtration and was recrystallized from ethanol-water to yield a white solid (0.14 g, 64%): mp 269.8-270.8 °C. ¹H NMR (acetone-*d*₆/300 MHz)
20 8.04 (s, 1H), 7.99 (d, 1H, *J* = 2.0 Hz), 7.88 (dd, 1H, *J* = 8.5 Hz 2.0 Hz), 7.79 (m, 2H), 7.68 (m, 1H), 7.57 (m, 1H), 7.23 (d, 1H, *J* = 8.6 Hz), 5.98 (q, 1H, *J* = 7.0 Hz). FABLRMS *m/z* 347 (M-H). ESHRMS *m/z* 347.0560 (M-H, Calc'd
25 347.0531). Anal. Calc'd for C₁₈H₁₁F₃O₄: C, 62.08; H, 3.18. Found: C, 61.48; H, 3.22.

EXAMPLE 97

5 **6-(4-Chlorobenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared analogous to the procedure described in Example 96: mp 268.3-269.4 °C. ¹H NMR (acetone-d₆/300 MHz) 8.03 (s, 1H), 7.99 (d, 1H, J = 2.0 Hz), 7.89 (dd, 1H, J = 8.5 Hz, 2.0 Hz), 7.81 (d, 2H, J = 8.5 Hz), 7.62 (d, 2H, J = 8.5 Hz), 7.23 (d, 1H, J = 8.5 Hz), 5.98 (q, 1H, J = 7.1 Hz). FABLRMS m/z 381 (M-H). ESHRMS m/z 381.0135 (M-H, Calc'd 381.0141). Anal. Calc'd for C₁₈H₁₀ClF₃O₄: C, 56.49; H, 2.63; Cl, 9.26. Found: C, 56.35; H, 2.66; Cl, 9.34.

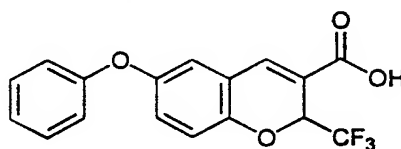
20

EXAMPLE 98

25 **6-(4-Hydroxybenzoyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared analogous to the procedure described in Example 96: mp 234.0-239.5 °C. ¹H NMR (acetone-*d*₆/300 MHz) 8.03 (s, 1H), 7.92 (d, 1H, *J* = 2.0 Hz), 7.83 (dd, 1H, *J* = 8.5 Hz 2.0 Hz), 7.74 (d, 2H, *J* = 8.7 Hz), 7.20 (d, 1H, *J* = 8.5 Hz), 7.00 (d, 1H, *J* = 8.7 Hz), 5.94 (q, 1H, *J* = 7.1 Hz). ESHRMS *m/z* 363.0471 (M-H, Calc'd 363.0480).

EXAMPLE 99

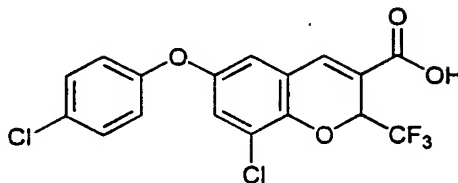


15
6-Phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

4-Phenoxyphenol was converted to 5-phenoxyphenylaldehyde by a similar procedure to that described in Example 2, Step 1. 5-Phenoxyphenylaldehyde was converted into the title compound by a similar procedure to that described in Example 11, Steps 2 & 3: mp 184.9-186.4 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.90 (s, 1H), 7.39 (m, 2H), 7.20 (d, 1H, *J* = 2.0 Hz), 7.08 (m, 3H), 7.02 (m, 2H), 5.98 (q, 1H, *J* = 7.2 Hz). FABLRMS *m/z* 335 (M-H). FABHRMS *m/z*

337.0663 (M+H, Calc'd 337.0687). Anal. Calc'd for $C_{17}H_{11}F_3O_4$: C, 60.72; H, 3.30. Found: C, 60.62; H, 3.29.

5

EXAMPLE 100

10 **8-Chloro-6-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

Step 1. Preparation of 5-phenoxybenzaldehyde.

Ethyl magnesium bromide (67.5 mL of an approximately 3.0 M solution in diethyl ether, 202.5 mmol) was added to toluene (50 mL). A solution of 4-phenoxyphenol (25.00 g, 134.26 mmol) in diethyl ether (35 mL) was added resulting in the evolution of gas. The reaction was heated to 80 °C causing distillation of the diethyl ether. Toluene (300 mL), HMPA (23.4 mL, 24.059 g, 134.26 mmol), and paraformaldehyde (10.07 g, 335.65 mmol) were added and the reaction was heated to 85 °C for 4 hours. The reaction was cooled to room temperature and was acidified with 2N HCl. The resulting layers were separated and the organic phase collected. The organic phase was

202

washed with brine. The combined aqueous phases were extracted with methylene chloride. The organic phases were combined, dried over MgSO_4 , filtered and concentrated *in vacuo* yielding a yellow oil. The oil was purified by silica flash chromatography (hexanes-ethyl acetate, 95:5). Concentration *in vacuo* of the desired fractions provided the salicylaldehyde as a pale yellow powder (12.0 g, 42%) of suitable purity to use in subsequent steps.

Step 2. Preparation of 3-chloro-5-(4-chlorophenoxy)salicylaldehyde.

To a stirred solution of the salicylaldehyde (Step 1) (0.981 g, 4.58 mmol) in acetic acid (20 mL) was added chlorine gas via a tube until the yellow color of chlorine persisted. After stirring for four hours at room temperature the reaction was sparged with nitrogen and diluted with water (50 mL). The resulting oily suspension was extracted with methylene chloride. The methylene chloride phase was washed with sodium bisulfite solution, dried over MgSO_4 , filtered and concentrated *in vacuo* providing the dichlorinated salicylaldehyde as a yellow oil (0.66 g, 51%) of suitable purity for use in subsequent steps without further purification.

Step 3. Preparation of ethyl 8-chloro-6-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A mixture of the dichlorinated
5 salicylaldehyde (Step 2) (0.66 g, 2.3 mmol),
triethylamine (0.49 g, 4.8 mmol), ethyl 4,4,4-
trifluorocrotonate (0.59 g, 3.5 mmol) in
dimethyl sulfoxide (5 mL) was heated to 85 °C
for 3.5 hours. The reaction was allowed to
10 cool to room temperature and was diluted with
ethyl acetate (50 mL). The resulting mixture
was washed with 3 N HCl (50 mL), aqueous
potassium carbonate solution (10 weight %, 2 X
30 mL), and brine. The organic phase was dried
15 over MgSO₄, filtered and concentrated *in vacuo*
yielding a brown oil. This oil was purified by
flash silica chromatography (hexanes-ethyl
acetate, 9:1) providing the substituted 2H-1-
benzopyran (0.39 g, 39%) of suitable purity to
20 use in subsequent steps without further
purification.

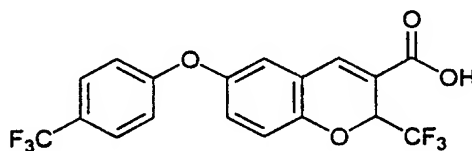
Step 4. Preparation of 8-chloro-6-(4-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

To a solution of the substituted 2H-1-
benzopyran ethyl ester (Step 3) (0.37 g, 0.85
mmol) in ethanol-THF (4 mL, 1:1) was added
sodium hydroxide solution (2 mL of 2.5 N, 5
30 mmol). After stirring for six hours the
mixture was concentrated *in vacuo*.

Acidification of the mixture with 3 N HCl yielded a solid which was collected by vacuum filtration. This solid was recrystallized from ethanol-water yielding the title compound as
5 yellow crystals (0.134 g, 38%): mp 227.8-228.9 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.93 (s, 1H), 7.42 (d, 2H, *J* = 8.9 Hz), 7.24 (s, 2H), 7.12 (d, 2H, *J* = 8.9 Hz), 5.97 (q, 1H, *J* = 7.1 Hz). FABLRMS *m/z* 403 (M-H). FABHRMS *m/z* 405.9790
10 (M+H, Calc'd 405.9801). Anal. Calc'd for C₁₇H₉Cl₂F₃O₄ + 2.33% H₂O: C, 49.22; H, 2.45. Found: C, 49.19; H, 2.27.

EXAMPLE 101

15

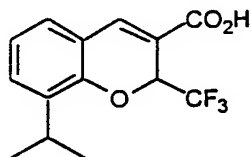


2-(Trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-
20 carboxylic acid

4-(4-Trifluoromethylphenyl)phenol was converted to 5-(4-trifluoromethylphenyl)salicylaldehyde via a
25 procedure similar to that described in Example 2, Step 1. The 5-(4-trifluoromethylphenyl)salicylaldehyde was converted to the title compound by a similar

procedure to that described in Example 11,
Steps 2 & 3: mp 153.5-154.4 °C. ¹H NMR
(acetone-d₆/300 MHz) 7.91 (s, 1H), 7.71 (d, 2H,
J = 8.9 Hz), 7.33 (s, 1H, *J* = 2.8 Hz), 7.15 (m,
5 4H), 5.86 (q, 1H, *J* = 7.1 Hz). FABLRMS *m/z* 403
(M-H). ESHRMS *m/z* 403.0399 (M-H, Calc'd
403.0405). Anal. Calc'd for C₁₈H₁₀F₆O₄: C, 53.48;
H, 2.49. Found: C, 53.52; H, 2.55.

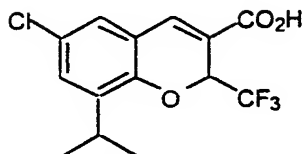
10

EXAMPLE 102

15

8-(1-Methylethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

4-(4-Methoxyphenyl)phenol was converted to
the title compound by a procedure similar to
that described in Example 2: mp 210.5-211.5 °C.
20 ¹H NMR (acetone-d₆/300 MHz) 7.86 (s, 1H), 7.35
(d, 1H, *J* = 7.7 Hz), 7.28 (s, 1H, *J* = 7.5 Hz),
7.04 (t, 1H, *J* = 7.7 Hz), 5.85 (q, 1H, *J* = 7.2
Hz), 3.33 (sept, 1H, *J* = 7.1 Hz), 1.25 (d, 6H,
J = 7.1 Hz). Anal. Calc'd for C₁₄H₁₃F₃O₃: C,
25 58.74; H, 4.58. Found: C, 58.65; H, 4.60.

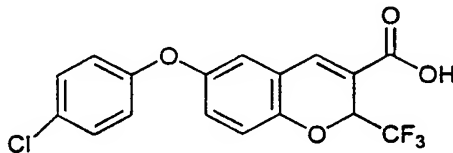
EXAMPLE 103

5 **6-Chloro-8-(1-methylethyl)-2-trifluoromethyl-
 2H-1-benzopyran-3-carboxylic acid**

8-(1-Methylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 6) was
10 converted to the title compound by a procedure similar to that described in Example 9. mp
185.4-189.2 °C. ¹H NMR (acetone-*d*₆/300 MHz)
7.87 (s, 1H), 7.38 (d, 1H, *J* = 2.4 Hz), 7.34
(d, 1H, *J* = 2.4 Hz), 5.90 (q, 1H, *J* = 7.3 Hz),
15 3.31 (m, 1H), 1.24 (d, 6H, *J* = 6.8 Hz). Anal.
Calc'd for C₁₅H₁₄ClF₃O₃: C, 52.43; H, 3.77; Cl,
11.05. Found: C, 52.58; H, 3.79; Cl, 10.96.

EXAMPLE 104

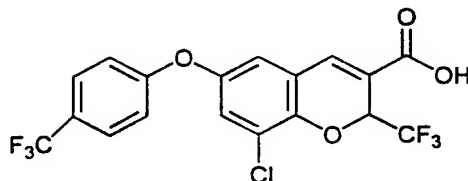
20



25 **6-(4-Chlorophenoxy)-2-(trifluoromethyl)-2H-1-
 benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared from 6-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 99) as the starting material by a procedure similar to that described in Example 9: mp 140.5-142.5 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.90 (s, 1H), 7.39 (d, 2H, *J* = 9.1 Hz), 7.25 (d, 1H, *J* = 2.6 Hz) 7.01-7.15 (m, 4H), 5.85 (q, 1H, *J* = 7.2 Hz). FABLRMS *m/z* 370 (M⁺). ESHRMS *m/z* 369.0130 (M-H, Calc'd 369.0141). Anal. Calc'd for C₁₇H₁₀ClF₃O₄ + 0.96% H₂O: C, 54.55; H, 2.80. Found: C, 54.38; H, 2.90.

EXAMPLE 105



8-Chloro-2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid

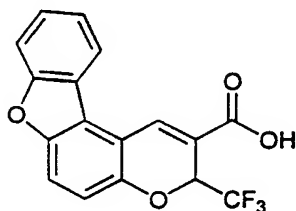
The benzopyran-3-carboxylic acid was prepared using 2-(trifluoromethyl)-6-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid (Example 101) as the starting material by a similar procedure to that described in Example 100: mp 223.7-226.0 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.94 (s, 1H), 7.74 (d, 2H, *J* = 8.5 Hz), 7.35 (m, 2H) 7.25 (d, 2H, *J* = 8.5 Hz), 6.00 (q, 1H, *J* = 7.0 Hz). FABLRMS *m/z* 437 (M-H).

ESHRMS m/z 437.0000 (M-H, Calc'd 437.0015).

Anal. Calc'd for $C_{18}H_9ClF_6O_4$: C, 49.28; H, 2.07;

Cl, 8.08. Found: C, 49.42; H, 2.12; Cl, 8.17.

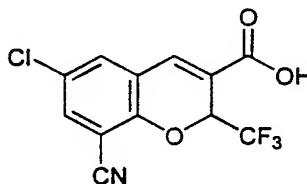
5

EXAMPLE 106

3-(Trifluoromethyl)-3H-benzofuro[3,2-f][1]benzopyran-2-carboxylic acid

10 2-Hydroxydibenzofuran was converted to the
title compound by a procedure similar to that
described in Example 2: mp 253.5-254.6 °C. 1H
NMR (acetone- d_6 /300 MHz) 8.54 (s, 1H), 8.23 (d,
1H, J = 7.5 Hz), 7.71 (s, 1H), 7.62 (m, 1H),
15 7.50 (m, 1H), 7.23 (d, 1H, J = 8.9 Hz), 5.95
(q, 1H, J = 7.3 Hz). FABLRMS m/z 333 (M-H).
ESHRMS m/z 333.0401 (M-H, Calc'd 333.0375).
Anal. Calc'd for $C_{17}H_9F_3O_4$: C, 61.09; H, 2.71.
Found: C, 60.95; H, 2.80.

20

EXAMPLE 107

6-Chloro-8-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl 6-chloro-8-(hydroxyiminomethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

5 Hydroxylamine hydrochloride (1.30 g, 18.7 mmol), sodium acetate (1.50 g, 19.4 mmol), and a mixture of ethanol-water (80:20, 15 mL) were stirred at room temperature for 0.4 hours. The aldehyde (Example 76, Step 3) (3.07 g, 9.0 mmol)
10 was dissolved in a solution of ethanol-water (4:1, 25 mL) and added to this mixture and stirred at 100 °C for 1 hour. The reaction was filtered hot and the filtrate allowed to cool to room temperature. An orange solid
15 crystallized in the filtrate which was collected by vacuum filtration. The solid was dissolved in ethyl acetate and the solution washed with water, brine, dried over MgSO₄, concentrated *in vacuo*. The resulting solid was
20 recrystallized from ethyl acetate-hexane yielding the oxime as a tan powder (1.50 g, 47%): mp 186.6-187.6 °C. ¹H NMR (acetone-*d*₆/300 MHz) 10.87 (s, 1H), 8.34 (s, 1H), 7.90 (s, 1H), 7.77 (d, 1H, *J* = 2.6 Hz), 7.60 (d, 1H, *J* = 2.6
25 Hz), 6.02 (q, 1H, *J* = 7.1 Hz), 4.35 (m, 2H), 1.34 (t, 3H, *J* = 7.0 Hz).

Step 2. Preparation of ethyl 6-chloro-8-cyano-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

30

The oxime from Step 1 (0.61 g, 1.7 mmol) and acetic anhydride (6 mL) were stirred at 140 °C for 6.3 hours. The reaction was poured into water, extracted with ethyl acetate, washed
5 with saturated NaHCO₃, brine, dried over MgSO₄, and concentrated *in vacuo* to give a brown oil (1.09 g). The oil was purified by flash chromatography (10:1; hexanes: ethyl acetate) yielding upon concentration the title compound
10 as a white solid (0.51 g, 88%): mp 114.6-115.6 °C. ¹H NMR (CDCl₃/300 MHz) 7.65 (s, 1H), 7.53 (d, 1H, *J* = 2.4 Hz), 7.44 (d, 1H, *J* = 2.4 Hz), 5.87 (q, 1H, *J* = 6.4 Hz), 4.36 (m, 2H), 1.37 (t, 3H, *J* = 6.5 Hz).

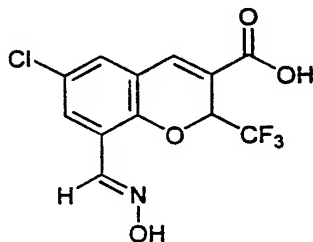
15

Step 3. Preparation of 6-chloro-8-cyano-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 2 (0.51 g 1.5 mmol)
20 was dissolved in THF (5 mL) and ethanol (5 mL), treated with 2.5N sodium hydroxide (1.2 mL, 3.0 mmol), and stirred at room temperature for 1.5 hours. The reaction mixture was concentrated *in vacuo*, acidified with 3N HCl, extracted with
25 ethyl acetate, washed with water, brine, dried over MgSO₄, concentrated *in vacuo*, and recrystallized from diethyl ether /hexane to give a white powder (0.10 g, 21%): mp 238.1-239.7 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.97 (s,
30 1H), 7.92 (d, 1H, *J* = 2.4 Hz), 7.89 (d, 1H, *J* = 2.4 Hz), 6.14 (q, 1H, *J* = 6.6 Hz). FABLRMS *m/z*

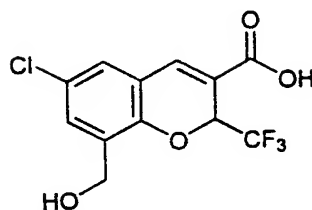
302 (M-H). ESHRMS m/z 301.9819 (M-H, Calc'd 301.9832). Anal. Calc'd for $C_{12}H_5ClF_3NO_3$: C, 47.47; H, 1.66; N, 4.61. Found: C, 47.41; H, 1.70; N, 4.55.

5

EXAMPLE 108

10 **6-Chloro-8-[(hydroxyimino)methyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared from the ethyl ester (Example 107, 15 Step 2) by a method similar to the procedure described in Example 1, Step 2: mp 246.9-247.9 °C. 1H NMR (acetone- d_6 /300 MHz) 10.90 (brs, 1H), 8.35 (s, 1H), 7.92 (s, 1H), 7.78 (d, 1H, J = 2.6 Hz), 7.61 (d, 1H, J = 2.6 Hz), 5.98 (q, 20 1H, J = 7.0 Hz). FABLRMS m/z 320 (M-H). ESHRMS m/z 319.9959 (M-H, Calc'd 319.9937). Anal. Calc'd for $C_{12}H_7ClF_3NO_4$: C, 44.81; H, 2.19; N, 4.35. Found: C, 44.92; H, 2.25; N, 4.26.

EXAMPLE 109

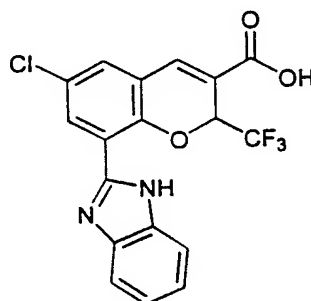
**6-Chloro-8-(hydroxymethyl)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

5

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 80 using the carboxylic acid (Example 76, step 4) as the starting

10 material: mp 174.6-178.9 °C. ¹H NMR (acetone-
d₆/300 MHz) 7.90 (s, 1H), 7.57 (d, 1H, J = 2.6
Hz), 7.47 (d, 1H, J = 2.6 Hz), 5.87 (q, 1H, J =
7.0 Hz), 4.70 (s, 2H). FABLRMS m/z 309 (M+H).
ESHRMS m/z 306.9981 (M-H, Calc'd 306.9985).

15 Anal. Calc'd for C₁₂H₈ClF₃O₃ (3.81 wt.% H₂O): C,
47.37; H, 3.08. Found: C, 47.33; H, 2.82.

EXAMPLE 110

20

**8-(1H-Benzimidazol-2-yl)-6-chloro-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic
acid**

213

Step 1. Preparation of ethyl 8-(1H-Benzimidazol-2-yl)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

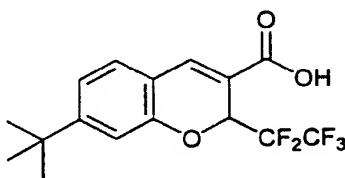
A solution of the aldehyde (Example 76, Step 3) (0.33 g, 0.99 mmol) and 1,2-phenylenediamine (0.11 g, 1.02 mmol) in nitrobenzene (20 mL) was heated to 150 °C for 1.8 hours. The reaction mixture was extracted with ethyl acetate, washed with brine, dried over MgSO₄, and concentrated *in vacuo* and purified by flash chromatography over silica gel (with 1:9 ethyl acetate/hexane as eluant) to give the ester as a brown solid (0.18 g, 43%) which was used in the next step without further purification.

Step 2. Preparation of 8-(1H-benzimidazol-2-yl)-6-chloro-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 1 (0.18 g 1.5 mmol) was dissolved in THF (5 mL) and ethanol (5 mL), treated with 2.5 N sodium hydroxide (2.6 mL, 6.5 mmol), and stirred at room temperature for 1.7 hours. The reaction mixture was concentrated *in vacuo*, acidified with 3 N HCl, filtered and recrystallized from ethanol-water to give a tan solid (0.09 g, 52%): mp >300 °C. ¹H NMR (acetone-*d*₆/300MHz) 8.59 (d, 1H, *J* = 2.6

Hz), 8.03 (s, 1H), 7.73 (d, 1H, $J = 2.6$ Hz),
 7.67 (brs, 2H), 7.28 (m, 2H), 6.13 (q, 1H, $J =$
 6.8 Hz). FABLRMS m/z 395 (M-H(^{37}Cl)). ESHRMS
 m/z 393.0262 (M-H, Calc'd 393.0254). Anal.
 5 Calc'd for $\text{C}_{18}\text{H}_{10}\text{ClF}_3\text{N}_2\text{O}_3$ (2.88 wt % H_2O): C,
 53.19; H, 2.80; N, 6.89. Found: C, 53.22; H,
 2.90; N, 6.80.

EXAMPLE 111



10

7-(1,1-Dimethylethyl)-2-(pentafluoroethyl)-2H- 1-benzopyran-3-carboxylic acid

15 Step 1. Preparation of ethyl 3-hydroxy- 4,4,5,5,5-pentafluoropentanoate.

A solution of ethyl 4,4,5,5,5-pentafluoro-
 3-oxo-pentanoate (41.32 g, 0.18 mole) in
 diethyl ether (70 mL) was cooled to 0 °C and
 20 treated with NaBH_4 (7.09 g, 0.19 mole). The
 reaction was allowed to warm to room
 temperature and stirred for 2 hours before
 quenching with 1 N HCl (200 mL). The layers
 were separated and the aqueous layer was
 25 extracted with diethyl ether. The combined
 organic layers were washed with 1 N HCl, brine,
 dried over MgSO_4 , and concentrated in vacuo to
 give the hydroxy ester as a clear oil (46.40 g)

which was used in the next step without further purification.

5 Step 2. Preparation of ethyl 4,4,5,5,5-pentafluoro-2-pentenoate.

The hydroxy ester from Step 1 (46.40 g, 0.18 mole) was stirred at 120 °C with P₂O₅ (25.59 g, 0.09 mole) for 2.6 hours then vacuum distilled (95 torr, 45-64 °C) to give the ester
10 as a clear oil (13.70 g, 35%): ¹H NMR (CDCl₃/300 MHz) 6.78 (m, 1H), 6.57 (dt, 1H, J = 15.9 Hz 2.0 Hz), 4.30 (q, 2H, J = 7.3 Hz), 1.34 (t, 3H, J = 7.1 Hz).

15 Step 3. Preparation of ethyl 7-(1,1-Dimethylethyl)-2-(pentafluoroethyl)-2H-1-benzopyran-3-carboxylate.

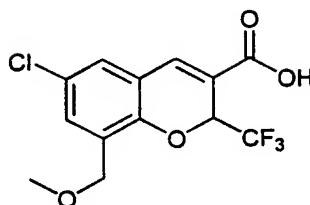
A mixture of 4-tert-butylsalicylaldehyde Example 8, step 1 (1.15 g, 6.4 mmol) and the
20 ethyl ester from Step 2 (1.59 g, 7.3 mmol) was dissolved in anhydrous DMF (4 mL). With stirring, K₂CO₃ (1.10 g, 9.0 mmol) was added causing the reaction to become deep red. The reaction was stirred at room temperature for
25 100 hours, acidified with 3 N HCl, diluted with ethyl acetate and washed with saturated NaHCO₃ solution, brine, dried over MgSO₄, filtered and concentrated in vacuo yielding a brown oil. This oil was purified by flash chromatography
30 over silica gel, eluting with 10% ethyl acetate/hexanes to afford a yellow oil (1.72 g,

70%): ^1H NMR (CDCl_3 /300 MHz) 7.76 (s, 1H), 7.14 (d, 1H, $J = 8.1$ Hz), 7.04 (dd, 1H, $J = 8.1$ Hz 1.8 Hz), 6.94 (s, 1H), 5.92 (dd, 1H, $J = 22.4$ Hz 3.0 Hz), 4.32 (m, 2H), 1.35 (t, 3H, $J = 7.2$ Hz), 1.30 (s, 9H).

Step 4. Preparation of 7-(1,1-Dimethylethyl)-2-(pentafluoroethyl)-2H-1-benzopyran-3-carboxylic acid.

10 The ester from Step 3 (1.58 g 4.20 mmol) was dissolved in THF (3 mL) and ethanol (3 mL), treated with 2.5 N sodium hydroxide (2 mL, 5 mmol), and stirred at room temperature for 23.3 hours. The reaction mixture was concentrated in
15 *vacuo*, acidified with 3 N HCl yielding a suspension. The solid was collected by filtration and was recrystallized from ethanol-water to yield a yellow solid (0.76 g, 52%): mp 171.0-173.5 °C. ^1H NMR (acetone- d_6 /300 MHz)
20 7.93 (s, 1H), 7.39 (d, 1H, $J = 8.1$ Hz), 7.18 (dd, 1H, $J = 8.1$ Hz 1.8 Hz), 7.02 (s, 1H), 6.01 (dd, 1H, $J = 23.1$ Hz 3.2 Hz), 1.32 (s, 9 H).
FABLRMS m/z 351 (M+H). EIHRMS m/z 350.0945 (M+, Calc'd 350.0941). Anal. Calc'd for $\text{C}_{16}\text{H}_{15}\text{F}_5\text{O}_3$: C, 54.86; H, 4.32. Found: C, 54.88; H, 4.32.
25

EXAMPLE 112



6-Chloro-8-(methoxymethyl)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl 6-chloro-8-
(hydroxymethyl)-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylate.

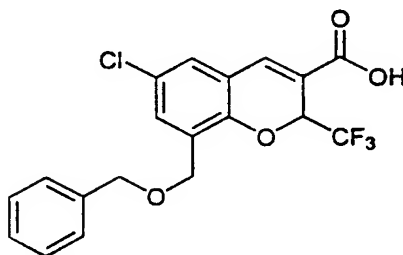
A suspension of the aldehyde (Example 75,
Step 1) (4.78 g, 14.3 mmol) was cooled to 0 °C
and treated with NaBH₄ (0.33 g, 4.8 mmol). The
solution was stirred for 10 minutes then
quenched with 3N HCl, extracted with ethyl
acetate, washed with saturated NaHCO₃, brine,
dried over MgSO₄, and concentrated *in vacuo* to
give a brown solid which was filtered through a
plug of silica gel to give the alcohol as a
brown solid (3.60 g, 75%). ¹H NMR (CDCl₃/300
MHz) 7.66 (s, 1H), 7.41 (d, 1H, *J* = 2.4 Hz),
7.17 (d, 1H, *J* = 2.4 Hz), 5.75 (q, 1H, *J* = 6.8
Hz), 4.71 (s, 2H), 4.33 (m, 2H), 1.85 (brs,
1H), 1.36 (t, 3H, *J* = 7.1). This solid was
used in the next step without further
purification.

Step 2. Preparation of ethyl 6-chloro-8-(methoxymethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

The alcohol from Step 1 (0.44 g, 1.3 mmol), silver triflate (0.36 g, 1.4 mmol) and 2,6-di-tert-butylpyridine (0.37 g, 1.9 mmol) were dissolved in methylene chloride (3 mL) cooled to 0 °C and treated with methyl iodide (0.40 g, 2.8 mmol). The reaction was allowed to warm and stirred at room temperature for 4.6 hours. The reaction was filtered through diatomaceous earth and the filtrate was washed with 3N HCl, saturated NaHCO₃, brine, dried over MgSO₄, and concentrated in vacuo yielding a brown oil. This oil was purified by flash chromatography over silica gel, eluting with 10% ethyl acetate-hexanes to afford the substituted 2H-1-benzopyran (0.19 g, 41%) as a white oily solid suitable

ESHRMS m/z 321.0141 (M-H, Calc'd 321.0141). Anal. Calc'd for $C_{13}H_{10}ClF_3O_4$: C, 48.39; H, 3.12. Found: C, 48.45; H, 3.11.

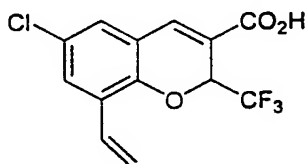
5

EXAMPLE 113

**6-Chloro-8-(benzyloxymethyl)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

10 The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 112: mp 133.8-135.4 °C. 1H NMR (acetone- d_6 /300 MHz) 7.90 (s, 1H), 7.54 (d, 1H, J = 2.6), 7.51 (d, 1H, J = 2.4 Hz), 7.42 (m, 5H), 5.91
15 (q, 1H, J = 7.1 Hz), 4.68 (s, 2H), 4.63 (s, 2H). FABLRMS m/z 399 (M+H). ESHRMS m/z 397.0454 (M-H, Calc'd 397.0461). Anal. Calc'd for $C_{19}H_{13}ClF_3O_4$: C, 57.23; H, 3.54; Cl, 8.89. Found: C, 57.34; H, 3.63; Cl, 8.77.

20

EXAMPLE 114

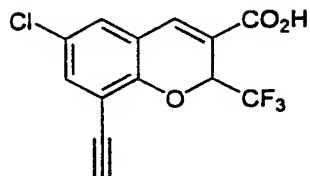
**6-Chloro-8-ethenyl-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid**

Step 1. Preparation of ethenyl-6-chloro-8-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

In a 100 mL round bottomed flask under N₂,
5 ethyl 8-bromo-6-chloro-2-trifluoromethyl-2H-benzopyran-3-carboxylate (Example 74, Step 1) (2.21 g, 5.73 mmol) was dissolved in toluene (30 mL of anhydrous reagent).
Tetrakis(triphenylphosphine)palladium(0) (0.132 g,
10 0.115 mmol) was added, followed by tributylethylenylstannane (2.0 g, 6.31 mmol). The resulting solution was heated to reflux for 5 hours. The reaction mixture was allowed to cool to room temperature, was poured into 50 mL of 20%
15 ammonium fluoride solution and stirred for one hour. Diethyl ether (100 mL) was added and the mixture was washed with water (2 x 50 mL). The organic phase was dried over MgSO₄, filtered, and evaporated to yield a yellow oil. The crude
20 material was purified by flash chromatography (0.5% ethyl acetate in hexanes) to afford the ester as a yellow solid (0.86 g, 45%): mp 75.9-77.2 °C. ¹H NMR (CDCl₃/300 MHz) 7.64 (s, 1H), 7.45 (d, 1H, J = 2.5 Hz), 7.12 (d, 1H, J = 2.6 Hz), 6.92 (dd, 1H, J = 17.7 Hz, 11.3 Hz), 5.81 (d, 1H, J = 17.7 Hz),
25 5.76 (q, 1H, J = 6.8 Hz), 5.41 (d, 2H, J = 11.1 Hz), 4.36-4.29 (m, 2H), 1.36 (t, 3H, J = 7.3 Hz). FABLRMS m/z 350.1 (M+NH₄⁺). ESHRMS m/z 350.0796 (M+NH₄⁺, Calc'd. 350.0771). Anal. Calc'd. for
30 C₁₅H₁₂ClF₃O₃ + 4.07% H₂O: C, 51.95; H, 3.94. Found: C, 51.67; H, 3.69.

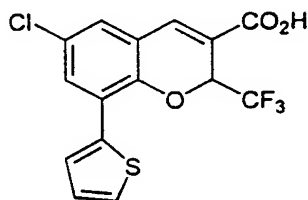
Step 2. Preparation of 6-chloro-8-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

5 The ester (Step 1) (0.350 g, 1.05 mmol) was dissolved in a solution of THF:ethanol:water(7:2:1; 10 mL), was treated with sodium hydroxide (0.46 mL, 1.05 mmol of a 2.5 N solution), and stirred at room temperature for 18 hours. The solvent was removed
10 in vacuo and the residue was dissolved in water (10 mL). Diethyl ether (10 mL) was added and the mixture acidified with concentrated HCl. The layers were separated, and the aqueous phase was extracted with diethyl ether (2 x 10 mL). The
15 organic phases were combined, dried over MgSO₄, filtered, and evaporated to yield a yellow solid, which was recrystallized in diethyl ether-hexane to afford the title compound as a yellow solid (0.288 g, 90%): mp 183.2-185.8 °C. ¹H NMR (CDCl₃/300 MHz)
20 7.77 (s, 1H), 7.49 (d, 1H, J = 2.2 Hz), 7.16 (d, 1H, J = 2.4 Hz), 6.93 (dd, 1H, J = 11.3, 17.7 Hz), 5.82 (d, 1H, J = 17.7 Hz), 5.74 (q, 1H, J = 6.9 Hz), 5.43 (d, 1H, J = 11.1 Hz). FABLRMS m/z 303 (M-H). ESHRMS m/z 303.0014 (M-H, Calc'd.
25 303.003582). Anal. Calc'd. for C₁₃H₈ClF₃O₃ + 1.58 % H₂O: C, 50.44; H, 2.78. Found: C, 50.42; H, 2.65.

EXAMPLE 115

5 **6-Chloro-8-ethynyl-2-(trifluoromethyl)-2H-1-**
 benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method
10 described in Example 114: mp 186.2-189.0 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.87 (s, 1H), 7.60 (d, 1H, *J* = 2.4 Hz), 7.51 (d, 1H, *J* = 2.4 Hz), 5.95 (q, 1H, *J* = 7.0 Hz), 4.02 (s, 1H). FABLRMS *m/z* 301 (M-H). ESHRMS *m/z* 300.9875 (M-H, Calc'd 300.9879). Anal.
15 Calc'd. for C₁₃H₆ClF₃O₃: C, 51.59; H, 2.00; Cl, 11.71. Found: C, 51.26; H, 2.06; Cl, 11.40.

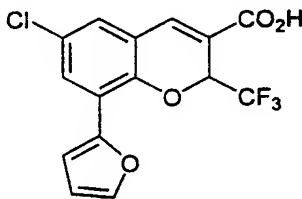
EXAMPLE 116

20

6-Chloro-8-(2-thienyl)-2-(trifluoromethyl)-2H-1-
 benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 114: mp 257.5-258.8 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.91 (s, 1H), 7.79 (d, 1H, *J* = 2.4 Hz), 7.74-7.72 (m, 1H), 7.62-7.61 (m, 1H), 7.51 (d, 1H, *J* = 2.4 Hz), 7.19-7.16 (m, 1H), 6.04 (q, 1H, *J* = 7.1 Hz). FABLRMS *m/z* 359 (M-H). ESHRMS *m/z* 358.9747 (M-H, Calc'd. 358.9756). Anal. Calc'd. for C₁₅H₈ClF₃O₃S: C, 49.94; H, 2.24; Cl, 9.83; S, 8.89. Found: C, 50.26; H, 2.45; Cl, 9.72; S, 9.00.

EXAMPLE 117

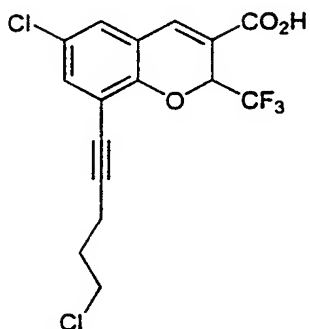


6-Chloro-8-(2-furanyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 114: mp 171.5-173.3 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.93 (s, 1H), 7.82 (d, 1H, *J* = 2.6 Hz), 7.72-7.71 (m, 1H), 7.50 (d, 1H, *J* = 2.6 Hz), 7.16 (d, 1H, *J* = 2.4 Hz), 6.65-6.63 (m, 1H), 6.11 (q, 1H, *J* = 7.1 Hz). FABLRMS *m/z* 343 (M-H). ESHRMS *m/z* 342.9995 (M-H, Calc'd. 342.9985). Anal.

Calc'd. for $C_{15}H_8ClF_3O_4 + 1.31 \% H_2O$: C, 51.59; H, 2.46; Cl, 10.15. Found: C, 51.57; H, 2.33; Cl, 10.14.

5

EXAMPLE 118

10 **6-Chloro-8-(5-chloro-1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

Step 1. Preparation of ethyl 6-chloro-8-(5-chloro-1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

15 Ethyl 6-chloro-8-iodo-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (Example 73, Step 2) (1.50 g, 3.47 mmol), tetrakis-(triphenylphosphine)palladium(0) (0.2 g, 0.174 mmol), copper(I)iodide (0.066 g, 0.347 mmol), and
20 triethylamine (1.05 g, 10.4 mmol) were dissolved in toluene (50 mL). 5-Chloro-1-pentyne (0.53 g, 5.20 mmol) was added via syringe and the mixture stirred for 18 hours at room temperature. The reaction was diluted with diethyl ether (50 mL), extracted with
25 0.5 N HCl (2 x 25 mL), and water (2 x 25 mL). The

organic phase was dried over MgSO_4 , filtered, and evaporated to yield an orange oil. The crude material was purified by flash chromatography in 2% ethyl acetate in hexane. Recrystallization from
5 hexane afforded the ester as a white solid (0.96 g, 68%): mp 84.8–85.9 °C. ^1H NMR (CDCl_3 /300 MHz) 7.61 (s, 1H), 7.33 (d, 1H, $J = 2.6$ Hz), 7.14 (d, 1H, $J = 2.6$ Hz), 5.79 (q, 1H, $J = 6.7$ Hz), 4.37–4.29 (m, 2H), 3.75 (t, 2H, $J = 6.7$ Hz), 2.67 (t, 2H, $J = 6.7$
10 Hz), 2.11–2.03 (m, 2H), 1.35 (t, 3H, $J = 7.2$ Hz). FABLRMS m/z 424.1 ($\text{M}+\text{NH}_4^+$). ESHRMS m/z 424.0694 ($\text{M}+\text{NH}_4^+$, Calc'd. 424.0694). Anal. Calc'd. for $\text{C}_{19}\text{H}_{15}\text{Cl}_2\text{F}_3\text{O}_3$: C, 53.09; H, 3.71; Cl, 17.41. Found: C, 53.02; H, 3.90; Cl, 17.63.

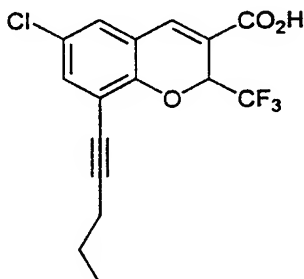
15

Step 2. Preparation of 6-chloro-8-(5-chloro-1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester (Step 1) (0.500 g, 1.23 mmol) was
20 dissolved in THF-ethanol-water(7:2:1; 10 mL). It was treated with sodium hydroxide (0.49 mL, 1.23 mmol of a 2.5 N solution), and stirred at room temperature for 18 hours. The solvent was
evaporated and the residue was dissolved in water
25 (10 mL). Diethyl ether (10 mL) was added and the mixture acidified with concentrated HCl. The organic layer was separated, and the aqueous phase was extracted with diethyl ether (2 x 10 mL). The combined extracts were dried over MgSO_4 , filtered,
30 and evaporated to yield a yellow solid, which was recrystallized in diethyl ether-hexane to afford

the title compound as a yellow solid (0.371 g, 80%): mp 154.4-156.4 °C. ¹H NMR (acetone-d₆/300 MHz) 7.88 (s, 1H), 7.53 (d, 1H, *J* = 2.4 Hz), 7.44 (d, 1H, *J* = 2.4 Hz), 5.94 (q, 1H, *J* = 7.1 Hz), 3.83 (t, 2H, *J* = 6.5 Hz), 2.68 (t, 2H, *J* = 6.8 Hz), 2.12-2.04 (m, 2H). ESLRMS *m/z* 377 (M-H). ESHRMS *m/z* 376.9930 (M-H, Calc'd. 376.9959). Anal. Calc'd. for C₁₆H₁₁Cl₂F₃O₃ + 1.18 % H₂O: C, 50.08; H, 3.02; Cl, 18.48. Found: C, 50.11; H, 2.73; Cl, 18.28.

EXAMPLE 119



15

6-Chloro-8-(1-pentynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

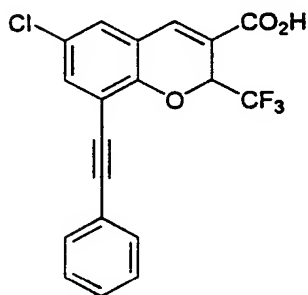
The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 118: mp 168.1-171.2 °C. ¹H NMR (CDCl₃/300 MHz) 7.75 (s, 1H), 7.37 (d, 1H, *J* = 2.6 Hz), 7.15 (d, 1H, *J* = 2.4 Hz), 5.77 (t, 1H, *J* = 6.7 Hz), 2.44 (t, 2H, *J* = 6.9 Hz), 1.68-1.61 (m, 2H), 1.07 (t, 3H, *J* = 7.25 Hz). FABLRMS *m/z* 345 (M+H). ESHRMS *m/z* 343.0373 (M-H, Calc'd.

25

343.0349). Anal. Calc'd. for $C_{16}H_{12}ClF_3O_3 + 0.69\%$
 H_2O : C, 55.36; H, 3.56. Found: C, 55.21; H, 3.62.

EXAMPLE 120

5

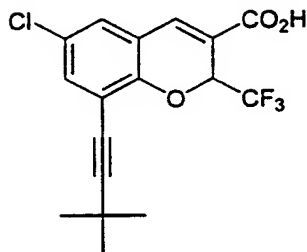


6-Chloro-8-(phenylethynyl)-2-(trifluoromethyl)-2H- 1-benzopyran-3-carboxylic acid

10

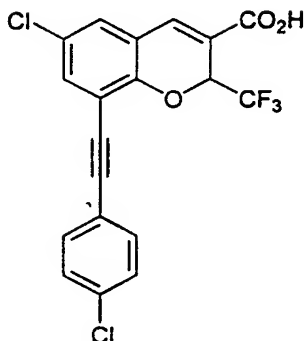
The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 118: mp 190.1-192.1 °C. 1H NMR ($CDCl_3$ /300 MHz) 7.92 (s, 1H), 7.61-7.57 (m, 4H), 7.47-7.44 (m, 3H), 6.01 (q, 1H, $J = 7.0$ Hz).
ESLRMS m/z 377 (M-H). ESHRMS m/z 377.0167 (M-H, Calc'd. 377.0192). Anal. Calc'd. for $C_{19}H_{10}ClF_3O_3$: C, 60.26; H, 2.66; Cl, 9.36. Found: C, 60.09; H, 2.73; Cl, 9.09.

20

EXAMPLE 121

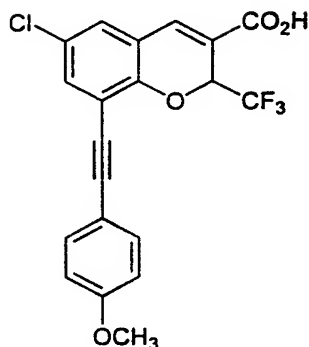
5 **6-Chloro-8-(3,3-dimethyl-1-butynyl)-2-**
 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method
10 described in Example 118: mp 218.3-222.4 °C. ¹H NMR (acetone-d₆/300 MHz) 7.87 (s, 1H), 7.51 (d, 1H, J = 2.4 Hz), 7.38 (d, 1H, J = 2.6 Hz), 5.92 (q, 1H, J = 6.9 Hz), 1.32 (s, 9H). FABLRMS m/z 359 (M+H). ESHRMS m/z 357.0490 (M-H, Calc'd. 357.0505). Anal.
15 Calc'd. for C₁₇H₁₄ClF₃O₃: C, 56.92; H, 3.93; Cl, 9.88. Found: C, 56.63; H, 3.94; Cl, 10.03.

EXAMPLE 122

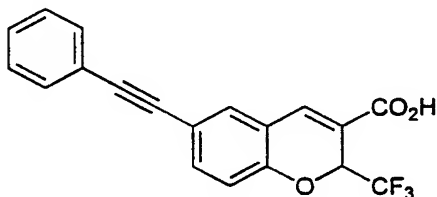
5 **6-Chloro-8-[(4-chlorophenyl)ethynyl]-2-**
 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method
10 described in Example 118: mp 210.4-211.4 °C. ¹H NMR (CDCl₃/300 MHz) 7.75 (s, 1H), 7.48-7.43 (m, 3H), 7.36 (s, 1H), 7.33 (s, 1H), 7.22 (d, 1H, J = 2.6 Hz), 5.82 (q, 1H, J = 6.6 Hz). FABLRMS m/z 411 (M-H). ESHRMS m/z 410.9802 (M-H, Calc'd. 410.980259).
15 Anal. Calc'd. for C₂₀H₁₂Cl₂F₃O₃: C, 55.23; H, 2.20; Cl, 17.16. Found: C, 55.22; H, 2.07; Cl, 17.39.

EXAMPLE 123

5 **6-Chloro-8-[(4-methoxyphenyl)ethynyl]-2-**
 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

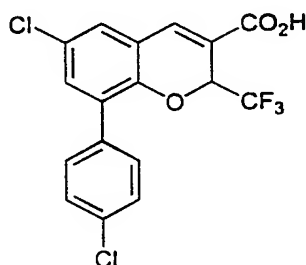
The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method
10 described in Example 118: mp 217.7-218.7 °C. ¹H NMR (CDCl₃/300 MHz) 7.75 (s, 1H), 7.51-7.47 (m, 3H), 7.18 (d, 1H, J = 2.4 Hz), 6.91-6.88 (m, 2H), 5.82 (1, 1H, J = 6.7 Hz). ESI/MS m/z 407 (M-H). ESI/MS m/z 407.0293 (M-H, Calc'd 407.0298). Anal. Calc'd
15 for C₂₀H₁₂ClF₃O₄: C, 58.77; H, 2.96; Cl, 8.67.
Found: C, 58.68; H, 2.85; Cl, 9.15.

EXAMPLE 124

20 **6-(Phenylethynyl)-2-(trifluoromethyl)-2H-1-**
 benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 118 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 240.1-241.3 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.94 (s, 1H), 7.70-7.69 (m, 1H), 7.61-7.53 (m, 3H), 7.44-7.41 (m, 3H), 7.10 (d, 1H, *J* = 7.1 Hz). ESHRMS *m/z* 343.0550 (M-H, Calc'd. 343.0582). Anal. Calc'd. for C₁₉H₁₁F₃O₃: C, 66.29; H, 3.22. Found: C, 66.26; H, 3.29.

EXAMPLE 125



6-Chloro-8-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

20

Step 1. Preparation of ethyl 6-chloro-8-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

Ethyl 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 73, Step 2) (1.3 g, 3.02 mmol), potassium carbonate (1.25 g, 9.06

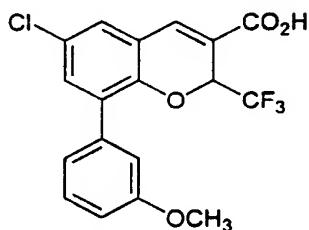
mmol), 4-chlorophenylboronic acid (0.52 g, 3.33 mmol), and tetrakis(triphenylphosphine)palladium(0) (0.174 g, 0.151 mmol) were added to toluene (30 mL) and the resulting solution was heated to reflux for 5 18 hours. After cooling to room temperature the reaction mixture was poured into ethyl acetate (50 mL). It was washed with 1 N HCl (2 x 25 mL), saturated aqueous sodium bicarbonate (2 x 25 mL), and water (2 x 25 mL). The organic phase was dried 10 over MgSO_4 , filtered, and concentrated *in vacuo* to yield a brown oil. The crude material was purified by flash chromatography using 1% ethyl acetate in hexane yielding a white solid. Recrystallization from hexane afforded the ester as a white solid 15 (0.79 g, 64%): mp 114.2-115.9 °C. ^1H NMR (CDCl_3 /300 MHz) 7.69 (s, 1H), 7.41 (s, 4H), 7.30 (d, 1H, J = 2.4 Hz), 7.22 (d, 1H, J = 2.6 Hz), 5.70 (q, 1H, J = 6.9 Hz), 4.37-4.29 (m, 2H), 1.35 (t, 3H, J = 7.1 Hz). ES/LRMS m/z 434 ($\text{M}+\text{NH}_4^+$). FAB/HRMS m/z 434.0574 20 ($\text{M}+\text{NH}_4^+$, Calc'd. 434.0538). Anal. Calc'd. for $\text{C}_{13}\text{H}_{13}\text{Cl}_2\text{F}_3\text{O}_3$: C, 54.70; H, 3.14; Cl, 17.00. Found: C, 54.79; H, 3.18; Cl, 16.65.

25 Step 2. Preparation of 6-chloro-8-(4-chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 1 (0.500 g, 1.20 mmol) was dissolved in a solution of THF:ethanol:water (7:2:1; 10 mL), treated with sodium hydroxide (0.48 30 mL, 1.20 mmol of a 2.5 N solution), and stirred at room temperature for 18 hours. The solvent was

removed *in vacuo* and the residue was dissolved in water (10 mL). Diethyl ether (10 mL) was added and the mixture acidified with concentrated HCl. The organic layer was separated, and the aqueous phase
5 was extracted with diethyl ether (2 x 10 mL). The combined extracts were dried over MgSO₄, filtered, and evaporated to yield a white solid, which was recrystallized in diethyl ether-hexane to afford the title compound as a white solid (0.40 g, 86%):
10 mp 205.5-207.3 °C. ¹H NMR (CDCl₃/300 MHz) 7.81 (s, 1H), 7.42 (s, 4H), 7.34 (d, 1H, *J* = 2.4 Hz), 7.25 (s, 1H), 5.69 (q, 1H, *J* = 6.8 Hz). FABLRMS *m/z* 387 (M-H). ESHRMS *m/z* 386.9788 (M-H, Calc'd. 386.980259). Anal. Calc'd. for C₁₇H₉Cl₂F₃O₃: C, 52.47; H, 2.33; Cl, 18.22. Found: C, 52.38; H, 2.47; Cl, 18.20.

EXAMPLE 126



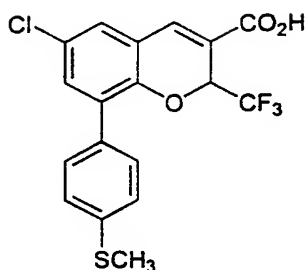
**6-Chloro-8-(3-methoxyphenyl)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

25 Step 1. Preparation of ethyl 6-chloro-8-(3-methoxyphenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

In a 100 mL round bottomed flask under nitrogen, ethyl 6-chloro-8-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 73, Step 2) (1.00 g, 2.31 mmol) and 3-methoxyphenylboronic acid (0.369 g, 2.43 mmol) were dissolved in 1-propanol (50 mL). The mixture was stirred at room temperature for 0.5 hours, allowing for the solids to dissolve. The resulting solution was treated with palladium (II) acetate (0.016 g, 0.0693 mmol), triphenylphosphine (0.055 g, 0.208 mmol), sodium carbonate (0.294 g, 2.77 mmol), and deionized water (10 mL). The reaction mixture was heated to reflux for 3 hours. After cooling to room temperature the mixture was extracted with ethyl acetate (1 X 150 mL, 2 x 25 mL). The combined organic phases were washed with saturated aqueous NaHCO₃ (50 mL) and brine (2 x 50 mL), dried over MgSO₄, filtered, and concentrated in vacuo to yield a yellow oil. The crude material was purified by flash chromatography in 0.5% ethyl acetate in hexane yielding a white solid. The solid was recrystallized from hexane yielding the desired ester as a white solid (0.60 g, 63%): mp 93.7-95.1 °C. ¹H NMR (CDCl₃/300 MHz) 7.69 (s, 1H), 7.35-7.32 (m, 2H), 7.22 (d, 1H, *J* = 2.6 Hz), 7.05-7.03 (m, 2H), 6.96-6.93 (m, 1H), 5.72 (q, 1H, *J* = 6.7 Hz), 4.34-4.31 (m, 2H), 1.35 (t, 3H, *J* = 7.1 Hz). FABLRMS *m/z* 413 (M+H). ESHRMS *m/z* 413.0765 (M+H, Calc'd. 413.076747). Anal. Calc'd. for C₂₀H₁₆ClF₃O₄: C, 58.19; H, 3.91; Cl, 8.59. Found: C, 58.33; H, 4.10; Cl, 8.61.

Step 2. Preparation of 6-chloro-8-(3-methoxyphenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

5 The ester from Step 1 (0.300 g, 0.727 mmol) was dissolved in THF-ethanol-water (7:2:1, 10 mL). It was treated with sodium hydroxide (0.29 mL of a 2.5 N solution, 0.727 mmol), and stirred at room temperature for 18 hours. The solvent was
10 evaporated and the residue was dissolved in water (10 mL). Ether (10 mL) was added, followed by a few drops of concentrated HCl. The ether layer was separated, and the aqueous phase was extracted with ether (2 x 10 mL). The ether extracts were
15 combined, dried over MgSO₄, filtered, and concentrated in vacuo to yield a white solid, which was recrystallized in diethyl ether-hexane to afford the title compound as a white solid (0.23 g, 81%): mp 173.1-177.4 °C. ¹H NMR (CDCl₃/300 MHz)
20 7.81 (s, 1H), 7.39-7.37 (m, 2H), 7.05-7.04 (m, 2H), 6.97-6.94 (m, 1H), 5.71 (q, 1H, *J* = 6.7 Hz), 3.85 (s, 3H). ESHRMS *m/z* 383.0278 (M-H, Calc'd. 383.029796). Anal. Calc'd. for C₁₈H₁₂ClF₃O₄: C, 56.20; H, 3.14; Cl, 9.21. Found: C, 55.90; H, 3.11; Cl, 9.48.
25

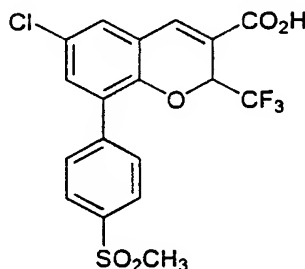
EXAMPLE 127

6-Chloro-8-[(4-methylthio)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

5

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 126: mp 211.4-212.5 °C ¹H NMR (acetone-*d*₆/300 MHz) 7.94 (s, 1H), 7.57 (d, 1H, *J* = 2.6 Hz), 7.53-7.50 (m, 2H), 7.45 (d, 1H, *J* = 2.6 Hz), 7.39-7.36 (m, 2H), 5.87 (q, 1H, *J* = 7.1 Hz), 2.55 (s, 3H). ESHRMS *m/z* 399.0051 (M-H, Calc'd. 399.0069). Anal. Calc'd. for C₁₈H₁₂ClF₃O₃S: C, 53.94; H, 3.02; Cl, 8.84; S, 8.00. Found: C, 53.86; H, 2.82; Cl, 8.91; S, 8.21.

15

EXAMPLE 128

6-Chloro-8-[(4-methylsulfonyl)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

20

Step 1. Preparation of ethyl-6-chloro-8-[(4-methylsulfonyl)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

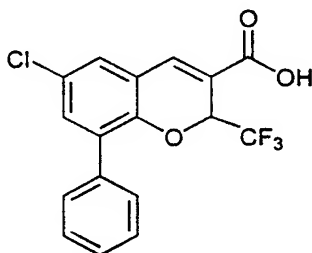
Oxone™ (1.44 g, 2.34 mmol) was dissolved in
5 H₂O (10 mL) and then chilled to 5 °C. A solution of
ethyl 6-chloro-8-[(4-methylthio)phenyl]-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylate
(Example 127, ethyl ester) (0.5 g, 1.17 mmol) in
methanol (20 mL) was slowly added to the reaction
10 mixture and the solution was stirred at room
temperature for 5 hours. The methanol was then
removed in vacuo. The remaining solution was
extracted with methylene chloride (2 x 50 mL). The
combined organic layers were dried over MgSO₄,
15 filtered, and evaporated to yield a yellow solid.
This solid was recrystallized in ether-hexane to
afford the sulfone as a white solid (0.46 g, 84%):
mp 139.2-146.2 °C. ¹H NMR (CDCl₃/300 MHz) 8.03(s,
1H), 8.00 (s, 1H), 7.70 (d, 2H, J = 2.4 Hz), 7.28
20 (d, 1H, J = 2.6 Hz), 5.71 (q, 1H, J = 6.9 Hz),
4.35-4.32 (m, 2H), 3.11(s, 3H), 1.35 (t, 3H, J =
7.2 Hz). FABLRMS m/z 467 (M+Li). ESHRMS m/z
478.0707 (M+NH₄⁺, Calc'd. 478.070281). Anal.
Calc'd. for C₂₀H₁₆ClF₃O₅S: C, 52.12; H, 3.50; Cl,
25 7.69. Found: C, 52.17; H, 3.36; Cl, 7.77.

Step 2. Preparation of 6-chloro-8-[(4-methylsulfonyl)phenyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

30 The sulfone from Step 1 (0.300 g, 0.651 mmol)
was dissolved in a solution of THF:ethanol:water

(7:2:1; 10 mL). It was treated with sodium hydroxide (0.26 mL, 0.651 mmol of a 2.5 N solution), and stirred at room temperature for 18 hours. The solvent was removed in vacuo and the residue was dissolved in water (10 mL). Diethyl ether (10 mL) was added and the mixture acidified with concentrated HCl. The organic layer was separated, and the aqueous phase was extracted with diethyl ether (2 x 10 mL). The combined organic extracts were dried over MgSO₄, filtered, and evaporated to yield a white solid. Recrystallization of this solid in ether-hexane afforded the title compound as a white solid (0.20 g, 73%): mp 286.5-287.8 °C. ¹H NMR (acetone-d₆/300 MHz) 8.07 (d, 2H, J = 6.7 Hz), 7.97 (s, 1H), 7.84 (d, 2H, J = 6.7 Hz), 7.67 (d, 1H, J = 2.6 Hz), 7.55 (d, 1H, J = 2.6 Hz), 5.92 (q, 1H, J = 7.1 Hz), 3.20 (s, 1H). ESHRMS m/z 430.9947 (M-H, Calc'd. 430.996782). Anal. Calc'd. for C₁₈H₁₂ClF₃O₅S: C, 49.95; H, 2.80; Cl, 8.19. Found: C, 50.04; H, 2.80; Cl, 8.25.

EXAMPLE 129



6-Chloro-8-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

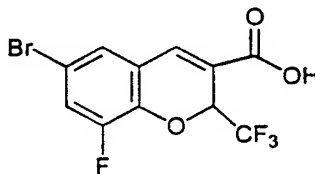
Step 1. Preparation of ethyl 6-chloro-8-phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A mixture of ethyl 6-chloro-8-bromo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 74, Step 1) (2.0 g, 5.2 mmol), tetrakis(triphenylphosphine) palladium(0) (2.15 g, 1.7 mmol), triphenylphosphine (0.013 g, 0.05 mmol), and tributylphenyltin (1.9 mL, 5.7 mmol) in toluene (60 mL) was heated to 110 °C for 3 days. The reaction mixture was allowed to cool to room temperature and filtered through a plug of silica gel eluting with 25% ethyl acetate in hexanes. The filtrate was concentrated *in vacuo* and then purified by flash chromatography (silica gel, ethyl acetate-hexanes, 1:9). The fractions containing desired product were combined and concentrated *in vacuo*. To remove the remaining tin impurities the mixture was taken up in THF (10 mL) and aqueous ammonium fluoride solution (10 wt %, 20 mL) and stirred at room temperature for 2 hours. The solution was extracted with ethyl acetate. The extracts were combined, dried over MgSO₄, filtered, and concentrated *in vacuo* to afford the ester as an oil (1.30 g, 65%). ¹H NMR(CDCl₃/300 MHz) 7.67 (s, 1H), 7.47-7.36 (m, 5H), 7.31 (d, 1H, *J* = 2.6 Hz), 7.18 (d, 1H, *J* = 2.4 Hz), 5.69 (q, 1H, *J* = 6.8 Hz), 4.30 (m, 2H), 1.33 (t, 3H, *J* = 7.1 Hz). ¹⁹F NMR (CDCl₃/282 MHz) δ -78.27 (d, *J* = 7.2 Hz). FAB/MS *m/z* 383 (M+H). ESI/MS *m/z* 400.0937 (M+NH₄⁺, Calc'd 400.0927)

Step 2. Preparation of 6-chloro-8-phenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

A solution of the ester from step 1 (1.0 g, 2.6 mmol) was dissolved in THF (5 mL) and methanol (5 mL) was treated with a 2.5 N NaOH solution (4.0 mL, 10.4 mmol). The resulting mixture was stirred at room temperature for 18 hours. The solvent was removed *in vacuo*, and the residue taken up in ethyl acetate and acidified with 3 N HCl. The solution was extracted with ethyl acetate. The extracts were combined, dried over MgSO₄, filtered, and concentrated *in vacuo* yielding a yellow solid. Recrystallization from ethyl acetate-hexanes afforded the title compound as a pale yellow solid (0.42 g, 46%): mp 196.3-197.7 °C. ¹H NMR (CDCl₃/300 MHz) d 7.65 (s, 1H), 7.40-7.23 (m, 6H), 7.15 (s, 1H), 5.63 (q, 1H, *J* = 6.5 Hz), 3.35 (broad s, 1H). ¹⁹F NMR (CDCl₃/282 MHz) d -78.71 (d, *J* = 5.8 Hz). FABLRMS *m/z* 355 (M+H). ESHRMS *m/z* 353.0198 (M-H, Calc'd 353.0192).

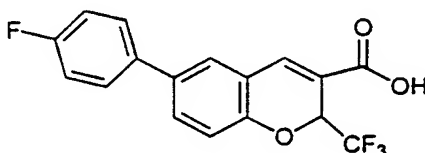
EXAMPLE 130



6-Bromo-8-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

4-Bromo-2-fluorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 206-208 °C. ¹H NMR (CD₃OD/300 MHz) 7.78 (s, 1H), 7.36-7.48 (m, 2H), 5.87 (q, 1H, *J* = 6.8 Hz). EIHRMS *m/z* 339.9349 (Calc'd 339.9358). Anal. Calc'd for C₁₁H₅BrF₄O₃: C 38.74, H 1.48; Found C 38.97, H, 1.60.

EXAMPLE 131

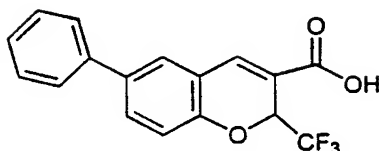


6-(4-Fluorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

15

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 125 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 207-210 °C. ¹H NMR (CD₃OD/300 MHz) 7.87 (s, 1H), 7.54-7.64 (m, 4H), 7.10-7.20 (m, 2H), 7.03 (d, 1H, *J* = 9.4 Hz), 5.77 (q, 1H, *J* = 7.0 Hz). EIHRMS *m/z* 338.0573 (Calc'd 338.0566) Anal. Calc'd for C₁₁H₆F₃IO₃ + 1.25% H₂O: C, 59.62; H, 3.08. Found C, 59.61; H, 3.09.

242

EXAMPLE 132

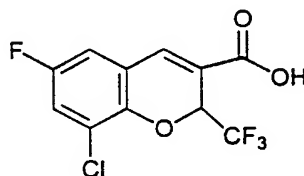
5 **6-Phenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 125 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 197-198 °C. ¹H NMR (CD₃OD/300 MHz) 7.87 (s, 1H), 7.28-7.64 (m, 7H), 7.03 (d, 1H, J = 6.8 Hz), 5.76 (q, 1H, J = 7.0 Hz). EIHRMS m/z 320.0604 (M⁺, Calc'd 320.0660). Anal. Calc'd for C₁₇H₁₁F₃O₃: C, 63.75; H 3.46. Found C, 63.56; H, 3.46.

10
15

EXAMPLE 133

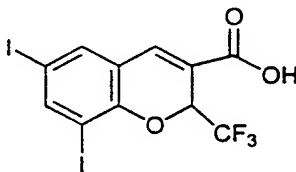
20



25 **8-Chloro-6-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

2-Chloro-4-fluorophenol was converted to the title compound by a procedure similar to that described in Example 2: mp 240-241 °C. ¹H NMR (CD₃OD/300 MHz) 7.77 (s, 1H), 7.26 (dd, 1H, J = 8.3, 2.9), 7.14 (dd, 1H, J = 8.1, 2.9), 5.87 (q, 1H, J = 6.8 Hz). EIHRMS m/z 295.9836 (Calc'd 295.9863). Anal. Calc'd for C₁₁H₅ClF₄O₃: C, 44.54; H, 1.70. Found C, 44.70; H, 1.73.

10

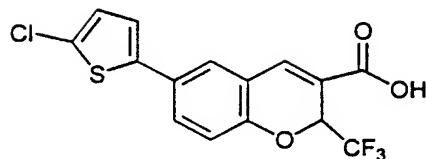
EXAMPLE 134

15

6,8-Diiodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 1: mp 243-244 °C. ¹H NMR (CD₃OD/300 MHz) 8.07 (d, 1H, J = 2.0 Hz), 7.71 (s, 1H), 7.70 (d, 1H, J = 2.0 Hz), 5.89 (q, 1H, J = 6.8 Hz). ESHRMS m/z 494.8174 (Calc'd for M-H 494.8202) Anal. Calc'd for C₁₁H₅F₃I₂O₃: C, 26.64; H, 1.02. Found C, 26.75; H, 1.06.

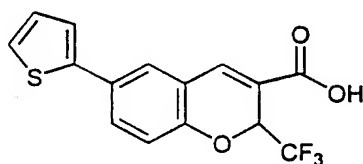
25

EXAMPLE 135

5 **6-(5-Chloro-2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 125 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 72, Step 3) as the starting material: mp 205-206 °C. ¹H NMR (CD₃OD/300 MHz) 7.83 (s, 1H), 7.50-7.58 (m, 2H), 7.14 (d, 1H, *J* = 4.0 Hz), 7.00 (d, 1H, *J* = 8.86 Hz), 6.93 (d, 1H, *J* = 4.0 Hz), 5.77 (q, 1H, *J* = 7.0 Hz). EIHRMS *m/z* 359.9810 (M⁺, Calc'd 359.9835). Anal. Calc'd for C₁₅H₈ClF₃O₃S: C, 49.94; H 2.24. Found C, 50.14; H, 2.29.

20

EXAMPLE 136

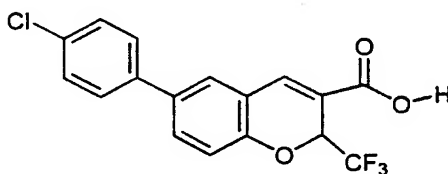
25

6-(2-thienyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

2AS

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 125 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 209-212 °C. ¹H NMR (CD₃OD/300 MHz) 7.83 (s, 1H), 7.58-7.62 (m, 2H), 7.30-7.38 (m, 2H), 6.80-7.09 (m, 2H), 5.76 (q, 1H, J = 7.0 Hz) FABHRMS m/z 325.0153 (Calc'd for M-H 325.0146)

EXAMPLE 137



15

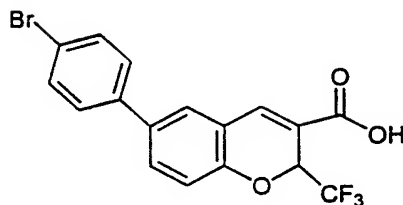
6-(4-Chlorophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 125 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 212-213 °C. ¹H NMR (CD₃OD/300 MHz) 7.89 (s, 1H), 7.56-7.66 (m, 4H), 7.40-7.48 (m, 2H), 7.04-7.10 (m, 1H), 5.77 (q, 1H, J = 7.0 Hz). ESHRMS m/z 353.0190 (Calc'd for M-H 353.0192). Anal. Calc'd for

$C_{17}H_{10}ClF_3O_3$: C, 57.56; H, 2.84. Found C, 57.41; H, 2.82.

EXAMPLE 138

5



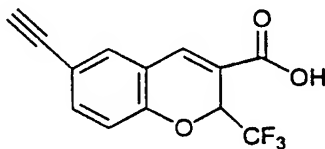
6-(4-Bromophenyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

10

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 126: using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 215-216 °C. 1H NMR ($CD_3OD/300$ MHz) 7.89 (s, 1H), 7.06-7.71 (m, 6H), 7.04-7.06 (m, 1H), 5.78 (q, 1H, $J = 6.8$ Hz). ESHRMS m/z 396.9681 (Calc'd for M-H 396.9687).

20

EXAMPLE 139

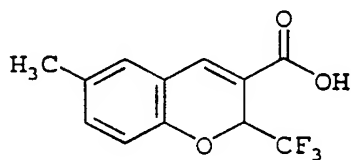


6-(Ethynyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

247-

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 118 using ethyl 6-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 24, Step 3) as the starting material: mp 198-200 °C. ¹H NMR (CD₃OD/300 MHz) 7.80 (s, 1H), 7.47 (dd, 1H, *J* = 8.5, 2.0 Hz), 7.41 (d, 1H, *J* = 2.0 Hz), 6.97 (d, 1H, *J* = 8.5 Hz), 5.71 (q, 1H, *J* = 6.8 Hz), 3.06 (s, 1H). ESHRMS *m/z* 267.0271 (Calc'd for M-H 267.0269) Anal. Calc'd for C₁₃H₇F₃O₃ + 1.06% H₂O: C, 57.60; H, 2.72. Found C, 57.59; H, 2.62.

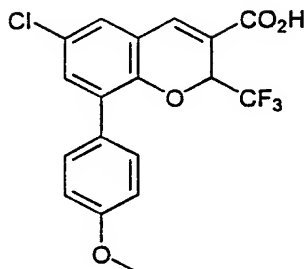
EXAMPLE 140



6-Methyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

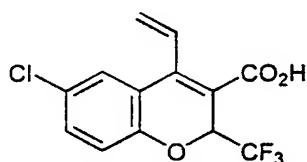
20

4-Methylsalicylaldehyde was converted to the title compound by a procedure similar to that described in Example 1: mp 191.8-193.0 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.80 (s, 1H), 7.72-7.73 (m, 2H), 6.90 (d, 1H, *J* = 8.4 Hz), 5.91 (q, 1H, *J* = 7.2 Hz). Anal. Calc'd for C₁₂H₉O₃F₃: C, 55.82; H, 3.51. Found: C, 55.89; H, 3.49.

EXAMPLE 141

5 **6-Chloro-8-(4-methoxyphenyl)-2-trifluoromethyl-2H-
1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 126: mp 194.0-196.0 °C. ¹H NMR (CDCl₃/300 MHz) 7.81 (s, 1H), 7.44 (s, 1H), 7.41 (s, 1H), 7.34 (d, 1H, J = 2.4 Hz), 7.21 (d, 1H, J = 2.4 Hz), 6.99 (s, 1H), 6.96 (s, 1H), 5.69 (q, 1H, J = 6.7 Hz), 3.86 (s, 3H). FABLRMS m/z 402.2 (M+NH₄).
15 ESHRMS m/z 383.0267 (M-H, Calc'd. 383.029796).
Anal. Calc'd. for C₁₈H₁₂ClF₃O₄: C, 56.20; H, 3.14; Cl, 9.21. Found: C, 56.08; H, 3.11; Cl, 9.13.

EXAMPLE 142

20

**6-Chloro-2-(trifluoromethyl)-4-ethenyl-2H-1-
benzopyran-3-carboxylic acid**

Step 1: Preparation of ethyl 3-(5-chloro-2-hydroxyphenyl)-3-oxo-propionate

A solution of lithium hexamethyldisilazide (800 mL of 1.0 M solution in THF, 800.0 mmol) was
5 chilled to -78 °C under a nitrogen atmosphere. A solution of 5-chloro-2-hydroxyacetophenone (45.493 g, 266.67 mmol) in THF (130 mL) was added dropwise to the stirred solution over 0.5 hour. The reaction was held at -78 °C for 1 hour, warmed to -
10 10 °C for 2 hours, warmed to 0 °C for 1 hour, then cooled to -78 °C. Diethyl carbonate (35.54 mL, 34.65 g, 29.34 mmol) was added via syringe in one portion. The temperature was maintained at -78 °C for 0.5 hour, warmed to room temperature over 0.5
15 hour, and stirred for 3 hours. The crude reaction mixture was carefully poured over a mixture of rapidly stirred ice (1200 mL)/conc HCl (222 mL). The layers were separated and the aqueous phase was extracted with ethyl acetate. The combined organic
20 phase was washed with brine, dried over Na₂SO₄, filtered and concentrated *in vacuo* yielding an oil that began to crystallize. Hexanes (150 mL) was added and crystallization proceeded. The crystalline product was collected by vacuum
25 filtration to afford the title compound (29.04 g, 45%) as tan crystalline needles: mp 71.8-73.1 °C. ¹H NMR (CDCl₃/300 MHz) 7.63 (d, 1H, *J* = 2.4 Hz), 7.45 (dd, 1H, *J* = 8.9, 2.6), 6.98 (d, 1H, *J* = 8.9 Hz), 4.25 (q, 2 H, *J* = 7.3 Hz), 3.98 (s, 2H), 1.29
30 (t, 3H, 7.3 Hz). FABLRMS *m/z* 249 (M+Li). EIHRMS *m/z* 242.0346 (M+, Calc'd 242.0346). Anal. Calc'd

for $C_{11}H_{11}ClO_4$: C, 54.45; H, 4.57. Found: C, 54.48; H, 4.62.

5 Step 2. Preparation of ethyl 2-(trifluoromethyl)-6-chloro-4-oxo-4H-1-benzopyran-3-carboxylate.

 The keto-ester (Step 1) (19.2 g, 79.1 mmol), was added to trifluoroacetic anhydride (67.2 mL, 49.9 g, 475.8 mmol), potassium carbonate (44 g, 318 mmol) and toluene (400 mL). This suspension was
10 stirred at room temperature for 36 hours, then heated to reflux for 4 hours. After cooling to room temperature, the suspension was poured over rapidly stirred (mechanical stirrer) ice (300 mL) and aqueous HCl (12 N, 50 mL). The resulting
15 organic phase was separated from the clear mixture, was washed with water (5 X 500 mL), brine (1 X 500 mL), dried over $MgSO_4$, filtered and concentrated in vacuo yielding tan solid which was dried under high vacuum. This sample was partially dissolved in
20 heptane (100 mL) and ethyl acetate (12 mL) with heating on a steam bath, was filtered to remove insoluble material. The filtrate was allowed to cool to room temperature yielding the desired 4-oxo-4H-1-benzopyran as a fluffy tan solid (14.17 g,
25 56%): mp 106.7-108.6 °C. This material was of suitable purity to use in the next step without further purification.

30 Step 3. Preparation of ethyl 2-(trifluoromethyl)-4-oxo-dihydro-1-benzopyran-3-carboxylate.

A stirred, chilled (0 °C) solution of the ketone (Step 2) (6.92 g, 21.58 mmol) in tetrahydrofuran (40 mL) and ethanol (50 mL) was treated portion-wise with sodium borohydride (NaBH₄, 0.41 g, 10.79 mmol). After 3 h additional sodium borohydride (0.30 g, 7.93 mmol) was added portion-wise over 1 hour. The reaction was poured into rapidly stirred cold aqueous HCl (15 mL of 12 N HCl diluted to 300 mL). During the addition a precipitate formed, that was collected by vacuum filtration and dried under high vacuum yielding the desired substituted 4-oxo-dihydro-1-benzopyran as a white powder (6.92 g, 99%): mp 80.2-84.9 °C. ¹H NMR (CDCl₃/300 MHz) 12.60 (br s, 1H), 7.69 (d, 1H, J = 2.6 Hz), 7.34 (dd, 1H, J = 2, 6, 8.7 Hz), 6.93 (d, 1H, J = 8.7 Hz), 5.59 (q, 1H, 6.6 Hz), 4.46-4.23 (m, 2H), 1.35 (t, 3H, J = 7.0 Hz). FABLRMS m/z 329 (M+Li). EIHRMS m/z 322.0213 (M+, Calc'd 322.0220). Anal. Calc'd for C₁₃H₁₀Cl₁F₃O₄ with 3.57% water: C, 46.67; H, 3.41. Found: C, 46.62; H, 3.14.

Step 4. Preparation of ethyl 6-chloro-4-(trifluoromethanesulfonyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A 50 mL Morton flask fitted with septa and addition funnel was charged with 2,6-di-tert-butylpyridine (1.782 g, 8.679 mmol), methylene chloride (15 mL), and trifluoromethanesulfonic anhydride (1.22 mL, 2.04 g, 7.23 mmol) followed by the dropwise addition of the chroman-4-one (Step 3) (2.145 g, 5.786 mmol) in methylene chloride (12

mL) over 0.33 hour. After stirring for 16 h at room temperature, the reaction was concentrated in vacuo and diluted with diethyl ether (50 mL) yielding a suspension. The suspension was vacuum
5 filtered and the filtrate washed with cold 2 N HCl and brine, dried over MgSO₄, filtered and concentrated in vacuo yielding the desired triflate as a light yellow powder (1.45 g, 55%) of suitable
10 purity to use without further purification: mp 79.2-80.4 °C. ¹H NMR (CDCl₃/300 MHz) 7.40 (s, 1H), 7.37 (d, 1H, *J* = 2.4 Hz), 7.02-6.99 (m, 1H), 5.92 (q, 1H, *J* = 6.6 Hz), 4.47-4.32 (m, 2H), 1.39 (t, 3H, *J* = 7.2 Hz).

15 Step 5. Preparation of ethyl 6-chloro-4-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

Ethyl 6-chloro-4-trifluoromethanesulfoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Step 4) (1.50 g, 3.30 mmol) was dissolved in
20 anhydrous THF (40 mL) in a 100 mL round bottomed flask under nitrogen. Tetrakis(triphenylphosphine)palladium(0) (0.267 g, 0.231 mmol) and lithium chloride (0.140 g, 3.3 mmol) were added, followed by
25 tributylethenylstannane (1.15 g, 3.6 mmol). The resulting solution was heated to reflux for 18 hours. GCMS analysis indicated the starting material had been consumed. The reaction mixture was allowed to cool to room temperature and was
30 poured into 20% ammonium fluoride solution (50 mL). After stirring for one hour, diethyl ether (100 mL)

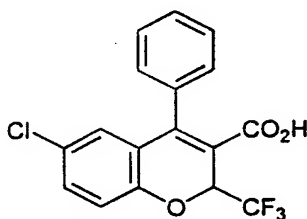
was added and the mixture was washed with water (2 x 50 mL). The organic phase was dried over MgSO_4 , filtered, and concentrated *in vacuo* yielding a brown oil. The crude material was purified by
5 flash column chromatography (hexane) to afford the ester as a yellow oil, which crystallized upon standing (0.760 g, 69%): mp 51.9-53.2 °C. ^1H NMR (CDCl_3 /300 MHz) 7.46 (d, 1H, $J = 2.4$ Hz), 7.28-7.14 (m, 2H), 6.96 (d, 1H, $J = 8.7$ Hz), 5.77-5.71 (m,
10 2H), 5.38 (dd, $J = 1.2, 17.9$ Hz), 4.32-4.26 (m, 2H), 1.33 (t, 2H, $J = 7.1$ Hz). FABLRMS m/z 333.2 (M+H). ESHRMS m/z 333.0510 (M+H, Calc'd.333.050532. Anal. Calc'd for $\text{C}_{15}\text{H}_{12}\text{ClF}_3\text{O}_3$ (1.14 wt % H_2O): C, 53.53; H, 3.72; Cl, 10.53.
15 Found: C, 53.46; H, 3.42; Cl, 10.70.

Step 6. Preparation of 6-chloro-4-ethenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 5 (0.300 g, 0.902 mmol)
20 was dissolved in a THF-EtOH- H_2O mixture (10 mL, 7:2:1) and treated with sodium hydroxide (0.360 mL, 0.902 mmol of a 2.5 N solution). This solution was stirred at room temperature for 18 hours. The solvent was evaporated and the residue was
25 dissolved in water (10 mL). Diethyl ether (10 mL) was added and the mixture acidified by the addition of concentrated HCl. The organic layer was separated, and the aqueous phase was extracted with diethyl ether (2 x 10 mL). The ether extracts were
30 combined, dried over MgSO_4 , filtered, and concentrated *in vacuo* yielding a yellow solid,

which was recrystallized in diethyl ether-hexane to afford the title compound as a white solid (0.163 g, 59%): mp 143.0-145.0 °C. ¹H NMR (CDCl₃/300 MHz) 7.49 (d, 1H, *J* = 2.6 Hz), 7.33-7.17 (m, 2H), 6.99 (d, 1H, *J* = 8.5 Hz), 5.82-5.72 (m, 2H), 5.42 (d, 1H, *J* = 17.9 Hz). ESHRMS *m/z* 303.00207 (M-H, Calc'd. 303.003582). Anal. Calc'd for C₁₃H₈ClF₃O₃ (1.10. wt % H₂O): C, 50.69; H, 2.74; Cl, 11.51. Found: C, 50.57; H, 2.37; Cl, 11.75.

10

EXAMPLE 143

15 **6-Chloro-2-(trifluoromethyl)-4-phenyl-2H-1-benzopyran-3-carboxylic acid**

The 2H-1-benzopyran-3-carboxylic acid was prepared from ethyl 6-chloro-4-(trifluoromethanesulfonyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 142, Step 4) using a procedure similar to that described in Example 142, Steps 5-6: mp 225.5-226.6 °C ¹H NMR (DMSO-*d*₆/300 MHz). 7.46-7.39 (m, 4H), 7.20-7.13 (m, 3H), 6.52 (d, 1H, *J* = 2.42 Hz), 6.12 (q, 1H, *J* = 7.1 Hz). FABLRMS *m/z* 355.1 (M+H). ESHRMS *m/z* 353.0215 (M-H, Calc'd. 353.019232). Anal. Calc'd. for

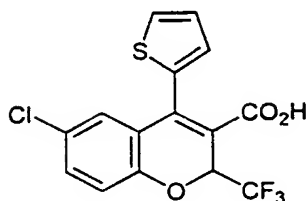
20

25

$C_{17}H_{10}ClF_3O_3$: C, 57.56; H, 2.84; Cl, 10.17. Found:
C, 57.18; H, 2.66; Cl, 10.17.

EXAMPLE 144

5

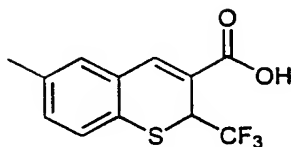


**6-Chloro-4-(2-thienyl)-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid**

10.

The 2H-1-benzopyran-3-carboxylic acid was prepared from ethyl 6-chloro-4-(trifluoromethanesulfonyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 142, Step 4) using a procedure similar to that described in Example 142, Steps 5-6: mp 200.8-206.7 °C. 1H NMR ($CDCl_3$ /300 MHz) 7.52 (dd, 1H, $J = 1.21, 5.04$ Hz), 7.28 (dd, 1H, $J = 2.42, 8.67$ Hz), 7.15 (dd, 1H, $J = 1.21, 3.42$ Hz), 6.98-6.93 (m, 2H), 5.83 (q, 1H, $J = 6.9$ Hz). FABLRMS m/z 378 ($M+NH_4$). Anal. Calc'd. for $C_{15}H_8ClF_3O_3S$: C, 49.94; H, 2.24; Cl, 9.83; S, 8.89. Found: C, 50.02; H, 1.98; Cl, 9.34; S, 8.89.

20

EXAMPLE 145

5 **6-Methyl-2-(trifluoromethyl)-2H-1-
 benzothiopyran-3-carboxylic acid**

Step 1. Preparation of 5-methyl-2-
mercaptobenzaldehyde.

10 Tetramethylethylenediamine (TMEDA) (12.6
 mL, 83.5 mmol) was added via syringe to n-BuLi
 (33 mL of 1.6 M in hexanes, 82.5 mmol) and the
 solution was chilled to 0 °C. A solution of p-
 thiocresol (4.53 g, 36.5 mmol) in cyclohexane
15 (40 mL) was added with stirring over 5 minutes.
 The resulting tan slurry was stirred overnight
 at room temperature, chilled to 0 °C, and DMF
 (4.0 mL, 3.77 g, 51.6 mmol) was added via
 syringe over 2 minutes. The resulting gummy
20 slurry was stirred at room temperature for 1.3
 hours. The reaction mixture was added to 3 N
 HCl (150 mL). This mixture was extracted with
 ethyl acetate. The combined organic layers were
 washed with brine, dried over MgSO₄, filtered
25 and concentrated *in vacuo* yielding a brown oil.
 This oil was purified by flash chromatography
 over silica gel, eluting with 10% ethyl

acetate-hexanes to afford 5-methyl-2-mercaptobenzaldehyde (4.47 g, 69%) as an intensely yellow solid suitable for use without further purification.

5

Step 2. Preparation of ethyl 6-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylate.

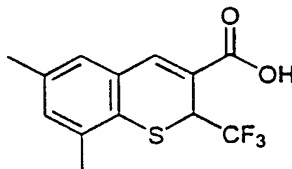
The 5-methyl-2-mercaptobenzaldehyde (Step
10 1) (3.25 g, 21.3 mmol) was added to DMF (5 mL) and ethyl 4,4,4-trifluorocrotonate (4.32 g, 25.7 mmol). With stirring, K_2CO_3 (3.78 g, 27.3 mmol) was added causing the reaction to become a deep red. The reaction was stirred at room
15 temperature for 20 hours, acidified with 3N HCl, diluted with ethyl acetate and washed with water, saturated $NaHCO_3$ solution, brine, dried over $MgSO_4$, filtered and concentrated in vacuo yielding an oil. The oil was crystallized from
20 diethyl ether-petroleum ether to give ethyl 6-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylate as a light yellow solid (4.47 g, 69%): mp 93.1-94.7 °C. 1H NMR (acetone- d_6 /300 MHz) 7.94 (s, 1H), 7.41 (s, 1H), 7.31 (d, 1H, J
25 = 7.9 Hz), 7.25 (d, 1H, J = 7.9 Hz), 4.96 (q, 1H, J = 8.5 Hz), 4.33 (m, 2H), 2.34 (s, 3H), 1.35 (t, 3H, J = 7.0 Hz). FABLRMS m/z 309 (M+Li).

Step 3. Preparation of 6-methyl-2-
(trifluoromethyl)-2H-1-benzothiopyran-3-
carboxylic acid.

The ester from Step 2 (0.55 g 1.8 mmol)
5 was dissolved in THF (1.5 mL) and ethanol (1.5
mL), treated with 2.5 N sodium hydroxide (1.5
mL, 3.8 mmol), and stirred at room temperature
for 88 hours. The reaction mixture was
concentrated *in vacuo*, acidified with 3 N HCl,
10 filtered, and recrystallized from diethyl
ether/petroleum ether to yield the title
compound as a yellow solid (0.14 g, 28%): mp
180.8-184.2 °C. ¹H NMR (acetone-*d*₆/300 MHz)
7.95 (s, 1H), 7.42 (s, 1H), 7.31 (d, 1H, *J* =
15 8.1 Hz), 7.25 (d, 1H, *J* = 8.1 Hz), 4.94 (q, 1H,
J = 8.7 Hz), 2.34 (s, 3H). FABLRMS *m/z* 281
(M+Li). EIHRMS *m/z* 274.0250 (M+, Calc'd
274.0275). Anal. Calc'd for C₁₂H₉F₃O₂S: C,
52.55; H, 3.31. Found: C, 52.54; H, 3.35.

20

EXAMPLE 146



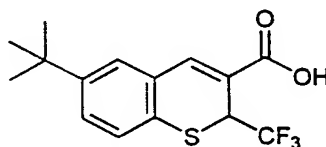
25

6,8-Dimethyl-2-(trifluoromethyl)-2H-1-
benzothiopyran-3-carboxylic acid

259

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 145: mp 220-225 °C (dec). ¹H NMR (acetone-d₆/300 MHz) 11.5 (brs, 1H), 7.94 (s, 1H), 7.26 (s, 1H) 7.14 (s, 1H), 4.98 (q, 1H, *J* = 8.7 Hz), 2.34 (s, 3H), 2.31 (s, 3H). FABLRMS *m/z* 295 (M+Li). EIHRMS *m/z* 288.0431 (M+, Calc'd 288.0432). Anal. Calc'd for C₁₃H₁₁F₃O₂S: C, 54.16; H, 3.85. Found: C, 54.10; H, 3.91.

EXAMPLE 147



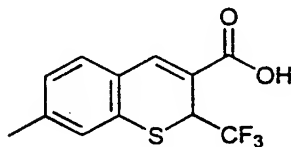
6-(1,1-Dimethylethyl)-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 145: mp 183.8-184.6 °C. ¹H NMR (acetone-d₆/300 MHz) 8.04 (s, 1H), 7.68 (d, 1H, *J* = 2.2 Hz), 7.46 (dd, 1H, *J* = 8.3 Hz 2.2 Hz), 7.37 (d, 1H, *J* = 8.3 Hz), 4.94 (q, 1H, *J* = 8.7 Hz), 1.34 (s, 9H). FABLRMS *m/z* 334 (M+NH₄). ESHRMS *m/z* 334.1087 300 MHz) 7.52 (dd, 1H, *J* = 1.21, 5.04 Hz), 7.28 (dd, 1H, *J* = 2.42, 8.67 Hz), 7.15 (dd, 1H, *J* = 1.21, 3.42 Hz),

6.98-6.93 (m, 2H), 5.83 (q, 1H, $J = 6.9$ Hz).
FABLRMS m/z 378 ($M+NH_4$). Anal. Calc'd. for
 $C_{15}H_8ClF_3O_3S$: C, 49.94; H, 2.24; Cl, 9.83; S,
8.89. Found: C, 50.02; H, 1.98; Cl, 9.34; S,
5 8.89. ($M+NH_4$, Calc'd 334.1089). Anal. Calc'd for
 $C_{15}H_{15}F_3O_2S$: C, 56.95; H, 4.78. Found: C, 57.03;
H, 4.83.

EXAMPLE 148

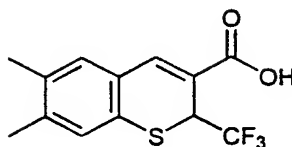
10



7-Methyl-2-(trifluoromethyl)-2H-1- benzothiopyran-3-carboxylic acid

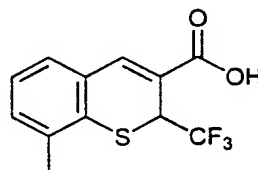
15

The 2H-1-benzothiopyran-3-carboxylic acid
was prepared by a procedure similar to the
method described in Example 145: mp 186.6-191.9
°C. 1H NMR (acetone- d_6 /300 MHz) 7.96 (s, 1H),
20 7.49 (dd, 1H, $J = 7.6$ Hz 2.82 Hz), 7.27 (s,
1H), 7.14 (d, 1H, $J = 7.6$ Hz), 4.96 (q, 1H, $J =$
5.3 Hz), 2.36 (s, 3H). ESHRMS m/z 273.0204 ($M-$
H, Calc'd 273.0197). Anal. Calc'd for $C_{12}H_9F_3O_2S$
(3.32 wt % H_2O): C, 50.81; H, 3.57. Found: C,
25 50.79; H, 3.44.

EXAMPLE 149

5 **6,7-Dimethyl-2-(trifluoromethyl)-2H-1-**
 benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid
was prepared by a procedure similar to the
10 method described in Example 145: mp 235-237 °C.
¹H NMR (acetone-d₆/300 MHz) 7.90 (s, 1H), 7.33
(s, 1H), 7.19 (s, 1H), 4.91 (q, 1H, J = 8.7
Hz), 2.28 (s, 3H), 2.26 (s, 3H). FABLRMS m/z
295 (M+Li). EIHRMS m/z 288.0439 (M+, Calc'd
15 288.0432). Anal. Calc'd for C₁₃H₁₁F₃O₂S: C,
54.16; H, 3.85. Found: C, 54.13; H, 3.85.

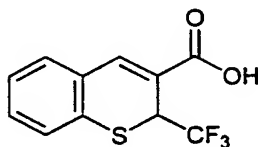
EXAMPLE 150

20

8-Methyl-2-(trifluoromethyl)-2H-1-
 benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 145: mp 224-225 °C. ¹H NMR (acetone-d₆/300 MHz) 11.60 (br s, 1H), 8.00 (s, 1H), 7.44 (d, 1H, *J* = 6.7 Hz), 7.31 (d, 1H, *J* = 6.8 Hz), 7.21 (m, 1H), 5.05 (q, 1H, *J* = 8.5 Hz), 2.38 (s, 3H). FABLRMS *m/z* 292 (M+NH₄). ESHRMS *m/z* 292.0591 (M+NH₄, Calc'd 292.0619). Anal. Calc'd for C₁₂H₉F₃O₂S: C, 52.55; H, 3.31. Found: C, 52.63; H, 3.38.

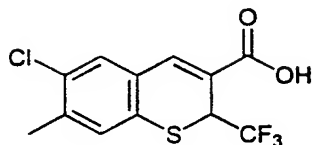
EXAMPLE 151



2-(Trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 145: mp 187-190 °C. ¹H NMR (acetone-d₆/300 MHz) 8.01 (s, 1H), 7.60 (d, 1H, *J* = 7.5 Hz), 7.45 (m, 2H), 7.31 (m, 1H), 4.98 (q, 1H, *J* = 8.7 Hz). ESHRMS *m/z* 259.0070 (M-H, Calc'd 259.0041). Anal. Calc'd for C₁₁H₇F₃O₂S: C, 50.77; H, 2.71. Found: C, 50.75; H, 2.78.

EXAMPLE 152



5

6-Chloro-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

Step 1. Preparation of N,N-dimethyl-O-(4-chloro-2-formyl-5-methylphenyl)thiocarbamate.

A mixture of 5-chloro-4-methylsalicylaldehyde (12.96 g, 76.0 mmol) and triethylamine (11.58 g, 114.4 mmol) was dissolved in anhydrous DMF (15 mL) treated with

15 *N,N*-dimethylthiocarbamoyl chloride (11.25 g, 91.0 mmol) and stirred at room temperature for 16 hours. The reaction was treated with 3 N HCl (50 mL) and filtered to give an orange solid. The solid was dissolved in ethyl acetate washed

20 with 3 N HCl, water, brine, dried over anhydrous MgSO₄, filtered and concentrated in vacuo to afford a brown solid (16.79 g) which was recrystallized from diethyl ether/hexane to give the O-aryl thiocarbamate as a tan solid

25 (4.92 g, 25%): ¹H NMR (acetone-d₆/300 MHz) 9.96 (s, 1H), 7.80 (s, 1H), 7.19 (s, 1H), 3.46 (s, 3H), 3.42 (s, 3H), 2.43 (s, 3H).

Step 2. Preparation of N,N-dimethyl-S-(4-chloro-2-formyl-5-methylphenyl)thiocarbamate.

The O-aryl thiocarbamate (Step 1) (4.92 g, 19.1 mmol) was dissolved in *N,N*-dimethylaniline (25 mL) and immersed in and stirred at 200 °C for 1.5 hours. The reaction mixture was cooled to room temperature and poured into a mixture of 3 N HCl (200 mL) and ice. Filtration gave a brown semisolid which was dissolved in ethyl acetate, washed with 3 N HCl, brine, dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to afford the S-arylthiocarbamate as a brown oil (3.80 g, 77%) which was used in the next step without further purification.

15

Step 3. Preparation of ethyl 6-chloro-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylate.

The S-arylthiocarbamate (Step 2) (3.80 g, 14.7 mmol) was dissolved in THF (10 mL) and ethanol (10 mL), treated with 2.5 N sodium hydroxide (16.5 mL, 34.2 mmol), and stirred at room temperature for 0.9 hours. The reaction was diluted with diethyl ether and washed with 3 N HCl, brine, dried over MgSO₄, filtered and concentrated *in vacuo* to yield the crude substituted 2-mercaptobenzaldehyde as a brown oil (2.82 g). This oil was added to DMF (10 mL) and ethyl 4,4,4-trifluorocrotonate (3.89 g, 23.1 mmol). With stirring, K₂CO₃ (3.23 g, 23.4

30

mmol) was added causing the reaction to become a deep red. The reaction was stirred at room temperature for 14.5 hours, acidified with 3 N HCl, extracted with ethyl acetate. The resulting organic phase was washed with brine, dried over MgSO_4 , filtered and concentrated in vacuo to give a yellow solid (6.36 g) which was used in the next step without further purification.

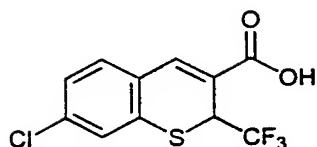
10

Step 4. Preparation of 6-chloro-7-methyl-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

The ester from Step 3 (2.02 g, 6.0 mmol) was dissolved in THF (10 mL) and ethanol (10 mL), treated with 2.5 N sodium hydroxide (5.5 mL, 13.8 mmol), and stirred at room temperature for 4.8 hours. The reaction mixture was concentrated in vacuo, acidified with 3 N HCl yielding a suspension. The solid was collected by filtration and was recrystallized from ethanol-water to yield the title compound as a yellow solid (0.20 g, 11%): mp 240.5-241.7 °C. ^1H NMR (acetone- d_6 /300MHz) 7.99 (s, 1H), 7.67 (s, 1H), 7.43 (s, 1H), 4.99 (q, 1H, $J = 8.5$ Hz), 2.39 (s, 3H). FABLRMS m/z 307 (M-H). FABHRMS m/z 306.9831 (M-H, Calc'd 306.9807). Anal. Calc'd for $\text{C}_{12}\text{H}_8\text{ClF}_3\text{O}_2\text{S}$: C, 46.69; H, 2.61; Cl, 11.48. Found: C, 46.78; H, 2.61; Cl, 11.41.

30

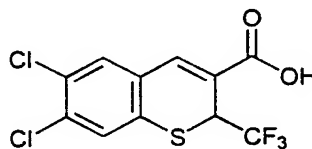
Z66

EXAMPLE 153

5

7-Chloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid
10 was prepared by a procedure similar to the
method described in Example 152: mp 225.7-227.3
°C. ¹H NMR (acetone-d₆/300 MHz) 8.02 (s, 1H),
7.63 (d, 1H, *J* = 8.3 Hz), 7.54 (d, 1H, *J* = 2.0
Hz), 7.36 (dd, 1H, *J* = 8.3 Hz 2.0 Hz), 5.04 (q,
15 1H, *J* = 8.5 Hz). ESHRMS *m/z* 292.9646 (M-H,
Calc'd 292.9651).

EXAMPLE 154

20

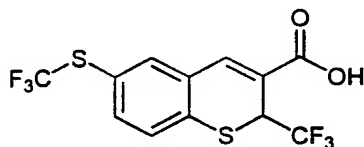
6,7-Dichloro-2-(trifluoromethyl)-2H-1-benzothiopyran-3-carboxylic acid

267

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 152: mp 262.5-263.5 °C. ¹H NMR (acetone-d₆/300 MHz) 8.04 (s, 1H),
5 7.90 (s, 1H), 7.74 (s, 1H), 5.09 (q, 1H, J = 8.5 Hz). ESHRMS m/z 326.9242 (M-H, Calc'd 326.9261).

EXAMPLE 155

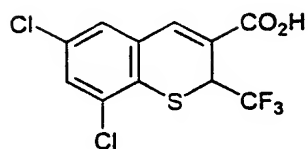
10



2-(Trifluoromethyl)-6-[(trifluoromethyl)thio]- 2H-1-benzothiopyran-3-carboxylic acid

15

The 2H-1-benzothiopyran-3-carboxylic acid was prepared by a procedure similar to the method described in Example 152: mp 129.3-132.4 °C. ¹H NMR (acetone-d₆/300 MHz) 8.10 (s, 2H),
20 8.00 (s, 2H), 7.71 (d, 2H, J = 8.1 Hz), 7.65 (d, 2H, J = 8.1 Hz), 5.09 (q, 1H, J = 8.5 Hz). ESHRMS m/z 358.9630 (M-H, Calc'd 358.9635).

EXAMPLE 156

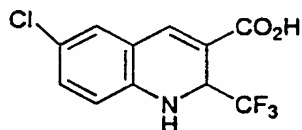
5 **6,8-Dichloro-2-trifluoromethyl-2H-1-**
 benzothiopyran-3-carboxylic acid

The 2H-1-benzothiopyran-3-carboxylic acid
was prepared by a procedure similar to the
10 method described in Example 152: mp 217.9-
220.3 °C. ¹H NMR (acetone-d₆/300 MHz) 12.50-
11.20 (br s, 1H exch.), 8.06 (s, 1H), 7.75 (d,
1H, *J* = 2.0 Hz), 7.64 (d, 1H, *J* = 2.2 Hz), 5.23
(q, 1H, *J* = 8.5

Hz). ESLRMS m/z 327 (M-H). ESHRMS m/z 326.9272 (M-H, Calc'd 326.9261).

EXAMPLE 157

5



6-Chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid

10

Step 1. Preparation of 2-amino-5-chlorobenzaldehyde.

2-Amino-5-chlorobenzyl alcohol (4.8 g, 30 mmol) and activated manganese (IV) oxide (21 g, 240 mmol) were refluxed in chloroform (100 mL) for 1 hour. The contents were allowed to cool, filtered through diatomaceous earth and concentrated *in vacuo* to afford the 2-amino-5-chlorobenzaldehyde as a dark solid (4.14 g, 81%): mp 74-76 °C. ¹H NMR (CDCl₃, 300 MHz) 9.80 (s, 1H), 7.42 (s, 1H), 7.23 (d, 1H, *J* = 7.0 Hz), 6.60 (d, 1H, *J* = 7.0 Hz).

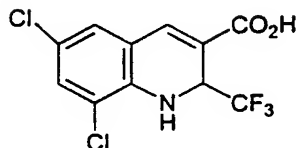
25 Step 2. Preparation of ethyl 6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate.

The 2-amino-5-chlorobenzaldehyde from Step 1 (15.0 g, 96 mmol), anhydrous potassium

carbonate (27.6 g, 200 mmol), and ethyl
4,4,4-trifluorocrotonate (34 mL, 200 mmol)
were mixed in anhydrous dimethylformamide (60
mL) and heated at 100 °C for 7 hours. The
5 contents were allowed to cool and partitioned
between ethyl acetate (200 mL) and water (200
mL). The aqueous layer was extracted with
ethyl acetate (1 x 100 mL). The ethyl
acetate extracts were combined and washed
10 with brine (1 x 200 mL), dried over MgSO₄, and
concentrated *in vacuo* leaving a dark oil
which solidified upon standing. The solid
was purified by flash chromatography (silica
gel; ethyl acetate-hexanes, 1:9). Fractions
15 containing the desired product were combined,
concentrated *in vacuo* and the residue
recrystallized from ethyl acetate-hexanes to
afford the ethyl 6-chloro-1,2-dihydro-2-
(trifluoromethyl)-3-quinolinecarboxylate as a
20 yellow solid (16.36 g, 56%): mp 132.6-134.2
°C. ¹H NMR (CDCl₃, 300 MHz) 7.61 (s, 1H), 7.10
(m, 2H), 6.55 (d, 1H, *J* = 8.0 Hz), 5.10 (q,
1H, *J* = 6.0 Hz), 4.55 (brs, 1H), 4.23 (m,
2H), 1.32 (t, 3H, *J* = 7.0 Hz). FABHRMS *m/z*
25 306.0468 (M+H⁺, Calc'd 306.0509). Anal.
Calc'd for C₁₃H₁₁NO₂F₃Cl: C, 51.08; H, 3.63; N,
4.58. Found: C, 50.81; H, 3.49; N, 4.72.

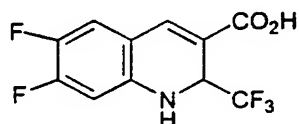
Step 3. Preparation of 6-chloro-1,2-dihydro-
30 2-(trifluoro-methyl)-3-quinolinecarboxylic
acid.

The ester from Step 2 (1.7 g, 5.6 mmol) and 2.5 N sodium hydroxide (4.4 mL, 11 mmol) were mixed in tetrahydrofuran (25 mL), methanol (10 mL), and water (25 mL). After stirring overnight, contents were concentrated *in vacuo* to remove the THF and methanol. The aqueous solution remaining was extracted with diethyl ether (2 x 100 mL). The resulting aqueous layer was acidified with 2 N HCl causing the precipitation of an oil. The oil was purified by flash chromatography on silica gel, eluting with ethyl acetate-hexanes (1:1). Fractions containing the desired product were combined, and concentrated *in vacuo*. The residue was triturated with dichloromethane, and filtered to afford the 6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid as a yellow solid (0.645 g, 41 %): mp 187.8-188.8 °C. ¹H NMR (acetone-d₆, 300 MHz) 7.69 (s, 1H), 7.36 (s, 1H), 7.15 (d, 1H, *J* = 8.0 Hz), 6.83 (d, 1H, *J* = 8.0 Hz), 6.60 (brs, 1H), 5.20 (m, 1H). ESHRMS *m/z* 276.0040 (M-H, Calc'd 276.0039). Anal. Calc'd for C₁₁H₇NO₂F₃Cl + 2.6% H₂O: C, 46.39; H, 2.98; N, 4.92. Found: C, 45.99; H, 2.54; N, 4.85.

EXAMPLE 158

5 **6,8-Dichloro-1,2-dihydro-2-(trifluoromethyl)-**
 3-quinolinecarboxylic acid

The 1,2-dihydro-3-quinolinecarboxylic acid was prepared by a procedure similar to
10 that described in Example 157: mp 223.4-225.7 °C. ¹H NMR (acetone-*d*₆, 300 MHz) 7.82 (s, 1H), 7.40 (m, 2H), 6.53 (brs, 1H), 5.40 (m, 1H). ESHRMS *m/z* 309.9657 (M-H, Calc'd 309.9649). Anal. Calc'd for C₁₁H₆NO₂F₃Cl₂: C, 42.34; H, 1.94; N, 4.49. Found: C, 42.20; H, 1.74; N, 4.52.

EXAMPLE 159

20

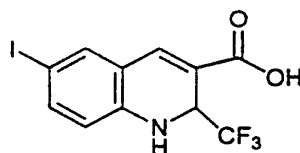
6,7-Difluoro-1,2-dihydro-2-(trifluoromethyl)-
 3-quinolinecarboxylic acid

25 The 1,2-dihydro-3-quinolinecarboxylic acid was prepared by a procedure similar to

that described in Example 157: mp 186.6-188.9 °C. ¹H NMR (acetone-d₆, 300 MHz) 7.79 (s, 1H), 7.32 (m, 1H), 6.71 (m, 1H), 6.64 (brs, 1H), 5.21 (m, 1H). ESHRMS m/z 278.0262 (M-H, Calc'd 278.0240). Anal. Calc'd for C₁₁H₆NO₂F₅ + 1.58% H₂O: C, 46.58; H, 2.31; N, 4.94. Found: C, 46.20; H, 2.07; N, 4.54.

EXAMPLE 160

10



6-Iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid

15

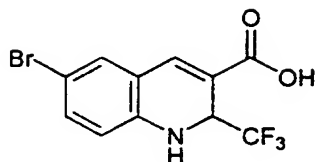
Step 1. Preparation of ethyl 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate.

A mixture of 5-iodo-2-aminobenzaldehyde (24.0 g, 96.7 mmol), diazobicyclo[2.2.2]-undec-7-ene (32.2 g, 212.0 mmol), and ethyl 4,4,4-trifluorocrotonate (35.7 g, 212.0 mmol) in 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidinone (48 mL) was heated at 60 °C for 8 hours. The solution was cooled to room temperature and the solution poured into ethyl acetate-hexanes (1:1, 500 mL). The solution was extracted with 2.5 N aqueous

hydrochloric acid (2 x 200 mL), saturated aqueous ammonium chloride (2 x 200 mL), dried over sodium sulfate, filtered and concentrated *in vacuo*. The resulting dark yellow oil was dissolved in hexanes (100 mL) and fine yellow crystals formed upon standing. Vacuum filtration of this suspension yielded ethyl 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate as fine yellow crystals (19.3 g, 50 % yield): mp 137-138 °C. ¹H NMR (CDCl₃, 300 MHz) 7.62 (s, 1H), 7.36-7.48 (m, 2H), 6.43 (d, *J* = 8.2 Hz), 5.36 (brs, 1H), 5.11 (q, 1H, *J* = 7.1 Hz), 4.25 -4.35 (m, 2H), 1.34 (t, 3H, *J* = 7.0 Hz). ESHRMS *m/z* 395.9716 (M-H, Calc'd 395.9708).

Step 2. Preparation of 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid

Hydrolysis of the ester (Step 1) was performed by a procedure similar to that described in Example 157, Step 3, yielding the carboxylic acid. mp 188-192 °C. ¹H NMR (CD₃OD/300 MHz) 7.668 (s, 1H), 7.46 (d, 1H, *J* = 2.2 Hz), 7.39 (dd, 1H, *J* = 8.4, 2.2 Hz), 6.52 (d, 1H, *J* = 8.4 Hz), 5.01 (q, 1H, *J* = 7.5 Hz). ESHRMS *m/z* 367.9401 (M, Calc'd 367.9395).

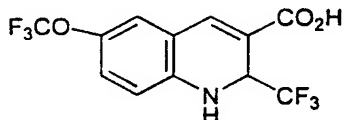
EXAMPLE 161

5 **6-Bromo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid**

The 1,2-dihydro-3-quinolinecarboxylic acid was prepared by a procedure similar to
10 that described in Example 160: mp 185-186 °C. ¹H NMR (CD₃OD/300 MHz) 7.68 (s, 1H), 7.31 (d, 1H, *J* = 2.2 Hz), 7.23 (dd, 1H, *J* = 8.7, 2.2 Hz), 6.64 (d, 1H, *J* = 8.7 Hz), 5.01 (q, 1H, *J* = 7.5 Hz). EIHRMS *m/z* 319.9519 (M, Calc'd 319.9534). Anal. Calc'd for C₁₁H₇BrF₃NO₂: C, 41.02; H, 2.19; N, 4.35; Found: C, 41.27, H, 2.23, N, 4.26.

EXAMPLE 162

20



1,2-Dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylic acid

25

Step 1. Preparation of 2-amino-5-
(trifluoromethoxy)benzoic acid.

5-(Trifluoromethoxy)isatin (15.0 g, 65 mmol) and potassium hydroxide pellets (4 g) were mixed in water (35 mL) and cooled to 0 °C. With vigorous stirring, a solution of 30% aqueous hydrogen peroxide (11.7 g), potassium hydroxide pellets (5.8 g), and water (80 mL) was added drop-wise keeping the temperature below 10 °C. After stirring 1 hour at 0 °C, glacial acetic acid (22 mL) was added drop-wise, causing foaming and formation of a precipitate. The contents were stirred overnight and filtered to afford the 2-amino-5-trifluoromethoxybenzoic acid as an amber solid (12.5 g, 87 %). A small amount was recrystallized from ethyl acetate-hexanes to afford amber needles for an analytical sample and the remaining compound was used without further purification: mp 142.5-144.2 °C. ¹H NMR (CDCl₃, 300 MHz) 7.98 (s, 1H), 7.18 (d, 1H, *J* = 8.0 Hz) 6.62 (d, 1H, *J* = 8.0 Hz), 6.40 (brs, 2H). Anal. Calc'd for C₈H₆NO₃F₃: C, 43.45; H, 2.73; N, 6.33. Found: C, 43.40; H, 2.65; N, 6.35.

Step 2. Preparation of 2-amino-5-
(trifluoromethoxy)benzyl alcohol.

The 2-amino-5-trifluoromethoxybenzoic acid (2.0 g, 9.0 mmol) in tetrahydrofuran (20 mL) was added dropwise to borane methyl

sulfide complex (1.5 mL, 15.0 mmol) in tetrahydrofuran (5 mL). The reaction was refluxed overnight and allowed to cool. A solution of 30 % aqueous hydrogen peroxide (0.5 mL), 2.5 N sodium hydroxide (0.5 mL) and water (10 mL) was added drop-wise and the reaction stirred 0.5 hours. After diluting with diethyl ether (50 mL), the organic layer was washed with 0.1 M aqueous sodium metabisulfite (2 x 10 mL) and 2.5 N aqueous sodium hydroxide (2 x 10 mL). The organic layer was diluted further with hexanes (50 mL) and washed with brine (2 x 20 mL), dried over anhydrous Na_2SO_4 , and concentrated in vacuo leaving an amber oil (1.9 g) which solidified. The solid was recrystallized from ethyl acetate-hexanes to afford the 2-amino-5-trifluoromethoxybenzyl alcohol as a light amber solid (1.44 g, 77%): mp 75.9-77.6 °C. ^1H NMR (CDCl_3 , 300 MHz) 7.00 (m, 2H), 6.65 (d, 1H, $J = 8.0$ Hz), 4.05 (s, 2H), 3.25 (brs, 3H). ESHRMS m/z 208.0592 ($\text{M}+\text{H}^+$, Calc'd 208.0585). Anal. Calc'd for $\text{C}_8\text{H}_8\text{NO}_2\text{F}_3$: C, 46.39; H, 3.89; N, 6.76. Found: C, 46.61; H, 3.79; N, 6.71.

Step 3. Preparation of 2-amino-5-(trifluoromethoxy)-benzaldehyde.

The 2-amino-5-trifluoromethoxybenzyl alcohol from Step 2 (9.7 g, 47 mmol) and manganese (IV) oxide (21 g, 240 mmol) were

refluxed in chloroform (200 mL) for 1 hour. The contents were allowed to cool and filtered. The filtrate was concentrated in vacuo leaving an amber oil (8.2 g) which
5 solidified. The oil was distilled (bulb to bulb apparatus) at 50 °C (0.1 mm) to afford a yellow solid (7.2 g). The solid was recrystallized from hexanes to afford the desired 2-amino-5-(trifluoromethoxy)-
10 benzaldehyde as yellow crystals (4.4 g, 46%): mp. 42-44 °C. ¹H NMR (CDCl₃, 300 MHz) 9.81 (s, 1H), 7.36 (s, 1H), 7.20 (d, 1H, *J* = 9.0 Hz), 6.64 (d, 1H, *J* = 9.0 Hz). EIHRMS *m/z* 205.0328 (M⁺, Calc'd 205.0350).

15

Step 4. Preparation of ethyl 1,2-dihydro-6-(trifluoro-methoxy)-2(trifluoromethyl)-3-quinolinecarboxylate.

The 2-amino-5-(trifluoromethoxy)benzaldehyde from Step 3
20 (5.3 g, 26 mmol), anhydrous potassium carbonate (6.9 g, 50 mmol), and ethyl 4,4,4-trifluorocrotonate (7.7 mL, 50 mmol) were mixed in anhydrous dimethylformamide (50 mL)
25 and heated at 90 °C for 6 hours. The reaction was allowed to cool to room temperature and was partitioned between ethyl acetate (200 mL) and water (200 mL). The aqueous layer was extracted with more ethyl acetate (100
30 mL). The ethyl acetate extracts were combined and washed with brine (200 mL),

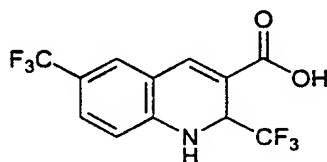
dried over MgSO_4 , and concentrated *in vacuo* yielding an oil (9.6 g). The oil was purified by flash chromatography on silica gel, eluting with ethyl acetate-hexanes (1:1). Fractions containing the desired product were combined, concentrated *in vacuo*, and the residue recrystallized from ethyl acetate-hexanes to afford the ethyl 1,2-dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylate as a yellow solid (4.05 g, 32%): mp. 123-125 °C. ^1H NMR (CDCl_3 , 300 MHz) 7.65 (s, 1H), 7.02 (m, 2H), 6.60 (m, 1H), 5.10 (m, 1H), 4.60 (brs, 1H), 4.28 (m, 2H), 1.32 (t, 3H, $J = 7.0$ Hz). ESHRMS m/z 356.0698 (M-H, Calc'd 356.0721). Anal. Calc'd for $\text{C}_{14}\text{H}_{11}\text{NO}_3\text{F}_6$: C, 47.34; H, 3.12; N, 3.94. Found: C, 47.37; H, 3.04; N, 3.93.

Step 5. Preparation of 1,2-dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylic acid.

The ethyl 1,2-dihydro-6-(trifluoromethoxy)-2-(trifluoromethyl)-3-quinolinecarboxylate from Step 4 (880 mg, 2.5 mmol) and 2.5 N aqueous sodium hydroxide (2 mL) were mixed in methanol (15 mL) and water (15 mL). The solution was heated on a steam bath for 2 hours. The reaction was allowed to cool to room temperature and was extracted with diethyl ether (50 mL). The aqueous

layer was acidified (pH = 1) with 3 N HCl and extracted with ethyl acetate (2 x 50 mL). The combined ethyl acetate extracts were dried over MgSO₄ and concentrated *in vacuo* leaving an oil. The oil was crystallized from cold dichloromethane-hexanes to afford the 1,2-dihydro-6-(trifluoromethoxy)-2(trifluoromethyl)-3-quinolinecarboxylic acid as yellow needles (0.727 g, 89%): mp 127.7-128.9 °C. ¹H NMR (CDCl₃, 300 MHz) 7.80 (s, 1H), 7.05 (m, 2H), 6.62 (d, 1H, *J* = 8.0 Hz), 5.13 (m, 1H), 4.62 (brs, 1H). ESHRMS *m/z* 326.0252 (M-H, Calc'd 326.0252). Anal. Calc'd for C₁₂H₇NO₃F₆: C, 44.05; H, 2.16; N, 4.28. Found: C, 43.89; H, 2.04; N, 4.24.

EXAMPLE 163



20

6-(Trifluoromethyl)-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid

Step 1. Preparation of N-(4-trifluoromethylphenyl)-2,2-dimethylpropanamide.

25

A solution of dichloromethane (200 mL), 4-aminobenzotrifluoride (32.0 g, 199 mmol)

and triethylamine (40 g, 396 mmol) was cooled to 0 °C under a dry nitrogen atmosphere. Trimethylacetyl chloride (32.9 g, 273 mmol) was added drop-wise over 2 hours, maintaining the temperature below 10 °C. After the addition, the contents were allowed to warm to room temperature for 2 hours. The reaction was washed with water (2 X 200 mL), saturated ammonium chloride solution (2 X 200 mL), dried over sodium sulfate and filtered. The solvent was removed in vacuo to afford a white solid, N-(4-trifluoromethylphenyl)-2,2-dimethylpropanamide (48.0 g, 98%): mp 157-159 °C. ¹H NMR (CDCl₃/300 MHz) 7.61 (ab, 4H, J = 8.7, Δv = 28.6 Hz), 7.47 (br s, 1H), 1.33 (s, 9H). ESHRMS m/z 246.1123 (M+H⁺, Calc'd 246.1106). Anal. Calc'd for C₁₂H₁₄F₃NO: C, 58.77; H, 5.75; N, 5.71. Found: C, 58.28; H, 5.79; N, 5.65.

Step 2. Preparation of N-[2-formyl-4-(trifluoromethyl) phenyl]-2,2-dimethylpropanamide.

A 1 liter three neck round bottom flask equipped with equalizing addition funnel, magnetic stirrer and temperature monitoring device was charged with N-(4-trifluoromethylphenyl)-2,2-dimethylpropanamide (10.13 g, 41.4 mmol) and anhydrous tetrahydrofuran (150 mL). The reaction was chilled to -78 °C under nitrogen

followed by slow addition of *n*-butyllithium (50 ml, 2.5 M in hexanes, 124 mmol) over 0.5 hours, such that the temperature of the reaction did not rise above -65°C. The contents were held at -78 °C for one hour, 0 °C for two hours, then chilled back to -78 °C. Excess *N,N*-dimethylformamide (100 mL, 1.37 mol) was added. The contents were warmed to room temperature and stirred for two hours.

10 Aqueous 1 N HCl was added to the reaction until the pH reached 1. The reaction was washed with water (2 X 200 mL), saturated ammonium chloride solution (2 X 200 mL), dried over sodium sulfate and filtered. The

15 filtrate was concentrated *in vacuo* to afford a yellow solid. The product was purified by flash chromatography (silica gel, 10% ethyl acetate, 90% hexanes) to yield, upon concentration of the appropriate fractions,

20 *N*-(2-formyl-4-trifluoromethylphenyl)-2,2-dimethylpropanamide as a solid (7.36 g, 65%): mp 69-73 °C. ¹H NMR (CDCl₃/300 MHz) 11.5 (br s, 1H), 9.99 (s, 1H), 8.67 (d, 1H, *J* = 8.8 Hz), 7.94 (d, 1H, *J* = 1.6 Hz), 7.83 (m, 1H),

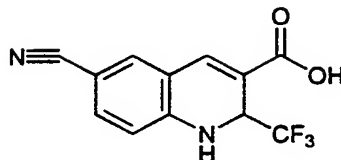
25 1.37 (s, 9H). ESHRMS *m/z* 274.1060 (*M* + *H*⁺, Calc'd 274.1055). Anal. Calc'd for C₁₃H₁₄F₃NO₂: C, 57.14; H, 5.16; N, 5.13. Found: C, 57.15; H, 5.43; N, 5.01.

Step 3. Preparation of ethyl 6-
(trifluoromethyl)-1,2-dihydro-2-
(trifluoromethyl)-3-quinolinecarboxylate.

To a suspension of N-(2-formyl-4-
5 (trifluoro-methylphenyl)-2,2-dimethyl
propanamide (Step 2) (921 mg, 3.7 mol) and
lithium hydride (115 mg, 14.5 mmol) in
dimethyl sulfoxide (10 mL) was added ethyl
4,4,4-trifluorocrotonate (2.83 g, 16.8 mmol)
10 and the contents warmed to 30 °C for 4 hours.
After the addition of ethyl acetate (50 mL),
the reaction was washed with water (2 X 30
mL), saturated ammonium chloride solution (2
X 30 mL), dried over sodium sulfate and
15 filtered. The filtrate was concentrated in
vacuo to afford a yellow solid. The product
was purified by flash chromatography (silica
gel, eluant: ethyl acetate-hexanes, 1:9) to
yield, upon concentration of the appropriate
20 fractions, ethyl 6-trifluoromethyl-1,2-
dihydro-2-(trifluoromethyl)-3-
quinolinecarboxylate as a yellow solid (65
mg, 5%): mp 138-139 °C. ¹H NMR (CDCl₃/300 MHz)
7.67 (s, 1H), 7.26 (s, 1H), 7.04 (d, 1H, J =
25 6.6 Hz), 6.62 (m, 1H), 5.14 (m, 1H), 4.60
(brs, 1H), 4.32 (m, 2H), 1.35 (t, 3H, J = 7.0
Hz). ESHRMS m/z 338.0592 (M-H, Calc'd
338.0616). Anal. Calc'd for C₁₃H₁₁F₃NO₂: C,
49.57; H, 3.27; N, 4.13; Found: C, 49.23; H,
30 2.81; N, 3.93.

Step 4. Preparation of ethyl 6-
trifluoromethyl-1,2-dihydro-2-
(trifluoromethyl)-3-quinolinecarboxylic acid.

Ethyl 6-trifluoromethyl-1,2-dihydro-2-
5 (trifluoromethyl)-3-quinolinecarboxylate from
Step 3 (45 mg, 0.13 mmol) was suspended in
methanol-tetrahydrofuran- water (10 mL,
7:2:1). Lithium hydroxide (24 mg, 0.52 mmol)
was added, and the mixture was gently heated
10 to reflux for two hours. The reaction was
cooled to room temperature and 1 N HCl added
until pH = 1. The organic solvent was removed
in vacuo to afford a suspension of a crude
yellow solid. Diethyl ether (20 mL) was
15 added, and the solution was washed with water
(2 X 20 mL), saturated ammonium sulfate (2 X
20 mL), dried over sodium sulfate and
filtered. The filtrate was concentrated *in*
vacuo to yield 6-trifluoromethyl-1,2-dihydro-
20 2-(trifluoromethyl)-3-quinolinecarboxylic
acid as a yellow solid, (0.041 g, 0.132 mmol,
99%): mp 150-156 °C. ¹H NMR (CD₃OD/300 MHz)
7.78 (s, 1H), 7.48 (s, 1H), 7.40 (m, 1H),
6.81 (m, 1H), 5.17 (m, 1H). ESHRMS *m/z*
25 310.0307 (M-H, Calc'd 310.0303).

EXAMPLE 164

5 **6-Cyano-1,2-dihydro-2-(trifluoromethyl)-3-
quinolinecarboxylic acid**

Step 1. Preparation of ethyl 6-cyano-1,2-
dihydro-2-(trifluoromethyl)-3-
10 quinolinecarboxylate.

N,N-Dimethylformamide (5 mL) was degassed with nitrogen for thirty minutes in a three neck round bottom flask equipped with a condenser, temperature monitoring, nitrogen
15 purge and heating mantle. Ethyl 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (Example 158) (0.522 g, 1.32 mmol) and zinc cyanide (0.102 g, 0.792 mmol) were added to the *N,N*-dimethylformamide
20 and stirred vigorously for ten minutes. Tetrakis(triphenyl-phosphine)palladium(0) (0.068 g, 0.53 mmol) was added and the contents gently warmed to 80 °C for 2 hours under a nitrogen atmosphere. Ethyl acetate
25 (20 mL) was added, followed by extraction with aqueous 2 N ammonium hydroxide (2 X 10 mL), water (2 X 10 mL), saturated ammonium chloride (2 X 10 mL), dried over sodium

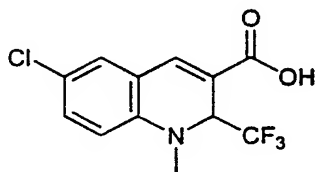
sulfate and solvent removed *in vacuo* to yield a yellow solid. The product was purified by flash chromatography (silica gel, ethyl acetate-hexanes, 3:1) to yield, upon
5 concentration of the appropriate fractions, ethyl 6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate as a yellow solid (188 mg, 48%): mp 211-212 °C. ¹H NMR (CDCl₃/300 MHz) 7.68 (s, 1H), 7.43 (m,
10 2H), 6.69 (d, 1H, *J* = 8.3 Hz), 5.22 (m, 1H), 4.98 (br s, 1H), 1.30 (m, 2H), 1.36 (t, 3H, *J* = 7.1 Hz). EIHRMS *m/z* 314.1147 (M+NH₄⁺, Calc'd 314.1116). Anal. Calc'd for C₁₄H₁₁F₃N₂O₂: C, 56.76; H, 3.74; N, 9.46. Found:
15 C, 56.44; H, 4.03; N, 9.29.

Step 2. Preparation of 6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid.

To a suspension of ethyl 6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (140 mg, 0.45 mmol) in methanol-tetrahydrofuran-water (10 mL, 7:2:1) was added lithium hydroxide (76 mg, 0.91 mmol) and the mixture gently heated to reflux
25 for two hours. The contents were cooled to room temperature and 1 N aqueous hydrochloric acid added until pH = 1. The organic solvent was removed *in vacuo* to afford a suspension of crude yellow solid. Diethyl ether (20 mL)
30 was added, and the solution was washed with water (2 X 20 mL), saturated ammonium sulfate

(2 X 20 mL), dried over sodium sulfate and filtered. The filtrate was concentrated in vacuo to yield 6-cyano-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid as a yellow solid, (116 mg, 95%): mp 238-240 °C. ¹H NMR (CD₃OD/300 MHz) 7.75 (s, 1H), 7.56 (m, 1H), 7.43 (m, 1H), 6.79 (d, 1H, J = 8.5 Hz) 5.19 (q, 1H, J = 7.1 Hz). EIHRMS m/z 267.0405 (M-H, Calc'd 267.0381). Anal. Calc'd for C₁₄H₁₁F₃N₂O₂: C, 53.74; H, 2.63; N, 10.45. Found: C, 53.99; H, 2.89; N, 10.19.

EXAMPLE 165



6-Chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid

Step 1. Preparation of ethyl 6-chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinolinecarboxylate.

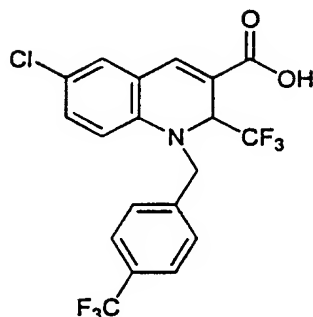
Ethyl 6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (Example 157, Step 2) (1.28 g, 4.21 mmol), tetrabutylammonium iodide (0.36 g, 0.92 mmol) and aqueous NaOH (50%, 2 mL) were stirred

vigorously in methylene chloride (40 mL). Dimethyl sulfate (2.12 g, 16.84 mmol) was added to the dark orange mixture via syringe over 2 hours. Hexane (5 mL) was added, and
5 the solution was washed with water (2 X 20 mL), saturated ammonium chloride solution (2 X 20 mL), dried over sodium sulfate and filtered. The filtrate was concentrated *in vacuo* to afford the crude ester as a yellow
10 solid. The solid was purified by flash chromatography (silica gel, 50 g; ethyl acetate-hexanes, 1:19) to yield, upon concentration of the appropriate fractions, ethyl 6-chloro-1,2-dihydro-1-methyl-2-
15 (trifluoromethyl)-3-quinoline-carboxylate (1.2 g, 90% yield): mp 118-120 °C. ¹H NMR (CD₃OD/300 MHz) 7.71 (s, 1H), 7.30-7.26 (m, 2H), 6.77-6.74 (m, 1H), 5.12 (q, 1H, *J* = 6.8 Hz), 4.44-4.22 (m, 2H), 3.18 (s, 3H), 1.35
20 (t, 3H, *J* = 7.0 Hz). EIHRMS *m/z* 320.0701 (M-H, Calc'd 320.0665) Anal. Calc'd for C₁₄H₁₃F₃NO₂Cl: C, 52.60; H, 4.10; N, 4.38. Found: C, 52.57; H, 4.14; N, 4.32.

25 Step 2. Preparation of 6-chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid

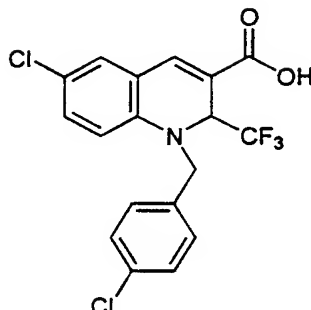
Ethyl 6-chloro-1,2-dihydro-1-methyl-2-(trifluoro-methyl)-3-quinolinecarboxylate (1.21 g, 3.78 mmol) was
30 suspended in methanol-tetrahydrofuran-water

(20 mL, 7:2:1). Lithium hydroxide (0.262 g, 6.24 mmol) was added, and the mixture was gently heated to reflux for two hours. The reaction was cooled to room temperature and 1
5 N HCl added until pH = 1. The organic solvent was removed *in vacuo* to afford a suspension of crude yellow solid. Diethyl ether (20 mL) was added, and the resulting solution was washed with water (2 X 20 mL), saturated
10 ammonium chloride (2 X 20 mL), dried over sodium sulfate and filtered. The filtrate was concentrated *in vacuo* to afford the product as a yellow solid, 6-chloro-1,2-dihydro-1-methyl-2-(trifluoromethyl)-3-quinoline-
15 carboxylic acid. (1.08 g, 98% yield): mp 208-209 °C. ¹H NMR (CD₃OD/300 MHz) 7.69 (d, 1H, *J* = 2.5 Hz), 7.28-7.24 (m, 2H), 6.73 (dd, 1H, *J* = 9.5, 2.5 Hz), 5.13 (q, 1H, *J* = 7.0), 3.16 (s, 3H). Anal. Calc'd for C₁₂H₉F₃NO₂Cl: C, 49.42; H, 3.11; N, 4.80; Cl, 12.16. Found: C, 49.88; H, 3.29; N, 4.59; Cl, 12.42
20

EXAMPLE 166

5 **6-Chloro-1,2-dihydro-2-(trifluoromethyl)-1-
 [[4-(trifluoromethyl)phenyl]methyl]-3-
 quinolinecarboxylic acid**

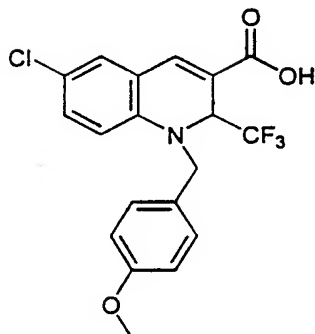
 The 1,2-dihydro-3-quinolinecarboxylic
10 acid was prepared by a procedure similar to
 that described in Example 165: mp 229-231 °C.
 ¹H NMR (CD₃OD/300 MHz) 7.77 (s, 1H), 7.58 (d,
 2H, *J* = 8.0 Hz), 7.39 (d, 2H, *J* = 8.0 Hz),
 7.30 (d, 1H, *J* = 2.4), 7.13 (dd, 1H, *J* = 8.9,
 2.4 Hz), 6.75 (d, 1H, *J* = 8.9 Hz), 5.27 (q,
15 1H, *J* = 7.0 Hz), 4.90 (ab, 2H, *J* = 16.7 Hz,
 Δ*v* = 95.2 Hz). EIHRMS *m/z* 434.0401 (Calc'd
 for M-H 434.0383) Anal. Calc'd for
 C₁₉H₁₄F₆NO₂Cl: C, 52.13; H, 3.22; N, 3.22;
20 Found: C, 52.36; H, 2.91; N, 3.21.

EXAMPLE 167

5 **6-Chloro-1-[(4-chlorophenyl)methyl]-1,2-**
 dihydro-2-(trifluoromethyl)-3-
 quinolinecarboxylic acid

 The 1,2-dihydro-3-quinolinecarboxylic
10 acid was prepared by a procedure similar to
 that described in Example 165: mp 250-253 °C.
 ¹H NMR (CD₃OD/300 MHz) 7.74 (s, 1H), 7.32-7.13
 (m, 6H), 6.76 (d, 1H, *J* = 8.7 Hz), 5.22 (q,
 1H, *J* = 7.0 Hz), 4.81 (ab, 2H, *J* = 16.3 Hz,
15 Δ*v* = 54.7 Hz). ESHRMS *m/z* 400.0105 (M-H,
 Calc'd 400.0119).

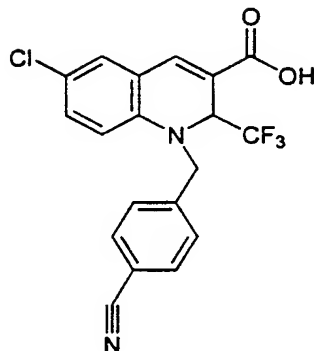
EXAMPLE 168



5 **6-Chloro-1,2-dihydro-2-(trifluoromethyl)-1-
 [[4-(methoxy)phenyl]methyl]-3-
 quinolinecarboxylic acid**

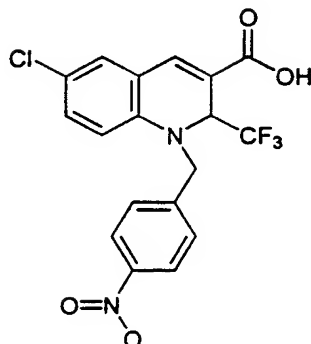
The 1,2-dihydro-3-quinolinecarboxylic
10 acid was prepared by a procedure similar to
that described in Example 165: mp 196-197 °C.
¹H NMR (CD₃OD/300 MHz) 7.71 (s, 1H), 7.27-7.26
(m, 1H), 7.18-7.12 (m, 3H), 6.85-6.81 (m,
3H), 5.16 (q, 1H, *J* = 7.1 Hz), 4.69 (ab, 2H,
15 *J* = 15.3 Hz, Δ*v* = 111.8 Hz), 3.73 (s, 3H).
ESHRMS *m/z* 396.0625 (M-H, Calc'd 396.0614).
Anal. Calc'd for C₁₉H₁₄F₃NO₂Cl: C, 52.13; H,
3.22; N, 3.22. Found: C, 52.36; H, 2.91; N,
3.21.

20

EXAMPLE 169

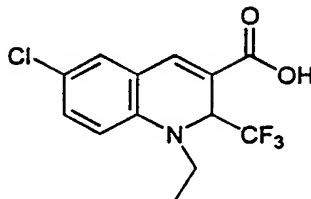
5 **6-Chloro-1-[(4-cyanophenyl)methyl]-1,2-**
 dihydro-2-(trifluoromethyl)-3-
 quinolinecarboxylic acid

10 The 1,2-dihydro-3-quinolinecarboxylic
 acid was prepared by a procedure similar to
 that described in Example 165: mp 258-260 °C.
 ¹H NMR (CD₃OD/300 MHz) 7.78 (s, 1H), 7.66 (d,
 2H, *J* = 8.2 Hz), 7.41 (d, 2H, *J* = 8.2 Hz),
 15 7.33 (d, 1H, *J* = 2.7 Hz), 7.15 (dd, 1H, *J* =
 8.7, 2.7 Hz), 6.71 (d, 1H, *J* = 8.7 Hz), 5.31
 (q, 1H, *J* = 7.0 Hz), 4.94 (ab, 2H, *J* = 17.1,
 Δ*v* = 91.8 Hz). ESHRMS *m/z* 391.0443 (M-H,
 Calc'd 391.0461). Anal. Calc'd for
 20 C₁₉H₁₂F₃N₂O₂Cl + 0.53 % H₂O: C, 57.79; H, 3.55;
 N, 7.09; Found: C, 57.26; H, 3.17; N, 6.78.

EXAMPLE 170

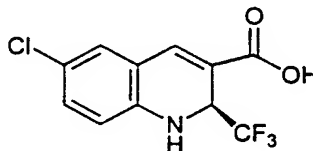
5 **6-Chloro-1,2-dihydro-1-[(4-nitrophenyl)methyl]-2-(trifluoromethyl)-3-quinolinecarboxylic acid**

The 1,2-dihydro-3-quinolinecarboxylic
10 acid was prepared by a procedure similar to that described in Example 165: mp 225-228 °C. ¹H NMR (CD₃OD-3% TFA/300 MHz) 8.14 (d, 2H, *J* = 8.8 Hz), 7.77 (s, 1H), 7.42 (d, 2H, *J* = 8.8 Hz), 7.29 (d, 1H, *J* = 2.4 Hz), 7.11 (dd, 1H,
15 *J* = 8.9, 2.4 Hz), 6.67 (d, 1H, *J* = 8.9 Hz), 5.27 (q, 1H, *J* = 6.8 Hz), 4.93 (ab, 2H, *J* = 17.2 Hz, Δ*v* = 95.0 Hz). ESHRMS *m/z* 411.0327 (M-H, Calc'd 411.0359).

EXAMPLE 171

5 **6-Chloro-1,2-dihydro-1-ethyl-2-**
(trifluoromethyl)-3-quinolinecarboxylic acid

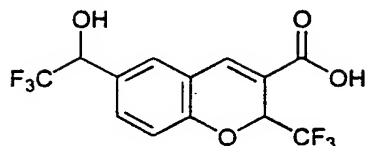
The 1,2-dihydro-3-quinolinecarboxylic acid was prepared by a procedure similar to
that described in Example 165: mp 201-202 °C.
10 ¹H NMR (CD₃OD/300 MHz) 7.67 (s, 1H), 7.25-7.22 (m, 2H), 6.86 (d, 1H, *J* = 8.7 Hz), 5.21 (q, 1H, *J* = 7.0 Hz), 3.81-3.71 (m, 1H), 3.47-3.39 (m, 1H), 1.20 (t, 3H, *J* = 7.2 Hz). ESHRMS *m/z*
15 304.0360 (M-H, Calc'd 304.0352).

EXAMPLE 172

20 **(S)-6-chloro-1,2-dihydro-2-(trifluoromethyl)-**
3-quinolinecarboxylic acid

To a solution of 6-chloro-1,2-dihydro-2-
25 (trifluoromethyl)-3-quinolinecarboxylic acid

(Example 157) (6.75 g, 24.3 mmol) in ethyl acetate (25 mL) was added (S)-(-)- α -methylbenzylamine (1.50 g, 12.2 mmol). To the resulting solution was added hexanes (50 mL) with mixing. Stirring was discontinued and the reaction held static at room temperature for 16 hours during which time yellow crystals formed. The crystals were collected and washed with ethyl acetate-hexanes (100 mL, 1:2). The resulting yellow solid (932 mg) was dissolved in ethyl acetate (20 mL) and extracted with 1 N HCl (3 x 10 mL). The organic layer was dried over sodium sulfate and solvent removed at reduced pressure. The (S)-6-chloro-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylic acid was obtained as a yellow solid (648 mg, 10% yield). mp 173-176 °C. ¹H NMR (acetone-d₆, 300 MHz) 7.80 (s, 1H), 7.35 (d, 1H, *J* = 2.2 Hz), 7.18 (d, 1H, *J* = 8.0, *J* = 2.2 Hz), 6.86 (d, 1H, *J* = 8.0 Hz), 6.60 (brs, 1H), 5.20 (m, 1H). Anal. Calc'd. for C₁₁H₇NO₂F₃Cl C, 47.40; H, 2.54; N, 5.40. Found C, 47.49; H, 2.60; N, 4.98. The compound was determined to have an optical purity greater than 90% ee. Optical purity was determined by HPLC as described in Example 66.

EXAMPLE 173

5 **6-(2,2,2-Trifluoro-1-hydroxyethyl)-2-**
 (trifluoromethyl)-2H-1-benzopyran-3-
 carboxylic acid

Step 1. Preparation of ethyl 6-(1-hydroxy-
10 2,2,2-trifluoroethyl)-2-(trifluoromethyl)-2H-
1-benzopyran-3-carboxylate.

The aldehyde (Example 75, Step 1) (0.89 g, 3.0 mmol) was cooled to 0 °C and treated with a 0.5 M solution of
15 trimethyl(trifluoromethyl)silane (8.4 mL, 4.2 mmol) and four drops of a 1.0M solution of tetrabutylammonium fluoride was added. The reaction was allowed to warm to room temperature and stirred for 21.1 hours. The
20 reaction was quenched with 3 N HCl, extracted with ethyl acetate, washed with water, brine, dried over MgSO₄, and concentrated *in vacuo* to give a brown oil (1.02 g). This oil was purified by flash chromatography over silica
25 gel, eluting with 10% ethyl acetate/hexanes to afford a brown oil (0.77 g, 58%): ¹H NMR (CDCl₃/300 MHz) 7.72 (d, 1H, *J* = 3.4 Hz), 7.34 (m, 2H), 6.99 (d, 1H, *J* = 8.5 Hz), 5.71 (q,

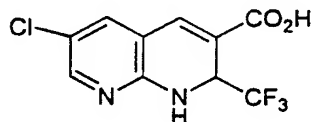
1H, $J = 6.8$ Hz), 4.83 (q, 1H, $J = 6.4$ Hz), 4.33 (m, 2H), 1.35 (t, 3H, $J = 7.1$ Hz), 0.11 (s, 9H). FABLRMS m/z 443 (M+H).

5 Step 2. Preparation of 6-(1-hydroxy-2,2,2-trifluoroethyl)-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 1 (0.15g 0.34 mmol) was dissolved in THF (2 mL) and ethanol (2 mL), treated with 2.5 N NaOH (1 mL, 2.5 mmol), and stirred at room temperature for 18.6 hours. The reaction mixture was concentrated *in vacuo*, acidified with 3 N HCl, extracted with ethyl acetate, washed with 3 N HCl, brine, dried over $MgSO_4$, and concentrated *in vacuo* to give a yellow oil which was recrystallized from ethyl acetate/hexane to yield a white solid (0.03 g, 25%): mp 114-120 °C. 1H NMR (acetone- d_6 /300 MHz) 7.94 (s, 1H), 7.65 (s, 1H), 7.60 (dd, 1H, $J = 8.2$ Hz 2.0 Hz), 7.11 (d, 1H, $J = 8.3$ Hz), 5.87 (q, 1H, $J = 7.0$ Hz), 5.24 (q, 1H, $J = 7.0$ Hz). FABLRMS m/z 341 (M-H). ESHRMS m/z 341.0241 (M-H, Calc'd 341.0249).

25

EXAMPLE 174



6-Chloro-2-(trifluoromethyl)-1,2-dihydro[1,8]naphthyridine-3-carboxylic acid

Step 1. Preparation of N-[5-chloropyridin-2-yl]-2,2-dimethylpropanamide.

To 2-amino-5-chloropyridine (10.0 g, 0.078 mol) (Aldrich) and triethylamine (12 mL, 0.086 mol) in methylene chloride (200 mL), at 0 °C, was added dropwise trimethylacetyl chloride in methylene chloride (15 mL). The reaction was allowed to warm to room temperature while stirring overnight. The resulting mixture was washed with water, brine, and was dried over MgSO₄ and filtered. Concentration of the filtrate in vacuo provided a colorless oil (19.2 g). The oil was dissolved in hexanes and cooled causing the precipitation of a solid. The solid was collected by filtration affording the amide as a white solid (14.96 g, 90%): mp 51.4 - 53.4 °C. ¹H NMR (CDCl₃/300 MHz) 8.25-8.15(m, 2H), 8.00 (br s, 1H), 7.68-7.60 (m, 1H), 1.28 (s, 9H). Anal. Calc'd for C₁₀H₁₃N₂OCl: C, 56.47; H, 6.16; N, 13.17 Found: C, 56.72; H, 6.34; N, 12.88.

Step 2. Preparation of N-[5-chloro-3-formylpyridin-2-yl]-2,2-dimethylpropanamide.

To a chilled (-78 °C), stirred solution of the amide (Step 1) (5.0 g, 0.024 mole) in tetrahydrofuran (100 mL) was added t-butyl

lithium (1.7M in pentane, 32.4 mL, 0.055 mole) dropwise. Dimethylformamide (2.3 mL, 0.03 mole) was added dropwise at -78 °C over 3 hours and the mixture allowed to warm room temperature. The reaction was quenched with ice water (200 mL) and extracted with ethyl acetate. The resulting organic phase was dried over MgSO₄ and was concentrated in vacuo to a volume of 20 mL. A white solid precipitated which was collected by filtration yielding the formylated product (3.24 g, 56%): mp 168.7-170.8 °C. ¹H NMR (CDCl₃/300 MHz) 10.60 (br s, 1H), 9.88 (s, 1H), 8.57 (s, 1H), 8.00 (s, 1H), 1.28 (s, 9H). Anal. Calc'd for C₁₁H₁₃N₂O₂Cl: C, 54.89; H, 5.44; N, 11.64 Found: C, 54.87; H, 5.42; N, 11.40.

20 Step 3. Preparation 2-amino-5-chloro-3-formylpyridine.

The product of Step 2 (2.7 g, 11 mmol) and 3 N HCl (50 mL) were heated at reflux for 2 hours. The reaction was allowed to cool to room temperature and was concentrated in vacuo yielding a light yellow solid (2.1 g). The solid was partitioned between ethyl acetate and 2.5 N NaOH solution. The ethyl acetate layer was dried over MgSO₄ and concentrated in vacuo providing a solid (1.7 g). The solid was recrystallized from ethyl acetate to give the desired substituted

pyridine as yellow needles (1.2 g, 68%): mp
176.1-177.3 °C. ¹H NMR (CDCl₃/300 MHz) 9.80
(s, 1H), 8.21 (s, 1H), 7.75 (s, 1H), 6.75 (br
s, 2H). Anal. Calc'd for C₆H₅N₂OCl: C, 46.03;
5 H, 3.22; N, 17.89 Found: C, 45.90; H, 3.24;
N, 17.80.

Step 4. Preparation of ethyl 6-chloro-2-
10 (trifluoromethyl)-1,2-
dihydro[1,8]naphthyridine-3-carboxylate.

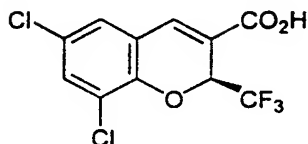
The substituted pyridine from Step 3
(1.7 g, 11 mmol), anhydrous potassium
carbonate (3.0 g, 22 mmol), and ethyl 4,4,4-
15 trifluorocrotonate (3.3 mL, 22 mmol) were
mixed in anhydrous dimethylformamide (20 mL)
and heated at 80 °C for 2 hours. The reaction
was allowed to cool to room temperature and
was partitioned between ethyl acetate (100
20 mL) and water (100 mL). The aqueous layer
was extracted with more ethyl acetate (100
mL). The combined organic extracts were
washed with brine (100 mL), dried over MgSO₄,
and concentrated *in vacuo* yielding a waxy
25 amber solid The solid was triturated with
diethyl ether providing the ester as a yellow
solid (613 mg, 18%). A small amount was
recrystallized from ethyl acetate for
analytical data: mp 180.1-181.9 °C. ¹H NMR
30 (CDCl₃/300 MHz) 7.99 (s, 1H), 7.61 (s, 1H),

7.39 (s, 1H), 6.00 (br s, 1H), 5.33-5.20 (m, 1H), 4.40-4.23 (m, 2H), 1.40-1.30 (m, 3H).

Anal. Calc'd for $C_{12}H_{10}N_2O_2F_3Cl$: C, 47.00; H, 3.29; N, 9.13 Found: C, 46.83; H, 3.03; N, 9.18.

Step 5. Preparation of 6-chloro-2-(trifluoromethyl)-1,2-dihydro[1,8]naphthyridine-3-carboxylic acid.

10 The ester from Step 4 (1.3 g, 4.4 mmol) and 2.5 N sodium hydroxide solution (3.5 mL, 9 mmol) were mixed in tetrahydrofuran (25 mL), methanol (10 mL), and water (25 mL). The mixture was heated at 50 °C for 4 hours, 15 allowed to cool to room temperature, and was concentrated *in vacuo* to remove the tetrahydrofuran and methanol. The resulting aqueous solution was washed with diethyl ether (2 x 100 mL). The aqueous phase was 20 acidified with 3 N HCl causing the precipitation of a yellow solid (1.1 g). The solid was triturated with ethanol-acetone and collected by vacuum filtration providing the title compound as a yellow solid (276 mg, 25 23%): mp 287.4-288.4 °C. 1H NMR (acetone- d_6 /300 MHz) 11.50 (br s, 1H), 8.03 (s, 1H), 7.83 (s, 1H), 7.75 (s, 1H), 7.28 (br s, 1H), 5.42-5.30 (m, 1H). Anal. Calc'd for $C_{10}H_6N_2O_2F_3Cl$: C, 43.11; H, 2.17; N, 10.05 30 Found: C, 42.88; H, 2.03; N, 10.06.

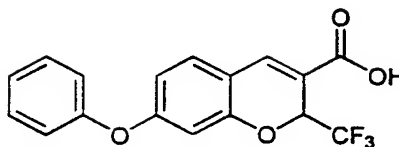
EXAMPLE 175

(S)-6,8-Dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

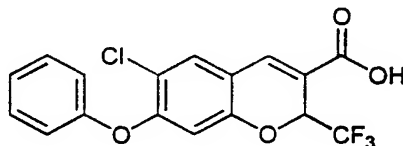
6,8-Dichloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid (Example 32) (300 g, 1.04 mol) was added to ethyl acetate (750 mL). The mixture was stirred for 5 minutes, warmed to 70 °C and held at this temperature for 5 minutes. The resulting solution was cooled to 50 °C and (s)-(-)- α -methylbenzylamine (58 g, 0.48 mol) was added. Heptane (1880 mL) was added and the mixture stirred for 0.5 hour, then stirring was discontinued. The reaction was allowed to cool to 22 °C and stand for 8 hours. The salt crystallized during this time and was collected by vacuum filtration. The solid was washed with ethyl acetate-heptane (1:3, 2 X 50 mL). The solid obtained was dried at 40 °C under vacuum (20 mm) for 24 hours to give the salt (35 g, 16 %).

A three-neck 2 L round bottom flask was purged with nitrogen and was charged with deionized water (750 mL) and the salt (103 g, 0.24 mole; This material was obtained using a

similar procedure to that described above). To the resulting stirred suspension was added concentrated HCl (37 mL) drop-wise over 0.5 hours with good stirring below 20 °C causing the free carboxylic acid to precipitate. After stirring for 2 hours, the suspension was vacuum filtered and the solid washed with deionized water (5 X 50 mL; until the washings were neutral). The solid was dried at 40 °C under vacuum (20 mm) for 12 hours yielding the title compound as a solid (74 g, 100%): mp 166.0-168.4 °C. ¹H NMR (acetone-*d*₆/300 MHz) 7.94 (s, 1H), 7.60 (s, 2H), 6.04 (q, 1H, *J* = 6.8 Hz). ESHRMS *m/z* 310.9489 (M-H, Calc'd 310.9450). This compound was determined to have an optical purity of greater than 90% ee. The optical purity was determined by the method described in Example 66.

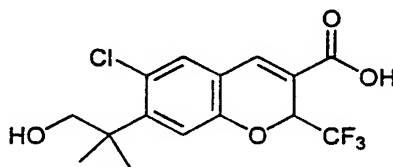
EXAMPLE 176**7-Phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

3-Phenoxyphenol was converted to the title compound by a procedure similar to that described in Example 2: mp 180.0-181.0 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.87 (s, 1H), 7.47 (m, 3H), 7.26 (dt, 1H, *J* = 7.6 Hz, 1.4 Hz), 7.15 (d, 2H, *J* = 7.7 Hz), 6.68 (dd, 1H, *J* = 8.5 Hz 2.3 Hz), 6.58 (s, 1H), 5.81 (q, 1H, *J* = 7.1 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.5 (d, *J* = 7.2 Hz). FABLRMS *m/z* 335 (M-H). ESHRMS *m/z* 335.0542 (M-H, Calc'd 335.0531). Anal. Calc'd for C₁₇H₁₁F₃O₄: C, 60.72; H, 3.30. Found: C, 60.55; H, 3.29.

EXAMPLE 177**6-Chloro-7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

3-Phenoxyphenol was converted to the title compound by a procedure similar to that described in Example 100: mp 220.2-221.8 °C. ¹H NMR (acetone-d₆/300MHz) 7.91 (s, 1H), 7.72 (s, 1H), 7.50 (m, 2H), 7.28 (t, 1H, *J* = 7.5 Hz), 7.15 (d, 2H, *J* = 7.7 Hz), 6.53 (s, 1H), 5.81 (q, 1H, *J* = 7.1 Hz). ¹⁹F NMR (acetone-d₆/282MHz) -79.4 (d, *J* = 7.2 Hz). FABLRMS *m/z* 369 (M-H). ESHRMS *m/z* 369.0157 (M-H, Calc'd 369.0141). Anal. Calc'd for C₁₇H₁₀ClF₃O₄: C, 55.08; H, 2.72; Cl, 9.56. Found: C, 54.94; H, 2.60; Cl, 9.84.

EXAMPLE 178



6-Chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of methyl 3-methoxyphenylacetate.

Trimethylsilyl chloride (182 g, 1.68 mol) was added dropwise to a solution of 3-methoxyphenylacetic acid (127 g, 0.77 mol) in methanol (1.0 L) over 1.1 hours. The reaction was stirred at room temperature for 17.25 hours, concentrated *in vacuo*, dissolved in ethyl acetate,

dried over MgSO_4 , and concentrated *in vacuo*, to give a brown oil (133 g, 97%): ^1H NMR (CDCl_3 /300MHz) 7.24 (t, 1H, $J = 7.5$ Hz), 6.83 (m, 3H), 3.80 (s, 3H), 3.69 (s, 3H), 3.60 (s, 2H).

Step 2. Preparation of methyl 2-(3-methoxyphenyl)-2-methylpropionate.

A 2L flask was charged with a 60% dispersion of sodium hydride in mineral oil (48.7 g, 1.22 mol) and THF (0.4 L). A solution of the ester from Step 1 (73.2 g, 0.41 mol) in THF (0.2 L) was added over one hour and the reaction was stirred for an additional 2.67 hours. The reaction was cooled in ice and treated with methyl iodide (136.8 g, 0.96 mol) over a period of 0.67 hours, and then stirred at room temperature for 14.75 hours. The reaction was poured into a mixture of 3N HCl (0.3 L) and ice (0.8 L), extracted with ethyl acetate (0.8 L), washed with 10% NaHSO_3 , water, brine, dried over MgSO_4 and concentrated *in vacuo*, to give a brown oil which was distilled under vacuum (2.2 mm Hg, 97-112 °C) to give a clear oil (50.9 g, 60%): ^1H NMR (CDCl_3 /300MHz) 7.25 (t, 1H, $J = 8.0$ Hz), 6.89 (m, 2H), 6.81 (dd, 1H, $J = 8.0$ Hz, 2.6 Hz), 3.81 (s, 3H), 3.66 (s, 3H), 1.57 (s, 6H).

Step 3. Preparation of methyl 2-(3-hydroxyphenyl)-2-methylpropionate.

The ester from Step 2 (29.1 g, 140 mmol) was dissolved in CH_2Cl_2 (170 mL) and cooled to -65 °C. The reaction was treated with a 1.0 M solution of

boron tribromide in CH_2Cl_2 (180 mL, 180 mmol). The reaction was stirred at -65°C for 1.6 hours, then the cooling bath was removed and the reaction was warmed to room temperature. The reaction was added to ice water and the layers were separated. The organic layer was concentrated *in vacuo*, dissolved in ethyl acetate, washed with saturated NaHCO_3 , brine, dried over MgSO_4 and concentrated *in vacuo*, to give methyl 2-(3-hydroxyphenyl)-2-methylpropionate as a brown oil (10.6 g, 42%): ^1H NMR (CDCl_3 /300MHz) 7.19 (m, 1H), 6.90 (d, 1H, $J = 7.9$ Hz), 6.84 (m, 1H), 6.75 (d, 1H, $J = 8.1$ Hz), 3.66 (s, 3H), 1.56 (s, 6H). The saturated NaHCO_3 layer was acidified with concentrated HCl, extracted with ethyl acetate, washed with brine, dried over MgSO_4 and concentrated *in vacuo*, to give 2-(3-hydroxyphenyl)-2-methylpropionic acid as a brown oil (15.8 g, 42 %): ^1H NMR (CDCl_3 /300MHz) 7.20 (m, 1H), 6.97 (dd, 1H, $J = 7.9$ Hz, 0.8 Hz), 6.88 (m, 1H), 6.75 (dd, 1H, $J = 8.1$ Hz, 0.8 Hz), 1.57 (s, 6H).

Step 4. Preparation of 2-(3-hydroxyphenyl)-2-methylpropionic acid.

The ester (Step 3) was hydrolyzed to the carboxylic acid via a method similar to that described in Example 1, Step 2: ^1H NMR (CDCl_3 /300MHz) 7.20 (m, 1H), 6.97 (dd, 1H, $J = 7.9$ Hz, 0.8 Hz), 6.88 (m, 1H), 6.75 (dd, 1H, $J = 8.1$ Hz, 0.8 Hz), 1.57 (s, 6H).

Step 5. Preparation of 3-(2-hydroxy-1,1-dimethylethyl)phenol.

The carboxylic acid obtained from several lots obtained from Step 3 and Step 4 (31.6 g, 175 mmol) was dissolved in THF (150 mL), cooled to 0 °C, and treated with borane methyl sulfide complex (33 mL, 348 mmol). The reaction was warmed to room temperature and stirred for 7.5 hours. The reaction was added to 3N HCl, extracted with ethyl acetate, washed with brine, dried over MgSO₄ and concentrated *in vacuo*, to give a brown oil (18.6 g, 64%): ¹H NMR (CDCl₃/300MHz) 7.21 (m, 1H), 6.96 (d, 1H, *J* = 7.9 Hz), 6.86 (t, 1H, *J* = 2.0 Hz), 6.71 (dd, 1H, *J* = 7.8 Hz, 2.4 Hz), 3.60 (s, 2H), 1.31 (s, 6H).

Step 6. Preparation of 4-(2-hydroxy-1,1-dimethylethyl)salicylaldehyde.

The phenol (Step 5) was converted to the salicylaldehyde via a method similar to that described in Example 2, Step 1: mp 80.4-81.9 °C. ¹H NMR (CDCl₃/300MHz) 11.00 (s, 1H), 9.87 (s, 1H), 7.51 (d, 1H, *J* = 8.3 Hz), 7.08 (dd, 1H, *J* = 8.3 Hz, 1.4 Hz), 7.02 (d, 1H, *J* = 1.4 Hz), 3.66 (s, 2H), 1.34 (s, 6H).

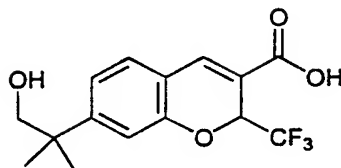
Step 7. Preparation of ethyl 6-chloro-7-(2-acetoxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

The salicylaldehyde (Step 6) was chlorinated via a method similar to that described in Example 9

yielding 5-chloro-4-(2-acetoxy-1,1-dimethylethyl)salicylaldehyde. This salicylaldehyde was converted to the substituted ethyl 2H-1-benzopyran-3-carboxylate via a method similar to that described in Example 1, Step 1: ^1H NMR ($\text{CDCl}_3/300\text{MHz}$) 7.63 (s, 1H), 7.22 (s, 1H), 7.02 (s, 1H), 5.70 (q, 1H, $J = 6.8$ Hz), 4.46 (s, 2H), 4.32 (m, 2H), 1.97 (s, 3H), 1.49 (s, 6H), 1.36 (t, 3H, $J = 7.0$ Hz). ^{19}F NMR ($\text{CDCl}_3/282\text{MHz}$) -78.8 (d, $J = 6.5$ Hz).

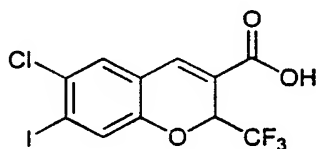
Step 8. Preparation of 6-chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester obtained from Step 7 was hydrolyzed to the carboxylic acid via a method similar to that described in Example 1, Step 2: mp 203.8-205.4 °C. ^1H NMR (acetone- $d_6/300\text{MHz}$) 7.87 (s, 1H), 7.50 (s, 1H), 7.21 (s, 1H), 5.83 (q, 1H, $J = 7.0$ Hz), 3.90 (s, 2H), 1.46 (s, 6H). ^{19}F NMR (acetone- $d_6/282\text{MHz}$) -79.4 (d, $J = 7.2$ Hz). ESHRMS m/z 349.0443 (M-H, Calc'd 349.0454). Anal. Calc'd for $\text{C}_{15}\text{H}_{14}\text{ClF}_3\text{O}_4$: C, 51.37; H, 4.02; Cl, 10.11. Found: C, 51.24; H, 4.02; Cl, 10.45.

EXAMPLE 179

7-(2-Hydroxy-1,1-dimethylethyl)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The salicylaldehyde from Example 178, Step 6 was converted to the title compound via a procedure similar to that described in Example 11, Steps 2 and 3: mp 228.9-230.0 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.86 (s, 1H), 7.37 (d, 1H, *J* = 8.1 Hz), 7.17 (dd, 1H, *J* = 8.1 Hz, 1.6 Hz), 7.08 (s, 1H), 5.81 (q, 1H, *J* = 7.2 Hz), 3.60 (s, 2H), 1.30 (s, 6H). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.5 (d, *J* = 7.2 Hz). ESHRMS *m/z* 315.0843 (M-H, Calc'd 315.0844). Anal. Calc'd for C₁₇H₁₀ClF₃O₄: C, 56.96; H, 4.78. Found: C, 56.91; H, 4.85.

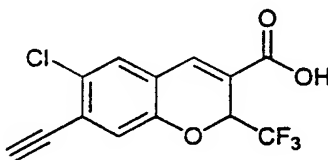
EXAMPLE 180

6-Chloro-7-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

3-Iodophenol was converted to the title compound by a procedure similar to that described in Example 100: mp 224.9-225.9 °C. ¹H NMR

(acetone- d_6 /300MHz) 7.92 (s, 1H), 7.72 (s, 1H), 7.68 (s, 1H), 5.91 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.3 (d, $J = 7.2$ Hz). FABLRMS m/z 403 (M-H). ESHRMS m/z 402.8836 (M-H, Calc'd 402.8846). Anal. Calc'd for $\text{C}_{11}\text{H}_5\text{ClF}_3\text{IO}_3$: C, 32.66; H, 1.25; Cl, 8.76. Found: C, 32.65; H, 0.97; Cl, 8.39.

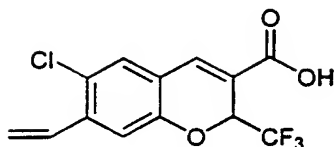
EXAMPLE 181



6-Chloro-7-ethynyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from ethyl 6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 180) via a procedure similar to that described in Example 118: mp >300 °C. ^1H NMR (acetone- d_6 /300MHz) 7.91 (s, 1H), 7.68 (s, 1H), 7.24 (s, 1H), 5.90 (q, 1H, $J = 7.0$ Hz), 4.23 (s, 1H). ^{19}F NMR (acetone- d_6 /282MHz) -79.3 (d, $J = 6.5$ Hz). FABLRMS m/z 301 (M-H). ESHRMS m/z 300.9884 (M-H, Calc'd 300.9879). Anal. Calc'd for $\text{C}_{13}\text{H}_6\text{ClF}_3\text{O}_3$: C, 51.59; H, 2.00; Cl, 11.71. Found: C, 51.63; H, 2.04; Cl, 11.84.

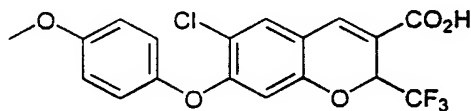
EXAMPLE 182



6-Chloro-7-ethenyl-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from ethyl 6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 180) via a procedure similar to that described in Example 114: mp >200 °C with decomposition. ¹H NMR (acetone-d₆/300MHz) 7.89 (s, 1H), 7.59 (s, 1H), 7.37 (s, 1H), 7.07 (dd, 1H, *J* = 17.5 Hz, 11.1 Hz), 6.07 (d, 1H, *J* = 17.3 Hz), 5.87 (q, 1H, *J* = 7.0 Hz), 5.57 (d, 1H, *J* = 11.1 Hz). ¹⁹F NMR (acetone-d₆/282MHz) -79.2 (d, *J* = 7.2 Hz). FABLRMS *m/z* 303 (M-H). ESHRMS *m/z* 303.0030 (M-H, Calc'd 303.0036). Anal. Calc'd for C₁₃H₈ClF₃O₃: C, 51.25; H, 2.65; Cl, 11.64. Found: C, 51.41; H, 2.66; Cl, 11.68.

EXAMPLE 183



6-Chloro-7-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of 5-chloro-4-fluorosalicylaldehyde.

4-Chloro-3-fluorophenol (Avocado product #16029) was converted to the salicylaldehyde via a method similar to that described in Example 2, Step 1: mp 102.7-103.7 °C. ¹H NMR (CDCl₃/300MHz) 11.22 (d, 1H, J = 1.6 Hz), 9.80 (s, 1H), 7.63 (d, 1H, J = 7.9 Hz), 6.77 (d, 1H, J = 10.3 Hz). ¹⁹F NMR (CDCl₃/282MHz) -100.3 (t).

Step 2. Preparation of ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

The salicylaldehyde (Step 1) was converted to the substituted ethyl 2H-1-benzopyran-3-carboxylate via a method similar to that described in Example 11 (Steps 2,3), Step 2: mp 97.6-98.6 °C. ¹H NMR (CDCl₃/300MHz) 7.63 (s, 1H), 7.29 (d, 1H, J = 9.7 Hz), 6.84 (d, 1H, J = 9.3 Hz), 5.69 (q, 1H, J = 6.6 Hz), 4.32 (m, 2H), 1.35 (t, 3H, J = 7.0 Hz). ¹⁹F NMR (CDCl₃/282MHz) -78.8 (d, J = 6.5 Hz), -106.7 (t, J = 8.7 Hz).

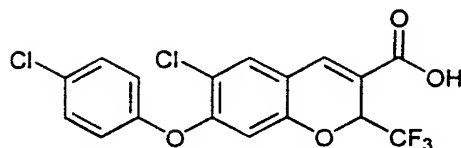
Step 3. Preparation of ethyl 6-chloro-7-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A 50 mL flask was charged with the substituted ethyl 2H-1-benzopyran-3-carboxylate from Step 2 (1.44 g, 4.43 mmol), 4-methoxyphenol (0.62g, 4.99 mmol), K₂CO₃ (0.84 g, 6.07 mmol), and DMSO (10 mL). The reaction was stirred at 105 °C for 18 hours,

poured into water extracted with ethyl acetate, washed with water, brine, dried over MgSO_4 and concentrated in *vacuo*, to give a brown oil which was passed through a column of silica gel with 5% ethyl acetate/hexanes to give a yellow solid (0.78 g, 41%): mp 115.8-117.0 °C. ^1H NMR (CDCl_3 /300MHz) 7.66 (s, 1H), 7.32 (s, 1H), 7.04 (d, 2H, $J = 9.3$ Hz), 6.98 (d, 2H, $J = 9.1$ Hz), 6.38 (s, 1H), 5.65 (q, 1H, $J = 6.8$ Hz), 4.33 (m, 2H), 3.86 (s, 3H), 1.37 (t, 3H, $J = 7.2$ Hz). ^{19}F NMR (CDCl_3 /282MHz) -78.9 (d, $J = 6.5$ Hz).

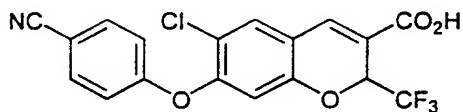
Step 4. Preparation of 6-chloro-7-(4-methoxyphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester (Step 3) was hydrolyzed to the carboxylic acid via a method similar to that described in Example 1, Step 2: mp 213.2-214.3 °C. ^1H NMR (acetone- d_6 /300MHz) 7.90 (s, 1H), 7.70 (s, 1H), 7.13 (d, 2H, $J = 9.3$ Hz), 7.09 (d, 2H, $J = 8.9$ Hz), 6.42 (s, 1H), 5.83 (q, 1H, $J = 7.0$ Hz), 3.87 (s, 3H). ^{19}F NMR (acetone- d_6 /282MHz) -79.5 (d, $J = 7.2$ Hz). FABLRMS m/z 401 (M+H). ESHRMS m/z 418.0799 (M+ NH_4 , Calc'd 418.0669). Anal. Calc'd for $\text{C}_{18}\text{H}_{12}\text{ClF}_3\text{O}_5$: C, 53.95; H, 3.02; Cl, 8.85. Found: C, 54.17; H, 3.03; Cl, 8.91.

EXAMPLE 184

**6-Chloro-7-(4-chlorophenoxy)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

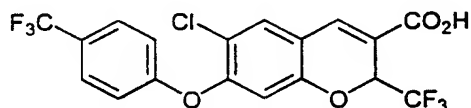
The title compound was prepared from 4-chlorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 205.4-206.5 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.93 (s, 1H), 7.75 (s, 1H), 7.50 (d, 2H, *J* = 8.9 Hz), 7.19 (d, 2H, *J* = 8.9 Hz), 6.69 (s, 1H), 5.87 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.4 (d, *J* = 7.2 Hz). FABLRMS *m/z* 403 (M-H). ESHRMS *m/z* 402.9773 (M-H, Calc'd 402.9752). Anal. Calc'd for C₁₇H₉Cl₂F₃O₄: C, 50.40; H, 2.24; Cl, 17.50. Found: C, 50.02; H, 1.98; Cl, 17.79.

EXAMPLE 185

**6-Chloro-7-(4-cyanophenoxy)-2-(trifluoromethyl)-2H-
1-benzopyran-3-carboxylic acid**

The title compound was prepared from 4-cyanophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 212.5-215.7 °C. ¹H NMR (acetone-d₆/300MHz) 7.96 (s, 1H), 7.90 (d, 2H, *J* = 8.9 Hz), 7.81 (s, 1H), 7.27 (d, 2H, *J* = 8.9 Hz), 6.96 (s, 1H), 5.90 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-d₆/282MHz) -79.4 (d, *J* = 7.2 Hz). FABLRMS *m/z* 396 (M+H). ESHRMS *m/z* 413.0544 (M+NH₄⁺, Calc'd 413.0516). Anal. Calc'd for C₁₈H₉ClF₃NO₄: C, 54.63; H, 2.29; N, 3.54; Cl, 8.96. Found: C, 54.40; H, 2.18; N 3.30; Cl, 9.20.

EXAMPLE 186

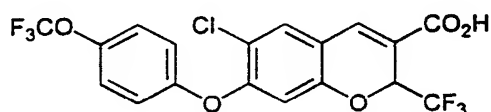


6-Chloro-2-(trifluoromethyl)-7-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 4-(trifluoromethyl)phenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 230.9-232.0 °C. ¹H NMR (acetone-d₆/300MHz) 7.93 (s, 1H), 7.77 (m, 3H), 7.27 (d, 2H,

$J = 8.7$ Hz), 6.86 (s, 1H), 5.87 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -62.8 (s, 3F), -79.4 (d, 3F, $J = 7.2$ Hz). ESLRMS m/z 437 (M-H). ESHRMS m/z 437.0008 (M-H, Calc'd 437.0015). Anal. Calc'd for $\text{C}_{18}\text{H}_9\text{ClF}_6\text{O}_4$: C, 49.28; H, 2.07; Cl, 8.08. Found: C, 49.25; H, 2.08; Cl, 8.24.

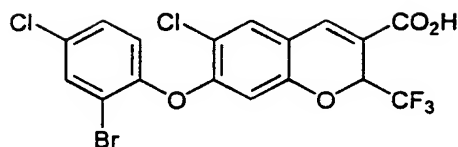
EXAMPLE 187



6-Chloro-2-(trifluoromethyl)-7-[4-(trifluoromethoxy)phenoxy]-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 4-(trifluoromethoxy)phenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 198.0-199.2 °C. ^1H NMR (acetone- d_6 /300MHz) 7.91 (s, 1H), 7.73 (s, 1H), 7.43 (d, 2H, $J = 9.1$ Hz), 7.25 (d, 2H, $J = 9.1$ Hz), 6.70 (s, 1H), 5.82 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -59.4 (s, 3F), -79.4 (d, 3F, $J = 7.2$ Hz). ESLRMS m/z 453 (M-H). ESHRMS m/z 452.9950 (M-H, Calc'd 452.9964). Anal. Calc'd for $\text{C}_{18}\text{H}_9\text{ClF}_6\text{O}_5$: C, 47.55; H, 2.00; Cl, 7.80. Found: C, 47.47; H, 1.89; Cl, 8.02.

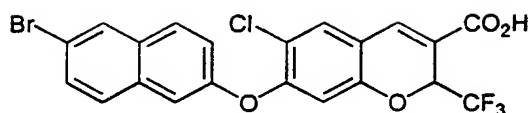
EXAMPLE 188



7-(2-Bromo-4-chlorophenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 2-bromo-4-chlorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 223.6-225.6 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.89 (s, 1H), 7.83 (d, 1H, *J* = 2.6 Hz), 7.73 (s, 1H), 7.52 (dd, 1H, *J* = 8.7 Hz, 2.4 Hz), 7.25 (d, 1H, *J* = 8.9 Hz), 6.55 (s, 1H), 5.83 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.4 (d, *J* = 7.2 Hz). FABLRMS *m/z* 481 (M-H). ESHRMS *m/z* 480.8856 (M-H, Calc'd 480.8857). Anal. Calc'd for C₁₇H₈BrCl₂F₃O₄: C, 42.18; H, 1.67; Br, 16.51; Cl, 14.65. Found: C, 42.15; H, 1.54; Br, 16.56; Cl, 14.39.

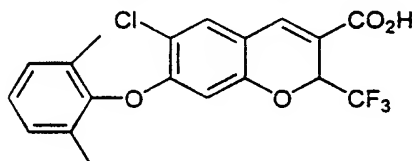
EXAMPLE 189



7-[(6-Bromo-2-naphthalenyl)oxy]-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 6-bromo-2-naphthol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 252.6-254.8 °C. ¹H NMR (acetone-*d*₆/300MHz) 8.19 (s, 1H), 8.04 (d, 1H, *J* = 9.1 Hz), 7.93 (s, 1H), 7.87 (d, 1H, *J* = 8.9 Hz), 7.75 (s, 1H), 7.65 (d, 1H, *J* = 8.8 Hz), 7.55 (s, 1H), 7.43 (d, 1H, *J* = 9.1 Hz), 6.72 (s, 1H), 5.83 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.4 (d, *J* = 6.5 Hz). FABLRMS *m/z* 497 (M-H). ESHRMS *m/z* 496.9421 (M-H, Calc'd 496.9403). Anal. Calc'd for C₂₁H₁₁BrClF₃O₄ with 1.75 wt % H₂O: C, 49.60; H, 2.38; Br, 15.71; Cl, 6.97. Found: C, 49.55; H, 2.03; Br, 15.56; Cl, 7.22.

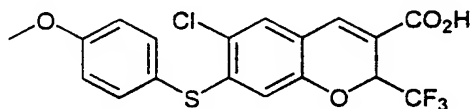
EXAMPLE 190



6-Chloro-7-(2,6-dimethylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 2,6-dimethylphenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 252.7-258.6 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.88 (s, 1H), 7.71 (s, 1H), 7.22 (m, 3H), 6.04 (s, 1H), 5.77 (q, 1H, *J* = 7.0 Hz), 2.13 (s, 6H). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.6 (d, *J* = 7.2 Hz). FABLRMS *m/z* 397 (M-H). ESHRMS *m/z* 397.0450 (M-H, Calc'd 397.0454). Anal. Calc'd for C₁₉H₁₄ClF₃O₄: C, 57.23; H, 3.54; Cl, 8.89. Found: C, 57.21; H, 3.56; Cl, 8.98.

EXAMPLE 191

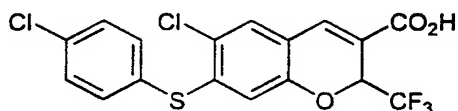


6-Chloro-7-[(4-methoxyphenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 4-methoxybenzenethiol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 239.4-242.3 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.84 (s, 1H), 7.55 (m, 3H), 7.16 (d, 2H, *J* = 8.7 Hz), 6.20 (s, 1H), 5.75 (q, 1H, *J* = 7.0 Hz), 3.90 (s, 3H).

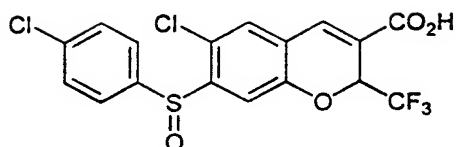
^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 6.5 Hz).
 ESLRMS m/z 415 (M-H). ESHRMS m/z 415.0005 (M-H,
 Calc'd 415.0019). Anal. Calc'd for $\text{C}_{18}\text{H}_{12}\text{ClF}_3\text{O}_4\text{S}$: C,
 51.87; H, 2.90; Cl, 8.51; S, 7.69. Found: C,
 51.76; H, 2.83; Cl 8.55; S, 7.77.

EXAMPLE 192



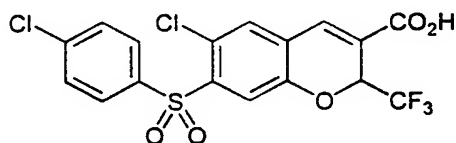
6-Chloro-7-[(4-chlorophenyl)thio]-2- (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 4-chlorobenzenethiol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 251.7-253.3 °C. ^1H NMR (acetone- d_6 /300MHz) 7.86 (s, 1H), 7.62 (m, 5H), 6.39 (s, 1H), 5.78 (q, 1H, J = 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 6.5 Hz). ESLRMS m/z 419 (M-H). ESHRMS m/z 418.9559 (M-H, Calc'd 418.9523). Anal. Calc'd for $\text{C}_{17}\text{H}_9\text{Cl}_2\text{F}_3\text{O}_3\text{S}$: C, 48.47; H, 2.15; Cl, 16.83; S, 7.61. Found: C, 48.38; H, 2.12; Cl, 17.09; S, 7.75.

EXAMPLE 193

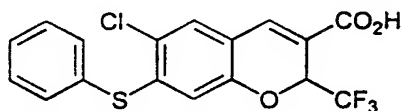
6-Chloro-7-[(4-chlorophenyl)sulfinyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

Following a procedure similar to that described in Example 128, ethyl 6-chloro-7-[(4-chlorophenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 192) was treated with OXONE® yielding the products labeled Example 193 and Example 194: mp 213.8-217.4 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.91 (s, 1H), 7.83 (d, 2H, *J* = 8.7 Hz), 7.62 (m, 4H), 5.98 (q, 1H, *J* = 6.8 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.3 (d, *J* = 7.2 Hz). ESRMS *m/z* 435 (M-H). ESHRMS *m/z* 434.9470 (M-H, Calc'd 434.9472). Anal. Calc'd for C₁₇H₉Cl₂F₃O₄S: C, 46.70; H, 2.07; Cl, 16.22; S, 7.33. Found: C, 46.80; H, 2.28; Cl, 16.75; S, 7.39.

EXAMPLE 194

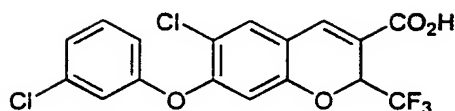
6-Chloro-7-[(4-chlorophenyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

mp 240.4-241.5 °C. ^1H NMR (acetone- d_6 /300MHz) 8.04 (d, 2H, J = 8.9 Hz), 7.92 (s, 2H), 7.75 (s, 1H), 7.72 (d, 2H, J = 8.7 Hz), 6.02 (q, 1H, J = 6.8 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.2 (d, J = 7.4 Hz). ESI/MS m/z 451 (M-H). ESI/MS m/z 450.9409 (M-H, Calc'd 450.9422). Anal. Calc'd for $\text{C}_{17}\text{H}_9\text{Cl}_2\text{F}_3\text{O}_5\text{S}$: C, 45.05; H, 2.00; Cl, 15.64. Found: C, 45.35; H, 2.05; Cl, 16.01.

EXAMPLE 195

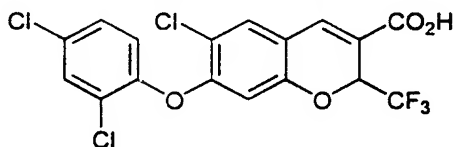
6-Chloro-7-(phenylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from benzenethiol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 271.7-274.4 °C. ^1H NMR (acetone- d_6 /300MHz) 7.86 (s, 1H), 7.60 (m, 6H), 6.29 (s, 1H), 5.74 (q, 1H, J = 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 7.2 Hz). FAB/MS m/z 385 (M-H). ESI/MS m/z 384.9904 (M-H, Calc'd 384.9913). Anal. Calc'd for $\text{C}_{17}\text{H}_{10}\text{ClF}_3\text{O}_5\text{S}$: C, 52.79; H, 2.61; Cl, 9.17; S, 8.29. Found: C, 52.80; H, 2.49; Cl, 9.71; S, 8.37.

EXAMPLE 196

**6-Chloro-7-(3-chlorophenoxy)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

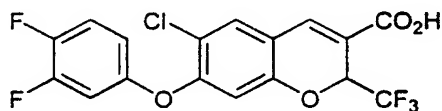
The title compound was prepared from 3-chlorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 174.1-176.1 °C. ¹H NMR (acetone-*d*₆/300MHz) 7.91 (s, 1H), 7.73 (s, 1H), 7.48 (m, 1H), 7.29 (d, 1H, *J* = 7.5 Hz), 7.17 (d, 1H, *J* = 2.0 Hz), 7.08 (d, 1H, *J* = 8.3 Hz), 6.73 (s, 1H), 5.85 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-*d*₆/282MHz) -79.4 (d, *J* = 6.5 Hz). FABLRMS *m/z* 403 (M-H). ESHRMS *m/z* 402.9752 (M-H, Calc'd 402.9729). Anal. Calc'd for C₁₇H₉Cl₂F₃O₄: C, 50.40; H, 2.24; Cl, 17.50. Found: C, 50.18; H, 2.00; Cl, 17.26.

EXAMPLE 197

6-Chloro-7-(2,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 2,4-dichlorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 233.0-234.0 °C. ^1H NMR (acetone- d_6 /300MHz) 7.90 (s, 1H), 7.73 (s, 1H), 7.69 (d, 1H, J = 2.6 Hz), 7.47 (dd, 1H, J = 8.7 Hz, 2.6 Hz), 7.28 (d, 1H, J = 8.7 Hz), 6.57 (s, 1H), 5.81 (q, 1H, J = 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 6.5 Hz). FABLRMS m/z 437 (M-H). ESHRMS m/z 436.9379 (M-H, Calc'd 436.9362). Anal. Calc'd for $\text{C}_{17}\text{H}_8\text{Cl}_3\text{F}_3\text{O}_4$: C, 46.45; H, 1.83; Cl, 24.19. Found: C, 46.38; H, 1.59; Cl, 24.46.

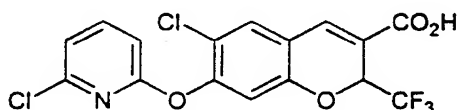
EXAMPLE 198



6-Chloro-7-(3,4-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

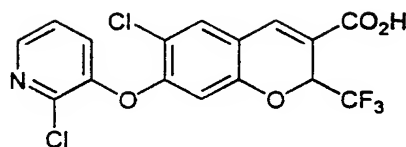
The title compound was prepared from 3,4-difluorobenzenethiol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 191.9-193.1 °C. ^1H NMR (acetone- d_6 /300MHz) 7.90 (s, 1H), 7.72 (d, 1H, J = 3.0 Hz), 7.45 (q, 1H, J = 9.1

Hz), 7.21 (m, 1H), 7.00 (m, 1H), 6.70 (d, 1H, $J = 2.6$ Hz), 5.84 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, 3F, $J = 6.5$ Hz), -136.4 (m, 1F), -145.1 (m, 1F). FABLRMS m/z 405 (M-H). ESHRMS m/z 404.9923 (M-H, Calc'd 404.9953). Anal. Calc'd for $\text{C}_{17}\text{H}_8\text{ClF}_3\text{O}_4$: C, 50.21; H, 1.98; Cl, 8.72. Found: C, 50.14; H, 1.73; Cl, 8.93.

EXAMPLE 199

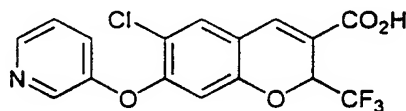
6-Chloro-7-[(6-chloro-2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 6-chloro-2-pyridinol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 226.4-227.4 °C. ^1H NMR (acetone- d_6 /300MHz) 7.99 (m, 1H), 7.96 (s, 1H), 7.75 (s, 1H), 7.30 (d, 1H, $J = 7.7$ Hz), 7.16 (d, 1H, $J = 8.0$ Hz), 7.10 (s, 1H), 5.90 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.3 (d, $J = 7.2$ Hz). FABLRMS m/z 404 (M-H). ESHRMS m/z 405.9853 (M+H, Calc'd 405.9861). Anal. Calc'd for $\text{C}_{16}\text{H}_8\text{Cl}_2\text{F}_3\text{NO}_4$: C, 47.32; H, 1.99; N, 3.45; Cl, 17.46. Found: C, 47.26; H, 1.93; N, 3.35; Cl, 17.19.

Example 200

6-Chloro-7-[(2-chloro-3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

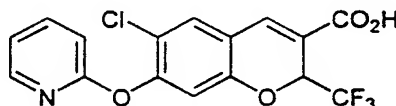
The title compound was prepared from 2-chloro-3-pyridinol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 213.4-216.2 °C. ¹H NMR (acetone-d₆/300MHz) 8.33 (m, 1H), 7.91 (s, 1H), 7.76 (s, 1H), 7.64 (m, 1H), 7.53 (m, 1H), 6.67 (s, 1H), 5.85 (q, 1H, J = 7.0 Hz). ¹⁹F NMR (acetone-d₆/282MHz) -79.4 (d, J = 7.2 Hz). ESHRMS m/z 405.9890 (M+H, Calc'd 405.9861). Anal. Calc'd for C₁₆H₈Cl₂F₃NO₄: C, 47.32; H, 1.99; N, 3.45; Cl, 17.46. Found: C, 47.36; H, 1.98; N, 3.40; Cl, 17.14.

EXAMPLE 201

6-Chloro-7-[3-pyridinyloxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 3-hydroxypyridine and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 237.8-238.8 °C. ¹H NMR (DMSO-*d*₆/300MHz) 13.3 (br s, 1H), 8.42 (m, 2H), 7.87 (s, 1H), 7.83 (s, 1H), 7.48 (m, 2H), 6.72 (s, 1H), 5.93 (q, 1H, *J* = 7.2 Hz). ¹⁹F NMR (DMSO-*d*₆/282MHz) -77.8 (d, *J* = 7.2 Hz). FABLRMS *m/z* 372 (M+H). ESHRMS *m/z* 370.0083 (M-H, Calc'd 370.0094). Anal. Calc'd for C₁₆H₉ClF₃NO₄: C, 51.70; H, 2.44; N, 3.77; Cl, 9.54. Found: C, 51.49; H, 2.47; N, 3.70; Cl, 9.69.

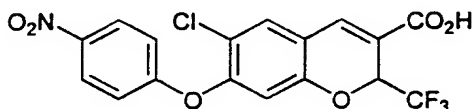
EXAMPLE 202



6-Chloro-7-[2-pyridinyloxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

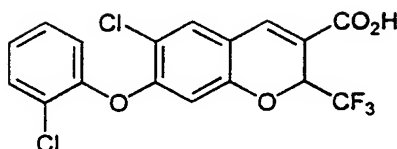
The title compound was prepared from 2-hydroxypyridine and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 234.2-235.4 °C. ¹H NMR (acetone-*d*₆/300MHz) 8.14 (dd, 1H, *J* = 4.8 Hz, 1.4 Hz), 7.92 (m, 2H), 7.70 (s, 1H), 7.14 (m, 2H), 6.99 (s, 1H), 5.88 (q, 1H, *J*

= 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 7.2 Hz). ESLRMS m/z 372 (M+H). ESHRMS m/z 372.0252 (M+H, Calc'd 372.0250). Anal. Calc'd for $\text{C}_{16}\text{H}_9\text{ClF}_3\text{NO}_4$: C, 51.70; H, 2.44; N, 3.77; Cl, 9.54. Found: C, 51.70; H, 2.44; N, 3.71; Cl, 9.65.

EXAMPLE 203

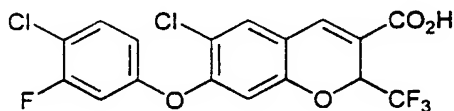
**6-Chloro-7-(4-nitrophenoxy)-2-(trifluoromethyl)-2H-
1-benzopyran-3-carboxylic acid**

The title compound was prepared from 4-nitrophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 246.4-248.3 °C. ^1H NMR (acetone- d_6 /300MHz) 8.34 (d, 2H, J = 9.3 Hz), 7.96 (s, 1H), 7.82 (s, 1H), 7.29 (d, 2H, J = 9.3 Hz), 7.02 (s, 1H), 5.90 (q, 1H, J = 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.3 (d, J = 7.2 Hz). ESLRMS m/z 433 (M+NH $_4$). ESHRMS m/z 433.0463 (M+NH $_4$, Calc'd 433.0414). Anal. Calc'd for $\text{C}_{17}\text{H}_9\text{ClF}_3\text{NO}_6$: C, 49.12; H, 2.18; N, 3.37; Cl, 8.53. Found: C, 48.94; H, 2.05; N, 3.32; Cl, 8.72.

EXAMPLE 204

**6-Chloro-7-(2-chlorophenoxy)-2-(trifluoromethyl)-
2H-1-benzopyran-3-carboxylic acid**

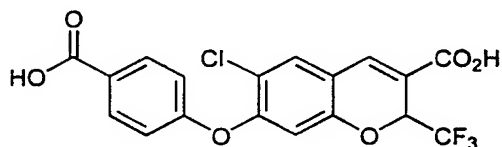
The title compound was prepared from 2-chlorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 199.0-199.1 °C. ¹H NMR (acetone-d₆/300MHz) 7.94 (s, 1H), 7.76 (s, 1H), 7.65 (m, 1H), 7.51 (m, 1H), 7.38 (m, 1H), 7.32 (m, 1H), 6.44 (s, 1H), 5.85 (q, 1H, *J* = 7.0 Hz). ¹⁹F NMR (acetone-d₆/282MHz) -79.5 (d, *J* = 6.5 Hz). ESHRMS *m/z* 402.9752 (M-H, Calc'd 402.9751). Anal. Calc'd for C₁₇H₉Cl₂F₃O₄: C, 50.40; H, 2.24; Cl, 17.50. Found: C, 50.42; H, 2.12; Cl, 17.72.

EXAMPLE 205

**6-Chloro-7-(4-chloro-3-fluorophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid**

The title compound was prepared from 4-chloro-3-fluorophenol and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4: mp 206.8-207.4 °C. ^1H NMR (acetone- d_6 /300MHz) 7.92 (s, 1H), 7.74 (s, 1H), 7.59 (m, 1H), 7.16 (m, 1H), 6.99 (m, 1H), 6.83 (s, 1H), 5.86 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, 3F, $J = 6.5$ Hz), -113.8 (m, 1F). Anal. Calc'd for $\text{C}_{17}\text{H}_8\text{Cl}_2\text{F}_4\text{O}_4$: C, 48.25; H, 1.91; Cl, 16.76. Found: C, 48.20; H, 1.97; Cl, 16.92.

EXAMPLE 206

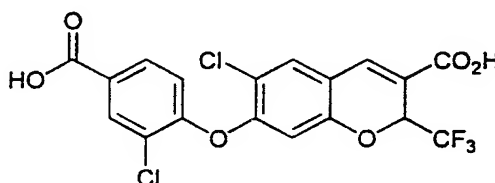


7-(4-Carboxyphenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 4-hydroxybenzoic acid and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4. (Note: In the step similar to step 3, an additional equivalent of potassium carbonate was added.) mp >300 °C. ^1H NMR (acetone- d_6 /300MHz)

8.16 (d, 2H, $J = 8.9$ Hz), 7.96 (s, 1H), 7.79 (s, 1H), 7.20 (d, 2H, $J = 8.9$ Hz), 6.87 (s, 1H), 5.90 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) - 79.4 (d, $J = 7.2$ Hz). ESHRMS m/z 413.0006 (M-H, Calc'd 413.0040). Anal. Calc'd for $\text{C}_{18}\text{H}_{10}\text{ClF}_3\text{O}_6$: C, 52.13; H, 2.43. Found: C, 51.81; H, 2.31.

EXAMPLE 207



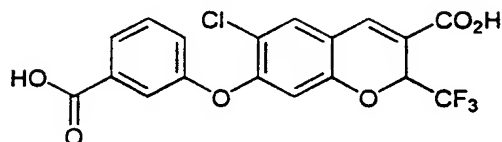
7-(4-Carboxy-2-chloro-phenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 3-chloro-4-hydroxybenzoic acid and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4.

(Note: In the step similar to Example 183, Step 3, an additional equivalent of potassium carbonate was added.) mp > 300 °C. ^1H NMR (acetone- d_6 /300MHz) 8.18 (d, 1H, $J = 2.0$ Hz), 8.03 (dd, 1H, $J = 8.5$ Hz, 2.0 Hz), 7.93 (s, 1H), 7.78 (s, 1H), 7.23 (d, 1H, $J = 8.7$ Hz), 6.77 (s, 1H), 5.87 (q, 1H, $J = 7.0$ Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, $J = 6.5$ Hz). ESHRMS m/z 446.9625 (M-H, Calc'd 446.9650). Anal.

Calc'd for $C_{18}H_9Cl_3F_3O_6$: C, 48.13; H, 2.02; Cl, 15.79. Found: C, 47.96; H, 1.90; Cl, 15.65.

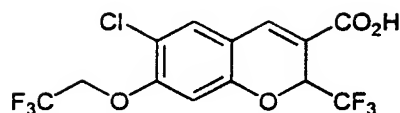
EXAMPLE 208



7-(3-Carboxyphenoxy)-6-chloro-2-(trifluoromethyl)- 2H-1-benzopyran-3-carboxylic acid

The title compound was prepared from 3-hydroxybenzoic acid and ethyl 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (Example 183, Step 2) via a procedure similar to that described in Example 183, Steps 3 and 4. (Note: In the step similar to step 3, an additional equivalent of potassium carbonate was added.) mp 288.1-289.2 °C. 1H NMR (acetone- d_6 /300MHz) 7.92 (m, 2H), 7.75 (s, 1H), 7.68 (d, 1H, J = 1.4 Hz), 7.62 (t, 1H, J = 7.9 Hz), 7.41 (m, 1H), 6.72 (s, 1H), 5.85 (q, 1H, J = 7.0 Hz). ^{19}F NMR (acetone- d_6 /282MHz) -79.4 (d, J = 6.5 Hz). ESHRMS m/z 413.0027 (M-H, Calc'd 413.0040). Anal. Calc'd for $C_{18}H_9ClF_3O_6$: C, 52.13; H, 2.43; Cl, 8.55. Found: C, 51.88; H, 2.48; Cl, 8.65.

EXAMPLE 209



6-Chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid

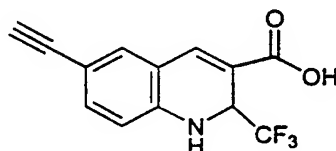
Step 1. Preparation of ethyl 6-Chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate.

A solution of 2,2,2-trifluoroethanol (1.3 g, 13 mmol) in 1,4-dioxane (2 mL) was added to a slurry of NaH (0.5 g, 13 mmol) in 1,4-dioxane (2 mL). The reaction was stirred for 10 minutes then a solution of 6-chloro-7-fluoro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylate (1.61 g, 5.0 mmol) in 1,4-dioxane (5 mL) was added and the reaction was heated to 85 °C for 22 hours. The reaction mixture was acidified with 3N HCl, extracted with ethyl acetate, washed with saturated NaHCO₃, brine, dried over MgSO₄, and concentrated *in vacuo*, to give a brown oil which was passed through a column of silica gel with 10% ethyl acetate/hexanes to give a white solid (0.37 g, 18%): ¹H NMR (CDCl₃/300MHz) 7.62 (s, 1H), 7.28 (s, 1H), 6.58 (s, 1H), 5.69 (q, 1H, *J* = 6.6 Hz), 4.42 (m, 2H), 4.31 (m, 2H), 1.34 (t, 3H, *J* = 7.1 Hz). ¹⁹F NMR (CDCl₃/282MHz) -74.2 (t, 3F, *J* = 8.0 Hz), -78.9 (d, 3F, *J* = 6.5 Hz).

Step 2. Preparation of 6-Chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid.

The ester (Step 1) was hydrolyzed to the carboxylic acid via a method similar to that described in Example 1, Step 2: mp 174.4-176.0 °C. ^1H NMR (acetone- d_6 /300MHz) 7.86 (s, 1H), 7.62 (s, 1H), 7.00 (s, 1H), 5.86 (q, 1H, $J = 7.0$ Hz), 4.90 (m, 2H). ^{19}F NMR (acetone- d_6 /282MHz) -75.1 (t, 3F, $J = 8.7$ Hz), -79.5 (d, 3F, $J = 7.2$ Hz). FABLRMS m/z 375 (M-H). ESHRMS m/z 374.9880 (M-H, Calc'd 374.9859).

EXAMPLE 210



1,2-Dihydro-6-ethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid

Step 1. Preparation of ethyl 6-trimethylsilylethynyl-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate.

The ethyl 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (Example 160, Step 1) (750 mg, 1.89 mmol), trimethylsilylacetylene (925 mg, 9.44 mmol), dichlorobis(triphenylphosphine)palladium(II) (35

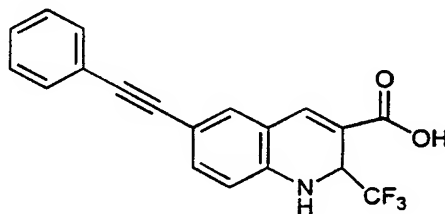
mg, 0.05 mmol), copper(I)iodide (9.5 mg, 0.05 mmol), and triethylamine (953 mg, 9.44 mmol) were mixed at room temperature for 2 h in acetonitrile (10 mL). The resulting black heterogeneous mixture was poured into ethyl acetate(50 mL) and extracted with water (2 x 25 mL), 1 N aqueous hydrochloric acid (25 mL), and saturated aqueous ammonium chloride (2 x 25 mL). The organic layer was dried over sodium sulfate and solvent removed at reduce pressure. The product was purified by flash chromatography (0-25% ethyl acetate/hexanes, silica gel). A black semisolid was isolated which upon trituration with hexanes afforded ethyl 6-trimethylsilylethynyl-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate as a yellow solid(267 mg,38%): mp 117-119 °C ¹H NMR (CDCl₃, 300 MHz) 7.65 (s, 1H), 7.24-7.32 (m, 2H), . 6.55 (d, 1H, J = 8.0 Hz), 5.08-5.18 (m, 1H), 4.66 (brs, 1H), 4.24-4.64 (m, 2H), 1.34 (t, 3H, J = 7.1 Hz) 0.22 (s, 9H). ESHRMS m/z 368.1306 (Calc'd for M+H 368.1293).

Step 2: Preparation of 1,2-dihydro-6-ethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid.

Hydrolysis of the ester and the trimethylsilyl moiety of ethyl 6-trimethylsilylethynyl-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (Step 1) was performed by a procedure similar to that described in Example 157, Step 3, yielding the

title compound: mp 259-268 °C ^1H NMR (CD_3OD , 300 MHz) 7.69 (s, 1H), 7.21-7.28 (m, 2H), 6.65 (d, 1H, $J = 8.4$ Hz), 5.11 (q, 1H, $J = 7.3$ Hz). ESHRMS m/z 268.0580 (Calc'd for $\text{M}+\text{H}$ 268.0586).

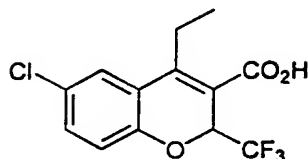
EXAMPLE 211



1,2-Dihydro-6-(phenylethynyl)-2-(trifluoromethyl)- 3-quinolinecarboxylic acid

Ethyl 6-iodo-1,2-dihydro-2-(trifluoromethyl)-3-quinolinecarboxylate (Example 160, Step 1) was converted to the title compound by a procedure similar to that described in Example 210: mp 218-219 °C ^1H NMR ($\text{CDCl}_3/\text{DMSO}-d_6$ (19:1), 400 MHz) 7.56 (s, 1H), 7.35 (d, 1H, $J = 8.5$ Hz), 7.18-7.24 (m, 6H), 6.57 (d, 1H, $J = 8.5$ Hz), 5.93 (bs, 1H), 5.03-5.05 (m, 1H). FABHRMS m/z 344.0898 (Calc'd for $\text{M}+\text{H}$ 344.0901).

EXAMPLE 212



6-Chloro-4-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid

Step 1. Preparation of ethyl-6-chloro-4-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate.

Ethyl-6-chloro-4-ethenyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylate (Example 142, Step 5) (0.433 g, 1.30 mmol) was dissolved in methanol (10 mL). Palladium (5 wt % on activated carbon) (0.150 g) was slowly added. The reaction vessel was charged with hydrogen (25 psi) and was stirred for 25 minutes. Gas chromatographic analysis indicated that the reaction was complete. The reaction mixture was filtered through Celite and evaporated to yield a orange oil. The oil was purified by flash column chromatography (100 % hexanes) to yield the desired product as a yellow oil (0.282 g, 65%): ^1H NMR (CDCl_3 /300 MHz) 7.43 (d, 1H, $J = 2.4$ Hz), 7.25 (dd, 1H, $J = 2.4, 8.7$ Hz), 6.92 (d, 1H, $J = 8.7$ Hz), 5.75 (q, 1H, $J = 7.0$ Hz), 4.38-4.24 (m, 2H), 3.05-2.95 (m, 2H), 1.35 (t, 3H, $J = 7.2$ Hz), 1.25, (t, 3H, $J = 7.6$ Hz).

Step 2. Preparation of 6-chloro-4-ethyl-2-trifluoromethyl-2H-1-benzopyran-3-carboxylic acid.

The ester from Step 1 (0.282 g, 0.842 mmol) was dissolved in a THF-EtOH- H_2O mixture (10 mL,

7:2:1). The resulting solution was treated with aqueous sodium hydroxide (0.340 mL, 0.842 mmol of a 2.5 N solution) and was stirred at room temperature for 18 hours. The solvent was removed *in vacuo* and the residue was dissolved in water (10 mL). Diethyl ether (10 mL) was added and the resulting mixture acidified by the addition of a few drops of concentrated HCl. The diethyl ether layer was separated, and the aqueous phase was extracted with additional ether (2 x 10 mL). The ether extracts were combined, dried over MgSO₄, filtered, and evaporated to yield a clear oil. Addition of hexane (10 mL) resulted in the formation of a white precipitate which was collected by vacuum filtration to afford the title compound (0.064 g, 25%) as a white powder: mp 170.9-172.1 °C ¹H NMR (CDCl₃/ 300 MHz) 7.47 (d, 1H, *J* = 2.4 Hz), 7.30 (dd, 1H, *J* = 2.4, 8.7 Hz), 6.96 (d, 1H, *J* = 8.7 Hz), 5.76 (q, 1H, *J* = 7.0 Hz), 3.08 (q, 2H, *J* = 7.6 Hz), 1.27 (t, 3H, *J* = 7.5 Hz). FABLRMS *m/z* 305.2 (M-H). ESHRMS *m/z* 305.0185 (M-H, Calc'd. 305.0192). Anal. Calc'd for C₁₃H₉ClF₃O₃: C, 50.92; H, 3.29; Cl, 11.56. Found: C, 50.68; H, 3.18; Cl, 11.64.

BIOLOGICAL EVALUATION

Rat Carrageenan Foot Pad Edema Test

5

The carrageenan foot edema test was performed with materials, reagents and procedures essentially as described by Winter, et al., (*Proc. Soc. Exp. Biol. Med.*, 111, 544 (1962)). Male Sprague-Dawley rats were selected in each group so that the average body weight was as close as possible. Rats were fasted with free access to water for over sixteen hours prior to the test. The rats were dosed orally (1 mL) with compounds suspended in vehicle containing 0.5% methylcellulose and 0.025% surfactant, or with vehicle alone. One hour later a subplantar injection of 0.1 mL of 1% solution of carrageenan/sterile 0.9% saline was administered and the volume of the injected foot was measured with a displacement plethysmometer connected to a pressure transducer with a digital indicator. Three hours after the injection of the carrageenan, the volume of the foot was again measured. The average foot swelling in a group of drug-treated animals was compared with that of a group of placebo-treated animals and the percentage inhibition of edema was determined (Otterness and Bliven, *Laboratory Models for Testing NSAIDs*, in *Non-steroidal Anti-Inflammatory Drugs*, (J. Lombardino, ed. 1985)). The % inhibition shows the % decrease from control paw volume determined in this procedure and the data for selected compounds in this invention are summarized in Table I.

TABLE I.

	Example	RAT PAW EDEMA % Inhibition	ANALGESIA % Inhibition
		30mg/ g body weight	30mg/ g body weight
5	1	57	58
	8	43	36
	12	30	37
	16	48	34
10	24	29	25
	31	39*	37*
	32	59*	73*
	38	47	55
15	39	18	8
	46	59	67
	65	43	32
	79	45	53
20	81	36	44
	83	56	65
	84	44	24
	105	75	69
25	115	52	62
	139	42	41
	156	54	66
	157	44*	36*
30	158	45*	55*
	161	53	56
	165	30*	33*
	172	57	67
30	177	59*	61*
	188	57	69

* - @ 10 mpk

35 Evaluation of COX-1 and COX-2 activity *in vitro*

The compounds of this invention exhibited inhibition *in vitro* of COX-2. The COX-2 inhibition activity of the compounds of this invention

illustrated in the Examples was determined by the following methods.

a. Preparation of recombinant COX baculoviruses

5 Recombinant COX-1 and COX-2 were prepared as described by Gierse et al, [*J. Biochem.*, 305, 479-84 (1995)]. A 2.0 kb fragment containing the coding region of either human or murine COX-1 or human or murine COX-2 was cloned into a BamH1 site of the
10 baculovirus transfer vector pVL1393 (Invitrogen) to generate the baculovirus transfer vectors for COX-1 and COX-2 in a manner similar to the method of D.R. O'Reilly et al (*Baculovirus Expression Vectors: A Laboratory Manual* (1992)). Recombinant
15 baculoviruses were isolated by transfecting 4 µg of baculovirus transfer vector DNA into SF9 insect cells (2×10^8) along with 200 ng of linearized baculovirus plasmid DNA by the calcium phosphate method. See M.D. Summers and G.E. Smith, *A Manual of*
20 *Methods for Baculovirus Vectors and Insect Cell Culture Procedures*, Texas Agric. Exp. Station Bull. 1555 (1987). Recombinant viruses were purified by three rounds of plaque purification and high titer (10^7 - 10^8 pfu/mL) stocks of virus were prepared. For
25 large scale production, SF9 insect cells were infected in 10 liter fermentors (0.5×10^6 /mL) with the recombinant baculovirus stock such that the multiplicity of infection was 0.1. After 72 hours the cells were centrifuged and the cell pellet
30 homogenized in Tris/Sucrose (50 mM: 25%, pH 8.0) containing 1% 3-[(3-cholamidopropyl)dimethylammonio]-1-propanesulfonate (CHAPS). The homogenate was centrifuged at 10,000xG for 30 minutes, and the resultant supernatant was
35 stored at -80°C before being assayed for COX activity.

b. Assay for COX-1 and COX-2 activity

COX activity was assayed as PGE₂ formed/ μ g protein/time using an ELISA to detect the prostaglandin released. CHAPS-solubilized insect cell membranes containing the appropriate COX enzyme were incubated in a potassium phosphate buffer (50 mM, pH 8.0) containing epinephrine, phenol, and heme with the addition of arachidonic acid (10 μ M). Compounds were pre-incubated with the enzyme for 10-20 minutes prior to the addition of arachidonic acid. Any reaction between the arachidonic acid and the enzyme was stopped after ten minutes at 37 °C/room temperature by transferring 40 μ l of reaction mix into 160 μ l ELISA buffer and 25 μ M indomethacin. The PGE₂ formed was measured by standard ELISA technology (Cayman Chemical). Results are shown in Table II.

c. Fast assay for COX-1 and COX-2 activity

COX activity was assayed as PGE₂ formed/ μ g protein/time using an ELISA to detect the prostaglandin released. CHAPS-solubilized insect cell membranes containing the appropriate COX enzyme were incubated in a potassium phosphate buffer (0.05 M Potassium phosphate, pH 7.5, 2 μ M phenol, 1 μ M heme, 300 μ M epinephrine) with the addition of 20 μ l of 100 μ M arachidonic acid (10 μ M). Compounds were pre-incubated with the enzyme for 10 minutes at 25 °C prior to the addition of arachidonic acid. Any reaction between the arachidonic acid and the enzyme was stopped after two minutes at 37 °C/room temperature by transferring 40 μ l of reaction mix into 160 μ l ELISA buffer and 25 μ M indomethacin. The PGE₂ formed was measured by standard ELISA technology (Cayman Chemical). Results are shown in Table II.

TABLE II.

	Example	COX-2*	COX-1*	COX-2	COX-1
		IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M
5	1	0.3	45		
	2	<0.1	78	<0.1	5.0
	6	<0.1	>100		
	7	0.1	16	<0.1	1.0
	8	<0.1	61	<0.1	21
10	9	<0.1	1.4	<0.1	<0.1
	12	7	55		
	13	.3	>100		
	14	>100	>100		
	15	>0.1	11	133.6	44
15	16	<0.1	24	1.4	51
	18	12	>100		
	21	11	3.5		
	22	>100	>100		
	23	7	>100	24	>100
20	25	>100	78		
	26	>100	20		
	27	67	>100		
	29	<0.1	>100		
	30	<0.1	1.2	16	3.8
25	31	<0.1	94		
	32	0.3	31	0.3	0.7
	33	<0.1	5.7	8.2	28
	35	2.2	8.9	1.7	11
	38	0.2	6.2	25.7	57
30	39	0.2	45	1.3	>100
	40	<0.1	24	74	43
	42	<0.1	2.3	<0.1	11
	43	99	85		
	44	0.3	72	21	>100
35	45	0.2	47	46	>100
	46	0.2	24	74	43
	47	1.9	31	1.7	>100
	49	24	>100	31	>100

TABLE II.

	Example	COX-2*	COX-1*	COX-2	COX-1
		IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M
5	50	79	>100		
	52	20	>100		
	53	8	13	6	>100
	54	19	>100		
	55	46	>100	53	>100
10	56	12	>100	29	>100
	57	21	10	21	>100
	59	43	>100		
	63	1.4	>100		
	65	<0.1	1.0		
15	66	82	38	<0.1	16.9
	67	<0.1	30	<0.1	6.7
	81	<0.1	10.5	<0.1	1.6
	82	<0.1	16	<0.1	5.6
	83	<0.1	9.6	<0.1	1.4
20	84	0.1	25	<0.1	2.8
	88	<0.1	12.4	<0.1	6.4
	91	<0.1	23	0.2	36
	96	0.2	>100	0.3	100
	97	0.2	78	0.1	25
25	98	2.0	>100	1.5	19
	99	0.2	36	<0.1	23
	101	<0.1	18	<0.1	16
	103	36	61		
	104	<0.1	24	<0.1	8.2
30	105	0.3	4.5	0.2	0.1

TABLE II. Cont.

	Example	COX-2*	COX-1*	COX-2	COX-1
		IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M
5	106	0.2	21	<0.1	5.7
	114	<0.1	<0.1	<0.1	<0.1
	115	<0.1	<0.1	<0.1	<0.1
	116	<0.1	<0.1	<0.1	<0.1
	120	<0.1	98	<0.1	33
10	125	<0.1	0.2	<0.1	<0.1
	129	0.2	2.6	<0.1	0.3
	138	0.3	42.5	<0.1	11.1
	152	<0.1	74	<0.1	10
	154	0.5	68.5	<0.1	37
15	155	<0.1	1.6	<0.1	<0.1
	156	<0.1	0.8	<0.1	0.1
	176	<0.1	<0.1	2.0	3.3
	177	<0.1	<0.1	<0.1	0.3
	178	8.4	>100	14.7	>100
20	179	15.6	>100	16.0	>100
	180	<0.1	43.4	<0.1	>100
	181	1.5	>100	>100	>100
	182	<0.1	>100	>100	>100
	183	<0.1	0.1	<0.1	4.1
25	184	<0.1	<0.1	<0.1	1.3
	185	<0.1	68.4	<0.1	>100
	186	<0.1	10.6	3.4	32.8
	187	2.9	48.9	33.7	>100
	188	<0.1	1.3	<0.1	3.7
30	189	<0.1	56.2	1.2	44.8
	190	0.5	0.5	0.5	37.4
	191	<0.1	39.7	42.2	>100
	192	<0.1	5.9	1.4	12.3
	193	1.2	>100	>100	>100
35	195	<0.1	0.6	<0.1	2.6
	196	<0.1	10.1	9.5	30.3
	197	<0.1	0.2	<0.1	0.4
	198	<0.1	3.8	<0.1	12.4

TABLE II. Cont.

Example		COX-2*	COX-1*	COX-2	COX-1
		IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M	IC ₅₀ μ M
5	199	<0.1	>100		
	200	<0.1	1.6	<0.1	28.5
	201	<0.1	9.4	0.1	63.8
	202			<0.1	42.8
	203	<0.1	78.1		
10	209	0.2	>100	>100	>100
	210	1.7	88.8	10.0	>100
	211	1.9	1.7	2.4	3.8

* fast assay

Also embraced within this invention is a class of pharmaceutical compositions comprising the active compounds of Formula I in association with one or more non-toxic, pharmaceutically-acceptable carriers and/or diluents and/or adjuvants (collectively referred to herein as "carrier" materials) and, if desired, other active ingredients. The active compounds of the present invention may be administered by any suitable route, preferably in the form of a pharmaceutical composition adapted to such a route, and in a dose effective for the treatment intended. The active compounds and composition may, for example, be administered orally, pulmonary, mucosally, intravascularly, intraperitoneally, subcutaneously, intramuscularly or topically.

The phrase "co-therapy" (or "combination-therapy"), in defining use of a cyclooxygenase-2 inhibitor agent and another pharmaceutical agent, is intended to embrace administration of each agent in a sequential manner in a regimen that will provide beneficial effects of the drug combination, and is intended as well to embrace co-administration of these agents in a substantially simultaneous manner,

such as in a single capsule having a fixed ratio of these active agents or in multiple, separate capsules for each agent.

The phrase "therapeutically-effective" is intended to qualify the amount of each agent which will achieve the goal of improvement in disease severity and the frequency of incidence over treatment of each agent by itself, while avoiding adverse side effects typically associated with alternative therapies.

For oral administration, the pharmaceutical composition may be in the form of, for example, a tablet, capsule, suspension or liquid. The pharmaceutical composition is preferably made in the form of a dosage unit containing a particular amount of the active ingredient. Examples of such dosage units are tablets or capsules. The active ingredient may also be administered by injection as a composition wherein, for example, saline, dextrose or water may be used as a suitable carrier.

The amount of therapeutically active compounds which are administered and the dosage regimen for treating a disease condition with the compounds and/or compositions of this invention depends on a variety of factors, including the age, weight, sex and medical condition of the subject, the severity of the disease, the route and frequency of administration, and the particular compound employed, and thus may vary widely. The pharmaceutical compositions may contain active ingredients in the range of about 0.1 to 2000 mg, preferably in the range of about 0.5 to 500 mg and most preferably between about 1 and 100 mg. A daily dose of about 0.01 to 100 mg/kg body weight, preferably between about 0.5 and about 20 mg/kg body weight and most preferably between about 0.1 to 10 mg/kg body weight, may be appropriate. The daily

dose can be administered in one to four doses per day.

In the case of psoriasis and other skin conditions, it may be preferable to apply a topical preparation of compounds of this invention to the affected area two to four times a day.

For inflammations of the eye or other external tissues, e.g., mouth and skin, the formulations are preferably applied as a topical ointment or cream, or as a suppository, containing the active ingredients in a total amount of, for example, 0.075 to 30% w/w, preferably 0.2 to 20% w/w and most preferably 0.4 to 15% w/w. When formulated in an ointment, the active ingredients may be employed with either paraffinic or a water-miscible ointment base. Alternatively, the active ingredients may be formulated in a cream with an oil-in-water cream base. If desired, the aqueous phase of the cream base may include, for example at least 30% w/w of a polyhydric alcohol such as propylene glycol, butane-1,3-diol, mannitol, sorbitol, glycerol, polyethylene glycol and mixtures thereof. The topical formulation may desirably include a compound which enhances absorption or penetration of the active ingredient through the skin or other affected areas. Examples of such dermal penetration enhancers include dimethylsulfoxide and related analogs. The compounds of this invention can also be administered by a transdermal device. Preferably topical administration will be accomplished using a patch either of the reservoir and porous membrane type or of a solid matrix variety. In either case, the active agent is delivered continuously from the reservoir or microcapsules through a membrane into the active agent permeable adhesive, which is in contact with the skin or mucosa of the recipient. If the active agent is absorbed through the skin, a controlled and predetermined flow of the active

agent is administered to the recipient. In the case of microcapsules, the encapsulating agent may also function as the membrane.

The oily phase of the emulsions of this invention may be constituted from known ingredients in a known manner. While the phase may comprise merely an emulsifier, it may comprise a mixture of at least one emulsifier with a fat or an oil or with both a fat and an oil. Preferably, a hydrophilic emulsifier is included together with a lipophilic emulsifier which acts as a stabilizer. It is also preferred to include both an oil and a fat. Together, the emulsifier(s) with or without stabilizer(s) make-up the so-called emulsifying wax, and the wax together with the oil and fat make up the so-called emulsifying ointment base which forms the oily dispersed phase of the cream formulations. Emulsifiers and emulsion stabilizers suitable for use in the formulation of the present invention include Tween 60, Span 80, cetostearyl alcohol, myristyl alcohol, glyceryl monostearate, and sodium lauryl sulfate, among others.

The choice of suitable oils or fats for the formulation is based on achieving the desired cosmetic properties, since the solubility of the active compound in most oils likely to be used in pharmaceutical emulsion formulations is very low. Thus, the cream should preferably be a non-greasy, non-staining and washable product with suitable consistency to avoid leakage from tubes or other containers. Straight or branched chain, mono- or dibasic alkyl esters such as di-isoadipate, isocetyl stearate, propylene glycol diester of coconut fatty acids, isopropyl myristate, decyl oleate, isopropyl palmitate, butyl stearate, 2-ethylhexyl palmitate or a blend of branched chain esters may be used. These may be used alone or in combination depending on the properties required. Alternatively, high melting

point lipids such as white soft paraffin and/or liquid paraffin or other mineral oils can be used.

Formulations suitable for topical administration to the eye also include eye drops
5 wherein the active ingredients are dissolved or suspended in suitable carrier, especially an aqueous solvent for the active ingredients. The antiinflammatory active ingredients are preferably present in such formulations in a concentration of
10 0.5 to 20%, advantageously 0.5 to 10% and particularly about 1.5% w/w.

For therapeutic purposes, the active compounds of this combination invention are ordinarily combined with one or more adjuvants appropriate to
15 the indicated route of administration. If administered *per os*, the compounds may be admixed with lactose, sucrose, starch powder, cellulose esters of alkanolic acids, cellulose alkyl esters, talc, stearic acid, magnesium stearate, magnesium
20 oxide, sodium and calcium salts of phosphoric and sulfuric acids, gelatin, acacia gum, sodium alginate, polyvinylpyrrolidone, and/or polyvinyl alcohol, and then tableted or encapsulated for convenient administration. Such capsules or tablets may
25 contain a controlled-release formulation as may be provided in a dispersion of active compound in hydroxypropylmethyl cellulose. Formulations for parenteral administration may be in the form of aqueous or non-aqueous isotonic sterile injection
30 solutions or suspensions. These solutions and suspensions may be prepared from sterile powders or granules having one or more of the carriers or diluents mentioned for use in the formulations for oral administration. The compounds may be dissolved
35 in water, polyethylene glycol, propylene glycol, ethanol, corn oil, cottonseed oil, peanut oil, sesame oil, benzyl alcohol, sodium chloride, and/or various buffers. Other adjuvants and modes of

administration are well and widely known in the pharmaceutical art.

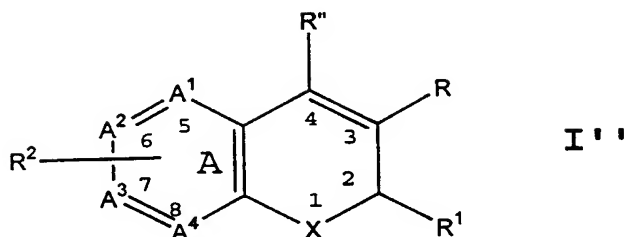
For pulmonary administration, the pharmaceutical composition may be administered in the form of an aerosol or with an inhaler including dry powder aerosol.

All mentioned references are incorporated by reference as if here written. The priority applications (Serial No. 60/044,485, filed April 21, 1997 and Serial No. 09/062,537, filed April 17, 1998) are also incorporated by reference.

Although this invention has been described with respect to specific embodiments, the details of these embodiments are not to be construed as limitations.

What is claimed is:

1. A compound of Formula I''



5

wherein X is selected from O, S, CR^cR^b and NR^a;
 wherein R^a is selected from hydrido, C₁-C₃-alkyl,
 (optionally substituted phenyl)-C₁-C₃-alkyl,
 acyl and carboxy-C₁-C₆-alkyl;

10

wherein each of R^b and R^c is independently
 selected from hydrido, C₁-C₃-alkyl, phenyl-
 C₁-C₃-alkyl, C₁-C₃-perfluoroalkyl, chloro, C₁-
 C₆-alkylthio, C₁-C₆-alkoxy, nitro, cyano and
 cyano-C₁-C₃-alkyl; or wherein CR^bR^c forms a 3-
 6 membered cycloalkyl ring;

15

wherein R is selected from carboxyl,
 aminocarbonyl, C₁-C₆-
 alkylsulfonylaminocarbonyl and C₁-C₆-
 alkoxycarbonyl;

20

wherein R'' is selected from hydrido, phenyl,
 thienyl, C₁-C₆-alkyl and C₂-C₆-alkenyl;
 wherein R¹ is selected from C₁-C₃-perfluoroalkyl,
 chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro,
 cyano and cyano-C₁-C₃-alkyl;

25

wherein R² is one or more radicals independently
 selected from hydrido, halo, C₁-C₆-alkyl, C₂-

C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl,
aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl, aryl-C₂-
C₆-alkenyl, C₁-C₆-alkoxy, methylenedioxy, C₁-
C₆-alkylthio, C₁-C₆-alkylsulfinyl, aryloxy,
5 arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-
alkoxy-C₁-C₆-alkyl, aryl-C₁-C₆-alkyloxy,
heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-
C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy,
C₁-C₆-haloalkylthio, C₁-C₆-haloalkylsulfinyl,
10 C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-
C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl,
hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino,
arylamino, aryl-C₁-C₆-alkylamino,
heteroarylamino, heteroaryl-C₁-C₆-alkylamino,
15 nitro, cyano, amino, aminosulfonyl, C₁-C₆-
alkylaminosulfonyl, arylaminosulfonyl,
heteroarylaminosulfonyl, aryl-C₁-C₆-
alkylaminosulfonyl, heteroaryl-C₁-C₆-
alkylaminosulfonyl, heterocyclylsulfonyl,
20 C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-alkylsulfonyl,
optionally substituted aryl, optionally
substituted heteroaryl, aryl-C₁-C₆-
alkylcarbonyl, heteroaryl-C₁-C₆-
alkylcarbonyl, heteroarylcarbonyl,
25 arylcarbonyl, aminocarbonyl, C₁-C₆-
alkoxycarbonyl, formyl, C₁-C₆-
haloalkylcarbonyl and C₁-C₆-alkylcarbonyl;
and

wherein the A ring atoms A¹, A², A³ and A⁴ are
30 independently selected from carbon and

nitrogen with the proviso that at least two
of A¹, A², A³ and A⁴ are carbon;
or wherein R² together with ring A forms a
radical selected from naphthyl, quinolyl,
5 isoquinolyl, quinoliziny, quinoxaliny and
dibenzofuryl;
or an isomer or pharmaceutically acceptable salt
thereof.

10 2. A compound of Claim 1 wherein X is
selected from O, S, CR^cR^b and NR^a; wherein R^a is
selected from hydrido, C₁-C₃-alkyl, (optionally
substituted phenyl)-C₁-C₃-alkyl, acyl and
carboxy-C₁-C₆-alkyl; wherein each of R^b and R^c
15 is independently selected from hydrido, C₁-C₃-
alkyl, phenyl-C₁-C₃-alkyl, C₁-C₃-perfluoroalkyl,
chloro, C₁-C₆-alkylthio, C₁-C₆-alkoxy, nitro,
cyano and cyano-C₁-C₃-alkyl; or wherein CR^bR^c
forms a cyclopropyl ring; wherein R is selected
20 from carboxyl, aminocarbonyl, C₁-C₆-
alkylsulfonylaminocarbonyl and C₁-C₆-
alkoxycarbonyl; wherein R'' is selected from
hydrido, phenyl, thienyl, C₁-C₄-alkyl and C₂-C₄-
alkenyl; wherein R¹ is selected from C₁-C₃-
25 perfluoroalkyl, chloro, C₁-C₆-alkylthio, C₁-C₆-
alkoxy, nitro, cyano and cyano-C₁-C₃-alkyl;
wherein R² is one or more radicals
independently selected from hydrido, halo, C₁-
C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-
30 C₆-alkynyl, aryl-C₁-C₃-alkyl, aryl-C₂-C₆-alkynyl,
aryl-C₂-C₆-alkenyl, C₁-C₆-alkoxy,

methylenedioxy, C₁-C₆-alkylthio, C₁-C₆-alkylsulfinyl, aryloxy, arylthio, arylsulfinyl, heteroaryloxy, C₁-C₆-alkoxy-C₁-C₆-alkyl, aryl-C₁-C₆-alkyloxy, heteroaryl-C₁-C₆-alkyloxy, aryl-C₁-C₆-alkoxy-C₁-C₆-alkyl, C₁-C₆-haloalkyl, C₁-C₆-haloalkoxy, C₁-C₆-haloalkylthio, C₁-C₆-haloalkylsulfinyl, C₁-C₆-haloalkylsulfonyl, C₁-C₃-(haloalkyl-C₁-C₃-hydroxyalkyl, C₁-C₆-hydroxyalkyl, hydroxyimino-C₁-C₆-alkyl, C₁-C₆-alkylamino, arylamino, aryl-C₁-C₆-alkylamino, heteroarylamino, heteroaryl-C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, C₁-C₆-alkylaminosulfonyl, arylaminosulfonyl, heteroarylamino, heteroaryl-C₁-C₆-alkylaminosulfonyl, heteroaryl-C₁-C₆-alkylaminosulfonyl, heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl, aryl-C₁-C₆-alkylsulfonyl, optionally substituted aryl, optionally substituted heteroaryl, aryl-C₁-C₆-alkylcarbonyl, heteroaryl-C₁-C₆-alkylcarbonyl, heteroarylcarbonyl, arylcarbonyl, aminocarbonyl, C₁-C₆-alkoxycarbonyl, formyl, C₁-C₆-haloalkylcarbonyl and C₁-C₆-alkylcarbonyl; and wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

3. A compound of Claim 2 wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, C₁-C₃-alkyl and (optionally substituted phenyl)methyl; wherein

5 R is carboxyl; wherein R" is selected from hydrido, C₁-C₃-alkyl and C₂-C₃-alkenyl; wherein R¹ is selected from C₁-C₃-perfluoroalkyl; wherein R² is one or more radicals independently selected from hydrido, halo, C₁-

10 C₆-alkyl, C₂-C₆-alkenyl, C₂-C₆-alkynyl, halo-C₂-C₆-alkynyl, optionally substituted phenyl-C₁-C₆-alkyl, optionally substituted phenyl-C₂-C₆-alkynyl, phenyl-C₂-C₆-alkenyl, C₁-C₃-alkoxy, methylenedioxy, C₁-C₃-alkoxy-C₁-C₃-alkyl, C₁-C₃-

15 alkylthio, C₁-C₃-alkylsulfinyl, optionally substituted phenyloxy, optionally substituted phenylthio, optionally substituted phenylsulfinyl, C₁-C₃-haloalkyl-C₁-C₃-hydroxyalkyl, phenyl-C₁-C₃-alkyloxy-C₁-C₃-alkyl,

20 C₁-C₃-haloalkyl, C₁-C₃-haloalkoxy, C₁-C₃-haloalkylthio, C₁-C₃-hydroxyalkyl, C₁-C₃-alkoxy-C₁-C₃-alkyl, hydroxyimino-C₁-C₃-alkyl, C₁-C₆-alkylamino, nitro, cyano, amino, aminosulfonyl, N-alkylaminosulfonyl, N-arylamino-

25 N-arylamino-sulfonyl, N-(phenyl-C₁-C₆-alkyl)aminosulfonyl, N-(heteroaryl-C₁-C₆-alkyl)aminosulfonyl, phenyl-C₁-C₃-alkylsulfonyl, 5- to 8-membered heterocyclylsulfonyl, C₁-C₆-alkylsulfonyl,

30 optionally substituted phenyl, optionally substituted 5- to 9-membered heteroaryl,

phenyl-C₁-C₆-alkylcarbonyl, phenylcarbonyl, 4-chlorophenylcarbonyl, 4-hydroxyphenylcarbonyl, 4-trifluoromethylphenylcarbonyl, 4-methoxyphenylcarbonyl, aminocarbonyl, formyl, and C₁-C₆-alkylcarbonyl; wherein the A ring atoms A¹, A², A³ and A⁴ are independently selected from carbon and nitrogen with the proviso that at least three of A¹, A², A³ and A⁴ are carbon; or wherein R² together with ring A forms a naphthyl, benzofurylphenyl, or quinolyl radical; or an isomer or pharmaceutically acceptable salt thereof.

4. A compound of Claim 3 wherein X is selected from O, S and NR^a; wherein R^a is selected from hydrido, methyl, ethyl, (4-trifluoromethyl)benzyl, (4-chloromethyl)benzyl, (4-methoxy)benzyl, and (4-cyano)benzyl, (4-nitro)benzyl; wherein R is carboxyl; wherein R" is selected from hydrido, ethyl and ethenyl; wherein R¹ is selected from trifluoromethyl and pentafluoroethyl; wherein R² is one or more radicals independently selected from hydrido, chloro, bromo, fluoro, iodo, methyl, tert-butyl, ethenyl, ethynyl, 5-chloro-1-pentynyl, 1-pentynyl, 3,3-dimethyl-1-butynyl, benzyl, phenylethyl, phenyl-ethynyl, 4-chlorophenyl-ethynyl, 4-methoxyphenyl-ethynyl, phenylethenyl, methoxy, methylthio, methylsulfinyl, phenyloxy, phenylthio, phenylsulfinyl, methylenedioxy,

benzyloxymethyl, trifluoromethyl,
difluoromethyl, pentafluoroethyl,
trifluoromethoxy, trifluoromethylthio,
hydroxymethyl, hydroxy-trifluoroethyl,
5 methoxymethyl, hydroxyiminomethyl, N-
methylamino, nitro, cyano, amino,
aminosulfonyl, N-methylaminosulfonyl, N-
phenylaminosulfonyl, N-furylaminosulfonyl, N-
(benzyl)aminosulfonyl, N-
10 (furylmethyl)aminosulfonyl, benzylsulfonyl,
phenylethylaminosulfonyl, furylsulfonyl,
methylsulfonyl, phenyl, phenyl substituted with
one or more radicals selected from chloro,
fluoro, bromo, methoxy, methylthio and
15 methylsulfonyl, benzimidazolyl, thienyl,
thienyl substituted with chloro, furyl, furyl
substituted with chloro, benzylcarbonyl,
optionally substituted phenylcarbonyl,
aminocarbonyl, formyl and methylcarbonyl;
20 wherein the A ring atoms A¹, A², A³ and A⁴ are
independently selected from carbon and nitrogen
with the proviso that at least three of A¹, A²,
A³ and A⁴ are carbon; or wherein R² together
with ring A forms a naphthyl, or quinolyl
25 radical; or an isomer or pharmaceutically
acceptable salt thereof.

5. A compound of Claim 4 selected from
compounds, and their isomers and
30 pharmaceutically-acceptable salts, of the group
consisting of

- 7-phenoxy-2-(trifluoromethyl)-2H-1-benzopyran-
3-carboxylic acid;
- 5 6-chloro-7-phenoxy-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 6-chloro-7-(2-hydroxy-1,1-dimethylethyl)-2-
(trifluoromethyl)-2H-1-benzopyran-3-
carboxylic acid;
- 7-(2-hydroxy-1,1-dimethylethyl)-2-
10 (trifluoromethyl)-2H-1-benzopyran-3-
carboxylic;
- 6-chloro-7-iodo-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 6-chloro-7-ethynyl-2-(trifluoromethyl)-2H-1-
15 benzopyran-3-carboxylic acid;
- 6-chloro-7-ethenyl-2-(trifluoromethyl)-2H-1-
benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-methoxyphenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-
20 carboxylic acid;
- 6-chloro-7-(4-chlorophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-
carboxylic acid;
- 6-chloro-7-(4-chlorophenoxy)-2-
25 (trifluoromethyl)-2H-1-benzopyran-3-
carboxylic acid;
- 6-chloro-7-(4-cyanophenoxy)-2-
(trifluoromethyl)-2H-1-benzopyran-3-
carboxylic acid;

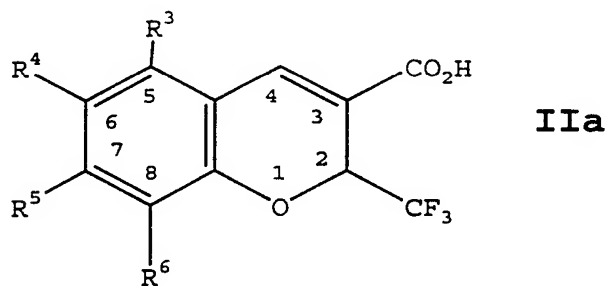
- 6-chloro-2-(trifluoromethyl)-7-[4-(trifluoromethyl)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-2-(trifluoromethyl)-7-[4-(trifluoromethoxy)phenoxy]-2H-1-benzopyran-3-carboxylic acid;
- 7-(2-bromo-4-chlorophenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 7-[(6-bromo-2-naphthalenyl)oxy]-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(2,6-dimethylphenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-
- 15 carboxylic acid;
- 6-chloro-7-[(4-methoxyphenyl)thio]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(4-chlorophenyl)thio]-2-
- 20 (trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(4-chlorophenyl)sulfinyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-[(4-chlorophenyl)sulfonyl]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(phenylthio)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;

- 6-chloro-7-(3-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 6-chloro-7-(2,4-dichlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(3,4-difluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-[(6-chloro-2-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[(2-chloro-3-pyridinyl)oxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 15 6-chloro-7-[3-pyridinyloxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-[2-pyridinyloxy]-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 20 6-chloro-7-(4-nitrophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 25 6-chloro-7-(2-chlorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 6-chloro-7-(4-chloro-3-fluorophenoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 30

- 7-(4-carboxyphenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 7-(4-carboxy-2-chlorophenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 5 7-(3-carboxyphenoxy)-6-chloro-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 10 6-chloro-7-(2,2,2-trifluoroethoxy)-2-(trifluoromethyl)-2H-1-benzopyran-3-carboxylic acid;
- 1,2-dihydro-6-ethynyl-2-((trifluoromethyl))-3-quinolinecarboxylic acid; and
- 15 1,2-dihydro-6-phenylethynyl-2-(trifluoromethyl)-3-quinolinecarboxylic acid.

6. A compound of Formula IIa

20



- wherein R^3 is selected from hydrido, C_1 - C_3 -alkyl, C_1 - C_3 -hydroxyalkyl, C_1 - C_3 -alkoxy and halo;
- 25 wherein R^4 is selected from hydrido, halo, C_1 - C_4 -alkyl, C_1 - C_3 -alkylthio, C_1 - C_3 -haloalkyl, amino,

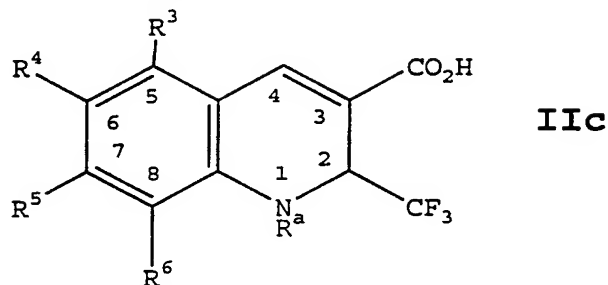
aminosulfonyl, C₁-C₃-alkylsulfonyl, C₁-C₃-alkylsulfinyl, C₁-C₃-alkoxy-C₁-C₃-alkyl, C₁-C₃-alkylcarbonyl, formyl, cyano, C₁-C₃-haloalkylthio, substituted or unsubstituted phenylcarbonyl, C₁-C₃-haloalkoxy, C₁-C₃-alkoxy, aryl-C₁-C₃-alkylcarbonyl, di-C₁-C₃-alkylaminosulfonyl, C₁-C₃-alkylaminosulfonyl, aryl-C₁-C₃-alkylaminosulfonyl, 5- or 6-heteroaryl-C₁-C₃-alkylaminosulfonyl, 5- or 6-membered heteroaryl, C₁-C₃-hydroxyalkyl, substituted or unsubstituted phenyl and 5- or 6-membered nitrogen-containing heterocyclylsulfonyl; wherein R⁵ is selected from hydrido, C₁-C₃-alkyl, halo, C₁-C₃-haloalkyl, C₁-C₄-hydroxyalkyl, C₂-C₃-alkynyl, C₂-C₃-alkenyl, C₁-C₃-alkoxy, phenoxy, phenoxy independently substituted with one or more radicals selected from C₁-C₃-haloalkyl, nitro, carboxy, C₁-C₃-haloalkoxy, C₁-C₃-alkoxy, cyano, C₁-C₃-alkyl and halo, naphthyloxy, naphthyloxy substituted with one or more halo radicals, phenylthio, phenylthio substituted with one or more halo radicals, phenylsulfinyl, phenylsulfinyl substituted with one or more halo radicals, phenylsulfonyl, phenylsulfonyl substituted with one or more halo radicals, pyridinyloxy, pyridinyloxy substituted with one or more halo radicals, and phenyl; and wherein R⁶ is selected from hydrido, halo, cyano, hydroxyiminomethyl, C₁-C₃-hydroxyalkyl, C₂-C₃-alkynyl, phenyl-C₂-C₃-alkynyl, C₁-C₃-alkyl, C₁-C₃-alkoxy, formyl and phenyl; or an isomer or pharmaceutically acceptable salt thereof.

7. Compound of Claim 6 wherein R³ is selected from hydrido, and chloro; wherein R⁴ is selected from chloro, methyl, tert-butyl, methylthio, trifluoromethyl, difluoromethyl, pentafluoromethyl, trifluoromethylthio, trifluoromethoxy, cyano, substituted or unsubstituted phenylcarbonyl, and substituted or unsubstituted phenyl; wherein R⁵ is selected from hydrido, methyl, tert-butyl, 2,2,2-trifluoroethoxy, 2-hydroxy-1,1-dimethylethyl, phenoxy, 4-methoxyphenoxy, 4-chlorophenoxy, 3-chlorophenoxy, 2-chlorophenoxy, 4-cyanophenoxy, 2,6-dimethylphenoxy, 2,4-dichlorophenoxy, 3,4-difluorophenoxy, 4-chloro-3-fluorophenoxy, 4-(trifluoromethyl)phenoxy, 4-nitrophenoxy, 4-carboxyphenoxy, 3-carboxyphenoxy, 2-chloro-4-carboxyphenoxy, 4-(trifluoromethoxy)phenoxy, 2-bromo-4-chlorophenoxy, (6-bromo-2-naphthalenyl)oxy, phenylthio, (4-methoxyphenyl)thio, (4-chlorophenyl)thio, (4-chlorophenyl)sulfinyl, (4-chlorophenyl)sulfonyl, (6-chloro-2-pyridinyl)oxy, (2-chloro-3-pyridinyl)oxy, (3-pyridinyl)oxy, (2-pyridinyl)oxy, iodo, ethenyl, ethynyl, chloro; and wherein R⁶ is selected from hydrido, chloro, thienyl, hydroxyiminomethyl, substituted or unsubstituted phenylethynyl, and substituted or unsubstituted phenyl; or an isomer or pharmaceutically acceptable salt thereof.

8. Compound of Claim 7 wherein R³ is
hydrido; wherein R⁴ is chloro or hydrido; and
wherein R⁶ is hydrido; or an isomer or
5 pharmaceutically acceptable salt thereof.

9. Compound of Claim 8 wherein R⁵ is
selected from 2,2,2-trifluoroethoxy, 2-hydroxy-
1,1-dimethylethyl, phenoxy, 4-methoxyphenoxy,
10 4-chlorophenoxy, 3-chlorophenoxy, 2-
chlorophenoxy, 4-cyanophenoxy, 2,6-
dimethylphenoxy, 2,4-dichlorophenoxy, 3,4-
difluorophenoxy, 4-chloro-3-fluorophenoxy, 4-
(trifluoromethyl)phenoxy, 4-nitrophenoxy, 4-
15 carboxyphenoxy, 3-carboxyphenoxy, 2-chloro-4-
carboxyphenoxy, 4-(trifluoromethoxy)phenoxy, 2-
bromo-4-chlorophenoxy, (6-bromo-2-
naphthalenyl)oxy, phenylthio, (4-
methoxyphenyl)thio, (4-chlorophenyl)thio, (4-
20 chlorophenyl)sulfinyl, (4-
chlorophenyl)sulfonyl, (6-chloro-2-
pyridinyl)oxy, (2-chloro-3-pyridinyl)oxy, (3-
pyridinyl)oxy, and (2-pyridinyl)oxy; or an
isomer or pharmaceutically acceptable salt
25 thereof.

10. A compound of Formula IIc



wherein R^a is selected from hydrido and lower
 5 aralkyl; wherein R^3 is selected from hydrido,
 lower alkyl, lower hydroxyalkyl, lower alkoxy
 and halo;
 wherein R^4 is selected from hydrido, halo,
 lower alkyl, lower alkylthio, lower haloalkyl,
 10 amino, aminosulfonyl, lower alkylsulfonyl,
 lower alkylsulfinyl, lower alkoxyalkyl, lower
 alkylcarbonyl, formyl, cyano, lower
 haloalkylthio, substituted or unsubstituted
 phenylcarbonyl, lower haloalkoxy, lower alkoxy,
 15 lower alkynyl, phenyl-lower alkynyl, lower
 aralkylcarbonyl, lower dialkylaminosulfonyl,
 lower alkylaminosulfonyl, lower
 aralkylaminosulfonyl, lower
 heteroaralkylaminosulfonyl, 5- or 6- membered
 20 heteroaryl, lower hydroxyalkyl, optionally
 substituted phenyl and 5- or 6- membered
 nitrogen containing heterocyclosulfonyl;
 wherein R^5 is selected from hydrido, lower
 alkyl, halo, lower haloalkyl, lower alkoxy, and
 25 phenyl; and wherein R^6 is selected from
 hydrido, halo, cyano, hydroxyiminomethyl, lower
 hydroxyalkyl, lower alkynyl, phenylalkynyl,

lower alkyl, lower alkoxy, formyl and phenyl;
or an isomer or pharmaceutically acceptable
salt thereof.

5 11. Compound of Claim 10 wherein R⁴ is
selected from chloro, methyl, tert-butyl,
methylthio, trifluoromethyl, difluoromethyl,
pentafluoromethyl, trifluoromethylthio,
trifluoromethoxy, ethynyl, phenylethynyl,
10 substituted or unsubstituted phenylcarbonyl,
and substituted or unsubstituted phenyl;
wherein R⁵ is selected from hydrido, methyl,
tert-butyl, chloro; and wherein R⁶ is selected
from hydrido, chloro, thienyl,
15 hydroxyiminomethyl, substituted or
unsubstituted phenylethynyl, and substituted or
unsubstituted phenyl; or an isomer or
pharmaceutically acceptable salt thereof.

20 12. Compound of Claim 11 wherein R³ is
hydrido or chloro; wherein R⁴ is selected from
ethynyl, and optionally substituted
phenylethynyl; or an isomer or pharmaceutically
acceptable salt thereof.

25

 13. A method of treating a cyclooxygenase-
2 mediated disorder in a subject, said method
comprising treating the subject having or
susceptible to said disorder with a
30 therapeutically-effective amount of a compound
of Claims 1-12; or a pharmaceutically-
acceptable salt thereof.

14. The method of Claim 13 wherein the cyclooxygenase-2 mediated disorder is inflammation.

5

15. The method of Claim 15 wherein the cyclooxygenase-2 mediated disorder is arthritis.

10 16. The method of Claim 15 wherein the cyclooxygenase-2 mediated disorder is pain.

17. The method of Claim 15 wherein the cyclooxygenase-2 mediated disorder is cancer.

15

18. A pharmaceutical composition comprising a therapeutically-effective amount of a compound, said compound selected from a family of compounds of Claims 1-12; or a
20 pharmaceutically-acceptable salt thereof.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 99/21460

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D311/58 C07D335/06 C07D215/54 C07D311/92 C07D493/04
C07D409/04 C07D491/04 A61K31/35 //(C07D491/04,311:00,
221:00),(C07D493/04,311:00,307:00),(C07D493/04,317:00,311:00)

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P,X	WO 98 47890 A (G. D. SEARLE & CO.) 29 October 1998 (1998-10-29) claims 1-38 ----	1-18
X	J. W. BUNTING, N. P. FITZGERALD: "Kinetic and thermodynamic control of pseudobase formation from C-3 substituted 1-methylquinolinium cations" CAN. J. CHEM., vol. 62, 1984, pages 1301-1307, XP002071264 cited in the application table 1 ----	1-4
A	US 4 761 424 A (M. E. CARETHERS ET AL.) 2 August 1988 (1988-08-02) claims 1-20 ----- -/-	1-18



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

7 February 2000

Date of mailing of the international search report

14/02/2000

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Herz, C

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 99/21460

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 4 046 778 A (H. ZINNES, N. A. LINDA) 6 September 1977 (1977-09-06) column 1, line 26 - line 32 ---	1-18
A	PATENT ABSTRACTS OF JAPAN vol. 1997, no. 04, 30 April 1997 (1997-04-30) & JP 08 337583 A (TAKEDA CHEM. IND., LTD.), 24 December 1996 (1996-12-24) abstract -----	1-18

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 99/21460

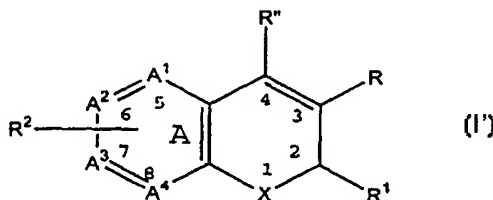
Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9847890 A	29-10-1998	AU 7125698 A	13-11-1998
US 4761424 A	02-08-1988	AU 605747 B	24-01-1991
		AU 6328586 A	02-04-1987
		DK 466486 A	06-04-1987
		EP 0221345 A	13-05-1987
		ES 2002398 A	01-08-1988
		JP 62081369 A	14-04-1987
		US 4921871 A	01-05-1990
		US 4874758 A	17-10-1989
		US 4868195 A	19-09-1989
		US 4868200 A	19-09-1989
		US 4868199 A	19-09-1989
		US 4868205 A	19-09-1989
US 4046778 A	06-09-1977	US 4007203 A	08-02-1977
JP 08337583 A	24-12-1996	NONE	



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification ⁷ : C07D 311/58, 335/06, 215/54, 311/92, 493/04, 409/04, 491/04, A61K 31/35 // (C07D 491/04, 311:00, 221:00) (C07D 493/04, 311:00, 307:00) (C07D 493/04, 317:00, 311:00)</p>	<p>A1</p>	<p>(11) International Publication Number: WO 00/23433</p> <p>(43) International Publication Date: 27 April 2000 (27.04.00)</p>
<p>(21) International Application Number: PCT/US99/21460</p> <p>(22) International Filing Date: 15 October 1999 (15.10.99)</p> <p>(30) Priority Data: 09/175,584 20 October 1998 (20.10.98) US</p> <p>(63) Related by Continuation (CON) or Continuation-in-Part (CIP) to Earlier Application US 09/062,531 (CON) Filed on 17 April 1998 (17.04.98)</p> <p>(71) Applicant (for all designated States except US): G.D. SEARLE & CO. [US/US]; Corporate Patent Dept., P.O. Box 5110, Chicago, IL 60680 (US).</p> <p>(72) Inventors; and (75) Inventors/Applicants (for US only): CARTER, Jeffrey, S. [US/US]; 15321 Grantley Drive, Chesterfield, MO 63017 (US). DEVADAS, Balekudru [-/US]; 2175 Parasol Drive, Chesterfield, MO 63017 (US). TALLEY, John, J. [US/US]; 4701 Clifton Avenue, St. Louis, MO 63109</p>		<p>(US). BROWN, David, L. [US/US]; 15504 Twingate, Chesterfield, MO 63017 (US). GRANETO, Matthew, J. [US/US]; 352 Hartwell Court, Chesterfield, MO 63017 (US). ROGIER, Donald, J., Jr. [US/US]; 1828 Westmeade Drive, Chesterfield, MO 63017 (US). NAGARAJAN, Srinivasan, R. [-/US]; 16209 Forest Meadows Drive, Chesterfield, MO 63017 (US). HANAU, Cathleen, E. [US/US]; 15551 Wendimill Drive, St. Louis, MO 63017 (US). HARTMANN, Susan, J. [US/US]; 449 Colony Woods Drive, Kirkwood, MO 63122 (US). LUDWIG, Cindy, L. [US/US]; 1312 Dautel Lane, St. Louis, MO 63017 (US). METZ, Suzanne [US/US]; 525 Westernmill Drive, Chesterfield, MO 63017 (US). KORTE, Donald, E. [US/US]; 5 Birch Court, Mundelein, IL 60062 (US). BERTENSHAW, Stephen, R. [US/US]; 43 Lansdown, Cheshire, CT 06410 (US). OBUKOWICZ, Mark, G. [US/US]; 655 North Kirkwood Road, Kirkwood, MO 63122 (US).</p> <p>(74) Agents: BULOCK, Joseph, W. et al.; G.D. Searle & Co., Corporate Patent Dept., P.O. Box 5110, Chicago, IL 60680 (US).</p> <p>(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).</p> <p>Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i></p>

(54) Title: SUBSTITUTED BENZOPYRAN ANALOGS FOR THE TREATMENT OF INFLAMMATION



(57) Abstract

A class of benzopyrans, benzothiopyrans, dihydroquinolines, dihydronaphthalenes, and analogs thereof, is described for use in treating cyclooxygenase-2 mediated disorders. Compounds of particular interest are defined by Formula (I') wherein X, A¹, A², A³, A⁴, R, R', R¹ and R² are as described in the specification.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						